

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 5:		(11) International Publication Number:	WO 91/05549
A61K.31/00, C07D 215/08 C07D 215/12, 223/16, 225/06 C07D 241/42, 243/14, 265/36	A1	(43) International Publication Date:	
C07D 241/42, 243/14, 203/30		(45) International Fublication Date:	2 May 1991 (02.05.91)

(21) International Application Number:

PCT/JP90/01340

(22) International Filing Date:

18 October 1990 (18.10.90)

(30) Priority data: 1/274338

2/66063

2/105580

2/181858

20 October 1989 (20.10.89) JP 15 March 1990 (15.03.90) JP 20 April 1990 (20.04.90) JP 9 July 1990 (09.07.90) JP

(71) Applicant (for all designated States except US): OTSUKA PHARMACEUTICAL COMPANY, LIMITED [JP/ JP]; 9, Kandatsukasa-cho 2-chome, Chiyoda-ku, Tokyo 101 (JP).

(72) Inventors; and

(75) Inventors/Applicants (for US only): OGAWA, Hidenori [JP/JP]; MIYAMOTO, Hisashi [JP/JP]; 21-3, Yoshinari-Aza-Todoroki, Ojin-cho, Tokushima-shi, Tokushima 771-11 (JP). KONDO, Kazumi [JP/JP]; 19-27, Okuwaji-ma-Aza-Suberiiwahama, Muya-cho, Naruto-shi, Tokushima 772 (JP). YAMASHITA, Hiroshi [JP/JP]; 57-1, Sasakino-Aza-Hachikami, Matsushige-cho, Itano-gun, Tokushima 771-02 (JP). NAKAYA, Kenji [JP/JP]; 48, Kamibetsukukita, Kawauchi-cho, Tokushima-shi, Tokushima 771-01 (JP). KOMATSU, Hajime [JP/JP]; TANAKA, Michinori [JP/JP]; KORA, Shinya [JP/JP]; 463-10, Kagasuno, Kawauchi-cho, Tokushima-shi, Tokushima 771-01 (JP). TOMINAGA, Michiaki [JP/JP];

310-6, Takaiso, Kamiita-cho, Itano-gun, Tokushima 771-13 (JP). YABUUCHI, Yoichi [JP/JP]; 900-25, Omatsu, Kawauchi-cho, Tokushima-shi, Tokushima 771-01 (JP).

(74) Agents: AOYAMA, Tamotsu et al.; Twin 21 Mid Tower, 1-61, Shiromi 2-chome, Chuo-ku, Osaka-shi, Osaka 540 (JP).

(81) Designated States: AT (European patent), BE (European patent), CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), KR, LU (European patent), NL (European patent), SE (European patent), US.

Published

With international search report.

(54) Title: BENZOHETEROCYCLIC COMPOUNDS

(57) Abstract

Novel benzoheterocyclic compounds of formula (I), wherein R^1 is H, halogen, alkyl, optionally substituted amino, alkoxy; R^2 is H, halogen, alkoxy, phenylalkoxy, OH, alkyl, optionally substituted amino, carbamoyl-alkoxy, optionally substituted amino-alkoxy, optionally substituted benzoyloxy; R^3 is a group: $-NR^4R^5$ or $-CO-NR^{11}R^{12}$; R^4 is H, optionally substituted benzoyl, alkyl; R^5 is a group α [R^{16} is halogen, optionally substituted alkyl, OH, alkoxy, alkanoyloxy, alkylthio, alkanoyl, carboxy, alkoxycarbonyl, CN, NO_2 , optionally substituted amino, phenyl, cycloalkyl, etc., or a group: $-O-A-NR^6R^7$; m is 0 to 3], phenyl-alkoxycarbonyl, alkanoyl, phenyl-alkanoyl, etc.; R^{11} is H or alkyl; R^{12} is cycloalkyl or optionally substituted phenyl; and W is a group: $-(CH_2)_p$ (p is 3 to 5) or $-CH = CH-(CH_2)_q$ (q is 1 to 3), the carbon atom of these groups being optionally replaced by 0, S, SO, SO₂ or a group: $-N(R^{13})$ - and further these groups having optionally 1 to 3 substituents of alkyl, alkoxycarbonyl, carboxy, OH, O, alkanoyloxy, etc., which have excellent vasopressin antagonistic activies and are useful as vasodilator, hypotensive agent, water diuretics, platelet agglutination inhibitor, and a vasopressin antagonistic composition containing the compound as the active ingredient.



FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	ES	Spain	MG	Madagascar
ΑU	Australia	FI	Finland	ML	Mali
BB	Barhados	FR	France	MR	Mauritania
BE	Belgium	GA	Gabon	MW	Malawi
BF	Burkina Faso	GB	United Kingdom	NL	Netherlands
BG	Bulgaria	GR	Greece	NO	Norway
BJ	Bunin	HU	Hungary	PL	Poland
BR	Brazil	IT	Italy	RO	Romania
CA	Canada	JP	Japan	SD	Sudan
CF	Central African Republic	KP	Democratic People's Republic	SE	Sweden
CG	Congo		of Korea	SN	Senegal
CH	Switzerland	KR	Republic of Korea	SU	Soviet Union
CI	Côte d'Ivoire	Li	Liechtenstein	TD	Chad
CM	Cameroon	LK	Sri Lanka	TC	Togo
DE	Germany	ŁU	Luxembourg .	US	United States of America
DK	Denmark	MC	Managa		

BENZOHETEROCYCLIC COMPOUNDS

Technical Field

This invention relates to novel benzoheterocyclic compounds which have excellent vasopressin antagonistic activities and are useful as vasodilator, hypotensive agent, water diuretics, platelet aggregation inhibitor.

Disclosure of the Invention

The benzoheterocyclic compounds of this invention have the following formula:

$$\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{N} \\
\mathbb{C}=0
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{2} \\
\mathbb{R}^{3}
\end{array}$$

wherein R¹ is hydrogen atom, a halogen atom, a lower alkyl, an amino having optionally a lower alkyl substituent, or a lower alkoxy,

R² is hydrogen atom, a halogen atom, a lower alkoxy, a phenyl(lower)alkoxy, hydroxy, a lower alkyl, an amino having optionally a lower alkyl substituent, a carbamoyl-substituted lower alkoxy, an amino-substituted lower alkoxy having optionally a lower alkyl substituent, or a benzoyloxy which has optionally a halogen substituent on the phenyl ring,

$$R^3$$
 is a group of the formula: $-N$
 R^4
or a group of the formula: $-N$
 R^5
the formula: $-C-N$
 R^{12}

 ${\ensuremath{\mathsf{R}}}^4$ is hydrogen atom, a benzoyl which has optionally a halogen substituent on the phenyl ring, or a lower alkyl,

 R^5 is a group of the formula: $-co \sqrt{(R^{16})_m}$ [wherein R^{16} is a halogen atom; a lower alkyl which has optionally a substituent selected from a halogen atom and hydroxy; hydroxy; a lower alkoxy; a lower alkanoyloxy; a lower alkylthio; a lower alkanoyl; carboxy; a lower alkoxycarbonyl; cyano; nitro; an amino which has optionally a substituent selected from a lower alkyl and a lower alkanoyl; phenyl; a cycloalkyl; a lower alkanoyloxysubstituted lower alkoxy; a carboxy-substituted lower alkoxy; a halogen-substituted lower alkoxy; a carbamoylsubstituted lower alkoxy; a hydroxy-substituted lower alkoxy; a lower alkoxycarbonyl-substituted lower alkoxy; a phthalimido-substituted lower alkoxy; an aminocarbonyl-lower alkoxy having a lower alkyl substituent; or a group of the formula: -0-A-N₂₇ (A is a lower alkylene, and R^6 and R^7 are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a lower alkanoyl, or benzoyl, or R^6 and R^7 may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocyclic group has optionally a substituent selected from piperidinyl and a lower alkyl); and m is an integer of 0 to 3], a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyllower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxy-lower alkanoyl, a naphthyloxy-lower alkanoyl, a halogen-substituted lower. alkanoyl, a group of the formula: $-CO-(N-R^8)$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula:

-CO-B-(CO)_n-N R⁹ (wherein B is a lower alkylene, n is an integer of 0 or 1, and R⁹ and R¹⁰ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyl-lower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxy-lower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower

alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R⁹ and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocyclic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl),

R¹¹ is hydrogen atom or a lower alkyl,

 ${\bf R}^{12}$ is a cycloalkyl, or a phenyl which has optionally 1 to 3 substituents selected from a lower alkoxy, a lower alkyl and a halogen atom,

W is a group of the formula: $-(CH_2)_p$ - (p is an integer of 3 to 5), or a group of the formula: $-CH=CH-(CH_2)_q$ - (q is an integer of 1 to 3), the carbon atom of these groups: $-(CH_2)_p$ - and $-CH=CH-(CH_2)_q$ - being optionally replaced by oxygen atom, sulfur atom, sulfinyl, sulfonyl, or a group of

the formula: -N- (R^{13} is hydrogen atom, a cycloalkyl, or a lower alkyl), and further said $-(CH_2)_p$ — and $-CH=CH-(CH_2)_q$ — groups having optionally 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxy, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl naving optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkyl sulfonyloxy-lower alkyl, an

azido-lower alkyl, a group of the formula: _____O, an aminocarbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-substituted lower alkoxy, a carboxy-substituted lower alkoxy, an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, an amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxy-imino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula: R81-N-CH2COO- (R81 is hydrogen atom or a lower alkyl),

hydrazino, pyrrolyl, an amino-lower alkanoyloxy having optionally a lower alkyl substituent, a group of the

formula: -O-A-CO-N $_{R83}^{82}$ (A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, or a pyridyl-lower alkyl, or R^{82} and R^{83} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula:

-(CO) $_{n}$ -N/ $_{p15}$ (wherein n is as defined above, and $_{R}^{14}$ and $_{R}^{15}$

and the figure and the state of the state of

are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having optionally 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyllower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or R¹⁴ and R¹⁵ may bind together with nitrogen atom to which they bond to form a 5or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen, wherein the heterocyclic group may of sionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl or a lower alkanoyl).

The benzoheterocyclic compounds of the formula (1) and

their salts have excellent vasopressin antagonistic activities and vasodilating activity, hypotensive activity, activity for inhibiting saccharide release in liver, activity for inhibiting growth of mesangium cells, water diuretic activity, platelet agglutination inhibitory activity and are useful as vasodilator, hypotensive agent, water diuretics, platelet agglutination inhibitor and are used for the prophylaxis and treatment of hypertension, edema, ascites, heart failure, renal function disorder, vasopressin parasecretion syndrome (SIADH), hepatocirrhosis, hyponatremia, hypokaliemia, diabetic, circulation disorder, and the like.

Each group in the above formula (1) includes specifically the following groups.

The "lower alkoxy" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms, for example, methoxy, ethoxy, propoxy, isopropoxy, butoxy, tertbutoxy, pentyloxy, hexyloxy, and the like.

The "lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, methyl, ethyl, propyl, isopropyl, butyl, tert-butyl, pentyl, hexyl, and the like.

The "halogen atom" includes fluorine atom, chlorine atom, bromine atom and iodine atome.

The "amino having optionally a lower alkyl substituent" includes an amino having optionally one or two substituents selected from a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, amino,

and the control of th

methylamino, ethylamino, propylamino, isopropylamino, butylamino, tert-butylamino, pentylamino, hexylamino, dimethylamino, diethylamino, dipropylamino, dibutylamino, dipentylamino, dihexylamino, N-methyl-N-ethylamino, N-ethyl-N-propylamino, N-methyl-N-butylamino, N-methyl-N-hexylamino, and the like.

The "lower arkenyl" includes a straight chain or branched chain alkenyl group having 2 to 6 carbon atoms, for example, vinyl, allyl, 2-butenyl, 3-butenyl, 1-methylallyl, 2-pentenyl, 2-hexenyl, and the like.

The "lower alkyl which has optionally a substituent selected from a halogen atom and hydroxy" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which may optionally have 1 to 3 substituents selected from a halogen atom and hydroxy, for example, in addition to the above-mentioned lower alkyl groups, hydroxymethyl, 2hydroxyethyl, 1-hydroxyethyl, 3-hydroxypropyl, 2,3dihyroxypropyl, 4-hydroxybutyl, 1,1-dimethyl-2-hydroxyethyl, 5,5,4-trihydroxypentyl, 5-hydroxypentyl, 6-hydroxyhexyl, 1hydroxyisopropyl, 2-methyl-3-hydroxypropyl, trifluoromethyl, trichloromethyl, chloromethyl, bromomethyl, fluoromethyl, iodomethyl, difluoromethyl, dibromomethyl, 2-chloroethyl, 2,2,2-trifluoroethyl, 2,2,2-trichloroethyl, 3-chloropropyl, 2,3-dichloropropyl, 4,4,4-trichlorobutyl, 4-fluorobutyl, 5chloropentyl, 3-chloro-2-methylpropyl, 5-bromohexyl, 5,6dichlorohexyl, and the like.

The "lower alkylene" includes a straight chain or

branched chain alkylene group having 1 to 6 carbon atoms, for example, methylene, ethylene, trimethylene, 2-methyltrimethylene, 2-methyltrimethylene, 1-methyltrimethylene, methylmethylene, ethylmethylene, tetramethylene, pentamethylene, hexamethylene, and the like.

The "lower alkanoyloxy" includes a straight chain or branched chain alkanoyloxy group having 1 to 6 carbon atoms, for example, formyloxy, acetyloxy, propionyloxy, butyryloxy, isobutyryloxy, pentanoyloxy, tert-butylcarbonyloxy, hexanoyloxy, and the like.

The "lower alkylthio" includes a straight chain or branched chain alkylthio group having 1 to 6 carbon atoms, for example, methylthio, ethylthio, propylthio, isopropylthio, butylthio, tert-butylthio, pentylthio, hexylthio, and the like.

The "lower alkanoyl" includes a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, formyl, acetyl, propionyl, butyryl, isobutyryl, pentanoyl, tert-butylcarbonyl, hexanolyl, and the like.

The "lower alkoxycarbonyl" includes a straight chain or branched chain alkoxycarbonyl group having 1 to 6 carbon atoms in the alkoxy moiety, for example, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, isopropoxycarbonyl, butoxycarbonyl, tert-butoxycarbonyl, pentyloxycarbonyl, hexyloxycarbonyl, and the like.

The "amino having optionally a substituent selected from a lower alkyl and a lower alkanoyl" includes an amino having optionally one or two substituents selected from a

straight chain or branched chain alkyl group having 1 to 6 carbon atoms and a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, amino, methylamino, ethylamino, propylamino, isopropylamino, butylamino, tert-butylamino, pentylamino, hexylamino, dimethylamino, diethylamino, dipropylamino, dibutylamino, dipentylamino, diethylamino, N-methyl-N-ethylamino, N-ethyl-N-propylamino, N-methyl-N-butylamino, N-methyl-N-hexylamino, N-methyl-N-acetylamino, N-acetylamino, N-formylamino, N-propionylamino, N-butyrylamino, N-isobutyrylamino, N-pentanoylamino, N-tert-butylcarbonylamino, N-hexanoylamino, N-ethyl-N-acetylamino, and the like.

The "cycloalkyl" includes a cycloalkyl having 3 to 8 carbon atoms, for example, cyclopropyl, cyclobutyl, cyclopetyl, cyclobutyl, cyclopetyl, cyclooctyl, and the like.

The "lower alkanoyloxy-substituted lower alkoxy" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms which is substituted by a straight chain or branched chain alkanoyloxy group having 2 to 6 carbon atoms, for example, acetyloxymethoxy, 2-propionyloxyethoxy, 1-butyryloxyethoxy, 3-acetyloxypropoxy, 4-acetyloxybutoxy, 4-isobutyryloxybutoxy, 5-pentanoyloxypentyloxy, 6-acetyloxyhexyloxy, 6-tert-butylcarbonyloxyhexyloxy, 1,1-dimethyl-2-hexanoyloxyethoxy, 2-methyl-3-acetyloxypropoxy, and he like.

The "carbamoyl-substituted lower alkoxy" includes a carbamoyl-substituted alkoxy group wherein the alkoxy moiety is a straight chain or branched chain alkoxy group having 1 to 6

ana water an ara gana kasa keterwa aliyega kataya jina artina ka tangan aliama 496,5m110,8g

carbon atoms, for example, carbamoylmethoxy, 2-carbamoylethoxy, 1-carbamoylethoxy, 3-carbamoylpropoxy, 4-carbamoylbutoxy, 5-carbamoylpentyloxy, 6-carbamoylhexyloxy, 1,1-dimethyl-2-carbamoylethoxy, 2-methyl-3-carbamoylpropoxy, and the like.

The "hydroxy-substituted lower alkoxy" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms and having 1 to 3 hydroxy substitutents, for example, hydroxymethoxy, 2-hydroxyethoxy, 1-hydroxyethoxy, 3-hydroxypropoxy, 2,3-dihydroxypropoxy, 4-hydroxybutoxy, 3,4-dihydroxybutoxy, 1,1-dimethyl-2-hydroxyethoxy, 5-hydroxypentyloxy, 6-hydroxyhexyloxy, 2-metnyl-3-hydroxypropoxy, 2,3,4-trihydroxybutoxy, and the like.

The "lower alkoxycarbonyl-substituted lower alkoxy" includes an alkoxycarbonyl-substituted straight chain or branched chain alkoxy group having 1 to 6 carbon atoms wherein the alkoxycarbonyl moiety is a straight chain or branched chain alkoxycarbonyl group having 1 to 6 carbon atoms, for example, methoxycarbonylmethoxy, 3-methoxycarbonylpropoxy, ethoxycarboxymethoxy, 3-ethoxycarbonylpropoxy, 4-ethoxycarbonylbutoxy, 5-isopropoxycarbonylpentyloxy, 6-propoxycarbonyl-hexyloxy, 1,1-dimethyl-2-butoxycarbonylethoxy, 2-methyl-3-tert-butoxycarbonylpropoxy, 2-pentyloxycarbonylethoxy, hexyloxy-carbonylmethoxy, and the like.

The "carboxy-substituted lower alkoxy" includes a carboxy-substituted alkoxy group wherein the alkoxy moiety is a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms, for example, carboxymethoxy, 2-carboxyethoxy, 1-

The first of the second of the

carboxyethoxy, 3-carboxypropoxy, 4-carboxybutoxy, 5-carboxy-pentyloxy, 6-carboxyhexyloxy, 1,1-dimethyl-2-carboxyethoxy, 2-methyl-3-carboxypropoxy, and the like.

The "phthalimido-substituted lower alkoxy" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms which is substituted by phthalimido group, for example, phthalimidomethoxy, 2-phthalimidoethoxy, 1-phthal-imidoethoxy, 3-phthalimidopropoxy, 4-phthalimidobutoxy, 5-phthalimidopentyloxy, 6-phthalimidohexyloxy, 1,1-dimethyl-2-phthalimidoethoxy, 2-methyl-3-phthalimidopropoxy, and the like.

The "5- or 6-membered saturated heterocyclic group which is formed by binding the groups R^6 and R^7 together with the nitrogen atom to which they bond with or without being intervened with nitrogen or oxygen atom" includes, for example, pyrrolidinyl, piperidinyl, morpholino, and the like.

The "heterocyclic group having a substituent selected from piperidinyl and a lower alkyl" includes a heterocyclic group having 1 to 3 substituents selected from piperidinyl and a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, 4-methylpiperiazinyl, 3,4-dimethylpiperazinyl, 3-ethylpyrrolidinyl, 2-propylpyrrolidinyl, 3,4,5-trimethylpiperidinyl, 4-butylpiperidinyl, 3-pentylmorpholino, 4-hexylpiperazinyl, 4-(1-piperidinyl)piperidinyl, 3-(1-piperidinyl)pyrrolidinyl, 3-(1-piperidinyl)-4-methylpiperazinyl, 3-(1-piperidinyl)morpholino, and the like.

The "phenyl(lower)alkanoyl" includes a phenylalkanoyl wherein the alkanoyl moiety is a straight chain or branched

A CONTRACTOR OF THE CONTRACTOR

chain alkanoyl group having 2 to 6 carbon atoms, for example, phenylacetyl, 3-phenylpropionyl, 2-phenylpropionyl, 4-phenylbutyryl, 2,2-dimethyl-3-phenylpropionyl, 5-phenylpentanoyl, 6-phenylhexanoyl, and the like.

The "cycloalkyl-lower alkanoyl" includes $C_3^-C_8$ cycloalkyl-alkanoyl group wherein the alkanoyl moiety is a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms, for example, cyclohexylacetyl, 3-cyclopropylpropionyl, 2-cyclopentylpropionyl, 4-cyclohexylbutyryl, 2,2-dimethyl-3-cycloheptylpropionyl, 5-cyclooctylpentanoyl, 6-cyclohexyl-hexanoyl, and the like.

The "cycloalkylcarbonyl" includes a cycloalkylcarbonyl having 3 to 8 carbon atoms, for example, cyclopropylcarbonyl, cyclobutylcarbonyl, cyclopentylcarbonyl, cyclohexylcarbonyl, cycloheptylcarbonyl, cyclooctylcarbonyl, and the like.

The "amino having optionally a lower alkanoyl substituent" includes an amino having optionally a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, amino, formylamino, acetylamino, propionylamino, butyrylamino, isobutyrylamino, pentanoylamino, tert-butylcarbonylamino, hexanoylamino, and the like.

The "phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optonally a lower alkanoyl substituent" includes a phenoxyalkanoyl group wherein the alkanoyl moiety is a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms and the phenyl ring has optionally 1

to 3 substituents selected from a straight chain or branched chain alkyl having 1 to 6 carbon atoms, a straight chain or branched chain alkoxy having 1 to 6 carbon atoms and an amino having optionally a straight chain or branched chain alkanovl having 1 to 6 carbon atoms, for example, phenoxyacetyl, 3phenoxypropionyl, 2-phenoxypropionyl, 4-phenoxybutyryl, 2,2dimethyl-3-phenoxypropionyl, 5-phenoxypentanoyl, 6-phenoxyhexanoyl, (2-aminophenoxy)acetyl, 3-(4-aminophenoxy)propionyl, (2-methylphenoxy)acetyl, (4-methylphenoxy)acetyl, (3-methylphenoxy)acetyl, (3-methoxyphenoxy)acetyl, (3-acetylaminophenoxy)acetyl, 4-(2-propionylaminophenoxy)butyryl, 2,2dimethyl-3-(4-butyrylaminophenoxy)propionyl, 5-(2-pentanoylaminophenoxy)pentanoyl, 6-(4-hexanoylaminophenoxy)hexanoyl, 3-(2-ethylphenoxy)propionyl, 2-(4-propylphenoxy)propionyl, 4-(4butylphenoxy)butyryl, 5-(3-pentylphenoxy)pentanoyl, 6-(4-hexylphenoxy)hexanoyl, (2,3-dimethylphenoxy)acetyl, (2,5-d. methylphenoxy)acetyl, (3,4-dimethylphenoxy)acetyl, (3,4,5-trimethylphenoxy)acetyl, 3-(4-::noxyphenoxy)propionyl, 2-(2-propoxyphenoxy)propionyl, 4-(3-butoxyphenoxy)butyryl, 5-(4-pentyloxyphenoxy)pentanoyl, 6-(4-hexyloxyphenoxy)hexanoyl, (3,4dimethoxyphenoxy)acetyl, (3,5-dimethoxyphenoxy)acetyl, (2,4dimethoxyphenoxy)acetyl, (3,4,5-trimethoxyphenoxy)acetyl, (2acetylamino-4-methylphenoxy)acetyl, (4-acetylamino-3-methoxyphene ')acetyl, and the like.

The "phthalimido-substituted lower alkanoyl" includes a straight chain or branched chain alkanoyl group havir 2 to 6 carbon atoms which is substituted by phthalimido group, for

example, 2-phthalimidoacetyl, 3-phthalimidopropionyl, 2-phthal-imidopropionyl, 4-phthalimidobutyryl, 2,2-dimethyl-3-phthal-imidopropionyl, 5-phthalimidopentanoyl, 6-phthalimidohexanoyl, 3-methyl-4-phthalimidobutyryl, and the like.

The "lower alkoxycarbonyl-lower alkanoyl" includes an alkoxycarbonyl-alkanoyl group wherein the alkoxy moiety is a straight chain or branched chain alkoxy having 1 to 6 carbon atoms and the alkanoyl moiety is a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms, for example, methoxycarbonylacetyl, 3-methoxycarbonylpropionyl, ethoxycarbonylacetyl, 3-ethoxycarbonylpropionyl, 4-ethoxycarbonylbutyryl, 3-propoxycarbonylpropionyl, 2-methoxycarbonylpropionyl, 6-propoxycarbonylpropionyl, 5-isopropoxycarbonylpropionyl, 2-methyl-3-tert-butoxycarbonylpropionyl, pentyloxycarbonylacetyl, hexyloxycarbonylacetyl, and the like.

The "carboxy-lower alkanoyl" includes a carboxyalkanoyl group wherein the alkanoyl moiety is a straight chain
or branched chain alkanoyl having 2 to 6 carbon atoms, for
example, carboxyacetyl, 3-carboxypropionyl, 2-carboxypropionyl,
4-carboxybutyryl, 2,2-dimethyl-3-carboxypropionyl, 5-carboxypentanoyl, 6-carboxyhexanoyl, and the like.

The "naphthyloxy-lower alkanoyl" includes a naphthyloxy-alkanoyl group wherein the alkanoyl moiety is a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms, for example, naphtyloxyacetyl, 3-naphtyloxypropionyl, 2-naphtyloxypropionyl, 4-naphthyloxybutyryl, 2,2-dimethyl-3-

naphthyloxypropionyl, 5-naphthyloxypentanoyl, 6-naphthyloxy-hexanoyl, and the like.

The "phenyl-lower alkoxycarbonyl" includes a phenyl-alkoxycarbonyl wherein the alkoxycarbonyl moiety is a straight chain or branched chain alkoxycarbonyl group having 1 to 6 carbon atoms, for example, benzyloxycarbonyl, 2-phenyl-ethoxycarbonyl, 1-phenylethoxycarbonyl, 3-phenylpropoxycarbonyl, 4-phenylbutoxycarbonyl, 5-phenylpentyloxycarbonyl, 6-phenylhexyloxycarbonyl, 1,1-dimethyl-2-phenylethoxycarbonyl, 2-methyl-3-phenylpropoxycarbonyl, and the like.

The "lower alkyl having optionally a hydroxy substituent" includes a straight chain or branched chain alkyl having 1 to 6 carbon atoms and having optionally 1 to 3 hydroxy substituents, for example, hydroxymethyl, 2-hydroxyethyl, 1-hydroxyethyl, 3-hydroxypropyl, 2,3-dihydroxyethyl, 4-hydroxybutyl, 3,4-dihydroxybutyl, 1,1-dimethyl-2-hydroxyethyl, 5-hydroxypentyl, 6-hydroxyhexyl, 2-methyl-3-hydroxypropyl, 2,3,4-trihydroxybutyl, and the like.

The "phenyl-lower alkyl" includes a phenylalkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, benzyl, 2-phenylethyl, 1-phenylethyl, 3-phenylpropyl, 4-phenylbutyl, 5-phenylpentyl, 6-phenylhexyl, 1,1-dimethyl-2-phenylethyl, 2-methyl-3-phenylpropyl, and the like.

The "phenoxy-lower alkyl" includes a phenoxyalkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example,

phenoxymethyl, 1-phenoxyethyl, 2-phenoxyethyl, 3-phenoxypropyl, 4-phenoxybutyl, 5-phenoxypentyl, 6-phenoxyhexyl, 1,1-dimethyl-2-phenoxyethyl, 2-methyl-3-phenoxypropyl, and the like.

The "phenyl which has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and a halogen atom" includes a phenyl group which has optionally 1 to 3 substituents selected from a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms and a halogen atom, for example, phenyl, 2-methoxyphenyl, 3-methoxyphenyl, 4methoxyphenyl, 2-ethoxyphenyl, 3-ethoxyphenyl, 4-ethoxyphenyl, 4-isopropoxyphenyl, 4-pentyloxyphenyl, 2,4-dimethoxyphenyl, 4hexyloxyphenyl, 3,4-dimethoxyphenyl, 3-ethoxy-4-methoxyphenyl, 2,3-dimethoxyphenyl, 3,4-diethoxyphenyl, 2,5-dimethoxyphenyl, 2,6-dimethoxyphenyl, 3,5-dimethoxyphenyl, 3,4-dipentyloxyphenyl, 3,4,5-trimethoxyphenyl, 2-chlorophenyl, 3-chlorophenyl, 4-chlorophenyl, 2-fluorophenyl, 3-fluorophenyl, 4-fluorophenyl, 2-bromophenyl, 3-bromophenyl, 4-bromophenyl, 2-iodophenyl, 3iodophenyl, 4-iodophenyl, 3,4-dichlorophenyl, 3,5-dichlorophenyl, 2,6-dichlorophenyl, 2,3-dichlorophenyl, 2,4-dichlorophenyl, 3,4-difluorophenyl, 3,5-dibromophenyl, 3,4,5-trichlorophenyl, 2-methoxy-3-chlorophenyl, 2-methylphenyl, 3-methylphenyl, 4-methylphenyl, 2-ethylphenyl, 3-ethylphenyl, 4ethylphenyl, 4-isopropylphenyl, 3-butylphenyl, 4-pentylphenyl, 4-hexylphenyl, 3,4-dimethylphenyl, 3,4-diethylphenyl, 2,4dimethylphenyl, 2,5-dimethylphenyl, 2,6-dimethylphenyl, 3,4,5trimethylphenyl, 3-chloro-4-methylphenyl, 3-methoxy-4-methyl-5-

rame important ega allenda elle i le elle i le rationa a le le le elle elle elle i le le elle elle elle elle e

iodophenyl, 3,4-dimethoxy-5-bromophenyl, 3,5-diiodo-4-methoxyphenyl, and the like.

The "amino-lower alkyl having optionally a lower alkyl substituent" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, aminomethyl, 2-aminoethyl, 1-aminoethyl, 3-aminopropyl, 4-aminobutyl, 5-aminopentyl, 6-aminohexyl, 1,1-dimethyl-2-aminoethyl, 2-methyl-3-aminopropyl, methylaminomethyl, 1-ethylaminoethyl, 2-propylaminoethyl, 3-isopropyl-aminopropyl, 4-butylaminobutyl, 5-pentylaminopentyl, 6-hexyl-aminohexyl, dimethylaminomethyl, (N-ethyl-N-propylamino)methyl, 2-(N-methyl-N-hexylamino)ethyl, and the like.

The "5- or 6-membered saturated heterocyclic group which is formed by binding the groups R⁹ and R¹⁰ together with the nitrogen atom to which they bond with or without being intervened with nitrogen or oxygen atom" includes, for example, pyrrolidinyl, piperidinyl, piperazinyl, morpholino, and the like.

The "heterocyclic group having a substituent selected from a lower alkyl, a lower alkoxycarbonyl and piperidinyl" includes a heterocyclic group having 1 to 3 substituents selected from a straight chain or branched chain alkyl c.cup having 1 to 6 carbon atoms, a straight chain or branched chain alkoxycarbonyl having 1 to 6 carbon atoms and piperidinyl, for example, in addition to the above-mentioned heterocyclic groups

having a substituent of a lower alkyl and piperidinyl, 4-methoxycarbonylpiperazinyl, 4-ethoxycarbonylpiperidinyl, 3-propoxycarbonylpyrrolidinyl, 2-pentyloxycarbonylmorpholino, 4-hexyloxycarbonylpiperidinyl, 4-ethoxycarbonyl-3-methyl-piperidinyl, 3-methyl-4-ethoxycarbonylpiperazinyl, and the like.

The "5- or 6-membered saturated heterocyclic group which is formed by binding the groups R^{14} and R^{15} together with the nitrogen atom to which they bond with or without being intervened with nitrogen or oxygen atom" includes, for example, pyrrolidinyl, piperidinyl, piperazinyl, morpholino, and the like.

The "heterocyclic group having a lower alkyl substituent" includes a heterocyclic group having 1 to 3 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, 4-methylpiperazinyl, 3,4-dimethylpiperazinyl, 3-ethylpyrrolidinyl, 2-propylpyrrolidinyl, 3,4,5-trimethylpiperidinyl, 4-butylpiperidinyl, 3-pentylmorpholino, 4-hexylpiperazinyl, and the like.

The heterocyclic ring in the formula (1) includes tetrahydroquinolyl, 2,3,4,5-tetrahydro-lH-benzazepinyl, 1,2,3,4,5,6-hexahydrobenzazocinyl, 1,2-dihydroquinolyl, 2,3-dihydro-lH-benzazepinyl, 1,2,3,4-tetrahydrobenzazocinyl, and the like.

The heterocyclic ring in the formula (1) wherein the carbon atom in the group of the formula: $-(CH_2)_p$ - or $-CH=CH-(CH_2)_q$ - for W is replaced by oxygen atom, sulfur atom,

R¹³

sulfinyl, sulfonyl, or a group of the formula: $-N-(R^{13})$ is hydrogen atom or a lower alkyl) includes a heterocylic group wherein the carbon atom in the group of the formula: $-(CH_2)_p$ -or $-CH=CH-(CH_2)_q$ - for W is replaced by oxygen atom, sulfur

 R^{13} atom, sulfinyl, sulfonyl, or a group of the formula: $-\dot{N}-$ (R¹³ is hydrogen atom or a straight chain or branched chain alkyl having 1 to 6 carbon atoms), for example, 3,4-dihydro-2H-1,4benzoxazinyl, 1,2,3,5-tetrahydro-4,1-benzoxazepinyl, 1,2,3,4tetrahydroquinoxalinyl, 1,2,3,4,5,6-hexahydro-1,5-benzodiazocinyl, 5-methyl-1,2,3,4,5,6-hexahydro-1,5-benzodiazocinyl, 4-methyl-1,2,3,4-tetrahydroquinoxalinyl, 1,2,3,4-tetrahydro-5,1-benzoxazepinyl, 3,4-dihydro-2H-1,4-benzothiazinyl, 2,3,4,5tetrahydro-1,5-benzothiazepinyl, 1,2,3,5-tetrahydro-4,1benzothiazepinyl, 4-ethyl-1,2,3,4-tetrahydroquinoxalinyl, 4propyl-1,2,3,4-tetrahydroquinoxalinyl, 4-butyl-1,2,3,4tetrahydroquinoxalinyl, 4-pentyl-1,2,3,4-tetrahydroquinoxalinyl, 4-hexyl-1,2,3,4-tetrahydroquinoxalinyl, 2,3,4,5tetrahydro-1H-1,4-benzodiazepinyl, 4-methyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepinyl, 4-ethyl-2,3,4,5-tetrahydro-1H-1,4benzodiazepinyl, 4-propyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepinyl, 4-butyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepinyl, 4-pentyl-2,3,4,5-tetrahydro-lH-1,4-benzodiazepinyl, 4-hexyl-2,3;4,5-tetrahydro-lH-1,4-benzodiazepinyl, 2,3,4,5-tetrahydro-1H-1,5-benzodiazepinyl, 5-methyl-2,3,4,5-tetrahydro-1H-1,5benzodiazepinyl, 5-ethyl-2,3,4,5-tetrahydro-1H-1,5-benzodiazepinyl, 5-propyl-2,3,4,5-tetrahydro-lH-1,5-benzodiazepinyl, 5-butyl-2,3,4,5-tetrahydro-lH-1,5-benzodiazepinyl, 5-pentyl-2,3,4,5-tetrahydro-lH-1,5-benzodiazepinyl, 5-hexyl-2,3,4,5-tetrahydro-lH-1,5-benzodiazepinyl, 3,4-dihydro-l-oxo-2H-1,4-benzothiazepinyl, 3,4-dihydro-l,1-dioxo-2H-1,4-benzothiazepinyl, 1-oxo-2,3,4,5-tetrahydro-l,5-benzothiazepinyl, 1,1-dioxo-2,3,4,5-tetrahydro-l,5-benzothiazepinyl, 4-oxo-1,2,3,5-tetrahydro-4,1-benzothiazepinyl, 4,4-dioxo-1,2,3,5-tetrahydro-4,1-benzothiazepinyl, and the like.

The "halogen-substituted lower alkoxy" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms which has 1 to 3 substituents of a halogen atom, for example, trifluoromethoxy, trichloromethoxy, chloromethoxy, bromomethoxy, fluoromethoxy, iodomethoxy, difluoromethoxy, dibromomethoxy, 2-chloroethoxy, 2,2,2-trifluoroethoxy, 2,2,2-trichloroethoxy, 3-chloropropoxy, 2,3-dichloropropoxy, 4,4,4-trichlorobutoxy, 4-fluorobutoxy, 5-chloropentyloxy, 3-chloro-2-methylpropoxy, 6-bromohexyloxy, 5,6-dichlorohexyloxy, and the like.

The "halogen-substituted lower alkanoyl" includes a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms which has 1 to 3 substituents of a halogen atom, for example, 2,2,2-trifluoroacetyl, 2,2,2-trichloroacetyl, 2-chloroacetyl, 2-bromoacetyl, 2-fluoroacetyl, 2-iodoacetyl, 2,2-difluoroacetyl, 2,2-dibromoacetyl, 3,3,3-trifluoropropionyl, 3,3,3-trichloropropionyl, 3-chloropropionyl, 2,3-dichloropropionyl, 4,4,4-trichlorobutyryl, 4-fluorobutyryl, 5-

chloropentanoyl, 3-chloro-2-methylpropionyl, 6-bromohexanoyl, 5,6-dibromohexanoyl, and the like.

The "aminocarbonyl-lower alkoxy having a lower alkyl substituent" includes a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms which is substituted by an aminocarbonyl group having 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, methylaminocarbonylmethoxy, 1-ethylaminocarbonylethoxy, 2-propylaminocarbonylethoxy, 3-isopropylaminocarbonylpropoxy, 4-butylaminocarbonylbutoxy, 5-pentylaminocarbonylpentyloxy, 6-hexylaminocarbonylpropoxy, dimethylaminocarbonylmethoxy, 3-diethylaminocarbonylpropoxy, diethylaminocarbonylmethoxy, (N-ethyl-N-propylamino)carbonylmethoxy, 2-(N-methyl-N-hexylamino)carbonylethoxy, and the like.

The "carbamoyl-lower alkyl" includes a carbamoyl-substituted alkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, carbamoylmethyl, 2-carbamoylethyl, 1-carbamoyl-ethyl, 3-carbamoylpropyl, 4-carbamoylbutyl, 5-carbamoylpentyl, 6-carbamoylhexyl, 1,1-dimethyl-2-carbamoylethyl, 2-methyl-3-carbamoylpropyl, and the like.

The "amino-lower alkanoyl having optionally a lower alkyl substituent" includes a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, 2-aminoacetyl, 3-aminopropionyl, 2-aminopropionyl,

4-aminobutyryl, 5-aminopentanoyl, 6-aminohexanoyl, 2,2-dimethyl-3-aminopropionyl, 2-methyl-3-aminopropionyl, 2-methyl-aminoacetyl, 2-ethylaminopropionyl, 3-propylaminopropionyl, 3-isopropylaminopropionyl, 4-butylaminobutyryl, 5-pentylaminopentanoyl, 6-hexylaminohexanoyl, 2-dimethylaminoacetyl, 2-diethylaminoacetyl, 2-(N-ethyl-N-propylamino)acetyl, 3-(N-methyl-N-hexylamino)propionyl, and the like.

The "amino-lower alkyl having optionally a lower alkanoyl substituent" includes a straight chain or branched chain alkyl having 1 to 6 carbon atoms which is substituted by an amino group having optionally a substituent of a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, aminomethyl, 2-aminoethyl, 1-aminoethyl, 3-aminopropyl, 4-aminobutyl, 5-aminopentyl, 6-aminohexyl, 1,1-dimethyl-2-aminoethyl, 2-methyl-3-aminopropyl, acetylaminomethyl, 1-acetylaminoethyl, 2-propionylaminoethyl, 3-isopropionylaminopropyl, 4-butyrylaminobutyl, 5-pentanoylaminopropyl, 6-hexanoylaminohexyl, formylaminomethyl, and the like.

The "anilinocarbonyl having optionally a lower alkyl substituent on the phenyl ring" includes an anilinocarbonyl group having optionally 1 to 3 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms on the phenyl ring, for example, anilinocarbonyl, 2-methylanilinocarbonyl, 3-methylanilinocarbonyl, 4-methylanilinocarbonyl, 2-ethylanilinocarbonyl, 3-ethylanilinocarbonyl, 4-ethylanilinocarbonyl, 4-ethylanilinocarbonyl, 4-pentylanilinocarbonyl, 4-hexylanilinocarbonyl, 3,4-dimethyl-

A CONTRACTOR CONTRACTOR

all expenses the grant makes the constitution

anilinocarbonyl, 3,4-diethylanilinocarbonyl, 2,4-dimethyl-anilinocarbonyl, 2,5-dimethylanilinocarbonyl, 2,6-dimethyl-anilinocarbonyl, 3,4,5-trimethylanilinocarbonyl, and the like.

The "phenylsulfonyl which has optionally a substituent selected from a halogen and a lower alkyl on the phenyl ring" includes a phenylsulfonyl group which has optionally 1 to 3 substitutents selected from a straight chain or branched chain alkyl group having 1 to 6 carbon atoms and a halogen atom, for example, phenylsulfonyl, 2-chlorophenylsulfonyl, 3-chlorophenylsulfonyl, 4-chlorophenylsulfonyl, 2-fluorophenylsulfonyl, 3-fluorophenylsulfonyl, 4-fluorophenylsulfonyl, 2-bromophenylsulfonyl, 3-bromophenylsulfonyl, 4-bromophenylsulfonyl, 2-iodophenylsulfonyl, 3-iodophenylsulfonyl, 4-iodophenylsulfonyl, 3,4-dichlorophenylsulfonyl, 3,5-dichlorophenylsulfonyl, 2,6-dichlorophenylsulfonyl, 2,3-dichlorophenylsulfonyl, 2,4-dichlorophenylsulfonyl, 3,4-difluorophenylsulfonyl, 3,5-dibromophenylsulfonyl, 3,4,5-trichlorophenylsulfonyl, 2-ethyl-3-chlorophenylsulfonyl, 2-methylphenylsulfonyl, 3-methylphenylsulfonyl, 4-methylphenylsulfonyl, 2-ethylphenylsulfonyl, 3-ethylphenylsulfonyl, 4-ethylphenylsulfonyl, 4-isopropylphenylsulfonyl, 3butylphenylsulfonyl, 4-pentylphenylsulfonyl, 4-hexylphenylsulfonyl, 3,4-dimethylphenylsulfonyl, 3,4-diethylphenylsulfonyl, 2,4-dimethylphenylsulfonyl, 2,5-dimethylphenylsulfonyl, 2,6-dimethylphenylsulfonyl, 3,4,6-trimethylphenylsulfonyl, 3,4,5-trimethylphenylsulfonyl, 3-chloro-4-methylphenylsulfonyl, 4-methyl-5-iodophenylsulfonyl, 3,4-dimethyl-5-bromophenylsulfonyl, 3,5-diiodo-4-methylphenylsulfonyl,

Subject to the first of the control of the control

and the like.

кон на 1916 г. Вънчина папура и ининерсиция вистему и поружения и постоя в пред постоя в пред постоя в 1910 г.

The "phthalimido-substituted lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by phthalimido group, for example, phthalimidomethyl, 2-phthalimidoethyl, 1-phthalimidoethyl, 3-phthalimidopropyl, 4-phthalimidobutyl, 5-phthalimidopentyl, 6-phthalimidohexyl, 1,1-dimethyl-2-phthalimidoethyl, 2-methyl-3-phthalimidopropyl, and the like.

The "lower alkynyl" includes a straight chain or branched chain alkynyl having 2 to 6 carbon atoms, for example, ethynyl, 2-propynyl, 2-butynyl, 3-butynyl, 1-methyl-2-propynyl, 2-pentynyl, 2-hexynyl, and the like. .

The "benzoyl which has optionally a halogen substituent on the phenyl ring" includes a benzoyl group which has optionally 1 to 3 substituents of a halogen atom on the phenyl ring, for example, benzoyl, 2-chlorobenzoyl, 3-chlorobenzoyl, 4-chlorobenzoyl, 2-fluorobenzoyl, 3-fluorobenzoyl, 4-fluorobenzoyl, 2-bromobenzoyl, 3-bromobenzoyl, 4-bromobenzoyl, 2-iodobenzoyl, 3-iodobenzoyl, 4-iodobenzoyl, 3,4-dichlorobenzoyl, 3,5-dichlorobenzoyl, 2,6-dichlorobenzoyl, 2,3-dichlorobenzoyl, 2,4-dichlorobenzoyl, 3,4-difluorobenzoyl, 3,5-dibromobenzoyl, 3,4,5-trichlorobenzoyl, and the like.

The "phenyl-lower alkoxy" includes a phenylalkoxy group wherein the alkoxy moiety is a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms, for

example, benzyloxy, 2-phenylethoxy, 1-phenylethoxy, 3-phenylpropoxy, 4-phenylbutoxy, 5-phenylpentyloxy, 6-phenyl-hexyloxy, 1,1-dimethyl-2-phenylethoxy, 2-methyl-3-phenyl-propoxy, and the like.

The "amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl" include a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents selected from a straight chain or branched chain alkyl group having 1 to 6 carbon atoms and a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, aminomethoxy, 2-aminoethoxy, 1-aminoethoxy, 3-aminopropoxy, 4-aminobutoxy, 5-aminopentyloxy, 6-aminohexyloxy, 1,1dimethyl-2-aminoethoxy, 2-methyl-3-aminopropoxy, acetylaminomethoxy, 1-acetylaminoethoxy, 2-propionylaminoethoxy, 3-isopropionylaminopropoxy, 4-butyrylaminobutoxy, 5pentanoylaminopentyloxy, 6-hexanoylaminohexyloxy, formylaminomethoxy, methylaminomethoxy, 1-ethylaminoethoxy, 2propylaminoethoxy, 3-isopropylaminopropoxy, 4-butylaminobútoxy, 5-pentylaminopentyloxy, 6-hexylaminohexyloxy, dimethylaminomethoxy, (N-ethyl-N-propylamino)methoxy, 2-(Nmethyl-N-hexylamino)ethoxy, and the like.

The "benzoyloxy which has optionally a halogen substituent on the phenyl ring" includes a benzoyloxy group which has optionally 1 to 3 substituents of a halogen atom on the phenyl ring, for example, benzoyloxy, 2-chloro-

benzoyloxy, 3-chlorobenzoyloxy, 4-chlorobenzoyloxy, 2-fluorobenzoyloxy, 3-fluorobenzoyloxy, 4-fluorobenzoyloxy, 2-bromobenzoyloxy, 3-bromobenzoyloxy, 4-bromobenzoyloxy, 2-iodobenzoyloxy, 3-iodobenzoyloxy, 4-iodobenzoyloxy, 3,4-dichlorobenzoyloxy, 3,5-dichlorobenzoyloxy, 2,6-dichlorobenzoyloxy, 2,3-dichlorobenzoyloxy, 2,4-dichlorobenzoyloxy, 3,4-difluorobenzoyloxy, 3,5-dibromobenzoyloxy, 3,4,5-trichlorobenzoyloxy, and the like.

The "lower alkanoyloxy-substituted lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by a straight chain or branched chain alkanoyloxy group having 2 to 6 carbon atoms, for example, acetyloxymethyl, 2-propion-yloxyethyl, 1-butyryloxyethyl, 3-acetyloxypropyl, 4-acetyloxybutyl, 4-isobutyryloxybutyl, 5-pentanoyloxypentyl, 6-acetyloxyhexyl, 6-tert-butylcarbonyloxyhexyl, 1,1-dimethyl-2-hexanoyloxyethyl, 2-methyl-3-acetyloxypropyl, and the like.

The "lower alkylsulfonyloxy-lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by a straight chain or branched chain alkylsulfonyloxy group having 1 to 6 carbon atoms, for example, methylsulfonyloxymethyl, 1-ethylsulfonyloxyethyl, 2-propylsulfonyloxyethyl, 3-isopropylsulfonyloxypropyl, 4-butylsulfonyloxybutyl, 5-pentylsulfoyloxypentyl, 6-hexylsulfonyloxyhexyl, 1,1-dimethyl-2-methylsulfoyloxypethyl, 2-methyl-3-ethylsulfonyloxypropyl, and the

мор правичный нумемем и нучуры почено одно открынию такражения нуму без одмужения од почено од 1900 година одно

like.

The "azido-lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by an azido group, for example, azidomethyl, 1-azidoethyl, 2-azidoethyl, 3-azidopropyl, 4-azidobutyl, 5-azidopentyl, 6-azidohexyl, 1,1-dimethyl-2-azidoethyl, 2-methyl-3-azidopropyl, and the like.

The "lower alkanoyloxyimino" includes a straight chain or branched chain alkanoyloxyimino group having 1 to 6 carbon atoms, for example, formyloxyimino, acetyloxyimino, propionyloxyimino, butyryloxyimino, isobutyryloxyimino, pentanoyloxyimino, tert-butylcarbonyloxyimino, hexanoyloxyimino, and the like.

The "lower alkylidene" includes a straight chain or branched chain alkylidene group having 1 to 6 carbon atoms, for example, methylidene, ethylidene, propylidene, isopropylidene, butylidene, pentylidene, hexylidene, and the like.

The "oxiranyl-substituted lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by oxiranyl group, for example, oxiranylmethyl, 1-oxiranylethyl, 2-oxiranylethyl, 3-oxiranylpropyl, 4-oxiranylbutyl, 5-oxiranylpentyl, 6-oxiranylhexyl, 1,1-dimethyl-2-oxiranylethyl, 2-methyl-3-oxiranylpropyl, and the like.

The "lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having

4. 19 (1995年) 19 (1995年) - 19 (1995年) 19 (1995年) - 19 (1995年) 19 (1995年) 19 (1995年) 19 (1995年) 19 (1995年)

optionally a lower alkyl substituent" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms and having 1 to 2 substituents selected from a straight chain or branched chain alkoxy group having 1 to 6 carbon atoms, hydroxy and an amino having optionally a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, methoxymethyl, 1-ethoxyethyl, 2propoxyethyl, 3-isopropoxypropyl, 4-butoxybutyl, 5-pentyloxypentyl, 6-hexyloxyhexyl; 1,1-dimethyl-2-methoxyethyl, 2methyl-3-ethoxypropyl, 3-methoxy-2-hydroxypropyl, hydroxymethyl, 2-hydroxyethyl, 1-hydroxyethyl, 3-hydroxypropyl, 2,3-dihydroxyethyl, 4-hydroxybutyl, 3,4-dihydroxybutyl, 1,1dimethyl-2-hydroxyethyl, 5,6-dihydroxyhexyl, 5-hydroxypentyl, 6-hydroxyhexyl, 6-(N-ethyl-N-methylamino)-5-methoxyhexyl, 2-methyl-3-hydroxypropyl, aminomethyl, 1-aminoethyl, 2-aminoethyl, 3-aminopropyl, 4-aminobutyl, 5-aminopentyl, 6aminohexyl, 1,1-dimethyl-2-aminoethyl, 2-methyl-3-aminopropyl, methylaminomethyl, ethylaminomethy, propylaminomethyl, isopropylaminomethyl, butylaminomethyl, tertbutylaminomethyl, pentylaminomethyl, hexylaminomethyl, dimethylaminomethyl, diethylaminomethyl, dipropylaminomethyl, dibutylaminomethyl, dipentylaminomethyl, dihexylaminomethyl, N-methyl-N-ethylaminomethyl, N-methyl-Npropylaminomethyl, N-methyl-N-butylaminomethyl, N-methyl-Nhexylaminomethyl, 1-methylaminoethyl, 2-ethylaminoethyl, 3propylaminopropyl, 4-butylaminobutyl, 1,1-dimethyl-2-pentylaminoethyl, 5-hexylaminopentyl, 6-dimethylaminohexyl, 4-

and with the first program water of MetMontage particles of the controlling good according to the controlling and the controll

Charles to the Bull of English Co.

dimethylaminobutyl, 2-diethylaminoethyl, 1-(N-methyl-N-hexylamino)ethyl, 3-dihexylaminopropyl, 6-diethylaminohexyl, 4-dibutylaminobutyl, 2-(N-methyl-N-pentylamino)ethyl, 2-hydroxy-3-diethylaminopropyl, 3-hydroxy-4-methylaminobutyl, 5-hydroxy-6-diethylaminohexyl, 4-hydroxy-5-dimethylaminopentyl, 4-hydroxy-5-diethylaminopentyl, 4-hydroxy-5-diethylaminopentyl, 5-hydroxy-6-ethylaminohexyl, 5-hydroxy-6-isopropylaminohexyl, 5-hydroxy-6-aminohexyl, and the like.

The "aminocarbonyloxy having optionally a lower alkyl substituent" includes an aminocarbonyloxy group having optionally 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, aminocarbonyloxy, methylaminocarbonyloxy, ethylaminocarbonyloxy, propylaminocarbonyloxy, isopropylaminocarbonyloxy, butylaminocarbonyloxy, tert-butylaminocarbonyloxy, pentylaminocarbonyloxy, hexylaminocarbonyloxy, dimethylaminocarbonyloxy, diethylaminocarbonyloxy, dipropylaminocarbonyloxy, dibutylaminocarbonyloxy, dipentylaminocarbonyloxy, dihexylaminocarbonyloxy, N-methyl-N-ethylaminocarbonyloxy, N-methyl-N-propylaminocarbonyloxy, N-methyl-N-butyl-aminocarbonyloxy, N-methyl-N-hexylaminocarbonyloxy, and the like.

The "lower alkanoyloxy having optionally a halogen substituent" includes a straight chain or branched chain alkanoyloxy group having 1 to 6 carbon atoms which has optionally 1 to 3 substituents . a halogen atom, for example, in addition to the above lower alkanoyl group, 2,2,2-trifluoroacetyloxy, 2,2,2-trichloroacetyloxy, 2-chloroacetyloxy, 2-bromoacetyloxy, 2-fluoroacetyloxy, 2-iodoacetyloxy, 2,2-difluoroacetyloxy, 2,2-fluoroacetyloxy, 2,2-

Carrier & Configuration of the State of State of the Stat

dibromoacetyloxy, 3,3,3-trifluoropropionyloxy, 3,3,3-trichloropropionyloxy, 3-chloropropionyloxy, 2,3-dichloropropionyloxy,
4,4,4-trichlorobutyryloxy, 4-fluorobutyryloxy, 5-chloropentanoyloxy, 3-chloro-2-methylpropionyloxy, 6-bromohexanoyloxy, 5,6dibromohexanoyloxy, and the like.

The "amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl" include a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents selected from a straight chain or branched chain alkyl group having 1 to 6 carbon atoms and a straight chain or branched chain alkanoyl group having 1 to 6 carbon atoms, for example, aminomethyl, 2-aminoethyl, 1-aminoethyl, 3-aminopropyl, 4-aminobutyl, 5-aminopentyl, 6-aminohexyl, 1,1-dimethyl-2-aminoethyl, 2-methyl-3-aminopropyl, acetylaminomethyl, 1-acetylaminoethyl, 2-propionylaminoethyl, 3-isopropionylaminopropyl, 4butyrylaminobutyl, 5-pentanoylaminopentyl, 6-hexanoylaminohexyl, formylaminomethyl, methylaminomethyl, 1-ethylaminoethyl, 2propylaminoethyl, 3-isopropylaminopropyl, 4-butylaminobutyl, 5pentylaminopentyl, 6-hexylaminohexyl, dimethylaminomethyl, (Nethyl-N-propylamino)methyl, 2-(N-methyl-N-hexylamino)ethyl, and the like.

The "amino-lower alkanoyloxy having optionally a lower alkyl substituent" includes a straight chain or branched chain alkanoyloxy having 2 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms,

for example, 2-aminoacetyloxy, 3-aminopropionyloxy, 2-aminopropionyloxy, 4-aminobutyryloxy, 5-aminopentanoyloxy, 6-aminohexanoyloxy, 2,2-dimethyl-3-aminopropionyloxy, 2-methyl-3-aminopropionyloxy, 2-methylaminoacetyloxy, 2-ethylaminopropionyloxy, 3-propylaminopropionyloxy, 3-isopropylaminopropionyloxy, 4-butyl-aminobutyryloxy, 5-pentylaminopentanoyloxy, 6-hexylaminohexanoyloxy, 2-dimethylaminoacetyloxy, 2-diethylaminoacetyloxy, 2-(N-ethyl-N-propylamino)acetyloxy, 3-(N-methyl-N-hexylamino)-propionyloxy, and the like.

The "pyridyl-lower alkyl" include a pyridylalkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, (4-pyridyl)-methyl, 1-(3-pyridyl)ethyl, 2-(2-pyridyl)ethyl, 3-(2-pyridyl)-propyl, 4-(3-pyridyl)butyl, 5-(4-pyridyl)pentyl, 6-(2-pyridyl)hexyl, 1,1-dimethyl-2-(3-pyridyl)ethyl, 2-methyl-3-(4-pyridyl)propyl, and the like.

The "5- or 6-membered saturated heterocyclic group which is formed by binding the groups R⁸² and R⁸³ together with the nitrogen atom to which they bond with or without being intervened with nitrogen, oxygen or sulfur atom" includes, for example, pyrrolidinyl, piperidinyl, piperazinyl, morpholino, thio-morpholino, and the like.

The above heterocyclic group which has a substituent selected from oxo, a lower alkyl, a lower alkanoyl and carbamoyl includes the above heterocyclic groups which have 1 to 3 substituents selected from oxo, a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, a straight chain or

branched chain alkanoyl group having 1 to 6 carbon atoms, and carbamoyl group, for example, 4-methylpiperazinyl, 3,4-dimethyl-piperazinyl, 3-ethylpyrrolidinyl, 2-propylpyrrolidinyl, 3,4,5-trimethylpiperidinyl, 4-butylpiperidinyl, 3-pentylmorpholino, 4-hexylpiperazinyl, 2-methylthiomorpholino, 4-acetylpiperazinyl, 2-propionylmorpholino, 3-butyrylthiomorpholino, 3-pentanoyl-pyrrolidinyl, 4-hexanoylpiperidinyl, 3-methyl-4-acetylpiperazin-yl, 2-carbamoylpyrrolidinyl, 4-carbamoylpiperazinyl, 3-carbamoyl-thiomorpholino, 2-carbamoylmorpholino, 3-carbamoylpiperidinyl, 1-oxo-thiomorpholino, 1,1-dioxothiomorpholino, and the like.

The "lower alkylsulfonyl" includes a straight chain or branched chain alkylsulfonyl group having 1 to 6 carbon atoms, for example, methylsulfonyl, ethylsulfonyl, propylsulfonyl, isopropylsulfonyl, butylsulfonyl, tert-butylsulfonyl, pentylsulfonyl, hexylsulfonyl, and the like.

The "aminocarbonyl having optionally a lower alkyl substituent" includes an aminocarbonyl group having optionally 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, aminocarbonyl, methylaminocarbonyl, ethylaminocarbonyl, propylaminocarbonyl, isopropylaminocarbonyl, butylaminocarbonyl, tert-butylaminocarbonyl, pentylaminocarbonyl, hexylaminocarbonyl, dimethylaminocarbonyl, diethylaminocarbonyl, dipropylaminocarbonyl, dibutyl-aminocarbonyl, dipentylaminocarbonyl, dihexylaminocarbonyl, N-methyl-N-ethylaminocarbonyl, N-ethyl-N-propylaminocarbonyl, N-methyl-N-butylaminocarbonyl, N-methyl-N-hexylaminocarbonyl, and the like.

and the first of the pateronal performance and part of the first of the contract of the contra

The "cyano-substituted lower alkyl" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by cyano group, for example, cyanomethyl, 2-1 cyanoethyl, 1-cyanoethyl, 3-cyanopropyl, 4-cyanobutyl, 5-cyanopentyl, 6-cyanohexyl, 1,1-dimethyl-2-caynoethyl, 2-methyl-3-cyanopropyl, and the like.

The "lower alkoxycarbonyl-substituted lower alkyl" includes an alkoxycarbonyl-substituted straight chain or branched chain alkyl group having 1 to 6 carbon atoms wherein the alkoxycarbonyl moiety is a straight chain or branched chain alkoxycarbonyl group having 1 to 6 carbon atoms, for example, methoxycarbonylmethyl, 3-methoxycarbonylpropyl, ethoxycarboxymethyl, 3-ethoxycarbonylpropyl, 4-ethoxycarbonylbutyl, 5-isopropoxycarbonylpentyl, 6-propoxycarbonylhexyl, 1,1-dimethyl-2-butoxycarbonylethyl, 2-methyl-3-tert-butoxycarbonylpropyl, 2-pentyloxycarbonylethyl, hexyloxycarbonylmethyl, and the like.

The "carboxy-substituted lower alkyl" includes a carboxy-substituted alkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, carboxymethyl, 2-carboxyethyl, 1-carboxy-ethyl, 3-carboxypropyl, 4-carboxybutyl, 5-carboxypentyl, 6-carboxyhexyl, 1,1-dimethyl-2-carboxyethyl, 2-methyl-3-carboxy-propyl, and the like.

The "tetrahydropyranyloxy-substituted lower alkyl" includes a tetrahydropyranyloxy-substituted straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, (2-tetrahydropyranyloxy)methyl, 2-(3-tetrahydropyranyl-

inativa teste attendida terban til samma meneral og delen til til en er er er enger delen er er er er er er er

oxy)ethyl, 1-(4-tetrahydropyranyloxy)ethyl, 3-(2-tetrahydropyranyloxy)propyl, 4-(3-tetrahydropyranyloxy)butyl, 5-(4-tetrahydropyranyloxy)pentyl, 6-(2-tetrahydropyranyloxy)hexyl, 1,1-dimethyl-2-(3-tetrahydropyranyloxy)ethyl, 2-methyl-3-(4-tetrahydropyranyloxy)propyl, and the like.

The "piperidinyl having optionally a phenyl-lower alkyl substituent" includes a piperidinyl which has optionally a substituent of a phenylalkyl group wherein the alkyl moiety is a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, piperidinyl, 1-benzyl-4-piperidinyl, 1-(2-phenylethyl)-3-piperidinyl, 1-(1-phenylethyl)-2-piperidinyl, 1-(3-phenylpropyl)-4-piperidinyl, 1-(4-phenylbutyl)-4-piperidinyl, 1-(5-phenylpentyl)-4-piperidinyl, 1-(6-phenylhexyl)-4-piperidinyl, 1-(1,1-dimethyl-2-phenylethyl)-3-piperidinyl, 1-(2-methyl-3-phenylpropyl)-2-piperidinyl, and the like.

The "imidazolyl-substituted lower alkanoyl" includes an imidazolyl-substituted alkanoyl group wherein the alkanoyl moiety is a straight chain or branched chain alkanoyl group having 2 to 6 carbon atoms, for example, (1-imidazolyl)acetyl, 3-(2-imidazolyl)propionyl, 2-(4-imidazolyl)propionyl, 4-(1-imidazolyl)butyryl, 2,2-dimethyl-3-(2-imidazolyl)propionyl, 5-(4-imidazolyl)-pentanoyl, 6-(1-imidazolyl)hexanoyl, and the like.

The "amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl" includes a straight chain or branched chain alkanoyl having 2 to 6 carbon atoms which is substituted by an amino group having optionally 1 to 2 substituents selected from a straight chain or

Company of the Alberta Company of

branched chain a kyl group having 1 to 6 carbon atoms and a straight chain or branched chain alkoxycarbonyl group having 1 to 6 carbon atoms, for example, 2-aminoacetyl, 3-aminopropionyl, 2-aminopropionyl, 4-aminobutyryl, 5-aminopentanoyl, 6-aminohexanoyl, 2,2-dimethyl-3-aminopropionyl, 2-methyl-3-aminopropionyl, 2-methylaminoacetyl, 2-ethylaminopropionyl, 3-propylaminopropionyl, 3-isopropylaminopropionyl, 4-butylaminobutyryl, 5-pentylaminopentanoyl, 6hexylaminohexanoyl, 2-dimethylaminoacetyl, 2-diethylaminoacetyl, 2-(N-ethyl-N-propylamino)acetyl, 3-(N-methyl-Nhexylamino)propionyl, 2-methoxycarbonylaminoacetyl, 2ethoxycarbonylaminoacetyl, 3-propoxycarbonylaminopropionyl, 4-butoxycarbonylaminobutyryl, 2-tert-butoxycarbonylaminoacetyl, 5-pentyloxycarbonylaminopentanoyl, 6-hexyloxycarbonylaminohexanoyl, 2-(N-methyl-N-tert-butoxycarbonylamino)acetyl, and the like.

The "aminocarbonyl-lower alkyl having a lower alkyl substituent" includes a straight chain or branched chain alkyl group having 1 to 6 carbon atoms which is substituted by an aminocarbonyl group having 1 to 2 substituents of a straight chain or branched chain alkyl group having 1 to 6 carbon atoms, for example, methylaminocarbonylmethyl, 1-ethyl-aminocarbonylethyl, 2-propylaminocarbonylethyl, 3-isopropyl-aminocarbonylpropyl, 4-butylaminocarbonylbutyl, 5-pentylaminocarbonylpentyl, 6-hexylaminocarbonylhexyl, dimethylaminocarbonylmethyl, 3-diethylaminocarbonylpropyl, diethylaminocarbonylmethyl, (N-ethyl-N-propylamino)carbonylmethyl, 2-(N-carbonylmethyl, (N-ethyl-N-propylamino)carbonylmethyl, 2-(N-carbonylmethyl, 2-(N

methyl-N-hexylamino)carbonylethyl, and the like.

The "amino-substituted lower alkoxy having optionally a lower alkyl substituent" includes an amino-substituted straight chain or branched chain alkoxy having 1 to 6 carbon atoms which has optionally 1 to 2 substituents of a straight chain or branched chain alkyl having 1 to 6 carbon atoms, such as aminomethoxy, 2-aminoethoxy, 1-aminoethoxy, 3-aminopropoxy, 4-aminobutoxy, 5-aminopentyloxy, 6-aminohexyloxy, 1,1-dimethyl-2-aminoethoxy, 2-methyl-3-aminopropoxy, methylaminomethoxy, 1-ethylaminoethoxy, 2-propyl-aminoethoxy, 3-isopropylaminopropoxy, 4-butylaminobutoxy, 5-pentylaminopentyloxy, 6-hexylaminohexyloxy, dimethylaminomethoxy, (N-ethyl-N-propylamino)methoxy, 2-(N-methyl-N-hexylamino)ethoxy, and the like.

The compounds of the present invention can be prepared by various processes, for example, by the processes shown in the following reaction schemes.

[Reaction Scheme-1]

$$\mathbb{R}^{1} \xrightarrow{\mathbb{N}} \mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{3}$$

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^3 , and W are the same as defined above. The process of Reaction Scheme-1 is carried out by

reacting a benzoheterocyclic compound of the formula (2) and a carboxylic acid compound of the formula (3) by a conventional amido bond forming reaction. The amido bond forming reaction can be carried out under the conditions for the conventional amido bond forming reaction, for example,

- (a) a mixed acid anhydride process, i.e. a process of reacting the carboxylic acid compound (3) with an alkylhalocarboxylic acid to form a mixed acid anhydride and reacting the resultant with the amine compound (2),
- (b) an activated ester process, i.e. a process of converting the carboxylic acid compound (3) into an activated ester, such as p-nitrophenyl ester, N-hydroxy-succinimide ester, 1-hydroxybenzotriazole ester, etc., and reacting the resultant with the amine compound (2),
- (c) a carbodiimide process, i.e. a process of condensing the carboxylic acid compound (3) and the amine compound (2) in the presence of an activating agent such as dicyclohexylcarbodiimide, carbonyldiimidazole, etc.,
- (d) other processes, i.e. a process of converting the carboxylic acid compound (3) into a carboxylic anhydride by treatment with a dehydrating agent such as acetic anhydride, and reacting the resultant with the amine compound (2); a process of reacting an ester of the carboxylic acid compound (3) with a lower alcohol and the amine compound (?) at a high temperature under high pressure; a process of reacting an acid halide compound of the carboxylic acid compound (3), i.e. a carboxylic acid

indicated by the Control of the property of the control of the con

halide, with the amine compound (2), and the like.

The mixed acid anhydride used in the above mixed acid anhydride process (a) is obtained by the known Schötten-Baumann reaction, and the reaction product is used without isolation from the reaction mixture for the reaction with the amine compound (2) to give the desired compound of the formula (1). The Schötten-Baumann reaction is usually carried out in the presence of a basic compound. The basic compound is any conventional compounds used for the Schötten-Baumann reaction and includes, for example, organic basic compounds such as triethylamine, trimethylamine, pyridine, dimethylaniline, N-methylmorpholine, 1,5-diazabicyclo[4.3.0]nonene-5 (DBN), 1,8-diazabicyclo[5.4.0]undecene-7 (DBU), 1,4-diazabicyclo[2.2.2]octane (DABCO), etc., and inorganic basic compounds such as potassium carbonate, sodium carbonate, potassium hydrogen carbonate, sodium hydrogen carbonate, etc. The reaction is usually carried out at a temperature of from about -20°C to about 100°C, preferably from about 0°C to about 50°C, for about 5 minutes to about 10 hours, preferably about 5 minutes to about 2 hours.

The reaction of the thus obtained mixed acid anhydride with the amine compound (2) is usually carried out at a temperature of from about -20°C to about 150°C, preferably about 10°C to about 50°C, for about 5 minutes to about 10 hours, preferably about 5 minutes to about 5 hours. The mixed acid anhydride process is usually carried

out in an appropriate solvent. The solvent is any conventional solvents which are usually used in the mixed acid anhydride process and includes, for example, halogenated hydrocarbons (e.g. chloroform, dichloromethane dichloroethane, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), ethers (e.g. diethyl ether, diisopropyl ether, tetrahycrofuran, dimethoxyethane, etc.), esters (e.g. methyl acetate, ethyl acetate, etc.), aprotic polar solvents (e.g. N, N-dimethylformamide, dimethylsulfoxide, hexamethylphosphoric triamide, etc.), or a mixture of these solvents. The alkylhalocarboxylic acid used in the mixed acid anhydride process includes, for example, methyl chloroformate, methyl bromoformate, ethyl chloroformate, ethyl bromoformate, isobutyl chloroformate, and the like. In said process, the carboxylic acid compound (3), the alkylhalocarboxylic acid and the amine (2) are usually used in each equimolar amount, but preferably, the alkylhalocarboxylic acid and the carboxylic acid compound (3) are used each in an amount of about 1 to 1.5 mole to 1 mole of the amine (2).

Among the above other processes (d), in case of the process of reacting the carboxylic acid halide with the amine compound (2), the reaction is usually carried out in the presence of a basic compound in an appropriate solvent. The basic compound is any conventional compounds and includes, in addition to the basic compounds used for the above-mentioned Schötten-Baumann reaction, sodium hydroxide,

potassium hydroxide, sodium hydride, potassium hydride. etc. The solvent includes, in addition to the solvents used for the above-mentioned mixed acid anhydride process, alcohols (e.g. methanol, ethanol, propanol, butanol, 3-methoxy-1-butanol, ethylcellosolve, methylcellosolve, etc.), aceto-nitrile, pyridine, acetone, water, and the like. The amount of the amine compound (2) and the carboxylic acid halide is not critical, but the carboxylic acid halide is usually used at least in equimolar amount, preferably about 1 to 5 moles to 1 mole of the amine compound (2). The reaction is usually carried out at a temperature of from about -20°C to about 180°C, preferably from about 0°C to about 150°C, for about 5 minutes to about 30 hours.

The amido bond forming reaction in the above Reaction Scheme-1 may also be carried out by reacting the carboxylic acid compound (3) and the amine (2) in the presence of a condensation agent, i.e. phosphoric compounds such as triphenylphosphine, diphenylphosphinyl chloride, phenyl-N-phenylphosphoramide chloridate, diethyl chlorophosphate, diethyl phosphorocyanidate, diphenylphosphoric azide, bis(2-oxo-3-oxazolidinyl)phosphinic chloride, etc. The reaction is usually carried out in the presence of the solvent and basic compound as used in the above reaction of the carboxylic acid halide and the amine (2) at a temperature of from about -20°C to about 150°C, preferably about 0°C to about 100°C, for about 5 minutes to about 30 hours. The condensation agent and the carboxylic acid compound (3)

are used at least in equimolar amount, preferably about 1 to 2 moles, to 1 mole of the amine (2).

[Reaction Scheme-2]

$$R^{1}$$
 N
 R^{5a}
 R^{1}
 R^{2}
 R^{4}
 R^{4}
 R^{5a}
 R^{4}
 R^{5a}
 R^{5a}
 R^{5a}
 R^{5a}
 R^{5a}
 R^{5a}

wherein R^1 , R^2 , R^4 and W are as defined above, R^{5a} is the same as R^5 as defined above except excluding an anilino-carbonyl having optionally a lower alkyl substituent on the phenyl ring, a phenylsulfonyl having optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring and quinolylsulfonyl.

The reaction of the compound (2b) and the compound (4) is carried out in the same manner as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-3]

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{N} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{2} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{N} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \\
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{1} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R}^{2} \longrightarrow \mathbb{R}^{2}$$

$$\begin{array}{c}
\mathbb{R$$

wherein R^1 , R^2 , R^{11} , R^{12} and W are as defined above.

The reaction of the compound (5) and the compound (6) is carried out under the same conditions as used in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-4]

wherein R^1 , R^2 , R^5 and W are as defined above, and R^{4a} is a lower alkyl, R^{17} and R^{18} are each hydrogen atom or a lower alkyl, and X is a halogen atom.

The reaction of the compound (7) and the compound (8) is usually carried out in an inert solvent in the

presence or absence of a basic compound. The inert solvent includes, for example, aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), ethers (e.g. tetrahydrofuran, dioxane, diethylene glycol dimethyl ether, etc.), halogenated hydrocarbons (e.g. dichloromethane, chloroform, carbon tetrachloride, etc.), lower alcohols (e.g. methanol, ethanol, isopropanol, butanol, tert-butanol, etc.), acetic acid, ethyl acetate, acetone, acetonitrile, pyridine, dimethylsulfoxide, dimethylformamide, hexamethylphosphoric triamide, etc., or a mixture of these solvents. The basic compound includes, for example, carbonates (e.g. sodium carbonate, potassium carbonate, sodium hydrogen carbonate, potassium hydrogen carbonate, etc.), metal hydroxides (e.g. sodium hydroxide, potassium hydroxide, etc.), sodium hydride, potassium, sodium, sodium amide, metal alcoholates (e.g. sodium methoxide, sodium ethoxide, etc.), and organic basic compounds (e.g. pyridine, N-ethyldiisopropylamine, dimethylaminopyridine, triethylamine, 1,5-diazabicyclo-[4.3.0]nonene-(5) (DBN), 1,8-diazabicyclo[5.4.0]undecene-7 (DBU), 1,4-diazabicyclo[2.2.2]octane (DABCO), etc.). The amount of the compound (7) and the compound (8) is not critical, but the compound (8) is usually used at least in equivalent amount, preferably 1 to 10 moles, to 1 mole of the compound (7). The reaction is usually carried out at a temperature of from about 0°C to about 200°C, preferably from about 0°C to about 170°C, for about 30 minutes to about In the reaction, an alkali metal halide (e.g.

sodium iodide, potassium iodide, etc.) may be added to the reaction system.

The reaction of the compound (7) and the compound (9) is carried out in an appropriate solvent or without solvent in the presence of a reducing agent. The solvent includes, for example, water, alcohols (e.g. methanol, ethanol, isopropanol, etc.), acetonitrile, formic acid, acetic acid, ethers (e.g. dioxane, diethyl ether, diglyme, tetrahydrofuran, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), or a mixture of these solvents. The reducing agent includes, for example, formic acid, fatty acid alkali metal salts (e.g. sodium formate, etc.), hydrogenating reducing agents (e.g. sodium boro hydride, sodium cyanoboro hydride, lithium aluminum hydride, etc.), catalystic reducing agents (e.g. palladium black, palladium-carbon, platinum oxide, platinum black, Raney nickel, etc.).

When formic acid is used as the reducing agent, the reaction is usually carried out at a temperature of from room temperature to about 200°C, peferably about 50°C to about 150°C, for about 1 to 10 hours. The formic acid is usually used in a large excess amount to the compound (7).

When a hydrogenating reducing agent is used, the reaction is usually carried out at a temperature of about -30°C to about 100°C, preferably about 0°C to about 70°C, for about 30 minutes to about 12 hours. The reducing agent is usually used in an amount of 1 to 20 moles, preferably 1

to 6 moles, to 1 mole of the compound (7). When lithium aluminum hydride is used as the reducing agent, it is preferable to use a solvent selected from ethers (e.g. diethyl ether, dioxane, tetrahydrofuran, diglyme, etc.) and aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.).

When a catalytic reducing agent is used, the reaction is usually carried out under atmospheric pressure to about 20 atm., preferably atmospheric pressure to about 10 atm. under hydrogen atmosphere or in the presence of a hydrogen donor (e.g. formic acid, ammonium formate, cyclo-hexene, hydrazine hydrate, etc.) at a temperature of about -30°C to about 100°C, preferably about 0°C to about 60°C, for about 1 to 12 hours. The catalytic reducing agent is usually used in an amount of about 0.1 to 40 % by weight, preferably about 1 to 20 % by weight, of the amount of the compound (7). The compound (9) is usually used at least in equivalent amount, preferably equivalent to a large excess amount, to the compound (7).

[Reaction Scheme-5A]

wherein R^1 , R^2 , R^{12} , R^{17} , R^{18} , X and W are as defined above, and R^{11a} is a lower alkyl.

[Reaction Scheme-5B]

wherein R^1 , R^2 , R^{11} , X and W are as defined above, and R^{12a} is a cycloalkyl.

The reaction of the compound (10) and the compound (11) in the Reaction Scheme-5A and the reaction of the compound (12) and the compound (13) in the Reaction Scheme-5B are carried out in the same manner as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

Besides, the reaction of the compound (10) and the compound (9) in the Reaction Scheme-5A is carried out in the same manner as in the reaction of the compound (7) and the compound (9) in the above Reaction Scheme-4.

¡Reaction Scheme-6A]

$$R^{1} \xrightarrow{W} N$$

$$CO$$

$$HN \xrightarrow{R^{6}} (14)$$

$$N-R^{4}$$

$$CO$$

$$(R^{16})_{2}$$

$$(O-A-X)_{2}$$

$$(1g)$$

$$R^{1} \xrightarrow{W} N$$

$$R^{2} \xrightarrow{N-R^{4}}$$

$$(O-A-N)_{R^{7}} (14)$$

wherein R^1 , R^2 , R^4 , R^{16} , R^6 , R^7 , X, W, and A are as defined above, ℓ is 0 or an integer of 1 to 3, ℓ' and ℓ'' are each an integer of 1 to 3, provided that $\ell + \ell'$ and $\ell + \ell''$ are each an integer not more than 3.

de las arganismentes en elegagarisma, es elegena arragaga, especies especies de la compressión del compressión de la com

W 12% 1

[Reaction Scheme-6B]

$$R^{1} \xrightarrow{N} R^{2} \xrightarrow{R^{19}H (15)} R^{2}$$

$$R^{1} \xrightarrow{N} R^{2}$$

$$R^{20}M (16)$$

$$R^{2$$

wherein R^1 , R^2 , R^4 , R^{16} , X, W, A, £, £', and £" are as defined above, and R^{19} is a lower alkanoyloxy, R^{20} is a lower alkanoyloxy, hydroxy or phthalimido, R^{21} is the same as as R^{19} and R^{20} , and M is an alkali metal (e.g. potassium, sodium, etc.).

The reaction of the compound (lg) and the compound (l4) in the Reaction Scheme-6A and the reaction of the compound (lg) and the compound (l5) or (l6) in the Reaction Scheme-6B can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. In the reaction, an alkali metal halide (e.g. sodium iodide, potassium iodide, etc.) may be added to the reaction system.

and the second

the term of the control of the

.

[Reaction Scheme-7]

wherein R^1 , R^2 , R^4 , R^{16} , W, ℓ , ℓ' , ℓ'' and A are as defined above.

The reaction of converting the compound (lj) into the compound (lk) can be carried out by reacting the compound (lj) with hydrazine in an appropriate solvent or by hydrolyzing the compound (lj). The solvent used in the reaction with hydrazine includes water and further the same solvent as used in the reaction of the compound (2b) and the compound (4) in the above Reaction Scheme-2. The reaction is usually carried out at a temperature of from room temperature to about 120°C, preferably about 0°C to about 100°C, for about 0.5 to 5 hours. Hydrazine is usually used in an amount of at least 1 mole, preferably about 1 to 5 moles, to 1 mole of the compound (lj).

et a tiger a teambre service a support timo di en a a a a transportation de grappio de transportation de la pr

The hydrolysis can be carried out in an appropriate solvent or without solvent in the presence of an acid or a basic compound. The solvent includes, for example, water, lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), ketones (e.g. acetone, methyl ethyl ketone, etc.), ethers (e.g. dioxane, tetrahydrofuran, ethylene glycol dimethyl ether, etc.), fatty acids (e.g. acetic acid, formic acid, etc.), or a mixture of these solvents. The acid includes, for example, mineral acids (e.g. hydrochloric acid, sulfuric acid, hydrobromic acid, etc.) and organic acids (e.g. formic acid, acetic acid, aromatic sulfonic acids, etc.). The basic compound includes, for example, metal carbonates (e.g. sodium carbonate, potassium carbonate, etc.), metal hydroxides (e.g. sodium hydroxide, potassium hydroxide, calcium hydroxide, etc.), and the like. The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from room temperature to about 150°C, for about 10 minutes to 25 hours.

Ĭŗ,

[Reaction Scheme-8]

$$R^{1}$$
 R^{2}
 R^{2

wherein R^1 , R^2 , R^4 , W, R^{16} , ℓ , ℓ , ℓ , ℓ , χ , and A are as defined above, and R^{22} is a lower alkanoyl.

The reaction of the compound (1%) and the compound (17) is carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the Reaction Scheme-4. In the reaction, an alkali metal halide (e.g. sodium iodide, potassium iodide, etc.) may be added to the reaction system.

The reaction of converting the compound (lm) into the compound (lt) can be carried out under the same condition as in the hydrolysis of the compound (lj) in the Reaction Scheme-7.

en en en koloniste de meneren kom engelen meneren en en en en koloniste paramanan en gregoriet. Det øge

[Reaction Scheme-9]

$$R^{1} \xrightarrow{W} \qquad \qquad R^{2} \xrightarrow{\mathbb{R}^{2}} \times (18)$$

$$R^{2} \xrightarrow{\mathbb{R}^{2}} \mathbb{R}^{2}$$

$$(OH)_{g} \qquad \qquad (O-\mathbb{R}^{23})_{g} \qquad (10)$$

wherein R^1 , R^2 , R^4 , W, R^{16} , ℓ , ℓ' , ℓ'' , ℓ''' , and X are as defined above, and R^{23} is a lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a halogen-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a phthalimido-substituted lower alkyl, an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, or a group of the formula: -A-N, R^6 and R^7 are as defined above).

The reaction of the compound (ln) and the compound (l8) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. In the reaction, an alkali metal halide (e.g. sodium iodide, potassium iodide, etc.) may be

added to the reaction system.

[Reaction Scheme-10]

wherein R^1 , R^2 , R^4 , W, R^{16} , R^{17} , R^{18} , ℓ , X, and A are as defined above, and R^6 is hydrogen atom, a lower alkyl

and the first angular to the control of the first of the control of the water grows and represent with a grown and the

having optionally a hydroxy substituent, a lower alkanoyl, or benzoyl, R^{7a} is a lower alkyl having optionally a hydroxy substituent, and R^{7b} is a lower alkanoyl or benzoyl.

The reaction of the compound (1p) and the compound (19) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4.

The reaction of the compound (1p) and the compound (20) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the Reaction Scheme-1.

Besides, the compound (lr) can also be obtained by reacting the compound (lp) with a compound of the formula: $(R^{7b})_{20}$ (R^{7b} is as defined above). The reaction can be carried out in an appropriate solvent or without solvent in the presence or absence, peferably presence, of a basic compound. The solvent includes, for example, the abovementioned aromatic hydrocarbons, lower alcohols (e.g. methanol, ethanol, propanol, etc.), dimethylformamide, dimethylsulfoxide, and further halogenated hydrocarbons (e.g. chloroform, methylene chloride, etc.), acetone, The basic compound includes, for example, pyridine, etc. tertiary amines (e.g. triethylamine, pyridine, etc.), sodium hydroxide, potassium hydroxide, sodium hydride, and the like. The above reaction can also be carried out in a solvent such as acetic acid or benzoic acid in the presence of a mineral acid (e.g. sulfuric acid, etc.). The acid

anhydride is usually used in an equimolar amount or more, preferably 1 to 10 moles, to 1 mole of the starting compound, and the reaction is usually carried out at a temperature of about 0°C to about 200°C, preferably from about 0°C to about 150°C, for about 0.5 to 15 hours.

[Reaction Scheme-11]

wherein R^1 , R^2 , R^4 , R^9 , R^{10} , W, and B are as defined above.

The reaction of the compound (1s) and the compound (21) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-12]

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 , \mathbb{W} , \mathbb{R}^9 , \mathbb{R}^{10} , \mathbb{X} , and \mathbb{B} are as defined above.

The reaction of the compound (lu) and the compound (21) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. In the reaction, an alkali metal halide (e.g. sodium iodide, potassium iodide, etc.) may be added to the reaction system.

[Reaction Scheme-13]

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 , W, and B are as defined above, and \mathbb{R}^{24} is a lower alkyl.

The reaction of the compound (2b) and the compound (22) can be carried out in an appropriate inert solvent.

The inert solvent includes, for example, aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), ethers (e.g. tetrahydrofuran, dioxane, diethylene glycol dimethyl

ether, etc.), lower alcohols (e.g. methanol, ethanol, isopropanol, butanol, etc.), halogenated hydrocarbons (e.g. dichloromethane, chloroform, carbon tetrachloride, etc.), acetic acid, ethyl acetate, acetonitrile, dimethylsulfoxide, dimethylformamide, hexamethylphosphoric triamide, and the like. The amount of the compound (2b) and the compound (22) is not critical, but the compound (22) is usually used in an amount of at least one mole, preferably 1 to 2 moles, to 1 mole of the compound (2b). The reaction is usually carried out at a temperature of from about 0°C to about 150°C, preferably from about 0°C to about 100°C, for about 30 minutes to about 10 hours.

The esterification of the compound (lw) is usually carried out by reacting the starting compound with an alcohol (e.g. methanol, ethanol, isopropanol, etc.) in the presence of a mineral acid (e.g. hydrochloric acid, sulfuric acid, etc.) and a halogenating agent (e.g. thionyl chloride, phosphorus oxychloride, phosphorus pentachloride, phosphorus trichloride, etc.) at a temperature of 0°c to 150°C, preferably 50°C to 100°C, for about 1 to 10 hours.

The hydrolysis of the compound (lx) can be carried out under the same conditions as in the hydrolysis of the compound (lj) in the Reaction Scheme-7.

[Reaction Scheme-14]

wherein R^1 , R^2 , R^4 , W, B, M, and X are as defined above, and R^{25} is a phenyl which has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, or naphthyl, and R^{25} is a phenoxy which has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, naphthyloxy or phthalimido.

The reaction of the compound (lu) and the compound (23) or (23a) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The compound (ly) wherein R^{25} is phthalimido can be converted into the compound (ly) wherein R^{25} is amino under the same conditions as in the reaction of converting the compound (lj) into the compound (lk) in the above

Reaction Scheme-7.

[Reaction Scheme-15]

wherein R^1 , R^2 and R^3 are as defined above, and R^{26} is oxo, R^{27} is hydroxy, and W' is the same as W, provided that the substituents on the group $-(CH_2)_p$ or $-CH=CH-(CH_2)_q$ are 0 to 2, and R^{28} and R^{29} are the same or different and are each hydrogen atom, a lower alkenyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a cyano-substituted

lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl which has optionally a phenyl-lower alkyl substituent, an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, or a lower alkyl, or R²⁸ and R²⁹ may bind together with the nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom, which heterocyclic ring may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl, or a lower alkanoyl.

The conversion of the compound (1A) into the compound (1B) is carried out by reduction thereof. The reducing reaction is preferably carried out by using a hydrogenating reducing agent (e.g. lithium aluminum hydride, sodium boro hydride, diborane, etc.). The reducing agent is usually used in an amount of at least one mole, preferably 1 to 15 moles, to 1 mole of the starting compound. The reducing reaction is usually carried out in an appropriate solvent, for example, water, alcohols (e.g. methanol, ethanol, isopropanol, etc.), ethers (e.g. tetrahydrofuran, diethyl ether, diis propyl ether, diglyme, etc.), or a mixture of these solvents, at a temperature of from about -60°C to about 150°C, peferably about -30°C to about 100°C, for about 10 minutes to 15 hours. When lithium aluminum

hydride or diborane is used as the reducing agent, it is preferable to use an anhydrous solvent such as tetrahydro-furan, diethyl ether, diisopropyl ether, diglyme, etc.

The reaction of converting the compound (1A) into the compound (1C) is usually carried out in an appropriate solvent or without solvent in the presence or absence of a dehydrating agent. The solvent includes, for example, lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), halogenated hydrocarbons (e.g. dichloromethane, dichloroethane, chloroform, carbon tetrachloride, etc.), aprotic polar solents (e.g. dimethylformamide, dimethylacetamide, N-methylpyrrolidone, etc.), or a mixture of these solvents. The dehydrating agent includes, for example, conventional drying agent used for dehydrating solvents (e.g. molecular sieves, etc.), mineral acids (e.g. hydrochloric acid, sulfuric acid, borone trifluoride, etc.), organic acids (e.g. p-toluenesulfonic acid, etc.), and the like. The reaction is usually carried out at a temperature of from room temperature to about 250°C, preferably from about 50°C to about 200°C, for about 1 to 48 hours. The amount of the compound (24) is not critical, but it is usually used at least in an equivalent amount, preferably equimolar to largely excess to the amount of the compound (1A). The dehydrating agent is preferably used in a largely excess amount in case of the drying agent and in a catalytic amount in case of the acid.

The subsequent reducing reaction can be carried out by various methods, for example by catalytically hydrogenating the compound in an appropriate solvent in the presence of a catalyst. The solvent includes, for example, water, acetic acid, alcohols (e.g. methanol, ethanol, isopropanol, etc.), hydrocarbons (e.g. hexane, cyclohexane, etc.), ethers (e.g. diethylene glycol dimethyl ether, dioxane, tetrahydrofuran, diethyl ether, etc.), esters (e.g. ethyl acetate, methyl acetate, etc.), aprotic polar solvents (e.g. dimethylformamide, etc.), or a mixture of these solvents. The catalyst includes, for example, palladium, palladium black, palladium-carbon, platinum, platinum oxide, copper chromite, Raney nickel, and the like. The catalyst is usually used in an amount of 0.02 to 1 part by weight to 1 part by weight of the starting compound. The reaction is usually carried out at a temperature of from about -20°C to about 100°C, peferably about 0°C to about 70°C, under a hydrogen atmospheric pressure of 1 to 10 atm. for about 0.5 to 20 hours.

Although the reducting reaction can be carried out under the above conditions, it is preferably carried out by using a hydrogenating reducing agent. The hydrogenating reducing agent includes, for example, lithium aluminum hydride, sodium borohydride, diborane, etc., and it is usually used in an amount of at least one mole, preferably 1 to 10 moles, to 1 mole of the compound (1A). The reaction is usually carried out in an appropriate solvent, such as

water, lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), ethers (e.g. tetrahydrofuran, diethyl ether, diglyme, etc.), dimethylformamide, or a mixture of these solvents, at a temperature of about -60°C to about 50°C, preferably about -30°C to room temperature, for about 10 minutes to about 5 hours. When lithium aluminum hydride or diborane is used as the reducing agent, it is preferable to use an anhydrous solvent such as diethyl ether, tetrahydrofuran, diglyme, etc.

The compound (1C) wherein at least one of R^{28} and R^{29} is hydrogen atom can be converted into the compound (1C) wherein at least one of R^{28} and R^{29} is a lower alkyl by reacting the compound (1C) with the compound (8) or the compound (9) under the same conditions as in the reaction of the compound (7) and the compound (8) or (9) in the above Reaction Scheme-4.

[Reaction Scheme-16]

$$R^1$$
 R^2
 R^3
 R^4
 R^2
 R^3
 R^4
 R^2
 R^3
 R^4
 R^4
 R^2
 R^3
 R^4
 R^2
 R^3
 R^4
 R^2
 R^3
 R^4
 R^4

$$R^{1}$$
 N
 CO
 R^{31}
 R^{2}
 R^{3}
 $R^$

(lJ)

wherein R^1 , R^2 , R^3 , R^{14} , R^{15} , W', and M are as defined above, and R^{31} is a phenyl-lower alkyl, and R^{30} is a lower alkoxycarbonyl.

The reaction of converting the compound (1D) into the compound (1E) can be carried out under the same conditions as in the reaction of converting the compound (1A) into the compound (1B) in the above Reaction Scheme-15.

The reaction of converting the compound (1D) into the compound (1F) can be carried out under the same conditions as in the hydrolysis reaction of the compound (1j) in the above Reaction Scheme-7.

The reaction of the compound (1F) and the compound (25) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The halogenation of the compound (1F) can be carried out under a conventional condition for halogenation of a carboxylic acid. The reaction of the thus-obtained carboxylic acid halide of the compound (1F) with the compound (26) is carried out in an appropriate solvent in the presence or absence of a basic compound. The solvent includes, for example, halogenated hydrocarbons (e.g. methylene chloride, chloroform, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), ethers (e.g. diethyl ether, tetrahydrofuran, dimethoxyethane, etc.), esters (e.g. methyl acetate, ethyl acetate, etc.), aprotic polar solvents

(e.g. N,N-dimethylformamide, dimethylsulfoxide, hexamethylphosphoric triamide, etc.), alcohols (e.g. methanol, ethanol, propanol, butanol, 3-methoxy-1-butanol, ethyl cellosolve, methyl cellosolve, etc.), pyridine, acetone, acetonitrile, water, or a mixture of these solvents. basic compound includes, for example, organic bases such as triethylamine, trimethylamine, pyridine, dimethylaniline, Nmethylmorpholine, DBN, DBU, DABCO, etc., inorganic bases such as potassium carbonate, sodium carbonate, potassium hydroxide, sodium hydroxide, potassium hydride, sodium hydride, silver carbonate, alcoholates (e.g. sodium methylate, sodium ethylate, etc.), and the like. compound (26) is usually used in an amount of at least 1 mole, preferably 1 to 1.5 mole, to 1 mole of the carboxylic acid halide of the compound (IF). The reaction is usually carried out at a temperature of from -30°C to about 180°C, preferably from about 0°C to about 150°C, for about 5 minutes to 30 hours.

The reaction of the compound (1H) and the compound (27) is carried out in an appropriate solvent or without solvent at a temperature of from about 0°C to about 200°C, preferably from room temperature to about 150°C. The solvent includes the same solvents as used in the above reaction of the carboxylic acid halide of the compound (1F) and the compound (26). The compound (27) is preferably used in an amount largely excess to the the compound (1H). The reaction is usually completed in a reaction time of about 1

to 5 hours.

The reaction of converting the compound (1I) into the compound (1J) can be carried out by reducing the compound. The reducing reaction is usually carried out by catalytically hydrogenating the compound in an appropriate solvent in the presence of a catalyst. The solvent includes, for example, water, acetic acid, alcohols (e.g. methanol, ethanol, isopropanol, etc.), hydrocarbons (e.g. hexane, cyclohexane, etc.), ethers (e.g. dioxane, tetrahydrofuran, diethyl ether, diethylene glycol dimethyl ether, etc.), esters (e.g. ethyl acetate, methyl acetate, etc.), aprotic polar solvents (e.g. N,N-dimethylformamide, etc.), acetic acid, or a mixture of these solvents. The catalyst includes, for example, palladium, palladium black, palladium-carbon, platinum, platinum oxide, copper chromite, Raney nickel, and the like. The catalyst is usually used in an amount of 0.02 to 1 part by weight to 1 part by weight of the starting compound. The reaction is usually carried out at a temperature of from about -20°C to about 100°C, peferably about 0°C to about 80°C, under a hydrogen atmospheric pressure of 1 to 10 atm. for about 0.5 to 20 hours.

[Reaction Scheme-17]

wherein R¹, R², R³, W', 1, R¹⁷, R¹⁸, and X are as defined above, and R^{14a} is hydrogen atom, a lower alkyl, a lower alkanoyl, a lower alkenyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl,

benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonylsubstituted lower alkyl, a carbamoyl-substituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxysubstituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent, a halogen-substituted lower alkanoyl, an imiazolyl-substituted lower alkanoyl, an aminolower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyllower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, R^{15a} is a lower alkyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoylsubstituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent, an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, or a lower alkenyl, and R^{15b} is a lower alkanoyl, a phenyl-lower alkoxycarbonyl, benzoyl, a lower alkoxycarbonyl, a halogen-substituted lower alkanoyl, an

imidazolyl-substituted lower alkanoyl, or an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl.

The reaction of the compound (1K) and the compound (28) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4.

The reaction of the compound (1K) and the compound (29) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1. The compound (1M) can also be obtained by reacting the compound (1K) with a compound of the formula $(R^{15b})_2O$ (wherein R^{15b} is as defined above). The reaction can be carried out under the same conditions as in the reaction of the compound (1p) and the compound of the formula: $(R^{7b})_2O$ as described hereinbefore.

The compound (1M) wherein R^{15b} is formyl can also be prepared by reacting the compound (1K) with a formate of the formula: HCOOR⁸² (R⁸² is a lower alkyl). The reaction is usually carried out in the solvent as used in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4 or without solvent, at a temperature of about 0°C to about 200°C, preferably about 0°C to about 170°C, for about 30 minutes to about 30 hours. The formate is preferably used in a largely excess amount to the compound (1K).

[Reaction Scheme-18]

wherein R^1 , R^2 , R^4 , R^{16} , W, ℓ , ℓ ' and ℓ " are as defined above, and R^{32} is a lower alkoxycarbonyl-substituted lower alkoxy, R^{33} is a carbamoyl-substituted lower alkoxy, R^{34} is

a carboxy-substituted lower alkoxy, R^{44} is an amino having optionally a lower alkyl substituent, and R^{45} is an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent.

The conversion of the compound (1N) into the compound (10) can be carried out by reacting the compound with aqueous ammonia in an appropriate solvent in an autoclave. The solvent includes the same solvents as used in the reaction of the carboxylic acid halide and the amine (2) in the above Reaction Scheme-1. The aqueous ammonia is used in a largely excess amount to the compound (1N). The reaction proceeds advantageously by adding an ammonium halide (e.g. ammonium chloride, etc.) to the reaction system. The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from room temperature to about 150°C, for about 1 to 10 hours.

The reaction of converting the compound (1N) into the compound (1P) can be carried out under the same conditions as in the hydrolysis of the compound (1j) in the above Reaction Scheme-7.

The reaction of the compound (1P) and the compound (30) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-19]

$$\begin{array}{c|c}
R^1 & W \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& & \\
& &$$

wherein R^1 , R^2 , R^4 , R^{16} , W, ℓ , ℓ' and ℓ'' are as defined above.

The reducing reaction in the above reaction scheme is usually carried out, for example, (i) with a reducing catalyst in an appropriate solvent or (ii) with a reducing agent such as a mixture of a metal or metal salt with an acid, or a mixture of a metal or metal salt with an alkali metal hydroxide, a sulfide or an ammonium salt in an appropriate inert solvent.

In case of using a reducing catalyst, the solvent includes, for example, water, acetic acid, alcohols (e.g. methanol, ethanol, isopropanol, etc.), hydrocarbons (e.g. hexane, cyclohexane, etc.), ethers (e.g. dioxane, tetrahydrofuran, diethyl ether, diethylene glycol dimethyl ether, etc.), esters (e.g. ethyl acetate, methyl acetate, etc.),

aprotic polar solvents (e.g. N,N-dimethylformamide, etc.), or a mixture of these solvents. The catalyst includes, for example, palladium, palladium black, palladium-carbon, platinum, platinum oxide, copper chromite, Raney nickel, and the like. The catalyst is usually used in an amount of 0.02 to 1 part by weight to 1 part by weight of the starting compound. The reaction is usually carried out at a temperature of from about -20°C to about 150°C, peferably about 0°C to about 100°C, under a hydrogen pressure of 1 to 10 atm. for about 0.5 to 10 hours. In the reaction, an acid such as hydrochloric acid may optionally added to the reaction system.

agent includes a mixture of iron, zinc, tin or stannous chloride and a mineral acid (e.g. hydrochloric acid, sulfuric acid, etc.), or a mixture of iron, ferrous sulfate, zinc or tin and an alkali metal hydroxide (e.g. sodium hydroxide, etc.), a sulfide (e.g. ammonium sulfide, etc.), aqueous ammonia, or an ammonium salt (e.g. ammonium chloride, etc.). The inert solvent includes, for example water, acetic acid, methanol, ethanol, dioxane, and the like. The reducing reaction conditions are determined depending on the kinds of the reducting agent, but in case of using a reducing agent comprising stannous chloride and hydrochloric acid, for example, it is preferably carried out at a temperature of about 0°C to room temperature for about 0.5 to 10 hours. The reducing agent is usually used in an

amount of at least one mole, preferably 1 to 5 moles, to 1 mole of the starting compound.

[Reaction Scheme-20]

wherein R^1 , R^2 , R^4 , R^{16} , R^{17} , R^{18} , ℓ , ℓ' , ℓ'' and W are as

defined above, and \mathbb{R}^{36} is a lower alkyl, \mathbb{R}^{37} is a lower alkanoyl, and \mathbb{R}^{35} is hydrogen atom, a lower alkyl or a lower alkanoyl.

The reaction of the compound (1S) and the compound (31) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4.

The reaction of the compound (1S) and the compound (32) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1. Besides, the compound (1U) can also be obtained by reacting the compound (1S) with a compound of the formula: $(R^{37})_2O$ $(R^{37})_3O$ is as defined above). The reaction is carried out under the same conditions as in the above reaction of the compound (1p) and a compound of the formula: $(R^{7b})_2O$.

The compound (1) wherein R⁸ is a phenyl-lower alkoxycarbonyl can be converted into the compound (1) wherein R⁸ is hydrogen atom in the same manner as in the reaction of converting the compound (11) into the compound (1J) in the above Reaction Scheme-16.

Other derivatives of the starting compound (2) can be prepared, for example, by the process shown in the following reaction scheme.

[Reaction Scheme-21]

wherein R^1 , R^2 , and W are as defined above.

The reaction of the compound (2) and the compound (33) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of converting the compound (34) into the compound (2a) can be carried out under the same conditions as in the reducing reaction in the above Reaction Scheme-19.

The starting compound (5) can be prepared, for example, by the process of the following reaction scheme.

[Reaction Scheme-22]

wherein \mathbb{R}^1 , \mathbb{R}^2 , and W are as defined above, and \mathbb{R}^{38} is a lower alkyl.

The reaction of the compound (2) and the compound (35) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of converting the compound (36) into the compound (5) can be carrie out under the same conditions as in the hydrolysic reaction in the above Reaction Scheme-7.

[Reaction Scheme-23]

$$R^{1}$$
 N
 R^{2}
 $R^{39}X$ (37)
 R^{2}
 R

wherein R^1 , R^2 , R^4 , R^{16} , ℓ , ℓ' , ℓ'' , χ , and W are as defined above, and R^{39} is a lower alkanoyl.

The reaction of the compound (1W) and the compound (37) can be carried out under the same conditions as in the reaction of the compound (1n) and the compound (18) in the above Reaction Scheme-9.

The hydrolysis reaction of the compound (1X) can be carried out under the same conditions as in the hydrolysis of the compound (1j) in the above Reaction Scheme-7.

[Reaction Scheme-24]

wherein R^1 , R^2 , R^4 , R^{16} , ℓ , ℓ' , ℓ'' , and W are as defined above, R^{40} is a lower alkanoyl, and R^{41} is a hydroxysubstituted lower alkyl.

The reaction of converting the compound (1Y) into the compound (1Z) can be carried out under the same conditions as in the reaction of converting the compound (1A) into the compound (1B) in the above Reaction Scheme-15.

[Reaction Scheme-25]

$$R^{1} \xrightarrow{\mathbb{N}} \mathbb{R}^{2}$$

$$= \mathbb{E}_{\text{Sterification}} \mathbb{R}^{2}$$

$$= \mathbb{E}_{\mathbb{R}^{4}} \mathbb{E}_{\mathbb{R}^{$$

wherein R^1 , R^2 , R^4 , R^{16} , ℓ , ℓ' , ℓ'' , and W are as defined above, R^{42} is a lower alkoxycarbonyl and R^{43} is carboxyl.

The reaction of converting the compound (laa) into the compound (lbb) can be carried out under the same conditions as in the hydrolysis of the compound (lj) in the above Reaction Scheme-7.

The esterification reaction of the compound (lbb) can be carried out under the same conditions as in the esterification of the compound (lw) in the above Reaction Scheme-13.

• -

[Reaction Scheme-26]

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 , and W are as defined above, and \mathbb{R}^{46} is a phenyl having optionally a lower alkyl substituent.

The reaction of the compound (2b) and the compound (38) is usually carried out in an appropriate solvent or without solvent in the presence or absence, preferably in the absence, of a basic compound. The solvent and basic compound are the same as those used in the reaction of the carboxylic acid halide and the amine (2) in the above Reaction Scheme-1.

The compound (38) is usually used in an amount of about 1 to 5 moles, preferably about 1 to 3 moles, to 1 mole of the compound (2b). The reaction is usually carried out at a temperature of from about 0°C to about 200°C, preferably from room temperature to about 150°C, for about 5 minutes to about 30 hours. In the reaction, a boron compound (e.g. boron trifluoride etherate, etc.) may be added to the reaction system.

[Reaction Scheme-27]

wherein R^1 , R^2 , R^4 , W, and X are as defined above, and R^{47} is a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, or quinolylsulfonyl.

The reaction of the compound (2b) and the compound (39) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

....

 $\Phi_{i}(t) = \{ (i,j) \in \mathcal{F}_{i} \mid (i,j) \in \mathcal{F}_{i} : i \in \mathcal{F}_{i} \text{ such that } i \in \mathcal$

wherein R^1 , R^2 , R^4 , W, R^{17} , R^{18} , and X are as defined above, R^{48} is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, and R^{49} is a lower alkyl or a carbamoyl-lower alkyl.

The reaction of the compound (lee) and the compound (40) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the

above Reaction Scheme-1.

The reaction of the compound (lee) and the compound (41) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4, provided that in the reaction product (lff) produced by the reaction of the compound (lee) and the compound (9), the group R⁴⁹ is a lower alkyl.

[Reaction Scheme-29]

$$R^{1}$$
 R^{1}
 R^{1}
 R^{1}
 R^{1}
 R^{1}
 R^{1}
 R^{2}
 R^{2}
 R^{2}
 R^{5}
 R^{5}
 R^{5}
 R^{5}
 R^{5}
 R^{5}

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^5 , and W are as defined above, and \mathbb{R}^{50} is a benzoyl having optionally a halogen substituent on the phenyl ring.

The reaction of the compound (7) and the compound (42) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-30]

wherein R^1 , W', R^{26} , R^2 , and R^3 are as defined above, R^{103} is hydroxy or sulfoxy, and R^{51} is hydroxyimino or sulfoxyimino.

The reaction of the compound (1A) and the compound (43) is usually carried out in an appropriate inert solvent in the presence or absence of a basic compound. The basic compound includes, for example, inorganic basic compounds such as sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, etc., and organic basic compounds such as piperidine, pyridine, triethylamine, 1,5-diazabicyclo[4.3.0]nonene-5 (DBN), 1,8-diazabicyclo-[5.4.0]undecene-7 (DBU), 1,4-diazabicyclo[2.2.2]octane The inert solvent includes, for example, (DABCO), etc. lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), ethers (e.g. dioxane, tetrahydrofuran, diethyl ether, ethylene glycol monomethyl ether, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), halogenated hydrocarbons (e.g. dichloromethane, dichloroethane, chloroform, carbon tetrachloride, etc.), pyridine,

dimethylformamide, dimethylsulfoxide, hexamethylphosphoric triamide, etc., or a mixture of these solvents. The compound (43) is usually used at least in equivalent amount, preferably 1 to 5 moles, to 1 mole of the compound (1A). The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from about 50°C to 150°C, for about 1 to 10 hours.

[Reaction Scheme-31]

R1 R27

Halogenation

$$R^1$$
 R^2
 R^3

(1B)

 R^2
 R^3
 R^3

wherein ${\rm R}^1$, W', ${\rm R}^{27}$, ${\rm R}^2$, M, and ${\rm R}^3$ are as defined above, and ${\rm R}^{52}$ is a halogen atom.

The halogenation of the compound (1B) is usually carried out in an appropriate solvent or without solvent by

reacting the compound (1B) with a halogenating agent.

The halogenating agent includes mineral acids (e.g. hydrochloric acid, hydrobromic acid, etc.), N,N-diethyl-1,2,2-trichlorovinylamide, phosphorus pentachloride, phosphorus pentabromide, phosphorus oxychloride, thionyl chloride, methanesulfonyl chloride, or a combination of a phenyl-lower alkyl halide (e.g. p-toluenesulfonyl chloride, etc.) and a basic compound. The basic compound includes the same compounds as used in the reaction of the compound (1A) and the compound (43) in the above Reaction Scheme-30. solvent includes, for example, ethers (e.g. dioxane, tetrahydrofuran, etc.), halogenated hydrocarbons (e.g. chloroform, methylene chloride, carbon tetrachloride, etc.), and the like. The amount of the halogenating agent may vary depending on the kinds of the halogenating agents, and in case of a combination of a phenyl-lower alkyl halide (e.g. p-toluenesulfonyl chloride, etc.) and a basic compound, it is used in an amount of at least 1 mole, preferably 1 to 2 moles, to 1 mole of the compound (1B), and in case of other halogenating agents, it is used at least in an equimolar amount, usually in a largely excess amount, to the compound The reaction is usually carried out at a temperature of from room temperature to about 150°C, preferably from room temperature to about 80°C, for about 1 to 80 hours.

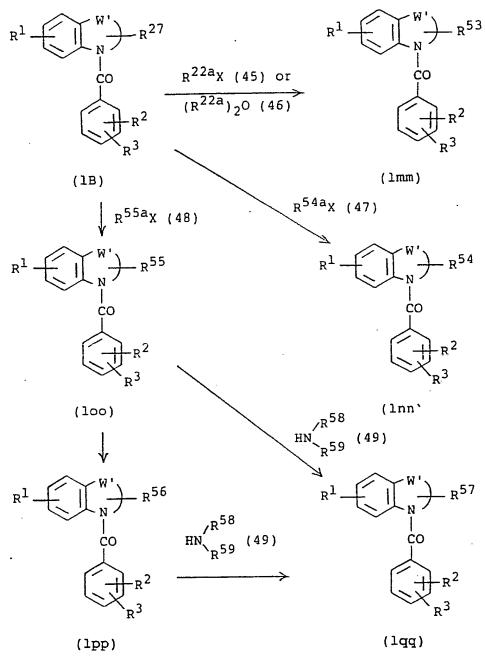
The reaction of the compound (ljj) and the compound (44) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the

above Reaction Scheme-4.

The reducing reaction of the compound (lkk) can be carried out under the same conditions as in the reducing reaction using a reducing catalyst for converting the compound (lA) into the compound (lC) in the above Reaction Scheme-15.

. - .

[Reaction Scheme-32A]



wherein R^1 , W^1 , R^2 , R^3 , R^{27} , X, and A are as defined above, R^{53} is a lower alkanoyloxy having optionally a halogen substituent, R^{54} is a lower alkoxy, an amino-lower alkanoyloxy having optionally a lower alkyl substituent, or a group of the formula:

[Reaction Scheme-32B]

 $-\text{O-A-CON} \frac{\text{R}^{82}}{\text{R}^{83}} \text{ (A, R}^{82} \text{ and R}^{83} \text{ are as defined above), R}^{55} \text{ is a}$ lower alkoxycarbonyl-substituted lower alkoxy, R}^{56} is a carboxy-substituted lower alkoxy, R}^{57} is an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, R}^{54a} is a lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a group of the formula: -A-CON R^{82} (A, R^{82}) and R^{83} are as defined above), R^{55a} is a lower alkoxy-carbonyl-substituted lower alkyl, R^{58} and R^{59} are the same or different and are each hydrogen atom or a lower alkyl, and R^{22a} is a lower alkanoyl having optionally a halogen substituent.

wherein R^1 , W', R^2 , R^3 , X, R^{27} , and A are as defined above, and R^{61} and R^{62} are the same or different and are each hydrogen atom, a lower alkyl or a lower alkanoyl.

The reaction of the compound (1B) and the compound (45) or the compound (46) in the Reaction Scheme-32A can be carried out under the same conditions as in the reaction of the compound (1n) and the compound (18) in the above Reaction Scheme-9.

The reaction of the compound (1B) and the compound (47) and the reaction of the compound (1B) and the compound (48) can be carried out under the same conditions as in the reaction of the compound (1n) and the compound (18) in the above Reaction Scheme-9.

The reaction of converting the compound (loo) into the compound (lpp) can be carried out under the same conditions as in the hydrolysis reaction of the compound (lj) in the above Reaction Scheme-7.

The reaction of the compound (loo) and the compound (49) and the reaction of the compound (lpp) and the compound (49) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of the compound (1B) and the compound (49a) in the Reaction Scheme-32B can be carried out under the same conditions as in the reaction of the compound (1n) and the compound (18) in the above Reaction Scheme-9.

[Reaction Scheme-33]

wherein R^1 , W^1 , R^2 , R^3 , R^{27} , R^{61} , R^{62} , M, and X are as defined above, R^{60} is a halogen-substituted lower alkyl, R^{64} is a phthalimido-substituted lower alkyl, R^{63} is an amino-

lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, or a phthalimido-substituted lower alkoxy, and R⁶⁵ is an amino-substituted lower alkyl.

The reaction of the compound (1B) and the compound (50) and the reaction of the compound (1B) and the compound (52) can be carried out under the same conditions as in the reaction of the compound (1n) and the compound (18) in the above Reaction Scheme-9.

The reaction of the compound (lrr) and the compound (51) or the compound (23a) can be carried out under the same conditions as in the reaction of the compound (lg) and the compound (l4) in the above Reaction Scheme-6.

The reaction of converting the compound (ltt) into the compound (luu) can be carried out under the same conditions as in the reaction of converting the compound (lj) into the compound (lk) in the above Reaction Scheme-7.

[Reaction Scheme-34]

wherein R^1 , R^2 , R^3 , R^{61} , W', A, R^{17} , R^{18} , and X are as defined above, R^{62a} is a lower alkyl, and R^{62b} is a lower alkanoyl.

___(lxx)

The reaction of the compound (1vv) and the compound (53) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4.

The reaction of the compound (lvv) and the compound (54) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the

above Reaction Scheme-1.

The reaction of the compound (lvv) and the compound (55) can be carried out under the same conditions as in the reaction of the compound (lp) and the compound of the formula: $(R^{7b})_20$ in the above Reaction Scheme-10. [Reaction Scheme-35]

$$R^{1} \xrightarrow{W'} CON_{R^{59}}$$

$$R^{1} \xrightarrow{W'} CH_{2}N_{R^{59}}$$

$$R^{2} \xrightarrow{R^{2}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

wherein R^1 , R^2 , R^3 , and W' are as defined above, R^{58} ' and R^{59} ' are the same or different and are each hydrogen atom, a lower alkyl, or a lower alkanoyl.

The reaction of converting the compound (lyy) into the compound (lzz) is usually carried out by reducing the compound (lyy).

The reducting reaction is preferably carried out by using a hydrogenating reducing agent. The hydrogenating reducing agent includes, for example, lithium aluminum hydride, sodium boro hydride, diborane, etc. The reducing agent is usually used in an amount of at least one make, preferably 1 to 15 moles, to 1 mole of the starting compound. The reducing reaction is usually carried out in an appropriate solvent, such as water, lower alcohols (e.g.

methanol, ethanol, isopropanol, etc.), ethers (e.g. tetra-hydrofuran, diethyl ether, diisopropyl ether, diglyme, etc.), or a mixture of these soslvents, at a temperature of about -60°C to about 150°C, preferably about -30°C to 100°C, for about 10 minutes to about 5 hours. When lithium aluminum hydride or diborane is used as the reducing agent, it is preferable to use an anhydrous solvent such as diethyl ether, tetrahydrofuran, diglyme, etc.

[Reaction Scheme-36]

R1
$$\xrightarrow{N}$$
 A-NHR^{58a} $\xrightarrow{R^{62a}X (53)}$ \xrightarrow{CO} $\xrightarrow{R^{62a}X (53)}$ \xrightarrow{CO} $\xrightarrow{R^{17}COR^{18} (9)}$ $\xrightarrow{R^{2}}$ $\xrightarrow{R^{3}}$ (1AA) (1CC) $\xrightarrow{R^{62b}OH (54) \text{ or } (R^{62b})_{2}O (55)}$ $\xrightarrow{R^{1}}$ $\xrightarrow{R^$

(1BB)

wherein R^1 , W', R^2 , R^3 , R^{62a} , R^{62b} , X, R^{17} , R^{18} , and A are as defined above, R^{58a} is hydrogen atom, a lower alkyl or a lower alkanoyl.

WO 91/05549 - 100 - PCT/JP90/01340

The reaction of the compound (1AA) and the compound (53) or the compound (9) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) or the compound (9) in the above Reaction Scheme-4.

The reaction of the compound (1AA) and the compound (54) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of the compound (1AA) and the compound (55) can be carried out under the same conditions as in the reaction of the compound (1p) and the compound of the formula: $(R^{7b})_{2}O$ in the above Reaction Scheme-10.

The compound (1BB) wherein R^{62b} is formyl can also be prepared by reacting the compound (1AA) with a formate of the formula: $HCOOR^{82}$ under the same conditions as in the reaction of the compound (1K) and the compound of the formula: $HCOOR^{82}$ as described hereinbefore.

The compounds of the formula (1) wherein W is sulfur atom or sulfinyl, or R⁸² and R⁸³ bind together with the nitrogen atom to which they bond to form thiomorpholino or 1-oxo-thiomorpholino can be converted into the corresponding compounds of the formula (1) wherein W is sulfinyl or sulfonyl, or R⁸² and R⁸³ bind together with the nitrogen atom to which they bond to form 1-oxo-thiomorpholino or 1,1-dioxo-thiomorpholino, respectively, by oxidation the sof.

The oxidation reaction is carried out in an

appropriate solvent in the presence of an oxidizing agent. The solvent includes, for example, water, organic acids (e.g. formic acid, acetic acid, trifluoroacetic acid, etc.), alcohols (e.g. methanol, ethanol, etc.), halogenated hydrocarbons (e.g. chloroform, dichloromethane, etc.), or a mixture of these solvents. The oxidizing agent includes, for example, peracids (e.g. performic acid, peracetic acid, trifluoro-peracetic acid, perbenzoic acid, m-chloroperbenzoic acid, o-carboxy-perbenzoic acid, etc.), hydrogen peroxide, sodium metaperiodate, dichromic acid, dichromates (e.g. sodium dichromate, potassium dichromate, etc.), permanganic acid, permanganates (e.g. potassium permanganate, sodium permanganate, etc.), lead salts (e.g. lead tetraacetate, etc.), and the like. The oxidizing agent is usually used in an amount of at least 1 mole, preferably 1 to 2 moles, to 1 mole of the starting compound. Besides, in cases of the oxidation of converting the sulfur atom into sulfonyl group, the oxidizing agent is usually used at least 2 moles, preferably 2 to 4 moles, to 1 mole of the starting compound. The above reaction is usually carried out at a temperature of about -10°C to about 40°C, preferably from about -10°C to room temperature, for about 1 to 100 hours.

The compound (1) wherein R^{16} or R^2 is a lower alkoxy can be converted into the correspond compound (1) wherein R^{16} or R^2 is hydroxy by heating the compound in a mixture of an acid (e.g. hydrobromic acid, hydrochloric acid, etc.) and a solvent (e.g. water, methanol, ethanol,

isopropyl alcohol, etc.) at 30 to 150°C, preferably at 50 to 120°C.

Besides, the compound (1) wherein ${\bf R}^{16}$ or ${\bf R}^2$ is hydroxy can also be prepared by hydrolysis of the above compound (1) wherein R^{16} or R^2 is a lower alkoxy. The hydrolysis can be carried out in an appropriate solvent in the presence of an acid. The solvent includes, for example, water, lower alcohols (e.g. methanol, ethanol, isopropyl alcohol, etc.), ethers (e.g. dioxane, tetrahydrofuran, etc.), halogenated hydrocarbons (e.g. dichloromethane, chloroform, carbon tetrachloride, etc.), polar solvents (e.g. acetonitrile, etc.), or a mixture of these solvents. The acid includes, for example, mineral acids (e.g. hydrochloric acid, hydrobromic acid, etc.), Lewis acids (e.g. boron trifluoride, aluminum chloride, boron tribromide, etc.), iodides (e.g. sodium iodide, potassium iodide, etc.), or a mixture of the above Lewis acid and iodide. The reaction is usually carried out at a temperature of from room temperature to about 150°C, preferably from room temperature to about 100C, for about 0.5 to 30 hours.

[Reaction Scheme-37]

$$R^{1}$$
 R^{51a}
 R^{62b}
 R^{1}
 R^{1}

wherein R^1 , R^2 , R^3 , R^{62b} , and W' are as defined above, R^{51a} is hydroxyimino, and R^{66} is a lower alkanoyloxyimino.

The reaction of the compound (lii') and the compound (54) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of the compound (lii') and the compound (55) can be carried out under the same conditions as in the reaction of the compound (lp) and the compound of the formula: $(R^{7b})_2O$ in the above Reaction Scheme-10.

[Reaction Scheme-38A]

(1GG)

(1FF)

[Reaction Scheme-38B]

(1EE)

$$R^1 \longrightarrow K^2$$
 R^2
 R^3
(1E)

 $R^{70}x$ (56)

 $R^1 \longrightarrow K^2$
 R^2
 R^3
(1JJ)

(1HH)

[Reaction Scheme-38C]

(1E)
$$\frac{R^{62b}OH (54)}{Or_{(R^{62b})_2}O (55)}$$
 $R^1 \longrightarrow CH_2OR^{62b}$ CO R^2 R^2 R^3 (1II)

[Reaction Scheme-38D]

(1JJ)

Reduction

Reduction

$$R^1$$
 CH_2NH_2
 R^2
 R^3

(1KK)

[Reaction Scheme-38E]

(1EE)
$$R^{1} \xrightarrow{W'} CH_{3}$$

$$CH_{2}OH$$

$$CO$$

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

wherein R^1 , R^2 , R^3 , W', R^{26} , R^{14} , R^{15} , R^{62b} , X and M are as defined above, R^{67} is methylidene, R^{68} is a group of the formula: O, and R^{69} is a group of the formula: $\begin{array}{c} CH_2OH \\ NR^{14}R^{15} \end{array}$ (R^{14} and R^{15} are as defined above), or $\begin{array}{c} CH_2R^{7D} \\ OH \end{array}$

amino having optionally a substituent selected from a lower alkyl and a lower alkanoyl, R^{70} is a lower alkylsulfonyl, and W" is the same as the above W, provided that the number of the substituent in the groups $-(CH_2)_p$ - and $-CH=CH-(CH_2)_q$ - is 0 or 1.

The reaction of converting the compound (1A) into the compound (1EE) is carried out in an appropriate solvent in the presence of a Wittig reagent and a basic compound. The Wittig reagent includes, for example, a phosphoric compound of the formula:

$$[(R^{71})_3P^+-CH_2-R^{72}]X^-$$
 (A)

wherein R^{71} is phenyl, R^{72} is hydrogen atom or a lower alkyl, and X is a halogen atom. The basic compound includes inorganic bases (e.g. metallic sodium, metallic potassium, sodium hydride, sodium amide, sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, sodium hydrogen carbonate, etc.), metal alcoholates (e.g. sodium methylate, sodium ethylate, potassium t-butoxide, etc.), alkyl or aryl lithiums or lithium amides (e.g. methyl lithium, nbutyl lithium, phenyl lithium, lithium diisopropylamide, etc.), organic bases (e.g. pyridine, piperidine, quinoline, triethylamine, N,N-dimethylaniline, etc.). The solvent includes any solvent which does not affect on the reaction, for example, ethers (e.g. diethyl ether, dioxane, tetrahydrofuran, monoglyme, diglyme, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), aliphatic hydrocarbons (e.g. n-hexane, heptane, cyclohexane, etc.), amines (e.g. pyridine, N,N-dimethylaniline, etc.), aprotic polar solvents (e.g. N,N-dimethylformamide, dimethylsulfoxide, hexamethylphosphoric triamide, etc.), alcohols (e.g. methanol, ethanol, isopropanol, etc.), and the like. reaction is usually carried out at a temperature of about -80°C to about 150°C, preferably about -80°C to about 120°C,

for about 0.5 to 15 hours.

The reaction of converting the compound (IEE) into the compound (ILL) can be carried out under the same conditions as in the catalytically hydrogenation reaction for converting the compound (IA) into the compound (IC) in the above Reaction Scheme-15.

The reaction of converting the compound (1EE) into the compound (1FF) is carried out under the same conditions as in the reaction of converting the compound (1) wherein W is sulfur atom or sulfinyl into the corresponding compound (1) wherein W is sulfinyl or sulfonyl respectively as described herebefore.

The reaction of the compound (1FF) and the compound (25) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The reaction of converting the compound (IEE) into the compound (IE) can be carried out by firstly subjecting it to hydroboration reaction and then to oxidation.

The hydroboration reaction is carried out in a solvent such as ethers (e.g. diethyl ether, tetrahydrofuran, dioxane, etc.) in the presence of a hydroborating agent at a temperature of from about 0°C to about 50°C, preferally about 0°C to room temperature, for about 1 to 10 hours. The hydroborating agent includes be fon hydride compounds, for example, BH3.tetrahydrofuran, BH3.S(CH3)2, BH2Cl, (CH3)2CHC(CH3)2BH2, (CH3)2CHCH(CH3)BH, (\sum_{2}-BH, (\sum_{2})2BH2, (CH3)2CHCH(CH3)BH, (\sum_{2}-BH, (\sum_{2})2BH2, (CH3)2CHCH(CH3)BH, (\sum_{2}-BH, (\sum_{2})2BH2, (CH3)2CHCH(CH3)BH, (\sum_{2}-BH2, (\sum_{2})2BH2, (\sum_{2})2BH2, (\sum_{2}-BH2, (\sum_{2})2BH2, (\sum_{2}-BH2, (\sum_{2})2BH2, (\sum_{2})2BH

 $^{\text{CH}_3}$, $^{\text{CH}_3}$, $^{\text{BHCl}_2}$, $^{\text{O}}$ B, and the like.

The subsequent oxidation is carried out in water in the presence of an oxidizing agent. The oxidizing agent includes, for example, alkaline hydrogen peroxides (e.g. hydrogen peroxide - sodium hydroxide, etc.), and air oxidation is also used. The reaction is usually carried out at a temperature of from room temperature to about 150°C, preferably from room temperature to about 100°C, for 0.5 to 7 hours.

The hydroborating agent and the oxidizing agent are each used in an amount of at least 1 mole, preferably 1 to 2 mole, to 1 mole of the compound (1EE).

The reaction of the compound (1E) and the compound (54) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The reaction of the compound (1E) and the compound (55) can be carried out under the same conditions as in the reaction of the compound (1p) and the compound of the formula: $(R^{7b})_{2}$ 0 in the above Reaction Scheme-10.

The reaction of the compound (1E) and the compound (56) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The reaction of the compound (1HH) and the compound (44) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The reducing reaction of the compound (1JJ) can be carried out under the same conditions as in the catalytic hydrogenation reaction for converting the compound (1A) into the compound (1C) in the above Reaction Scheme-15.

The reaction of converting the compound (IEE) into the compound (IMM) can be carried out by reacting with an oxidizing agent in an appropriate solvent in the presence of a co-oxidizing agent.

The solvent used for the reaction with an oxidizing agent includes, for example, pyridine, ethers (e.g. dioxane, tetrahydrofuran, diethyl ether, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), halogenated hydrocarbons (e.g. dichloromethane, dichloroethane, chloroform, carbon tetrachloride, etc.), esters (e.g. ethyl acetate, etc.), water, alcohols (e.g. methanol, ethanol, isopropanol, t-butanole, etc.), or a mixture of these solvents. The cooxidizing agent includes, for example, organic amine Noxides (e g. pyridine N-oxide, N-ethyldiisopropylamine Noxide, N-methylmorpholine N-oxide, trimethylamine N-oxide, triethylamine N-oxide, etc.). The oxidizing agent includes, for example, osmium tetraoxide, and the like. The oxidizing agent is usually used in an amount of at least 1 mole, preferably 1 to 5 moles, to 1 mole of the starting compound. The reaction is usually carried out at a temperature of from -20°C to 150°C, preferably from room temperature to 100°C, for about 1 to 10 hours.

[Reaction Scheme-39]

wherein R^1 , R^2 , R^3 , R^{27} , W', M, and X are as defined above, R^{73} is an aminocarbonyl having optionally a lower alkyl substituent, R^{74} is an aminocarbonyloxy having optionally a lower alkyl substituent, R^{74} is a lower alkyl.

(lNN')

The reaction of the compound (1A) and the compound (57) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The reaction of the compound (1A) and the compound (59) is carried out in an appropriate solvent in the presence of an acid. The solvent includes the same solvent

as used in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. The acid includes, for example, mineral acids (e.g. hydrochloride acid, sulfuric acid, etc.), sulfonic acids (e.g. methanesulfonic acid, p-toluenesulfonic acid, etc.), alkanoic acids (e.g. trifluoroacetic acid, etc.), and the like. The compound (59) is used in an amount of at least 1 mole, preferably 1 to 5 moles, to 1 mole of the compound (1A). The reaction is usually carried out at a temperature of from room temperature to about 150°C, preferably from room temperature to about 150°C, preferably from room temperature

The reaction of the compound (1A) and the compound (58) can be carried out under the same conditions as in the reaction of the compound (2b) and the compound (38) in the above Reaction Scheme-26.

[Reaction Scheme-40]

$$R^{75}$$
 R^{76}
 CH
 CH
 CH
 $(CH_2)_q$
 $(CH_2)_q$

wherein R^1 , R^2 , R^3 , X, and q are as defined above, and R^{75} , R^{76} and R^{77} are each a lower alkyl, and the carbon atom in the formula: $-(CH_2)_q$ - may be substituted by oxygen atom,

sulfur atom, sulfinyl, sulfonyl, or a group of the formula: _R13 -N- (R¹³ is as defined above), and further the group: $-(CH₂)_{\alpha}$ may optionally have 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxyl, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkylsulfonyloxy-lower alkyl, an azido-lower alkyl, a. group of the formula: 0, an aminocarbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-substituted lower alkoxy, a carboxy-substituted lower alkoxy, an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, an amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxyimino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula: R^{81} -N-CH₂COO- (R^{81} is

hydrogen atom or a lower alkyl), hydrazino, pyrrolyl, an aminolower alkanoyloxy having optionally a lower alkyl substituent,

a group of the formula: -O-A-CO-N $_{R83}^{R82}$ (A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, or a pyridyl-lower alkyl, or R^{82} and R^{83} may bind together with nitrogen atom to which they

bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula: -(CO) $_{n}\text{-N}_{-15}^{\text{R}^{14}}$ (n is as defined above, and \mathbf{R}^{14} and \mathbf{R}^{15} are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or R^{14} and R^{15} may bind together : in the nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom, which

heterocyclic group may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl and a lower alkanoyl.

The reaction of the compound (100) and the compound (60) is carried out in an appropriate solvent in an autoclave. The solvent includes any solvent as used in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from room temperature to about 150°C, for about 1 to 7 hours.

The subsequent deamination reaction is carried out in an appropriate solvent in the presence of a basic compound. The solvent includes the same solvent as used in the above reaction of the compound (100) and the compound (60). The basic compound includes any basic compound as used in the reaction of converting the compound (1A) into the compound (1EE) in the above Reaction Scheme-38. The reaction is usually carried out at a temperature of from room temperature to about 150°C, preferably from room temperature to about 100°C, for about 1 to 10 hours.

[Reaction Scheme-41]

$$R^{1}$$
 N
 N
 R^{14}
 $R^{$

wherein R^1 , R^2 , R^3 , R^{14} , M, and W' are as defined above, R^{78} is

an oxiranyl-substituted lower alkyl, R⁷⁹ is a lower alkoxy, or an amino having optionally a lower alkyl substituent, and R⁸⁰ is a lower alkyl having 2 substituents selected from hydroxy, a lower alkoxy, and an amino having optionally a lower alkyl substituent.

The reaction of the compound (1QQ) and the compound (61) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

The reaction of the compound (1QQ) and the compound (62) can be carried out by firstly reacting them in trifluoro-acetic acid at a temperature of about 0°C to about 100°C, preferably about 0°C to about 50°C, for about 1 to 7 hours, followed by hydrolysis of the resultant.

The hydrolysis is carried out in an appropriate solvent or without solvent in the presence of an acid or a basic compound. The solvent includes, for example, water, lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), ketones (e.g. acetone, methyl ethyl ketone, etc.), ethers (e.g. dioxane, tetrahydrofuran, ethylene glycol dimethyl ether, etc.), fatty acids (e.g. acetic acid, formic acid, etc.), or a mixture of these solvents. The acid includes, for example, mineral acids (e.g. hydrochloric acid, sulfuric acid, hydrobromic acid, etc.), organic acids (e.g. formic acid, acetic acid, aromatic sulfonic acid, etc.), and the like. The basic compound includes, for example, metal carbonates (e.g. sodium carbonate, potassium carbonate, etc.), metal hydroxides (e.g. sodium hydroxide, potassium hydroxide, calcium hydroxide,

etc.). The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from room temperature to about 150°C, for about 0.5 to 25 hours.

[Reaction Scheme-42]

$$R^1$$
 R^{81}
 R^2
 R^3
 R^{1}
 R^{1}
 R^{1}
 R^{1}
 R^{2}
 R^{2}
 R^{3}
 R^{3}
 R^{1}
 R^{2}
 R^{2}
 R^{3}
 R^{3}
 R^{2}
 R^{3}

wherein R^1 , R^2 , R^3 , and W' are as defined above, and R^{81} is hydroxyimino or a lower alkanoyloxyimino.

The reaction of converting the compound (1SS) into the compound (122) is carried out by catalytically hydrogenating the compound (1SS) in an appropriate solvent in the presence of a catalyst. The solvent includes, for example, water, acetic acid, alcohols (e.g. methanol, ethanol, isopropanol, etc.), hydrocarbons (e.g. hexane, cyclohexane, etc.), ethers (e.g. diethylene glycol dimethyl ether, dioxane, tetrahydrofuran, diethyl ether, etc.), esters (e.g. ethyl acetate, methyl acetate, etc.), aprotic polar solvents (e.g. dimethylformamide, etc.), or a mixture of these solvents. The catalyst includes, for example, palladium, palladium black, palladium-carbon, platinum, platinum oxide, copper chromate, Raney nickel, and the like. The catalyst is usually used in an amount of 0.02 to 1 part by weight to 1 part by weight of the compound (1SS).

about -20°C to about 100°C, peferably about 0°C to about 70°C, under a hydrogen atmospheric pressure of 1 to 10 atm. for about 0.5 to 20 hours.

Alternatively, the reducing reaction can also be carried out by using a hydrogenating reducing agent. The hydrogenating reducing agent includes, for example, lithium aluminum hydride, sodium boro hydride, diborane, etc. The reducing agent is usually used in an amount of at least one mole, preferably 1 to 10 moles, to 1 mole of the compound (1SS). The reaction is usually carried out in an appropriate solvent, such as water, lower alcohols (e.g. methanol, ethanol, isopropanol, etc.), ethers (e.g. tetrahydrofuran, diethyl ether, diglyme, etc.), acetic acid, and the like, at a temperature of about 0°C to about 200°C, preferably about 0°C to 170°C, for about 10 minutes to about 10 hours. When lithium aluminum hydride or diborane is used as the reducing agent, it is preferable to use an anhydrous solvent such as diethyl ether, tetrahydrofuran, diglyme, etc.

[Reaction Scheme-43]

wherein R^1 , R^2 , R^3 , W^* , ℓ , R^{14a} are as defined above, and R^{83} is phenyl or a lower alkyl.

The reaction of the compound (1K) and the compound (63) can be carried out under the same conditions as in the reaction of the compound (2b) and the compound (38) in the above Reaction Scheme-26.

[Reaction Scheme-44]

wherein R^1 , R^2 , R^3 , W', 1, R^{14a} are as defined above.

The reaction of the compound (1K) and the glyconitrile (64) can be carried out in an appropriate solvent. The solvent includes the same solvent as used in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4. The reaction is usually carried out at a temperature of from about 0°C to about 150°C, preferably about 0°C to about 100°C, for about 1 to 10 hours. The glyconitrile (64) is used in an amount of at least 1 mole, preferably 1 to 2 moles, to 1 mole of the compound (1K).

[Reaction Sch. 2-45]
$$R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R84} \qquad R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R86} \qquad R^{1} \xrightarrow{R} R^{2} \qquad R^{2} \qquad R^{3} \qquad R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R87} \qquad R^{1} \xrightarrow{R} R^{2} \qquad R^$$

wherein R^1 , R^2 , R^3 , W', ℓ , R^{14a} are as defined above, R^{84} is a lower alkoxycarbonyl-substituted lower alkyl, R^{85} is an amino having optionally a lower alkyl substituent, R^{86} is an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, and R^{87} is a carboxy-substituted lower alkyl.

The reaction of the compound (1VV) and the compound (65) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

The hydrolysis reaction of the compound (1.7V) can be carried out under the same conditions as in the hydrolysis reaction of the compound (1QQ) and the compound (62) in the above Reaction Scheme-41.

[Reaction Scheme-46]

$$R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R88} \qquad R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R89} \qquad R^{14a}$$

$$CO \qquad R^{91} \times (66) \qquad CO \qquad R^{2}$$

$$R^{2} \qquad R^{3} \qquad (122)$$

$$R^{1} \xrightarrow{W'} (CO)_{\varrho} - N_{R90} \qquad R^{14a}$$

wherein R^1 , R^2 , R^3 , W^1 , ℓ , X, and R^{14a} are as defined above, R^{88} is a tetrahydropyranyloxy-substituted lower alkyl, R^{89} is a lower alkanoyloxy-substituted lower alkyl, R^{90} is a hydroxy-substituted lower alkyl, and R^{91} is a lower alkanoyl.

The reaction of the compound (1YY) and the compound (66) can be carried out in a solvent such as acetic acid at a temperature of about 0°C to about 200°C, preferably about 0°C to about 150°C, for about 0.5 to 15 hours.

The hydrolysis reaction of the compound (1YY) can be carried out under the same conditions as in the hydrolysis reaction of the compound (1QQ) and the compound (62) in the above Reaction Scheme-41, wherein a pyridinium salt (e.g. pyridinium p-toluenesulfonate, etc.) may be used as the acid.

[Reaction Scheme-47]

wherein R^1 , R^2 , R^3 , W', and R^{26} are as defined above.

The reaction of converting the compound (1A) into the compound (1bbb) can be carried out under the same conditions as in the reaction of converting the compound (1A) into the compound (1C) in the above Reaction Scheme-15.

[Reaction Scheme-48]

wherein R^1 , R^2 , R^3 and W' are as defined above, R^{92} and R^{93} are each a lower alkoxy.

The reaction of the compound (122) and the compound (68) is carried out in an appropriate solvent in the presence of an acid. The solvent includes, for example, water, alcohols (e.g. methanol, ethanol, isopropanol, etc.), ketones (e.g. acetone, methyl ethyl ketone, etc.), ethers (e.g. dioxane, tetrahydrofuran, ethylene glycol dimethyl ether, etc.), fatty

acids (e.g. acetic acid, formic acid, etc.), or a mixture of these solvents. The acid includes, for example, mineral acids (e.g. hydrochloric acid, sulfuric acid, hydrobromic acid, etc.), organic acids (e.g. formic acid, acetic acid, aromatic sulfonic acids, etc.). The reaction is usually carried out at a temperature of from room temperature to about 200°C, preferably from room temperature to about 150°C, for about 0.5 to 5 hours. The compound (68) is usually used in an amount of at least 1 mole, preferably 1 to 2 moles, to 1 mole of the compound (192).

[Reaction Scheme-49]

wherein R^1 , R^2 , R^3 , W', and R^{14a} are as defined above, R^{94} is a halogen-substituted lower alkanoyl, R^{95} is an imidazolyl-substituted lower alkanoyl or an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, and R^{96} is imidazolyl, or an amino having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl.

The reaction of the compound (1ddd) and the compound (69) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above

Reaction Scheme-4.

[Reaction Scheme-50]

wherein R^1 , R^2 , R^3 , and W' are as defined above, R^{97} is a lower alkanoyloxy having a halogen substituent, R^{98} is an amino having optionally a lower alkyl substituent, and R^{99} is an amino-lower alkanoyloxy having optionally a lower alkyl substituent.

The reaction of the compound (lfff) and the compound (70) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

[Reaction Scheme-51]

wherein R^1 , R^2 , R^3 , W', R^{82} , and R^{83} are as defined above, R^{100} is a carboxy-substituted lower alkoxy, and R^{101} is a

WO 91/05549 - 125 - PCT/JP90/0134

group of the formula: $-O-A-CON \frac{R^{82}}{R^{83}}$ (A, R^{82} and R^{83} are as defined above).

The reaction of the compound (1hhh) and the compound (71) can be carried out under the same conditions as in the reaction of the compound (2) and the compound (3) in the above Reaction Scheme-1.

[Reaction Scheme-52]

wherein R^1 , R^2 , R^3 , W'', X, and R^{82} are as defined above, and R^{102} is hydrogen atom or a lower alkyl, provided that in the compound (ljjj), the groups of the formulae: $-NH-R^{102}$ and -OH are substituted at the positions adjacent each other.

The reaction of the compound (ljjj) and the compound (72) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

[Reaction Scheme-53]

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^3 , \mathbb{W}^1 , \mathbb{R}^{26} and X are as defined above, and \mathbb{R}^{104} is a lower alkyl.

The reaction of the compound (1A) and the compound (75) can be carried out in an appropriate solvent. The solvent includes, for example, ethers (diethyl ether, dioxane, tetrahydrofuran, etc.), aromatic hydrocarbons (e.g. benzene, toluene, xylene, etc.), saturated hydrocarbons (e.g. pentane, hexane, heptane, cyclohexane, etc.), or a mixture of these solvents. The reaction is usually carried out at a temperature of from about -70°C to about 50°C, preferably from about -30°C to room temperature, for about 1 to 6 hours. The compound (73) is used in an amount of at least 1 mole, preferably 1 to 5 moles, to 1 mole of the compound (1A).

[Reaction Scheme-54]

$$R^{1} \xrightarrow{N} A-R^{105}$$

$$R^{1} \xrightarrow{N} A-N \xrightarrow{R^{58}}$$

$$R^{1} \xrightarrow{N} A-N \xrightarrow{R^{59}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{1} \xrightarrow{N} A-N \xrightarrow{R^{58}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{1} \xrightarrow{N} A-N \xrightarrow{R^{59}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

$$R^{1} \xrightarrow{N} A-N \xrightarrow{R^{59}}$$

$$R^{2} \xrightarrow{R^{3}}$$

$$R^{3} \xrightarrow{R^{3}}$$

wherein R^1 , R^2 , R^3 , W', R^{58} ', R^{59} ', and A are as defined above, and R^{105} is a lower alkylsulfonyloxy.

The reaction of the compound (1mmm) and the compound (74) can be carried out under the same conditions as in the reaction of the compound (7) and the compound (8) in the above Reaction Scheme-4.

Among the active compounds (1) of this invention, the compounds having an acidic group can easily be converted into salts by treating with a pharmaceutically acceptable basic compound. The basic compound includes, for example, metal hydroxides such as sodium hydroxide, potassium hydroxide, lithium hydroxide, calcium hydroxide, etc., alkali metal carbonates or hydrogen carbonates such as sodium carbonate, sodium hydrogen carbonate, etc., alkali metal alcoholates such as sodium methylate, potassium ethylate, etc. Besides, among the active compounds (1) of this invention, the compounds having a basic group can easily be converted into acid addition salts thereof by treating with a pharmaceutically acceptable acid. The acid includes, for example, inorganic acids such as sulfuric acid, nitric acid, hydrochloric acid, hydrobromic acid, etc., and organic acids such as acetic acid, p-toluenesulfonic acid, ethanesulfonic acid, oxalic acid, maleic acid, fumaric acid, citric acid, succinic acid, benzoic acid, etc. These salts are useful as an active ingredient as like as the compounds (1) in the free form.

In addition, the compounds (1) of this invention include stereoisomers and optical isomers, and these isomers are also useful as the active ingredient in this invention.

The compounds of this invention thus obtained can easily be isolated and purified by conventional isolation methods. The isolation methods are, for example, distillation method, recrystallization method, column chromatography, ion exchange chromatography, gel chromatography, affinity chromtography, preparative thin layer chromatography, extraction with a solvent, and the like.

The compounds and their salts of this invention are useful as a vasopressin antagonist and are used in the form of a conventional pharmaceutical preparation. The preparation is prepared by using conventional dilutents or carriers such as fillers, thickening agents, binders, wetting agents, disintegrators, surfactants, lubricants, and the like. The pharmaceutical preparations may be selected from various forms in accordance with the desired utilities, and the representative forms are tablets, pills, powders, solutions, suspensions, emulsions, granules, capsules, suppositories, injections (solutions, suspensions, etc.), and the like. In order to form in tablets, there are used carriers such as vehicles (e.g. lactose, white sugar, sodium chloride, glucose, urea, starches, calcium carbonate, kaolin, crystalline cellulose, silicic acid, etc.), binders (e.g. water, ethanol, propanol, simple syrup, glucose solution, starch solution, gelatin solution, carboxymethyl cellulose, s. lac, methyl cellulose, potassium phosphate, polyviny grrolidone, etc.), disintegrators (e.g. dry starch, sodium arginate, agar powder, laminaran powder, sodium hydrogen carbonate, calcium carbonate, polyoxyethylene sorbitan fatty acid esters, sodium laurylsulfate, stearic

monoglyceride, starches, lactose, etc.), disintegration inhibitors (e.g. white sugar, stearin, cacao butter, hydrogenated oils, etc.), absorption promoters (e.g. quaternary ammonium base, sodium laurylsulfate, etc.), wetting agents (e.g. glycerin, starches, etc.), adsorbents (e.g. starches, lactose, kaolin, bentonite, colloidal silicates, etc.), lubricants (e.g. purified talc, stearates, boric acid powder, polyethylene glycol, etc.), and the like. Moreover, the tablets may also be in the form of a conventional coated tablet, such as sugar-coated tablets, gelatin-coated tablets, enteric coated tablets, film coating tablets, or double or multiple layer tablets. In the preparation of pills, the carriers include vehicles (e.g. glucose, lactose, starches, cacao butter, hydrogenated vegetable oils, kaolin, talc, etc.), binders (e.g. gum arabic powder, tragacanth powder, gelatin, ethanol, etc.), disintegrators (e.g. laminaran, agar, etc.), and the like. In the preparation of suppositories, the carriers include, for example, polyethylene glycol, cacao butter, higher alcohols, higher alcohol esters, gelatin, semisynthetic glycerides, and the like. Capsules can be prepared by charging a mixture of the compound of this invention with the above carriers into hard gelatin capsules or soft capsules in a usual manner. In the preparation of injections, the solutions, emulsions or suspendions are sterilized and are preferably made isotonic with the blood. In the preparation of these solutions, emulsions and suspensions, there are used conventional diluents, such as water, ethyl alcohol, macrogol

(propylene glycol), ethoxylated isostearyl alcohol, polyoxylated isostearyl alcohol, polyoxyethylene sorbitan fatty acid
esters, and the like. In this case, the pharmaceutical
preparations may also be incorporated with sodium chloride,
glucose, or glycerin in an amount sufficient to make them
isotonic, and may also be incorporated with conventional
solubilizers, buffers, anesthetizing agents. Besides, the
pharmaceutical preparations may optionally be incorporated with
coloring agents, preservatives, perfumes, flavors, sweeting
agents, and other medicaments, if required.

The amount of the active compound of this invention (active ingredient) to be incorporated into the anti-vaso-pressin preparations is not specified but may be selected from a broad range, but usually, it is preferably in the range of 1 to 70 % by weight, more preferably 5 to 50 % by weight.

The anti-vasopressin preparation of this invention may be administered in any method, and suitable method for administration may be determined in accordance with various forms of preparation, ages, sexes and other conditions of the patients, the degree of severity of diseases, and the like. For instance, tablets, pills, solutions, suspensions, emulsions, granules and capsules are administered orally. The injections are intraveneously administered alone or together with a conventional auxiliary liquid (e.g. glucose, amino acid solutions), and further are optionally administered alone in intramuscular, intracutaneous, subcutaneous, or intraperitoneal route, if required. Suppositories are

administered in intrarectal route.

The dosage of the anti-vasopressin agent of this invention may be selected in accordance with the usage, ages, sexes and other conditions of the patients, the degree of severity of the diseases, and the like, but is usually in the range of about 0.6 to 50 mg of the active compound of this invention per 1 kg of body weight of the patient per day. The active compound is preferably contained in an amount of 10 to 1000 mg per the dosage unit.

Brief Description of Drawing

Fig. 1 to Fig. 4 show a chart of NMR (CDCl $_3$) of the compounds in Examples 978 and 979.

Best Mode for Carrying Out the Invention

The present invention is illustrated by the following Preparations of anti-vasopressin agent, Reference Examples of processes for preparing the starting compounds to be used for preparing the active compounds, Examples of processes for preparing the active compounds, and Experiments of the activities of the active compounds of this invention.

Preparation 1

Components

Film coated tablets are prepared from the following components.

Componencs	Amount
-	
4-Methylamino-1-[4-(3,5-dichlorobenzoyl-	
amino)benzoyl]-1,2,3,4-tetrahydroquinoline	150 g

Avicel (tradename of microcrystalline cellulose,	
manufactured by Asahi Chemical Industry Co., Ltd., Japan)	40 g
Corn starch	30 g
COIN States	
Magnesium stearate	2 g
•	10 g
Hydroxypropyl methylcellulose	9
	3 g
Polyethylene glycol-6000	
Castor oil	40 g
Castor off	40 -
Ethanol	40 g

The active component of this invention, Avicel, corn starch and magnesium stearate are mixed and kneaded and the mixture is tabletted using a conventional pounder (R 10 mm) for sugar coating. The tablets thus obtained are coated with a film coating agent consisting of hydroxypropyl methylcellulose, polyethylene glycol-6000, castor oil and ethanol to give film coated tablets.

Preparation 2

Tablets are prepared from the following components.

Components	Amount
<pre>1-[4-(N-Butylanilinoacetylamino)benzoyl]- 2,3,4,5-tetrahydroy-lH-benzazepine</pre>	150 g
Citric acid	1.0 g
Lactose	33.5 g
Dicalcium phosphate	70.0 g
Pullonic F-68	30.0 g
Sodium laurylsulfate	3.0 g
Polyvinylpyrrolidone	.⇒.0 g
Polyethylene glycol (Carbowax 1500)	4.5 g

in the entropy of the contract of the entropy of the contract of the contract

Polyethylene glycol (Carbowax 6000)	45.0	9
Corn starch	30.0	g
Dry sodium stearate	3.0	9
Dry magnesium stearate	3.0	9
Ethanol	q.s.	•

The active compound of this invention, citric acid, lactose, dicalcium phosphate, Pullonic F-68 and sodium laurylstearate are mixed. The mixture is screened with No. 60 screen and is granulated with an alcohol solution containing polyvinylpyrrolidone, carbowax 1500 and 6000. If required, an alcohol is added thereto so that the powder mixture is made a paste-like mass. Corn starch is added to the mixture and the mixture is continuously mixed to form uniform particles. The resulting particles are passed through No. 10 screen and entered into a tray and then dried in an oven at 100°C for 12 to 14 hours. The dried particles are screened with No. 16 screen and thereto are added dry sodium laurylsulfate and dry magnesium stearate, and the mixture is tabletted to form the desired shape.

The core tablets thus prepared are vanished and dusted with talc in order to guard from wetting.

Undercoating is applied to the core tablets. In order to administer the tablets orally, the core tablets are vanished several times. In order to give round shape and smooth surface to the tablets, further undercoating and coating with lubricant are applied thereto. The tablets are further coated with a coloring coating material until the desired

Burner and the transfer of the contract of the

colored tablets are obtained. After drying, the coated tablets are polished to obtain the desired tablets having uniform gloss.

Preparation 3

An injection preparation is prepared from the following components.

Components	Amount
4-Methyl-l-[4-(2,3-dimethylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-l,4-benzodiazepine	5 g
Polyethylene glycol (molecular weight: 4000)	0.3 g
Sodium chloride	0.9 g
Polyoxyethylene sorbitan monooleate	0.4 g
Sodium metabisulfite	0.1 g
Methyl-paraben	0.18 g
Propyl-paraben	0.02 g
Distilled water for injection	10.0 ml

The above parabens, sodium metabisulfite and sodium chloride are dissolved in distilled water of half volume of the above with stirring at 80°C. The solution thus obtained is cooled to 40°C, and the active compound of this invention and further polyethylene glycol and polyoxyethylene sorbitan monocleate are dissolved in the above solution. To the solution is added distilled water for injection to adjust to the desired volume, and the solution i terilized by filtering with an appropriate filter paper to give an injection preparation.

Reference Example 1

To a solution of 1,2,3,4-tetrahydroquinoline (28.7 g) in acetone (400 ml) and water (200 ml) is added potassium carbonate (38.8 g), and thereto is added p-nitrobenzoyl chloride (40 g) under ice-cooling and the mixture is stirred at room temperature overnight. To the reaction mixture is added a suitable amount of water. The precipitated crystal is collected by filtration and dried to give 1-(4-nitrobenzoyl)-1,2,3,4-tetrahydroquinoline (40.8 g) as white powder, m.p. 86 - 88°C.

Reference Example 2

To a solution of 10 % Pd-C (5 g) in ethanol (500 ml) is added 1-(4-nitrobenzoyl)-1,2,3,4-tetrahydroquinoline (53.4 g) and the mixture is subjected to catalytic reduction at ordinary temperature under atmospheric pressure of hydrogen. After the reduction, 10 % Pd-C is removed by filtration, and the filtrate is concentrated under reduced pressure to give 1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline (46.7 g) as yellow powder, m.p. 185 - 188°C.

Reference Example 3

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

1-(3-Nitrobenzoy1)-1,2,3,4-tetrahydroquinoline, white powder, m.p. $134-136^{\circ}C$

1-(2-Nitrobenzoy1)-1,2,3,4-tetrahydroquinoline,

yellow powder, m.p. 152 - 154°C

3-Methyl-l-(4-nitrobenzoyl)-l,2,3,4-tetrahydro-

quinoline, yellow powder, m.p. 109 - 110°C

4-Methyl-l-(4-nitrobenzoyl)-1,2,3,4-tetrahydro-

quinoline, yellow powder, m.p. 134 - 136°C

2-Methyl-l-(4-nitrobenzoyl)-1,2,3,4-tetrahydro-

quinoline, yellow powder, m.p. 143 - 145°C

1-(4-Nitrobenzoy1)-2,3,4,5-tetrahydro-1H-

benzazepine, yellow powder, m.p. 143 - 145°C

1-(3-Methyl-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-

benzazepine, white powder, m.p. 100 - 102°C

1-(3-Methoxy-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-

benzazepine, yellow powder, m.p. 146 - 148°C

1-(4-Nitrobenzoyl)-1,2,3,4,5,6-hexahydrobenz-

azocine, white powder, m.p. 83 - 85°C

1-(4-Nitrobenzoyl)-3,4-dihydro-2H-1,4-benzoxazine,

yellow powder, m.p. 167 - 169°C

1-(4-Nitrobenzoyl)-1,2,3,5-tetrahydro-4,1-

benzoxazepine, yellow powder, m.p. 196 - 198°C

l-(4-Nitrobenzoyl)-4-methyl-1,2,3,4-tetrahydro-

quinoxaline, brown powder

 1_{H-NMR} (CDC1₃) δ : 3.03 (3H, s), 3.54 (2H, t, J=5.7

Hz), 4.06 (2H, t, J=5.7 Hz), 6.2-6.5 (2H, m), 6.70 (1H, d,

J=8.2 Hz), 6.9-7.1 (1H, m), 7.54 (2H, d, J=8.8 Hz), 8.13

(2H, d, J = 8.8 Hz)

1-(4-Nitrobenzoyl)-5-methyl-2,3,4,5-tetrahydro-1H-

1,5-benzodiazepine, yellow oil

 1 H-NMR (CDCl₃) δ : 1.7-2.0 (1H, m), 2.0-2.3 (1H, m), 2.8-3.0 (1H, m), 2.98 (3H, s), 3.0-3.2 (1H, m), 3.4-3.6 (1H, m), 4.6-4.8 (1H, m), 6.5-6.7 (2H, m), 6.94 (1H, d, J=8.1 Hz), 7.1-7.2 (1H, m), 7.33 (2H, d, J=8.9 Hz), 7.97 (2H, d, J=8.9 Hz)

1-(4-Nitrobenzoyl)-4-methyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine, brown oil

¹H-NMR (CDCl₃) δ : 2.44 (3H, s), 3.0-3.3 (3H, m),
3.77 (1H, d, J=13.7 Hz), 4.06 (1H, d, J=13.6 Hz), 4.9-5.1
(1H, m), 6.59 (1H, d, J=7.7 Hz), 6.97 (1H, t, J=7.6 Hz),
7.15 (1H, t, J=7.4 Hz), 7.2-7.5 (3H, m), 8.03 (2H, d, J=8.8 Hz)

1-(3-Methoxy-4-nitrobenzoyl)-4-methyl-2,3,4,5tetrahydro-1H-1,4-benzodiazepine, yellow powder, m.p. 146 -

1-(4-Nitrobenzoyl)-4-n-propyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine, yellow powder, m.p. 131 - 133°C

1-(4-Nitrobenzoyl)-5-chloro-1,2,3,4-tetrahydro-quinoline, white powder, m.p. 134 - 136°C

l-(4-Nitrobenzoyl)-6-methoxy-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 149 - 151°C

l-(4-Nitrobenzoyl)-6-methyl-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 109 - 110°C

l-(4-Nitrobenzoyl)-7-methoxy-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 139 - 141°C 1-(4-Nitrobenzoyl)-3-(4-methyl-1-piperazinyl)1,2,3,4-tetrahydroquinoline, yellow amorphous

 $^{1}\text{H-NMR}$ (CDCl₃) δ : 2.29 (3H, s), 2.35-3.20 (11H, m), 3.86-4.15 (2H, m), 6.48-6.63 (1H, m), 6.89 (1H, t, J=7.4 Hz), 7.05 (1H, t, J=7.4 Hz), 7.22 (1H, d, J=7.4 Hz), 7.52 (2H, d, J=8.8 Hz), 8.11 (2H, d, J=8.8 Hz)

1-(4-Nitrobenzoyl)-3-(1-pyrrolidinyl)-1,2,3,4tetrahydroquinoline, yellow amorphous

 l_{H-NMR} (CDCl₃) δ : 1.70-1.95 (4H, m), 2.52-3.30 (7H, m), 3.80-4.22 (2H, m), 6.52 (1H, brs), 6.88 (1H, t, J=7.6 Hz), 6.96-7.11 (1H, m), 7.20 (2H, d, J=7.6 Hz), 7.54 (2H, d, J=8.8 Hz), 8.12 (2H, d, J=8.8 Hz)

1-(4-Nitrobenzoyl)-4-oxo-1,2,3,4-tetrahydro-quinoline, yellow powder, m.p. 189 - 190°C

1-(4-Nitrobenzoyl)-3-hydroxymethyl-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 97 - 100°C

1-(4-Nitrobenzoyl)-3-ethoxycarbonyl-1,2,3,4-tetrahydroquinoline, pale yellow powder, m.p. 162 - 163°C

1-(4-Nitrobenzoyl)-4-dimethylamino-1,2,3,4-tetrahydroquinoline, light brown oil

 ${}^{1}_{H-NMR}$ (CDCl₃) δ : 1.80-2.02 (1H, m), 2.20-2.50 (7H, m), 3.47 (1H, t, J=4.9 Hz), 3.70-3.88 (1H, m), 4.06-4.25 (1H, m), 6.46 (1H, d, J=7.5 Hz), 6.89 (1H, t, J=7.5 Hz), 7.05 (1H, t, J=7.5 Hz), 7.34 (1H, d, J=7.5 Hz), 7.50 (2H, d, J=7.0 Hz), 8.10 (2H, d, J=7.0 Hz)

Reference Example 4

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

1-(3-Aminobenzoyl)-1,2,3,4-tetrahydroquinoline, white powder, m.p. 128 - 130°C

1-(2-Aminobenzoyl)-1,2,3,4-tetrahydroquinoline,
yellow powder

 1 H-NMR (CDCl₃) δ : 2.01 (2H, quint, J=6.6 Hz), 2.81 (2H, t, J=6.6 Hz), 3.86 (2H, t, J=6.4 Hz), 4.6-4.8 (2H, m), 6.43 (1H, t, J=7 Hz), 6.66 (1H, d, J=8 Hz), 6.79 (1H, dd, J=1.4 Hz, J=7.6 Hz), 6.8-7.2 (5H, m)

3-Methyl-1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 197 - 200°C

4-Methyl-1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 197 - 199°C

2-Methyl-1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline, yellow powder, m.p. 204 - 206°C

1-(4-Aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, yellow powder, m.p. 172 - 174°C

1-(3-Methyl-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 156 - 158°C

1-(3-Methoxy-4-aminobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 165 - 167°C

1-(4-Aminobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, white powder, m.p. 177 - 179°C

1-(4-Aminobenzoy1)-3,4-dihydro-2H-1,4-benzoxazine,

white powder, m.p. 192 - 194°

1-(4-Aminobenzoyl)-1,2,3,5-tetrahydro-4,1-benz-oxazepine, yellow powder, m.p. 196 - 198°C

l-(4-Aminobenzoyl)-4-methyl-1,2,3,4-tetrahydroquinoxaline, yellow powder, m.p. 210 - 212°C

1-(4-Aminobenzoyl)-5-methyl-2,3,4,5-tetrahydro-1H-1,5-benzodiazepine, white powder, m.p. 159 - 161°C

1-(4-Aminobenzoyl)-4-methyl-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine, brown powder, m.p. 169 - 171°C

1-(3-Methoxy-4-aminobenzoy1)-4-methyl-2,3,4,5tetrahydro-1H-1,4-benzodiazepine, yellow oil

 1 H-NMR (CDCl₃) $_{6}$: 2.41 (3H, s), 2.9-3.2 (3H, m), 3.61 (3H, s), 3.6-4.2 (4H, m), 4.8-5.2 (1H, m), 6.38 (1H, d, J=8.1 Hz), 6.6-6.8 (3H, m), 6.9-7.2 (2H, m), 7.2-7.4 (1H, m)

l-(4-Aminobenzoyl)-4-n-propyl-2,3,4,5-tetrahydrolH-1,4-benzazepine, brown powder, m.p. 151 - 153°C

1-(4-Aminobenzoyl)-5-chloro-1,2,3,4-tetrahydro-quinoline, white powder, m.p. 174 - 175°C

1-(4-Aminobenzoyl)-6-methoxy-1,2,3,4-tetrahydroquinoline, pale yellow powder, m.p. 159 - 160°C

1-(4-Aminobenzoyl)-6-methyl-1,2,3,4-tetrahydroquinoline, white powder, m.p. 145 - 146°C

1-(4-Aminobenzoyl)-7-methoxy-1,2,3,4-tetrahydroquinoline, pale yellow powder, m.p. 150 - 2 °C

1-(4-Aminobenzoyl)-3-(4-methyl-1-p_perazinyl)1,2,3,4-tetrahydroquinoline, light beige powder, m.p. 157 -

159°C

1-(4-Aminobenzoy1)-3-(1-pyrrolidiny1)-1,2,3,4tetrahydroquinoline, pale yellow powder, m.p. 173 - 174.5°C

1-(4-Aminobenzoyl)-2,3-dihydro-4(lH)-quinolinone,
pale yellow powder, m.p. 178 - 180°C

1-(4-Aminobenzoy1)-3-hydroxymethyl-1,2,3,4-tetrahydroquinoline, white powder, m.p. 179 - 181°C

1-(4-Aminobenzoyl)-3-ethoxycarbonyl-1,2,3,4-tetrahydroquinoline, pale yellow amorphous

 1 H-NMR (CDCl₃) δ : 1.21 (3H, t, J=7.1 Hz), 3.00-3.24 (3H, m), 3.70-4.30 (6H, m), 6.48 (2H, d, J=8.5 Hz), 6.69 (1H, d, J=7.9 Hz), 6.77-7.30 (5H, m)

1-(4-Aminobenzoyl)-4-dimethylamino-1,2,3,4-tetrahydroguinoline, brown oil

 1 H-NMR (CDCl₃) δ : 1.83-2.05 (1H, m), 2.13-2.30 (1H, m), 2.34 (6H, m), 3.55-3.83 (2H, m), 3.89 (1H, brs), 3.97-4.18 (1H, m), 6.47 (2H, d, J=7.0 Hz), 6.68 (1H, d, J=7.9 Hz), 6.85-7.05 (2H, m), 7.20 (2H, d, J=7.0 Hz), 7.37 (1H, d, J=7.4 Hz)

Reference Example 5

To terephthalic acid monomethyl ester (15 g) is added thionyl chloride (100 ml) and the mixture is refluxed for 2 hours. The thionyl chloride is distilled off under reduced pressure to give terephthalic acid chloride monomethyl ester. Separately, to a solution of 1,2,3,4-tetrahydroquinoline (14.4 g) in dichloromethane (200 ml) is

added triethylamine (16.9 g) and further thereto is added slowly terephthalic acid chloride monomethyl ester obtained above under ice-cooling. Then, the mixture is stirred at room temperature for 1 hour. After completion of the reaction, water is added to the reaction mixture. The mixture is extracted with dichloromethane and dried over magnesium sulfate. The solvent is distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane) to give 1-(4-methoxycarbonylbenzoyl)-1,2,3,4-tetrahydroquinoline (22.7 g) as white powder, m.p. 72 - 74°C.

Reference Example 6

To a solution of 1-(4-methoxycarbonylbenzoyl)
1,2,3,4-tetrahydroquinoline (22.7 g) in methanol (300 ml) is

added 5 % aqueous sodium hydroxide solution (150 ml) and the

mixture is refluxed for 2 hours. Methanol is distilled off

under reduced pressure and the resulting residue is

acidified with diluted hydrochloric acid, extracted with

diethyl ether, and dried over magnesium sulfate. The

solvent is distilled off under reduced pressure and the

resulting crystal is collected by filtration to give 1-(4
carboxybenzoyl)-1,2,3,4-tetrahydroquinoline (13.2 g) as

white powder, m.p. 181 - 183°C.

Reference Example 7

Using the suitable starting materials, the

following compounds are obtained in the same manner as in Reference Example 1.

5-Dimethylamino-l-(4-nitrobenzoyl)-2,3,4,5tetrahydro-lH-benzazepine, pale yellow powder, m.p. 139 -

5-Dimethylamino-1-(3-methoxy-4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 139 141°C

4-(N-Methyl-N-ethylamino)-l-(4-nitrobenzoyl)
1,2,3,4-tetrahydroquinoline, pale yellow oil

¹H-NMR (CDCl₃) δ: 1.11 (3H, t, J=7.1 Hz), 1.90-2.25 (2H, m), 2.30 (3H, s), 2.57 (2H, q, J=7.1 Hz), 3.55-3.85 (2H, m), 4.00-4.21 (1H, m), 6.35-6.60 (1H, m), 6.80-6.98 (1H, t, J=7.9 Hz), 7.00-7.15 (1H, m), 7.33-7.60 (3H, m), 8.10 (2H, d, J=8.8 Hz)

4-Dimethylamino-1-(3-methoxy-4-nitrobenzoyl)1,2,3,4-tetrahydroquinoline, brown oil

¹H-NMR (CDCl₃) δ : 1.80-2.05 (1H, m), 2.33 (6H, s), 2.30-2.50 (1H, m), 3.40-3.52 (1H, m), 3.78 (3H, s), 3.70-3.88 (1H, m), 4.04-4.24 (1H, m), 6.52 (1H, d, J=8.2 Hz), 6.85-7.13 (4H, m), 7.28-7.38 (1H, m), 7.71 (1H, d, J=8.2 Hz)

1-(4-Nitrobenzoyl)-4-ethyl-2,3,4,5-tetrahydro-1H1,4-benzodiazepine, yellow oil

 1 H-NMR (CDCl₃) δ : 1.16 (3H, t, J=7.1 Hz), 2.5-2.7 (2H, m), 3.0-3.3 (3H, m), 3.98 (2H, q, J=14 Hz), 4.8-5.0 (1H, m), 6.59 (1H, d, J=7.7 Hz), 6.96 (1H, t, J=7.7 Hz),

```
7.14 (1H, t, J=7.4 Hz), 7.2-7.4 (3H, m), 8.02 (2H, d, J=8.8
Hz)
         1-(4-Nitrobenzoyl)-4-isopropyl-2,3,4,5-tetrahydro-
1H-1,4-benzodiazepine, yellow powder, m.p. 222 - 223°C
          l-(4-Nitrobenzoyl)-4-cyclohexyl-2,3,4,5-tetrahydro-
1H-1,4-benzodiazepine, brown oil
          ^{1}\text{H-NMR} (CDCl<sub>3</sub>) \delta : 1.0-1.5 (5H, m), 1.5-2.1 (5H,
m), 2.4-2.7 (1H, m), 2.9-3.3 (3H, m), 3.94 (2H, s), 4.9-5.1
(1H, m), 6.57 (1H, d, J=7.7 Hz), 6.8-7.0 (1H, m), 7.0-7.2
(1H, m), 7.2-7.4 (3H, m), 8.01 (2H, d, J=8.8 Hz)
          l-(4-Nitrobenzoyl)-5-methyl-1,2,3,4,5,6-hexahydro-
1,5-benzodiazocine, yellow oil
          1_{H-NMR} (CDCl<sub>3</sub>) \delta : 1.5-2.1 (2H, m), 2.40 (3H, s),
2.3-2.6 (lH, m), 2.8-3.2 (2H, m), 3.50 (lH, d, J=13.4 Hz),
3.84 (1H, d, J=13.4 Hz), 4.8-5.0 (1H, m), 7.0-7.3 (4H, m),
7.41 (2H, d, J=8.9 Hz), 8.00 (2H, d, J=8.9 Hz)
          1-(4-Nitrobenzoy1)-1,2,3,4-tetrahydro-5,1-
benzoxazepine, white powder, m.p. 144.5 - 145.5°C
          1-(2-Nitrobenzoyl)-4-methyl-2,3,4,5-tetrahydro-1H-
```

1,4-benzodiazepine, yellow powder, m.p. 177 - 180°C

1-(3-Nitrobenzoyl)-4-methyl-2,3,4,5-tetrahydro-lH
1,4-benzodiazepine, yellow powder, m.p. 145 - 146°C

6-Fluoro-1-(4-nitrobenzoyl)-1,2,3,4-tetrahydroquinoline, yellow needles, m.p. 145 - 146°C

Reference Example 8

Using the suitable starting materials, the

Hz, 1.5 Hz)

following compounds are obtained in the same manner as in Reference Example 2.

5-Dimethylamino-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder, m.p. 120 - 122°C

5-Dimethylamino-1-(3-methoxy-4-amino)-2,3,4,5tetrahydro-1H-benzazepine, white powder, m.p. 121 - 123°C

4-(N-Methy-N-ethylamino)-l-(4-aminobenzoyl)-

1,2,3,4-tetrahydroquinoline, orange amorphous

¹H-NMR (CDCl₃) δ : 1.11 (3H, t, J=7.1 Hz), 1.90-2.20 (2H, m), 2.28 (3H, s), 2.26 (2H, q, J=7.1 Hz), 3.60-4.25 (5H, m), 6.48 (2H, d, J=8.5 Hz), 6.69 (1H, d, J=7.9 Hz), 6.80-7.05 (2H, m), 7.24 (2H, d, J=8.5 Hz), 7.46 (1H, d, J=6.2 Hz)

4-Dimethylamino-1-(3-methoxy-4-aminobenzoyl)1,2,3,4-tetrahydroquinoline, pale yellow amorphous

¹H-NMR (CDCl₃) δ : 1.83-2.04 (1H, m), 2.15-2.32
(1H, m), 2.33 (6H, s), 3.50-3.82 (2H, m), 3.64 (3H, s),
3.95-4.18 (3H, m), 6.50 (1H, d, J=7.9 Hz), 6.65 (1H, dd,
J=7.9 Hz, 1.1 Hz), 6.78-7.03 (4H, m), 7.34 (1H, dd, J=7.5

1-(4-Aminobenzoyl)-4-ethyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine, white powder, m.p. 186 - 188°C

1-(4-Aminobenzoyl)-4-isopropyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine, white powder, m.p. 191 - 192°C

1-(4-Aminobenzoyl)-4-cyclohexyl-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine, white powder, m.p. 149.5 - 150.5°C l-(4-Aminobenzoyl)-5-methyl-1,2,3,4,5,6-hexahydro1,5-benzodiazocine, yellow powder, m.p. 143 - 145°C

l-(4-Aminobenzoyl)-1,2,3,4-tetrahydro-5,1
benzoxazepine, yellow powder, m.p. 163.5 - 164.5°C

l-(2-Aminobenzoyl)-4-methyl-2,3,4,5-tetrahydro-1H
1,4-benzodiazepine, yellow powder, m.p. 144 - 146°C

l-(3-Aminobenzoyl)-4-methyl-2,3,4,5-tetrahydro-1H
1,4-benzodiazepine, white powder, mp. 153 - 155°C

6-Fluoro-1-(4-aminobenzoyl)-1,2,3,4-tetrahydro
quinoline, white powder, m.p. 160.5 - 161.5°C

Reference Example 9

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

1-(2-Chloro-4-nitrobenzoyl)-4-methyl-2,3,4,5tetrahydro-1H-1,4-benzodiazepine

 l_{H-NMR} (CDCl₃) δ : 2.40 (3H, s), 2.96-3.33 (3H, m), 3.60-3.79 (1H, m), 3.96-4.23 (1H, m), 4.70-4.91 (1H, m), 6.80-7.43 (5H, m), 7.80-7.99 (1H, m), 8.08-8.21 (1H, m) $l_{L-1}(3-Methyl-4-nitrobenzoyl)-4-methyl-2,3,4,5-$

tetrahydro-1H-1,4-benzodiazepine

 1 H-NMR (CDCl₃) δ : 2.43 (3H, s), 2.48 (3H, s), 2.92-3.28 (3H, m), 3.91 (2H, AB-q, J=13.9 Hz, 45.5 Hz), 4.77-5.01 (1H, m), 6.54-6.70 (1H, m), 6.88-7.37 (5H, m), 7.62-7.78 (1H, m)

5-Dimethylamino-1-(2-chloro-4-nitrobenzoyl)-

```
2,3,4,5-tetrahydro-lH-benzazepine
```

 $^{1}\text{H-NMR}$ (CDCl₃) 6 : 1.23-2.57 (10H, m), 2.68-5.15 (3H, m), 6.79-7.45 (4H, m), 7.49-8.39 (3H, m)

5-Oxo-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder (ethyl acetate/n-hexane), m.p. 147-148°C

5-Hydroxy-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, white powder (ethyl acetate/n-hexane), m.p. 148 - 150°C

5-Methoxy-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, colorless amorphous

¹H-NMR (CDCl₃) 6: 1.47-2.48 (4H, m), 2.70-3.10 (1H, m), 3.26-3.64 (3H, m), 4.29-5.12 (2H, m), 6.60 (1H, d, J=7.7 Hz), 6.88-7.67 (5H, m), 7.92-8.12 (2H, m)

5-Ethoxycarbonylmethoxy-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 107 - 108°C (recrystallized from ethyl acetate/n-hexane)

5-(4-Bromobutoxy)-1-(4-nitrobenzoy1)-2,3,4,5tetrahydro-1H-benzazepine, colorless oil

 1 H-NMR (CDCl₃) δ : 1.49-2.55 (8H, m), 2.72-3.07 (1H, m), 3.24-3.77 (4H, m), 4.40-5.15 (2H, m), 6.53-6.66 (1H, m), 6.91-7.06 (1H, m), 7.07-7.80 (4H, m), 7.94-8.13 (2H, m)

5-(4-Dimethylaminobutoxy)-1-(4-nitrobenzoyl)- 2,3,4,5-tetrahydro-lH-benzazepine, colorless oil $$^{1}\text{H-NMR}$$ (CDCl3) δ : 1.51-1.88 (6H, m), 2.23-2.61

(4H, m), 2.27 (3H, s), 2.35 (3H, s), 2.74-3.14 (1H, m), 3.55-3.77 (2H, m), 4.48-5.11 (2H, m), 6.54-6.66 (1H, m), 6.91-7.04 (1H, m), 7.06-7.80 (4H, m), 7.93-8.11 (2H, m) 5-[4-(Phthalimid-1-yl)propoxy-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, colorless amorphous l_{H-NMR} (CDCl₃) δ : 1.48-2.56 (6H, m), 2.71-3.05 (1H, m), 3.40-4.05 (4H, m), 4.47-5.11 (2H, m), 6.50-6.64 (lH, m), 6.84-7.03 (lH, m), 7.03-7.20 (lH, m), 7.20-7.57 (2H, m), 7.57-7.93 (5H, m), 7.97-8.20 (2H, m) 5-Chloro-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-1Hbenzazepine, light brown powder 1_{H-NMR} (CDCl₃) δ : 1.75-3.3 (4H, m), 4.6-6.25 (3H, m), 6.45-6.7 (lH, m), 6.8-7.5 (4H, m), 7.55-7.7 (lH, m), 7.9-8.1 (2H, m) 5-Oxo-1-(2-chloro-4-nitorobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, pale yellow amorphous 1_{H-NMR} (CDCl₃) δ : 1.95-2.45 (2H, m), 2.94 (1H, t, J=6 Hz), 3.05-5.3 (2H, m), 6.96-7.1 (1H, m), 7.12-7.5 (3H, m), 7.75-7.85 (1H, m), 7.95-8.1 (1H, m), 8.14 (1H, s) 4-Dimethylaminomethyl-1-(4-nitrobenzoyl)-1,2,3,4tetrahydroquinoline, white powder, m.p. 117 - 119°C 3-Dimethylamino-l-(4-nitrobenzoyl)-2,3,4,5tetrahydro-lH-benzazepine, yellow oil 1_{H-NMR} (CDCl₃) δ : 1.5-1.7 (1H, m), 2.1-2.4 (1H, m), 2.42 (6H, s), 2.6-2.7 (1H, m), 2.8-3.0 (3H, m,, 5.1-5.3(1H, m), 6.62 (1H, d, J=7.8 Hz), 6.95 (1H, t, J=7.7 Hz),

7.14 (lH, t, J=7.5 Hz), 7.2-7.4 (3H, m), 8.00 (2H, d, J=8.9 Hz)

3-Dimethylamino-1-(3-methoxy-4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, yellow oil

 1 H-NMR (CDCl₃) & : 1.5-1.7 (1H, m), 2.0-2.3 (1H, m), 2.41 (6H, s), 2.5-2.8 (1H, m), 2.8-3.0 (3H, m), 3.75 (3H, s), 5.1-5.3 (1H, m), 6.6-6.8 (2H, m), 6.9-7.3 (4H, m), 7.59 (1H, d, J=8.3 Hz)

4-(4-Nitrobenzoyl)-3,4-dihydro-2H-1,4-benzothiazine, yellow powder, m.p. 180 - 182°C

5-(4-Nitrobenzoyl)-2,3,4,5-tetrahydro-1,5-benzothiazepine, yellow powder, m.p. 162 - 163°C

Reference Example 10

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

l-(2-Chloro-4-aminobenzoyl)-4-methyl-2,3,4,5tetrahydro-lH-1,4-benzodiazepine, white powder
(recrystallized from methanol/diethyl ether), m.p. 194.5 195.5°C

1-(3-Methyl-4-aminobenzoyl)-4-methyl-2,3,4,5tetrahydro-1H-1,4-benzodiazepine

¹H-NMR (CDCl₃) 6 : 2.01 (3H, s), 2.41 (3H, s), 2.82-3.21 (3H, m), 3.50-4.21 (4H, m), 4.78-5.14 (1H, m), 6.24-6.40 (1H, m), 6.59-6.82 (2H, m), 6.90-7.18 (3H, m), 7.19-7.34 (1H, m) 5-Dimethylamino-1-(2-chloro-4-aminobenzoyl)2,3,4,5-te_rahydro-1H-benzazepine, white powder
(recrystallized from dichloromethane/diethyl ether), m.p.
162 - 164°C

5-Dimethylamino-1-(2-methoxy-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (recrystallized from methanol/diethyl ether)

 l_{H-NMR} (CDCl₃) δ : 1.23-2.80 (llH, m), 2.90-3.38 (lH, m), 3.50-5.19 (6H, m), 5.87-6.41 (2H, m), 6.65-7.56 (5H, m)

5-Methoxy-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, white powder (recrystallized from ethyl acetate/n-hexane), m.p. 154 - 155°C

5-Ethoxycarbonylmethoxy-1-(4-aminobenzoy1)-2,3,4,5-tetrahydro-lH-benzazepine, white powder (recrystallized from ethyl acetate/n-hexane), m.p. 231 - 232°C

5-(4-Dimethylaminobutoxy)-l-(4-aminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, colorless oil

 l_{H-NMR} (CDCl₃) δ : 1.47-1.83 (6H, m), 1.83-2.54 (4H, m), 2.29 (6H, s), 2.61-3.00 (1H, m), 3.36-3.76 (2H, m), 4.35-5.20 (2H, m), 6.27-6.48 (2H, m), 6.57-6.76 (1H, m), 6.90-7.61 (5H, m)

```
(2H, m), 6.57-6.78 (1H, m), 6.87-7.57 (5H, m), 7.62-7.76 (2H, m), 7.76-7.97 (2H, m)
```

5-Chloro-1-(4-aminobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, pale yellow amorphous

¹H-NMR (CDCl₃) δ : 1.35-4.3 (7H, m), 4.55-6.7 (2H, m), 6.3-6.55 (2H, m), 6.6-6.8 (1H, m), 6.85-7.45 (5H, m)

5-Oxo-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-

benzazepine, pale yellow amorphous

¹H-NMR (CDCl₃) δ: 1.95-2.35 (2H, m), 2.89 (2H, t, J=6.3 Hz), 3.0-5.3 (4H, m), 6.35-6.47 (2H, m), 6.72-6.83 (1H, m), 7.0-7.15 (2H, m), 7.18-7.32 (2H, m), 7.81-7.93 (1H, m)

5-Oxo-1-(2-chloro-4-aminobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, white powder

 $^{1}\text{H-NMR}$ (CDCl₃) δ : 1.85-2.3 (2H, m), 2.87 (2H, t, J=6.2 Hz), 3.1-4.75 (4H, m), 6.15-7.5 (6H, m), 7.65-7.9 (1H, m)

4-Dimethylaminomethyl-1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline, white powder, m.p. 123 - 125°C

3-Dimethylamino-1-(4-aminobenzoyl)-2,3,4,5tetrahydro-1H-benzazepine, white powder, m.p. 175 - 177°C

3-Dimethylamino-1-(3-methoxy-4-aminobenzoyl)-

2,3,4,5-tetrahydro-lH-benzazepine, yellow oil

 $^{1}\text{H-NMR}$ (CDCl₃) δ : 1.5-1.7 (1H, m), 2.1-2.3 (1H, m), 2.3-2.6 (1H, m), 2.40 (6H, s), 2.7-3.0 (3H, m), 3.60 (3H, s), 3.8-4.0 (2H, br), 5.2-5.4 (1H, m), 6.37 (1H, d,

Reference Example 11

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

5-Carbamoyloxy-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 243 - 244°C (recrystallized from ethyl acetate/diisopropyl ether)

5-Methylaminocarbonyloxy-1-(4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 207 208°C (recrystallized from ethyl acetate/n-hexane)

5-Dimethylaminocarbonyloxy-1-(4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 155 156°C (recrystallized from ethyl acetate/diisopropyl
ether/n-hexane)

5-Methylidenyl-1-(4-nitrobenzoyl)-2,3,4,5tetrahydro-1H-benzazepine, colorless prisms, m.p. 133.5 134°C (recrystallized from ethyl acetate/diisopropyl ether)

5-Oxo-6-methyl-1-(2-chloro-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, colorless prisms, m.p. 90 - 92°C (recrystallized from ethanol)

1-(4-Nitrobenzoyl)-1,2,3,5-tetrahydro-4,1-

```
benzothiazepine, yellow powder, m.p. 185 - 187°C
(recrystallized from dichloromethane/diethyl ether)
         5-Dimethylamino-1-(2-dimethylamino-4-nitrobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, yellow powder, m.p. 123
125°C (recrystallized from diethyl ether/dichloromethane)
         5-Oxo-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-1,4-
benzodiazepine, white powder, m.p. 201.5 - 202.5°C
(recrystallized from diethyl ether/dichloromethane)
         5-0xo-4-methyl-1-(4-nitrobenzoyl)-2,3,4,5-
tetrahydro-1H-1,4-benzodiazepine, white powder, m.p. 136 -
138°C (recrystallized from diethyl ether/dichloromethane)
         5-Dimethylamino-1-(3-methyl-4-nitrobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, yellow-oil-
         ^{1}H-NMR (CDCl<sub>3</sub>) \delta : 1.16-3.18 (11H, m), 2.18 (3H,
s), 3.40-5.15 (2H, m), 6.50-7.68 (6H, m), 7.70-7.84 (1H, m)
         5-Dimethylamino-1-(2-methyl-4-nitrobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, colorless amorphous
         ^{1}\text{H-NMR} (CDCl<sub>3</sub>) \delta : 1.19-2.86 (11H, m), 2.20 (3H,
s), 2.94-3.24 (1H, m), 3.36-5.18 (1H, m), 6.49-8.20 (7H, m)
         5-Dimethylamino-1-(2-fluoro-4-nitrobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, yellow oil
         ^{1}H-NMR (CDCl<sub>3</sub>) & : 1.21-2.66 (10H, m), 2.66-5.11
(3H, m), 6.63-8.25 (7H, m)
         5-Dimethylamino-1-(3-fluoro-4-nitrobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 152 -
152.5°C (recrystallized from chloroform/diethyl ether)
```

will be a second to the contract of the contra

Reference Example 12

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

5-Carbamoyloxy-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder, m.p. 215 - 216°C (recrystallized from ethyl acetate/n-hexane)

5-Methylaminocarbonyloxy-1-(4-aminobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 192 195°C (recrystallized from ethyl acetate/n-hexane)

5-Dimethylaminocarbonyloxy-1-(4-aminobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 228 230°C (recrystallized from ethyl acetate/diisopropyl ether)

5-Methyl-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 155 - 156°C (recrystallized from ethyl acetate/n-hexane)

5-Oxo-6-methyl-1-(2-chloro-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 193 - 195°C (recrystallized from ethanol)

l-(4-Aminobenzoyl)-1,2,3,5-tetrahydro-4,1-benzothiazepine, white powder, m.p. 179 - 180°C (recrystallized
from dichloromethane/diethyl ether)

5-Dimethylamino-1-(2-dimethylamino-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 163 - 165°C (recrystallized from diethyl ether/dichloromethane)
5-Oxo-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-

or writing a process of the control of the control

```
benzazepine, yellow powder, m.p. 195 - 197°C (recrystallized
from diethyl ether/dichloromethane)
         5-Oxo-4-methyl-l-(4-aminobenzoyl)-2,3,4,5-tetra-
hydro-1H-1,4-benzazepine, yellow powder, m.p. 190 - 192°C
(recrystallized from diethyl ether/dichloromethane)
         5-Dimethylamino-1-(2-ethoxy-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. lll -
114°C (recrystallized from diethyl ether)
         5-Dimethylamino-1-(3-methyl-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, yellow oil
         ^{1}H-NMR (CDCl<sub>3</sub>) \delta : 0.66-2.56 (14H, m), 2.93-5.22
(4H, m), 6.23-7.80 (7H, m)
         5-Dimethylamino-1-(2-methyl-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 154 -
156°C (recrystallized from methanol/diethyl ether)
         5-Dimethylamino-1-(2-fluoro-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 161 -
163°C (recrystallized from dichloromethane/diethyl ether)
         5-Dimethylamino-1-(3-fluoro-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 156 -
157°C (recrystallized from methanol/diethyl ether)
         5-Oxo-1-(2-methoxy-4-aminobenzoyl)-2,3,4,5-tetra-
hydro-1H-benzazepine, colorless prisms, m.p. 160 - 160.5°C
(recrystallized from methanol/diethyl ether)
         Example 1
```

region from the control of the contr

To a solution of 1,2,3,4-tetrahydroquinoline (28.7

g) in acetone (400 ml) and water (200 ml) is added potassium carbonate (38.8 g) and further thereto is added 4-benzoyl-aminobenzoyl chloride (56 g) under ice-cooling. The mixture is stirred at room temperature overnight. Water is added to the reaction mixture, and the mixture is extracted with dichloromethane. The extract is dried over magnesium sulfate, and the solvent is distilled off under reduced pressure. The resulting residue is purified by silica gel column chromatography and recrystallized from methanol to give 1-[4-(benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (57 g) as white powder, m.p. 202.5 - 203.5°C.

Using the suitable starting materials, the compounds as shown in the following Table 1 are obtained in the same manner as in Example 1.

•

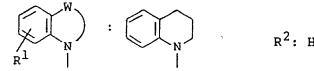
•

Table 1

$$\begin{array}{c|c}
 & W \\
 & N \\
 & R^1 \\
 & C=0 \\
 & R^2 \\
 & R^3
\end{array}$$

Example 2

Structure



Crystalline form: Light yellow powder

Recrystallization solvent: Methanol

Melting Point: 198.5 - 199.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : H

Crystalline form: White powder

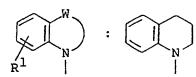
Recrystallization solvent: Methanol

Melting Point: 200.5 - 201.5°C

Form: Free

Example 4

Structure



_R2. н

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 206 - 207°C

Structure

$$\mathbb{R}^{1}$$

R²: н

Crystalline form: Yellow powder

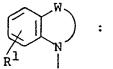
Recrystallization solvent: Methanol

Melting Point: 216 - 217°C

Form: Free

Example 6

Structure



 R^2 : H

$$R^3$$
: 4-NHC-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 202 - 203°C

Form: Free

_

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

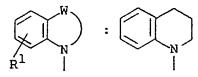
Recrystallization solvent: Methanol

Melting Point: 212 - 213°C

Form: Free

Example 8

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 167.5 - 168.5°C

Form: Free

e de currente para la la marte e el respector republicado, e per el cura de portente e de la companya de la cultura

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: Н

Crystalline form: White powder

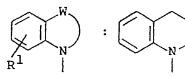
Recrystallization solvent: Methanol

Melting Point: 205 - 206°C

Form: Free

Example 10

Structure



R²: Н

$$R^3: 4-NHC-OH$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: >300°C

NMR analysis: 1)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : H

$$R^3: 4-NHC \longrightarrow OCH_3$$

Crystalline form: Yellow powder

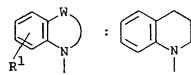
Recrystallization solvent: Methanol

Melting Point: 176 - 177°C

Form: Free

Example 12

Structure



 R^2 ; H

$$R^3$$
: 4-NHC- \sim -OCH₂CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol --

Melting Point: 219 - 220°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

$$R^3: 4-NHC \longrightarrow O(CH_2)_3CH_3$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 193 - 194°C

Form: Free

Example 14

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

$$R^3$$
: 4-NHC- $\sqrt{}$ -OCOCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 232 - 233°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

$$R^3: 4-NHC \longrightarrow SCH_3$$

Crystalline form: White powder

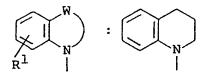
Recrystallization solvent: Methanol

Melting Point: 209 - 210°C

Form: Free

Example 16

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 184.5 - 185.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 224.5 - 225.5°C

Form: Free

Example 18

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 220.5 - 221.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

$$R^3$$
: 4-NHC- $\left(\begin{array}{c} O \\ \parallel \end{array}\right)$ -COCH₃

Crystalline form: Yellow powder

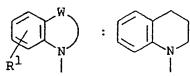
Recrystallization solvent: Methanol

Melting Point: 231 - 232°C

Form: Free

Example 20

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: >300°C

NMR analysis: 2)

Form: Free

. <u>.</u>

F

Structure

$$\mathbb{R}^{\mathbb{N}}$$
 : \mathbb{N}

 R^2 : H

Crystalline form: White powder

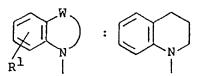
Recrystallization solvent: Methanol

Melting Point: 208 - 209°C

Form: Free

Example 22

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 234.5 - 235.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} :

к²: н

Crystalline form: Yellow powder

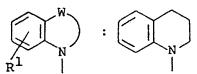
Recrystallization solvent: Methanol

Melting Point: 263.5 - 264.5°C

Form: Free

Example 24

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 237 - 238°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

Crystalline form: White powder

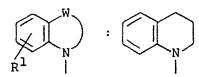
Recrystallization solvent: Methanol

Melting Point: 234 - 235°C

Form: Free

Example 26

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 236.5 - 237.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

$$R^3: 4-NHC-$$

Crystalline form: White powder

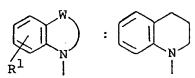
Recrystallization solvent: Methanol

Melting Point: 206.5 - 207.5°C

Form: Free

Example 28

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 210 - 211°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : H

Crystalline form: White powder

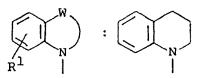
Recrystallization solvent: Methanol

Melting Point: 210.5 - 211.5°C

Form: Free

Example 30

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 178 - 179°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

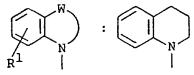
Recrystallization solvent: Methanol

Melting Point: 192 - 193°C

Form: Free

Example 32

Structure



 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 217 - 218°C

Form: Free

and the POTE of the electropic parameters of the gradient of the control of the c

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: н

$$R^3$$
: 4-NHC- \bigcirc OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 143 - 144°C

Form: Free

Example 34

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 170.5 - 171.5°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

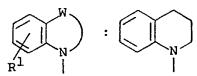
Recrystallization solvent: Methanol

Melting Point: 169.5 - 170.5°C

Form: Free

Example 36

Structure



R²: Н

$$R^3: 4-NHC \xrightarrow{O}_{SCH_3}$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 174.5 - 175.5°C

general and the adjets of the later of the endings and the segment regal, also a second of a garden

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 148.5 - 149.5°C

Form: Free

Example 38

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 165 - 166°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

к²: н

Crystalline form: White powder

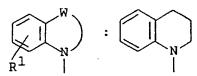
Recrystallization solvent: Methanol

Melting Point: 243 - 244°C

Form: Free

Example 40

Structure



R²: н

and a paragraph with the complete EAA and the Common of Equation of the Complete Common of the Complete Common

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 199 - 200°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

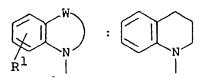
Recrystallization solvent: Methanol

Melting Point: 232.5 - 233.5°C

Form: Free

Example 42

Structure



R²: Н

$$R^3: 4-NHC NO_2$$

Crystalline form: White powder

Recrystallization solvent: Methanol ----

Melting Point: 178.5 - 179.5°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 205.5 - 206.5°C

Form: Free

Example 44

Structure





R²: H

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 234 - 235°C

Form: Free

4

agagarda fazigi eri eta elikaria eri kalangara kalangarakan eta gaga daga eta daga baranga baranga gagagahngag

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 225 - 226°C

Form: Free

Example 46

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

化双氯化物 解放 医囊胚后丛

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 224 - 225°C

		·	

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

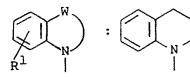
Recrystallization solvent: Methanol

Melting Point: 236 - 237°C

Form: Free

Example 48

Structure



R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 175.5 - 176.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} :

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 231 - 232°C

Form: Free

Example 50

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: Н

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 204 - 205°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

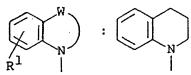
Recrystallization solvent: Methanol

Melting Point: 190 - 191°C

Form: Free

Example 52

Structure



 R^2 : H

an ang manang manang pangkan ng kalamang pangkan na manggan and Mala Ang penandahan ang pangkan ang manang man

$$R^3$$
: 4-NHC- OCH_3

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 156 - 157°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 200 - 201°C

Form: Free

Example 54

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

$$R^3$$
: 4-NHC OCH₃ OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 206 - 207°C

Form: Free

Same and the second of the sec

BODE SERVICE AND DESCRIPTION OF

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

$$R^3: 4-NHC \longrightarrow OCH_3$$

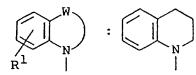
Crystalline form: Colorless amorphous

NMR analysis: 3)

Form: Free

Example 56

Structure



 R^2 : H

and the contracting the distribution of the contraction of the contrac

$$R^3: 4-NHC \xrightarrow{O CH_3} CH_3$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 215.5 - 216.5°C

Form: Free

Ł

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

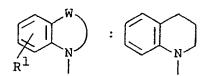
Recrystallization solvent: Methanol

Melting Point: 189 - 190°C

Form: Free

Example 58

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 203.5 - 204.5°C

Form: Free

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

 R^2 : H

$$R^3: 4-NHC \xrightarrow{CH_3}$$

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 254.5 - 255.5°C

Form: Free

Example 60

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} :

R²: н

Crystalline form: Brown powder

Recrystallization solvent: Methanol

Melting Point: 182.5 - 183.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: н

$$R^3$$
: 4-NHC- CH_3

Crystalline form: Colorless amorphous

NMR analysis: 4)

Form: Free

Example 62

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

$$R^3: 4-NHC \longrightarrow_{NO_2}^{NO_2}$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 263 - 264°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: White powder

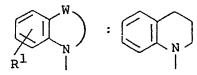
Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 217 - 218°C

Form: Free

Example 64

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 183 - 184°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

Crystalline form: Yellow powder

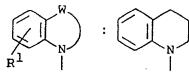
Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 207.5 - 208.5°C

Form: Free

Example 66

Structure



R². н

Crystalline form: Yellow powder

Recrystallization solvent: Dichloromethane/ethanol

and grower than the state of the contraction of the contraction of the state of the

Melting Point: 251 - 252°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

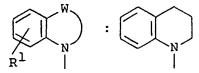
Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 208.5 - 209.5°C

Form: Free

Example 68

Structure



R²: Н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 231 - 232°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: н

O || R³: 4-NHCCH₃

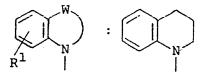
Crystalline form: Colorless amorphous

NMR analysis: 5)

Form: Free

Example 70

Structure



к²: н

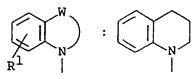
O || R³: 4-NHC(CH₂)₂CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 134 - 135°C

Structure



 R^2 : H

 R^3 : 4-NHC(CH₂)₄CH₃

Crystalline form: Yellow powder

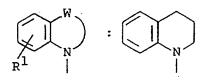
Recrystallization solvent: Methanol

Melting Point: 115 - 116°C

Form: Free

Example 72

Structure



R²: н

R³: 4-NHCCH₂C(CH₃)₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 178.5 - 179.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

O II R³: 4-NHCCH(CH₃)₂

Crystalline form: White powder

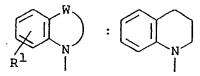
Recrystallization solvent: Methanol

Melting Point: 182.5 - 183.5°C

Form: Free

Example 74

Structure



R²: H

R³: 4-NHCC(CH₃)₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 164 - 165°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

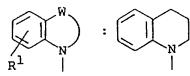
Crystalline form: Colorless amorphous

NMR analysis: 6)

Form: Free

Example 76

Structure



к²: н

Crystalline form: Yellow amorphous

NMR analysis: 7)

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1

 R^2 : H

Crystalline form: White powder

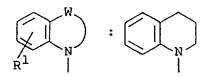
Recrystallization solvent: Methanol

Melting Point: 155 - 156°C

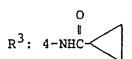
Form: Free

Example 78

Structure



R²: н



Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 182.5 - 183.5°C

Form: Free

reflect and arrange facilities follow

The second of th

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

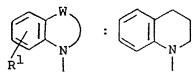
Recrystallization solvent: Methanol

Melting Point: 164.5 - 165.5°C

Form: Free

Example 80

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 165 - 167°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: H

Crystalline form: White powder

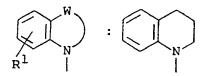
Recrystallization solvent: Methanol

Melting Point: 124 - 125°C

Form: Free

Example 82

Structure



R²: H

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 140.5 - 141.5°C

Form: Free

a vertada ministropa por calabia est projetar est referencia e a calabia estrega de per-

Structure

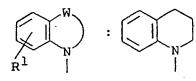
Crystalline form: Colorless amorphous

NMR analysis: 8)

Form: Free

Example 84

Structure



Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 211 - 212°C

Structure

R²: н

Crystalline form: White powder

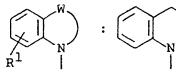
Recrystallization solvent: Methanol

Melting Point: 178 - 179°C

Form: Free

Example 86

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 212.5 - 213.5°C

Form: Free

gring general province and province in the community of the province of the province of the community of the

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

к²: н

Crystalline form: White powder

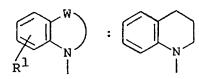
Recrystallization solvent: Methanol

Melting Point: 193 - 194°C

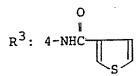
Form: Free

Example 88

Structure



к²: н



Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 203 - 204°C

Form: Free

Control of the Contro

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 9)

Form: Free

Example 90

Structure





R²: H

Crystalline form: Colorless amorphous

to the war it is a marker of the consequence of the

NMR analysis: 10)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

к²: н

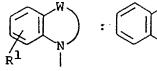
Crystalline form: Colorless amorphous

NMR analysis: 11)

Form: Free

Example 92

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 156.5 - 157.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

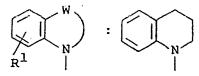
Crystalline form: Colorless amorphous

NMR analysis: 12)

Form: Free

Example 94

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

an an an air an an an an aig an galaig an air an an galain ag galain an an an an an an air galain air an an an

Melting Point: 203.5 - 204.5°C

Structure

R²: н

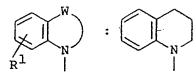
Crystalline form: Colorless amorphous

NMR analysis: 13)

Form: Free

Example 96

Structure



R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 126 - 127°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

$$R^3$$
: 2-NHC- \sim -OCH₃

Crystalline form: White powder

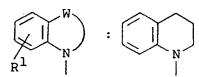
Recrystallization solvent: Methanol

Melting Point: 158.5 - 159.5°C

Form: Free

Example 98

Structure



R²: н

$$R^3: 2-NHC \xrightarrow{O}_{OCH_3}$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 129 - 130°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 131.5 - 132.5°C

Form: Free

Example 100

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: н

$$R^3$$
: 2-NHC-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 140 - 141°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 138.5 - 139.5°C

Form: Free

Example 102

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

_D2. _U

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 128 - 129°C

Form: Free

200 aliente de la companya de la co

Structure

к²: н

Crystalline form: White powder

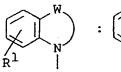
Recrystallization solvent: Methanol

Melting Point: 160 - 161°C

Form: Free

Example 104

Structure



: CH₃

R²: F

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 175 - 176°C

- 209 -

Example 105

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$: $\mathbb{C}^{\mathbb{N}}$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 197 - 198°C

Form: Free

Example 106

Structure

$$\mathbb{R}^{1} \longrightarrow \mathbb{R}^{N} : \mathbb{R}^{1}$$

R²: н

or ephylacometal bid

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 204 - 205°C

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^2 :

к²: н

$$R^3: 4-NHC \xrightarrow{O}_{CH_3}$$

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 174 - 175°C

Form: Free

Example 108

Structure

к²: н

Crystalline form: Yellow powder

Recrystallization solvent: Methanol

Melting Point: 202 - 203°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$: $\mathbb{C}^{\mathbb{N}}$

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 203 - 204°C

Form: Free

Example 110

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{H_{3}}$

R²: E

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 170.5 - 171.5°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$ $\mathbb{C}^{\mathbb{N}}$

R³: 4-NHC-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 149 - 150°C

Form: Free

Example 112

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 185 - 186°C

Form: Free

C

Structure

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 225 - 226°C

Form: Free

Example 114

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 234 - 235°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 149.5 - 150.5°C

Form: Free

Example 116

Structure

$$\begin{array}{c}
\text{re} \\
\text{W} \\
\text{R}^{1}
\end{array}$$
: $\begin{array}{c}
\text{Ch}_{3} \\
\text{N}
\end{array}$

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 197 - 198°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$ \mathbb{R}^{2}

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 204 - 205°C

Form: Free

Example 118

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$ \mathbb{R}^{2} : \mathbb{R}^{2} : \mathbb{R}^{2} : \mathbb{R}^{2}

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 224.5 - 225.5°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$ $\mathbb{C}^{\mathbb{N}}$

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 189.5 - 190.5°C

Form: Free

Example 120

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 221.5 - 222.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: Н

Crystalline form: Colorless needles

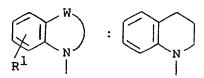
Recrystallization solvent: Methanol

Melting Point: 154 - 155°C

Form: Free

Example 122

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 165 - 166°C

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

 R^2 : H

$$R^3: 4-C-NH$$

CH₃

Crystalline form: Colorless needles

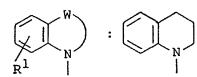
Recrystallization solvent: Methanol

Melting Point: 141 - 142°C

Form: Free

Example 124

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 165.5 - 166.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: н

$$R^3: 4-C-NH \xrightarrow{CH_3} CH_3$$

Crystalline form: Colorless needles

Recrystallization solvent: Methanol

Melting Point: 164 - 165°C

Form: Free

Example 126

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 203.5 - 204.5°C

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^N :

к²: н

$$R^3: 4-C-NH$$

Crystalline form: White powder

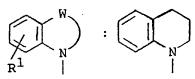
Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 236.5 - 237.5°C

Form: Free

Example 128

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 206.5 - 207.5°C

Form: Free

а арбинация ремети набродник и не 1974 г. Маничи стандром начины и брезов 1984. В Сем и стане и начувает истор

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 271 - 272°C

Form: Free

Example 130

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 246 - 247°C

Form: Free

(大学) [35] ([36] [36] (47) (47) (47) (47) (47) (48) (48) (48) (48)

. . . .

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: Н

Crystalline form: White powder

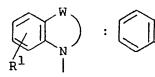
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 210 - 211°C

Form: Free

Example 132

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 230.5 - 231.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

$$R^3$$
: 4-NHC- CH_3

Crystalline form: White powder

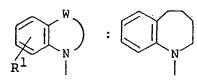
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 203 - 204°C

Form: Free

Example 134

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 170 - 171°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 225.5 - 226.5°C

Form: Free

Example 136

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

$$R^3$$
: 4-NHC- \bigcirc OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 210.5 - 211.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R} \mathbb{R}

R²: н

$$R^3: 4-NHC \longrightarrow_{OCH_3}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 183 - 184°C

Form: Free

Example 138

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 191.5 - 192.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 203.5 - 204.5°C

Form: Free

Example 140

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 215.5 - 216.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 211.5 - 212.5°C

Form: Free

Example 142

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

$$R^3$$
: 4-NHC- \sim -CN

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 280.5 - 281.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 235.5 - 236.5°C

Form: Free

Example 144

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/dichloromethane

Melting Point: 249.5 - 250.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 217 - 218°C

Form: Free

Example 146

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 201.5 - 203°C

stilled have been a sufficiently as well as the control of the con

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 $R^2: 3-CH_3$

Crystalline form: White powder

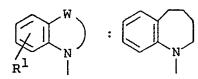
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 221 - 222°C

Form: Free

Example 148

Structure



R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 193 - 194°C

Form: Free

.

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R} \mathbb{R}

 R^2 : 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 176 - 177°C

Form: Free

Example 150

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^N \mathbb{R}^N

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 188 - 189.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 227 - 228°C

Form: Free

Example 152

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: 3-СН₃

$$R^3$$
: 4-NHC- CH_3

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 186 - 187°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1}

R²: 3-осн₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 135 - 136°C

Form: Free

Example 154

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 173 - 174°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} : \mathbb{N}

R²: 3-OCH₃

Crystalline form: White powder

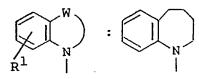
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 174.5 - 175.5°C

Form: Free

Example 156

Structure



R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 156 - 157°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 153 - 154°C

Form: Free

Example 158

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 169 - 170°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 185 - 186°C

Form: Free

Example 160

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 213 - 214°C

Structure

$$\bigcap_{\mathbb{R}^1} \bigvee_{\mathbb{N}} : \bigcap_{\mathbb{N}} \bigvee_{\mathbb{R}^2 \colon H}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 240 - 241°C

Form: Free

Example 162

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{2} : \mathbb{R}^{2} :

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 225 - 226°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: Н

$$R^3: 4-NHC \xrightarrow{O}_{CH_3}$$

Crystalline form: White powder

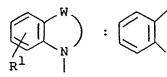
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 209.5 - 210.5°C

Form: Free

Example 164

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 198 - 199°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{2} :

$$R^3$$
: 4-NHC-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 214.5 - 215.5°C

Form: Free

Example 166

Structure

$$R^3: 4-NHC-$$
OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 196.5 - 197.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 194 - 195°C

Form: Free

Example 168

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 191 - 192°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

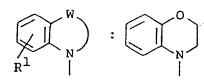
Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 227 - 228°C

Form: Free

Example 170

Structure



R²: I

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 182 - 183°C

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \end{array} \right) : \left(\begin{array}{c} \\ \\ \\ \end{array} \right)$$

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 222 - 223°C

Form: Free

Example 172

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 204 - 205°C

Structure

$$\left(\begin{array}{c} \left(\begin{array}{c} W \\ \end{array}\right) \end{array}\right) : \left(\begin{array}{c} O \\ \end{array}\right)$$

R²: н

$$R^3: 4-NHC \xrightarrow{O}_{CH}$$

Crystalline form: White powder

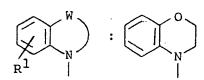
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 194 - 195°C

Form: Free

Example 174

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 213 - 214°C

Structure

$$\left(\begin{array}{c} \left(\begin{array}{c} W \\ R^1 \end{array} \right) \end{array} \right) : \left(\begin{array}{c} \left(\begin{array}{c} W \\ N \end{array} \right) \end{array} \right)$$

 R^2 : H

Crystalline form: White powder

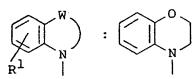
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 201 - 202°C

Form: Free

Example 176

Structure



R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 173 - 174°C

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \end{array} \right) : \left(\begin{array}{c} \\ \\ \\ \\ \end{array} \right)$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 150.5 - 151.5°C

Form: Free

Example 178

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

$$R^3$$
: 4-NHC CH_3

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 207.5 - 208.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: E

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 256.5 - 257.5°C

Form: Free

Example 180

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 199.5 - 200.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{0} \mathbb{R}^{0}

к²: н

Crystalline form: White powder

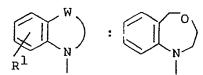
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 211 - 212°C

Form: Free

Example 182

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 189.5 - 190.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : E

$$R^3: 4-NHC \xrightarrow{O}_{CH_3}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 176.5 - 177.5°C

Form: Free

Example 184

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 202 - 203°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 219 - 220°C

Form: Free

Example 186

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl_ether

Melting Point: 272 - 273°C

Structure

R²: H

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 146 - 147°C

Form: Free

Example 188

Structure

R²: Н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 229.5 - 230.5°C

Structure

R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 119.5 - 120.5°C

Form: Free

Example 190

Structure

$$\begin{array}{c}
\text{re} \\
\text{CH}_{2} \\
\text{N}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3} \\
\text{N}
\end{array}$$

R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 189 - 190°C

Structure

R²: H

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 207 - 208°C

Form: Free

Example 192

Structure

R²: H

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 196.5 - 197.5°C

Structure

R²: н

$$R^3$$
: 4-NHC- \bigcirc OCH₃

Crystalline form: Yellow powder

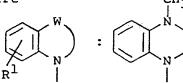
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 182 - 183°C

Form: Free

Example 194

Structure



R²: н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 172 - 173°C

Structure

к²: н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 197.5 - 198.5°C

Form: Free

Example 196

Structure

_Б2. н

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 227 - 228°C

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 216.5 - 217.5°C

Form: Free

Example 198

Structure

$$\begin{array}{c}
\text{CH} \\
\text{CH} \\
\text{N}
\end{array}$$

$$\begin{array}{c}
\text{CH} \\
\text{N}
\end{array}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 207 - 208°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 236 - 237°C

Form: Free

Example 200

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 199.5 - 200.5°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3} \\
\text{N} \\
\text{N}
\end{array}$$

R²: н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 171.5 - 172.5°C

Form: Free

Example 202

Structure

$$\begin{array}{c}
\text{CH}_{3} \\
\text{N} \\
\text{R}^{1} \\
\text{N}
\end{array}$$

в². н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 222.5 - 223.5°C

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 209.5 - 210.5°C

Form: Free

Example 204

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 14)

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 15)

Form: Hydrochloride

Example 206

Structure

R²: н

$$R^3$$
: 4-NHC- CH_3

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 16)

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 17)

Form: Hydrochloride

Example 208

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$
: $\begin{array}{c}
\text{N} \\
\text{N}
\end{array}$

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 18)

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$
: $\begin{array}{c}
\text{CH}_{3} \\
\text{N}
\end{array}$

R²: E

$$R^3$$
: 4-NHC-CH₃

Crystalline form: Yellow powder

Recrystallization solvent: Ethanol/water

NMR analysis: 19)

Form: Hydrochloride

Example 210

Structure

$$\begin{array}{c}
\text{R}^{1} \\
\text{R}^{1}
\end{array}$$
: $\begin{array}{c}
\text{R}^{1} \\
\text{N}
\end{array}$

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 20)

Structure

 R^2 : E

$$R^3: 4-NHC \longrightarrow OCH_3$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 159.5 - 160.5°C

Form: Free

Example 212

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 189.5 - 190.5°C

and fighter was a protection of grey and high protection are the black as a fight first interpretable of the c

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 170.5 - 171.5°C

Form: Free

Example 214

Structure

$$R^3$$
: 4-NHC- \bigcirc OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Company of the American State of the Company of the

Melting Point: 165 - 166°C

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 173.5 - 174.5°C

Form: Free

Example 216

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 182 - 183°C

Structure

к²: н

Crystalline form: White powder

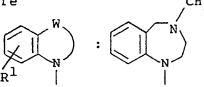
Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 225.5 - 226.5°C

Form: Free

Example 219

Structure



R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 21)

Structure

 $R^2: 3-OCH_3$

Crystalline form: White powderr

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 147.5 - 148.5°C

Form: Free

Example 221

Structure

R²: 3-ОСН₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 136 - 137°C

Form: Free

¥

Structure

R²: 3-ОСН₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 191.5 - 192.5°C

Form: Free

Example 223

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array}\right)$$

R²: 3-осн₃

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 145 - 146°C

Structure

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 22)

Form: Hydrochloride

Example 225

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{R}^{2}: \text{ F}
\end{array}$$

$$R^3$$
: 4-NHC- CH_3

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 23)

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 24)

Form: Hydrochloride

Example 227

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 25)

Structure

$$(Cn_2)_2cn_3$$

$$(Cn_2)_2cn_3$$

$$R^2: I$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 26)

Form: Hydrochloride

Example 229

Structure

$$\begin{array}{c}
(CH_2)_2CH_3 \\
N \\
R^2: E
\end{array}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

NMR analysis: 27)

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 206 - 207°C

Form: Free

Example 231

Structure

$$\begin{array}{c}
\text{R1} \\
\text{N}
\end{array}$$

к²: н

Crystalline form: White powder

Recrystallization solvent: Chloroform/methanol

Melting Point: 211 - 213°C

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Chloroform/methanol

Melting Point: 228.5 - 229.5°C

Form: Free

Example 233

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{Ch}_{3} \\
\text{N} \\
\text{N}
\end{array}$$

_в2. н

$$R^3: 4-NHC \xrightarrow{O}_{C1}^{C1}$$

Crystalline form: White powder

Recrystallization solvent: Chloroform/methanol

Melting Point: 237 - 238°C

Structure

$$\mathbb{R}^1$$
 $\mathbb{C}^{\mathbb{N}}$ $\mathbb{C}^{\mathbb{N}}$

R²: Н

Crystalline form: White powder

Recrystallization solvent: Chloroform/methanol

Melting Point: 226 - 228°C

Form: Free

Example 235

Structure

$$\bigcap_{\mathbb{R}^1} \bigvee_{\mathbb{N}} : \bigcap_{\mathbb{N}} \bigvee_{\mathbb{N}}$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Chloroform/methanol

Melting Point: 220 - 222°C

Structure

re
$$CO_2C_2H_5$$
 R^2

Crystalline form: Colorless amorphous

NMR analysis: 28)

Form: Free

Example 237

Structure

re
$$\mathbb{C}^{H_2OH}$$
 : \mathbb{R}^2

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 162 - 165°C

Structure

Crystalline form: Light brown amorphous

NMR analysis: 29)

Form: Free

Example 239

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{2} : H

Crystalline form: Light brown amorphous

NMR analysis: 30)

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 215 - 217°C

Form: Free

Example 241

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 221 - 223°C

R²: н

Example 242

Structure

re
$$\mathbb{N}$$
 : \mathbb{N} : \mathbb{N}

O C1

Crystalline form: Colorless amorphous

NMR analysis: 31)

Form: Free

Example 243

Structure

$$\begin{array}{c}
\text{re} \\
\text{NH}_{2}
\end{array}$$

R²: E

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 207 - 210°C

Form: Free

agrand make

Structure

$$R^3$$
: 4-NHC $C1$

Crystalline form: Colorless amorphous

NMR analysis: 32)

Form: Free

Example 245

Structure

re
$$\mathbb{R}^{1}$$
 \mathbb{R}^{1}
 \mathbb{R}^{2} :
 \mathbb{R}^{2} :
 \mathbb{R}^{2} :
 \mathbb{R}^{2}

Crystalline form: Colorless amorphous

NMR analysis: 33)

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 34)

Form: Free

Example 247

Structure

are
$$N(CH_3)_2$$
 R^1 N

 \mathbb{R}^2 : H

$$R^3: 4-NHC \xrightarrow{O} C1$$

Crystalline form: Colorless amorphous

NMR analysis: 35)

Form: Free

u systyma ur grangi i i negyven elektrik i

Structure

re
$$CON(CH_3)_2$$
 R^1 $|$

к²: н

Crystalline form: Light yellow powder

Recrystallization solvent: Ethanol

Melting Point: 186 - 187°C

Form: Free

Example 249

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 190 - 191°C

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

²: H

Crystalline form: Light yellow scales

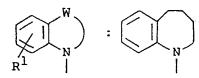
Recrystallization solvent: Ethanol/water

Melting Point: 230 - 231°C

Form: Free

Example 251

Structure



R²: н

Crystalline form: Light yellow needles

Recrystallization solvent: Ethanol

Melting Point: 227 - 228°C

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

к²: н

O || R³: 4-NHCCH₂CH₂COOH

Crystalline form: Colorless needles

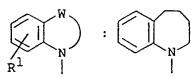
Recrystallization solvent: Ethyl acetate

Melting Point: 192°C

Form: Free

Example 253

Structure



 R^2 : H

 $\begin{array}{c} \text{O} \\ \text{} \\ \text{R}^3 \colon \text{ 4-NHCCH}_2\text{CH}_2\text{CH}_2\text{COOH} \end{array}$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 186.5 - 189°C

Form: Free

.

*

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

$$\begin{array}{ccc} & & \text{O} & & \text{O} \\ & & \parallel & & \parallel \\ \text{R}^3 \colon & 4-\text{NHC}(\text{CH}_2)_2\text{CN}(\text{C}_2\text{H}_5)_2 \end{array}$$

Crystalline form: Light yellow scales

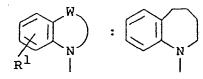
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 165 - 167°C

Form: Free

Example 255

Structure



 R^2 : E

 R^3 : 4-NHC(CH₂)₂CNH(CH₂)₃CH₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 169 - 170°C

Form: Free

a digital on the Matin professionary consistency any argument and the control of the consequence and the consequence of the con

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

Crystalline form: Colorless scales

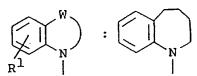
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 174 - 177°C

Form: Free

Example 257

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 114 - 118°C

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 170 - 172°C

Form: Free

Example 259

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : F

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 179 - 181°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: Н

O O | 1 R³: 4-NHC(CH₂)₂CNH₂

Crystalline form: White powder

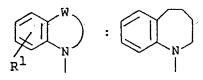
Recrystallization solvent: Ethyl acetate

Melting Point: 118 - 121°C

Form: Free

Example 261

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 144 - 148°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{2} : \mathbb{R}^{2} : \mathbb{R}^{2} : \mathbb{R}^{2}

R³: 4-NHC(CH₂)₃CNH(CH₂)₃CH₃

Crystalline form: Colorless scales

Recrystallization solvent: Ethyl acetate

Melting Point: 156 - 157°C

Form: Free

Example 263

Structure

$$R^3: 4-NHC(CH_2)_3CNH$$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 204 - 206°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{2} :

R³: 4-NHCCH₂Cl

Crystalline form: Light yellow powder

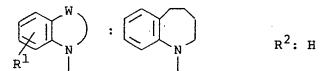
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 165 - 167°C

Form: Free

Example 265

Structure



R³: 4-NHCCH₂CH₂Cl

Crystalline form: Light yellow amorphous

NMR analysis: 36)

Form: Free

•

*

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

 R^3 : 4-NHC(CH₂)₃Cl

Crystalline form: White powder

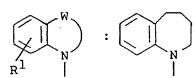
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 122 - 124°C

Form: Free

Example 267

Structure



 R^2 : H

 R^3 : 4-NHCCH₂CO₂C₂H₅

Crystalline form: Light yellow powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 116 - 117°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

к²: н

R³: 4-NHC(CH₂)₂CO₂C₂H₅

Crystalline form: White powder

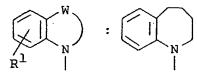
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 121 - 123°C

Form: Free

Example 269

Structure



R²: Н

O Br | | R³: 4-NHC-CH-C₂H₅:

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate

Melting Point: 186 - 187°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} :

R²: н

Crystalline form: White powder

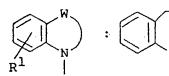
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 139 - 142°C

Form: Free

Example 271

Structure



R²: н

Crystalline form: Light yellow amorphous

NMR analysis: 37)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

R³: 4-NHCCH₂NHCH(CH₃)₂

Crystalline form: White powder

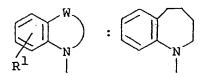
Recrystallization solvent: Ethyl acetate

Melting Point: 149.5 - 152.5°C

Form: Free

Example 273

Structure



R²: н

R³: 4-NHCCH₂NHC(CH₃)₃

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 150 - 152.5°C

Form: Free

1

£

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

R²: Н

 \mathbb{R}^3 : 4-NHCCH₂NH(CH₂)₂OH

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 150°C

Form: Free

Example 275

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: н

 R^3 : 4-NHCCH₂N(C₂H₅)₂

Crystalline form: Colorless needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 101 - 104°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 120 - 122°C

Form: Free

Example 277

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: H

$$R^3: 4-NHCCH_2N-CH_2$$

Crystalline form: Light yellow amorphous

NMR analysis: 38)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless needles

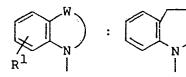
Recrystallization solvent: Ethanol

Melting Point: 183 - 186°C

Form: Free

Example 279

Structure



R²: E

Crystalline form: Light brown powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 139 - 142°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: E

Crystalline form: Light yellow powder

Recrystallization solvent: Ethanol

Melting Point: 162 - 165°C

Form: Free

Example 281

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Light yellow scales

Recrystallization solvent: Ethyl acetate

Melting Point: 224 - 227°C

Structure

$$\mathbb{R}^{1}$$
 :

R²: H

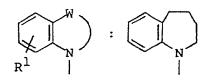
Crystalline form: Light yellow amorphous

NMR analysis: 39)

Form: Free

Example 283

Structure



R²: н

Crystalline form: Light yellow powder

Recrystallization solvent: Ethanol/water

Melting Point: 162 - 164°C

Structure

$$\mathbb{Q}^{\mathbb{Q}}$$
 : $\mathbb{Q}^{\mathbb{Q}}$

R²: н

O || R³: 4-NHCCH₂NH₂

Crystalline form: Light yellow powder

Recrystallization solvent: Ethanol

Melting Point: 238 - 241°C (decomposed)

Form: Hydrochloride

Example 285

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

O O II II II R3: 4-NHCCH2NHCCH3

Crystalline form: Light yellow amorphous

NMR analysis: 40)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 41)

Form: Free

Example 287

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 168 - 169°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

Crystalline form: Light brown powder

Recrystallization solvent: Ethanol

Melting Point: 189 - 191°C

Form: Free

Example 289

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 200 - 202°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless scales

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 143 - 146°C

Form: Free

Example 291

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3$$
: $4-NHCCH_2NH-CH_3$

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 117 - 117.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

Crystalline form: Light brown powder

Recrystallization solvent: Diethyl ether/ethyl acetate

Melting Point: 225 - 226°C

Form: Free

Example 293

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethanol

Melting Point: 175 - 176.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 234 - 236°C

Form: Free

Example 295

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

_R2. н

Crystalline form: Colorless scales

Recrystallization solvent: Ethyl acetate

Melting Point: 172 - 174°C

Structure

$$\mathbb{R}^1$$
 :

R²: н

Crystalline form: White powder

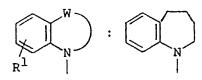
Recrystallization solvent: Ethyl acetate

Melting Point: 154 - 155°C

Form: Free

Example 297

Structure



R²: Н

Crystalline form: Light yellow needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 181.5 - 183.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHCCH_2NH \xrightarrow{CH_3} -CH_3$$

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 173 - 175°C

Form: Free

Example 299

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHCCH_2N$$

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 137 - 138°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

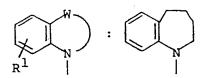
Crystalline form: Light yellow amorphous

NMR analysis: 42)

Form: Free

Example 301

Structure



 R^2 : H

Crystalline form: Colorless needles

Recrystallization solvent: Diethyl ether/ethyl acetate

Melting Point: 129 - 130°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHCCH_2N$$

Crystalline form: Colorless needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 181 - 183°C

Form: Free

Example 303

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 248 - 249°C

Structure

R²: Н

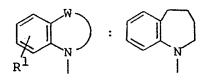
Crystalline form: Light yellow amorphous

NMR analysis: 43)

Form: Free

Example 305

Structure



R²: н

Crystalline form: Light yellow needles.

Recrystallization solvent: Ethanol

Melting Point: 94 - 96°C

Structure

$$\mathbb{R}^{1}$$
 :

к²: н

Crystalline form: Light brown powder

Recrystallization solvent: Ethyl acetate

Melting Point: 159 - 161°C

Form: Free

Example 307

Structure

$$\mathbb{R}^{1}$$
 :

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 180 - 183°C

Structure

$$\mathbb{R}^1$$
 :

R²: Н

Crystalline form: Light brown powder

Recrystallization solvent: Ethanol

Melting Point: 177 - 180°C

Form: Free

Example 309

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

$$R^3$$
: 4-NHC(CH₂)₃N

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 91 - 93°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

Crystalline form: Light brown scales

Recrystallization solvent: Ethanol

Melting Point: 155 - 156.5°C

Form: Free

Example 311

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

Crystalline form: Colorless scales

Recrystallization solvent: Ethyl acetate

Melting Point: 172.5 - 175°C

Structure

$$\mathbb{R}^1$$
:

R²: н

Crystalline form: White powder

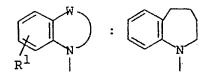
Recrystallization solvent: Ethanol

Melting Point: 148 - 150.5°C

Form: Free

Example 313

Structure



R²: Н

$$R^3$$
: 4-NHCCH₂O- CH_3

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 172 - 173°C

Form: Free

?

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless scales

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 133 - 135°C

Form: Free

Example 315

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 217 - 219°C

Structure

$$\mathbb{R}^{1}$$
 :

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate

Melting Point: 226 - 227.5°C

Form: Free

Example 317

Structure

$$\mathbb{R}^1$$
 :

R²: н

$$R^3$$
: 4-NHC-CHNH

Crystalline form: Colorless amorphous .

NMR analysis: 44)

Form: Free

?

÷

æį

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: н

Crystalline form: White powder

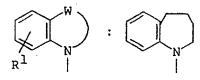
Recrystallization solvent: Dichloromethane

Melting Point: 234 - 235°C

Form: Free

Example 319

Structure



R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol

Melting Point: 218 - 218.5°C

Structure

$$\mathbb{R}^{1} \quad | \quad |$$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 202.5 - 206°C

Form: Free

Example 321

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 174 - 176°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: H .

$$R^3: 4-NHC \xrightarrow{O} OCH_3$$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 216 - 218°C

Form: Free

Example 323

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHC \longrightarrow_{NO_2}^{O}$$

Crystalline form: White powder

Melting Point: >300°C

NMR analysis: 45)

Structure

$$\mathbb{R}^{1}$$
 :

R²: н

Crystalline form: Colorless prisms

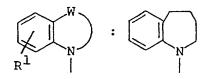
Recrystallization solvent: Ethanol

Melting Point: 250.5 - 251°C

Form: Free

Example 325 ···

Structure



 R^2 : H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 223 - 225°C

Form: Free

?

. 1

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: н

$$R^3: 4-NHC$$

Crystalline form: Colorless prismsr

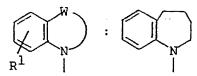
Recrystallization solvent: Methanol

Melting Point: 213 - 214°C

Form: Free

Example 327

Structure



R²: Н

$$R^3: 4-NHC$$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 246 - 247°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Colorless prisms

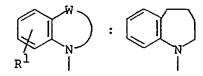
Recrystallization solvent: Methanol

Melting Point: 248 - 251°C

Form: Free

Example 329

Structure



R²: H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 268.5 - 270.5°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHC \longrightarrow O(CH_2)_6 N C_2 H_5$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 174 - 176°C

Form: Hydrochloride

Example 331

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 130 - 134°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 214 - 217°C

Form: Hydrochloride

Example 333

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 218 - 220°C

Form: Hydrochloride

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: н

$$R^3$$
: $4-NHC - NH_2$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 222 - 225°C

Form: Free

Example 335

Structure

$$\mathbb{R}^{1}$$
 :

 R^2 : H

Crystalline form: Colorless needles

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 171 - 172°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{2} : \mathbb{R}^{2} : \mathbb{R}^{2}

$$R^3: 4-NHC \longrightarrow O(CH_2)_6N \longrightarrow N$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 235.5 - 236°C

Form: Dihydrochloride

Example 337

Structure

$$\mathbb{R}^1$$
:

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 241 - 243°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: White powder

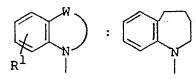
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 187 - 191°C

Form: Free

Example 339

Structure



R²: н

$$R^3: 4-NHC \longrightarrow O(CH_2)_6N \xrightarrow{CH_3}$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 240 - 244°C

Form: Hydrochloride

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 181 - 182°C

Form: Free

Example 341

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHC$$

O O(CH₂)₄N

N-CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 188 - 190°C

Form: Dihydrochloride

Structure

$$\mathbb{R}^{1}$$
 :

R²: н

Crystalline form: White powder

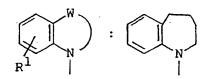
Recrystallization solvent: Isopropyl alcohol

Melting Point: 218 - 218.5°C

Form: Hydrochloride

Example 343

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 243 - 245.5°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHC \longrightarrow -O(CH_2)_6NH_2$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 130 - 133°C

Form: Free

Example 345

Structure

$$\mathbb{R}^1$$
 :

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 155 - 158°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 208 - 210°C

Form: Hydrochloride

Example 347

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 154 - 155°C

Form: Hydrochloride

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

$$R^3: 4-NHC \longrightarrow O(CH_2)_4NHCOCH_3$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 142 - 143°C

Form: Free

Example 349

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 120 - 125°C

Form: Hydrochloride

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 91 - 95°C

Form: Hydrochloride

Example 351

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 145 - 146.5°C

к²: н

Example 352

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 105 - 105.5°C

Form: Free

Example 353

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 151 - 155°C

Form: Dihydrochloride

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: Н

$$R^3: 4-NHC-$$

O O(CH₂)6NH₂

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 135.5 - 137.5°C

Form: Free

Example 355

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

$$R^3: 4-NHC \longrightarrow R^3$$

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 178 - 178.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane

Melting Point: 266.5 - 268°C

Form: Free

Example 357

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 123 - 124°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: н

$$\begin{array}{c} \text{O OCH}_2\text{CO}_2\text{C}_2\text{H}_5\\ \text{R}^3\colon \text{ 4-NHC} \end{array}$$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 212 - 213.5°C

Form: Free

Example 359

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: Colorless scales

Recrystallization solvent: Ethyl acetate

Melting Point: 160.5 - 162°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: H

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 103 - 105°C

Form: Free

Example 361

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 145 - 146°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 247 - 250°C

Form: Free

Example 363

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

$$R^3: 4-NHC \xrightarrow{O. OCH_2CONH_2}$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 198 - 199°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

Crystalline form: White powder

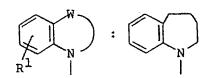
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 181.5 - 182.5°C

Form: Free

Example 365

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 170 - 170.5°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 156 - 158°C

Form: Hydrochloride

Example 367

Structure

Crystalline form: White powder

Recrystallization solvent: Diethyl ether

Melting Point: 168.5 - 170.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 177 - 181.5°C

Form: Hydrochloride

Example 369

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: Н

$$R^3: 4-NHC- N-CCH_3$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 211 - 213°C

Structure

$$\mathbb{R}^{1}$$
 :

R²: H

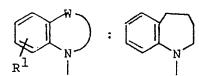
Crystalline form: White powder

NMR analysis: 46)

Form: Free

Example 371

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol/ethyl acetate

Melting Point: 166 - 167°C

Structure

$$\mathbb{R}^1$$
 \mathbb{N} :

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 127 - 131°C

Form: Free

Example 373

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 170 - 171°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

_R2. н

$$R^3: 4-C-N$$
 OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 125 - 126°C

Form: Free

Example 375

Structure

 R^2 : H

Crystalline form: Light yellow amorphous

NMR analysis: 47)

Form: Hydrochloride

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 48)

Form: Hydrochloride

*

- l) lH-NMR (CDCl₃) 6: 1.92 (lH, t, J=6.2 Hz), 1.98 (lH, t, J=6.4 Hz), 2.8 (2H, t, J=6.4 Hz), 3.76 (2H, t, J=6.2 Hz), 6.75 (lH, d, J=7.6 Hz), 6.86 (2H, d, J=8.6 Hz), 6.8-7.1 (2H, m), 7.20 (lH, d, J=7 Hz), 7.30 (2H, d, J=8.6 Hz), 7.72 (2H, d, J=8.6 Hz), 7.84 (2H, d, J=8.6 Hz), 10.13 (lH, s)

- 1 H-NMR (CDCl₃) δ : 1.9-2.1 (2H, m), 2.37 (6H,
 s), 2.84 (2H, t, J=6.6 Hz), 3.90 (2H, t, J=6.6 Hz),
 6.71 (1H, d, J=7.9 Hz), 6.8-7.2 (4H, m), 7.35 (2H,
 d, J=8.6 Hz), 7.44 (2H, s), 7.56 (2H, d, J=8.6 Hz),
 8.00 (1H, s)

(1H, s)

- 1 H-NMR (CDCl₃) δ: 0.8-1.3 (6H, m), 1.6-2.3 (9H, m), 2.83 (2H, t, J=6.6 Hz), 3.89 (2H, t, J=6.5 Hz), 6.72 (1H, d, J=7.9 Hz), 6.8-7.1 (2H, m), 7.15 (1H, d, J=7.4 Hz), 7.28 (2H, d, J=8.3 Hz), 7.44 (2H, d, J=8.4 Hz), 7.9-8.1 (1H, m)
- 1 H-NMR (CDCl₃) δ : 1.7-2.1 (17H, m), 2.83 (2H, t, J=6.7 Hz), 3.90 (2H, t, J=6.6 Hz), 6.68 (1H, d, J=8.1 Hz), 6.8-7.1 (2H, m), 7.14 (1H, d, J=7 Hz), 7.32 (2H, d, J=8.7 Hz), 7.39 (1H, s), 7.46 (2H, d, J=8.7 Hz)
- 9) ¹H-NMR (CDCl₃) δ: 1.99 (2H, quint, J=6.5 Hz), 2.82 (2H, t, J=6.6 Hz), 3.82 (2H, t, J=6.5 Hz), 6.8-7.1 (4H, m), 7.1-7.3 (2H, m), 7.4-7.6 (3H, m), 7.67 (1H, s), 7.8-8.0 (3H, m), 8.42 (1H, s)
- 11) ¹H-NMR (CDCl₃) δ : 1.98 (2H, quint, J=6.5 Hz),
 2.82 (2H, t, J=6.5 Hz), 3.81 (2H, t, J=6.5 Hz),

.1

- 3.84 (3H, s), 6.8-7.5 (10H, m), 7.68 (1H, s), 7.95 (1H, d, J=8.2 Hz), 8.52 (1H, s)

- 16) 1 H-NMR (DMSO- 1 d₆) 6 : 2.20 (3H, s), 2.27 (3H, s), 2.5-3.8 (6H, m), 4.3-5.3 (3H, m), 6.82 (1H, d, J=7.2 Hz), 7.1-7.4 (7H, m), 7.5-7.8 (3H, m), 10.43 (1H, s), 11.0-12.2 (1H, br)
- 18) $^{1}H-NMR (DMSO-d_{6}) \delta : 2.38 (3H, s), 2.5-3.8 (6H, m),$

4.3-5.3 (3H, m), 6.81 (1H, d, J=7.0 Hz), 7.1-7.5 (6H, m), 7.5-7.8 (5H, m), 10.35 (1H, s), 10.9-12.2 (1H, br)

- ¹H-NMR (DMSO-d₆) δ : 2.5-3.8 (6H, m), 4.3-5.2 (3H, m), 6.82 (1H, d, J=7.4 Hz), 7.2-7.3 (4H, m), 7.5-7.8 (5H, m), 7.75 (1H, d, J=1.8 Hz), 10.70 (1H, s), 10.8-12.2 (1H, br)
- 21) ¹H-NMR (DMSO-d₆) δ : 2.5-3.8 (9H, m), 4.3-4.7 (1H, m), 4.7-5.1 (2H, m), 6.8-7.1 (3H, m), 7.1-7.4 (2H, m), 7.5-7.7 (2H, m), 7.8-8.0 (3H, m), 9.79 (1H, s), 10.8-12.2 (1H, br)
- 1 H-NMR (DMSO-d₆) δ : 0.8-1.2 (3H, m), 1.7-2.2 (2H, m), 2.5-3.8 (5H, m), 4.3-5.2 (3H, m), 6.80 (1H, d, J=7.2 Hz), 7.1-7.3 (4H, m), 7.6-7.7 (3H, m), 7.85 (1H, s), 7.96 (2H, d, J=1.8 Hz), 10.62 (1H, s), 10.8-12.0 (1H, br)
- ¹H-NMR (DMSO-d₆) δ : 0.8-1.2 (3H, m), 1.7-2.1 (2H,

				·	
			·		
		c			
	·				
	·				

- m), 2.38 (3H, s), 2.6-3.8 (5H, m), 4.3-5.2 (3H, m), 6.81 (1H, d, J=7.0 Hz), 7.2-7.5 (6H, m), 7.6-7.8 (5H, m), 10.33 (1H, s), 10.5-11.7 (1H, br)
- ¹H-NMR (DMSO-d₆) δ : 0.8-1.2 (3H, m), 1.7-2.1 (2H, m), 2.6-3.8 (5H, m), 3.8-5.2 (3H, m), 6.82 (1H, d, J=7.2 Hz), 7.1-7.5 (8H, m), 7.5-7.7 (3H, m), 10.42 (1H, s), 10.7-12.0 (1H, br)
- ¹H-NMR (DMSO-d₆) δ : 0.8-2.0 (15H, m), 2.2-2.5 (1H, m), 2.6-3.7 (5H, m), 4.3-5.2 (3H, m), 6.76 (1H, d, J=7.0 Hz), 7.1-7.4 (4H, m), 7.46 (2H, d, J=8.6 Hz), 7.61 (1H, d, J=6.4 Hz), 10.03 (1H, s), 10.5-11.8 (1H, br)
- ¹H-NMR (DMSO-d₆) δ : 0.8-1.1 (3H, m), 1.7-2.0 (2H, m), 2.20 (3H, s), 2.29 (3H, s), 2.6-3.7 (5H, m), 4.3-5.2 (3H, m), 6.82 (1H, d, J=7.0 Hz), 7.2-7.4 (7H, m), 7.5-7.7 (3H, m), 10.41 (1H, s), 10.6-12.0 (1H, br)
- 29) ¹H-NMR (CDCl₃) δ : 2.29 (3H, s), 2.32 (3H, s), 2.34 (3H, s), 2.50-3.15 (11H, m), 3.79 (1H, dd, J=13.2 Hz, 7.3 Hz), 4.05 (1H, dd, J=13.2 Hz, 5.7 Hz), 6.62 (1H, d, J=7.7 Hz), 6.82-7.48 (8H, m), 7.53 (2H, d,

J=8.4 Hz), 8.05 (1H, brs) 1 H-NMR (CDCl₃) δ : 1.65-2.01 (4H, m), 2.31 (3H, 30) s), 2.35 (3H, s), 2.55-3.02 (6H, m), 3.09 (1H, dd, J=15 Hz, 5 Hz), 3.70 (1H, dd, J=12.5 Hz, 8.0 Hz), 4.22 (1H, dd, J=12.5 Hz, 5 Hz), 6.67 (1H, d, J=7.8 Hz), 6.80-7.32 (7H, m), 7.37 (2H, d, J=8.6 Hz), 7.53 (1H, d, J=8.3 Hz), 7.66 (1H, brs) 1 H-NMR (CDCl₃) δ : 2.80 (1H, dd, J=16.1 Hz, 5.3 31) Hz), 3.16 (1H, dd, J=15.8 Hz, 5.3 Hz), 3.75-4.50 (3H, m), 4.87-5.10 (3H, m), 6.80-7.60 (14H, m), 7.74 (2H, d, J=1.9 Hz), 8.47 (1H, brs) 1 H-NMR (CDCl₃) δ : 2.35 (6H, s), 2.72-3.10 (3H, 32) m), 3.65-3.78 (1H, m), 4.06-4.18 (1H, m), 6.60-7.62 (9H, m), 7.74 (2H, d, J=1.8 Hz), 8.52 (1H, brs) 1 H-NMR (CDCl₃) δ : 1.87 (3H, s), 2.68 (1H, dd, 33) J=5.6 Hz, 16 Hz), 3.14 (1H, dd, J=5.6 Hz, 16 Hz), 3.70-3.95 (2H, m), 4.32-4.50 (1H, m), 6.29 (1H, d, J=7.6 Hz, 6.90-7.80 (11H, m), 9.16 (1H, brs) 1 H-NMR (CDCl₃) δ : 1.62 (1H, brs), 1.90-2.25 (2H, 34) m), 2.55 (3H, s), 3.78 (1H, t, J=5.1 Hz), 3.95 (2H,

t, J=6.7 Hz), 6.69 (1H, t, J=7.9 Hz), 6.90-7.13
(2H, m), 7.23-7.40 (3H, m), 7.42-7.56 (3H, m), 7.77
(2H, d, J=1.9 Hz), 8.53 (1H, brs)

1H-NMR (CDCl₃) 6: 1.80-2.02 (1H, m), 2.20-2.35

- J=7.8 Hz), 6.81-7.10 (2H, m), 7.16-7.50 (6H, m), 7.80 (2H, d, J=1.8 Hz), 9.13 (1H, brs)
- 36) ¹H-NMR (CDCl₃) δ : 1.35-1.60 (1H, m), 1.65-2.20 (3H, m), 2.65-3.20 (5H, m), 3.81 (2H, d, J=6.5 Hz), 4.90-5.10 (1H, m), 6.60 (1H, d, J=8.0 Hz), 6.90 (1H, t, J=8.0 Hz), 7.00-7.50 (6H, m)
- 37) ¹H-NMR (CDCl₃) δ : 1.30-2.25 (4H, m), 2.55-3.20 (3H, m), 3.35 (2H, s), 3.80 (2H, s), 4.90-5.10 (1H, m), 6.62 (1H, d, J=8.0 Hz), 6.85-7.45 (12H, m), 9.27 (1H, brs)
- 38) 1 H-NMR (CDCl₃) $_{\delta}$: 1.35-2.25 (4H, m), 2.33 (3H, s), 2.60-3.20 (3H, m), 3.12 (2H, s), 3.61 (2H, s), 5.00 (1H, brs), 6.50-7.60 (13H, m), 9.14 (1H, brs)
- 39) ¹H-NMR (CDCl₃) δ: 1.27 (3H, t, J=7.1 Hz), 1.25-2.50 (12H, m), 2.70-3.10 (4H, m), 3.05 (2H, s), 4.15 (2H, q, J=7.0 Hz), 4.90-5.10 (1H, m), 6.63 (1H, d, J=7.5 Hz), 6.91 (1H, t, J=7.5 Hz), 7.00-7.50 (6H, m), 9.14 (1H, brs)
- 1H-NMR (CDCl₃) 6: 1.30-1.65 (1H, m), 1.80-2.25 (5H, m), 2.70-3.20 (3H, m), 4.01 (2H, d, J=5.0 Hz), 4.90-5.10 (1H, m), 6.61 (1H, d, J=7.7 Hz), 6.89 (1H, t, J=7.0 Hz), 7.00-7.45 (6H, m), 9.05 (1H, brs)
- 1H-NMR (CDCl₃) 6: 1.18 (6H, s), 1.30-2.20 (4H, m), 2.60-3.20 (3H, m), 3.30 (2H, s), 3.73 (2H, s), 4.90-5.10 (1H, m), 6.61 (1H, d, J=7.3 Hz), 6.70-

7.45 (12H, m), 9.50 (1H, brs)

- 1H-NMR (CDCl₃) δ : 1.19 (3H, t, J=7.0 Hz), 1.30-1.70 (1H, m), 1.75-2.20 (3H, m), 2.65-3.15 (3H, m), 3.46 (2H, q, J=7.0 Hz), 3.88 (2H, s), 4.90-5.10 (1H, m), 6.55-7.45 (13H, m), 8.36 (1H, brs)
- ¹H-NMR (CDCl₃) δ: 1.08 (3H, t, J=7.2 Hz), 1.05-2.25 (14H, m), 2.25-3.25 (10H, m), 4.90-5.10 (1H, m), 6.64 (1H, d, J=7.6 Hz), 6.90 (1H, t, J=7.2 Hz), 6.94-7.50 (6H, m), 11.50 (1H, brs)
- ¹H-NMR (CDCl₃) δ : 1.06 (3H, t, J=7.5 Hz), 1.30-2.20 (6H, m), 2.60-3.20 (3H, m), 3.65 (1H, m), 3.95 (1H, brs), 4.90-5.10 (1H, m), 6.50-6.75 (3H, m), 6.75-7.05 (2H, m), 7.05-7.55 (8H, m), 8.67 (1H, brs)
- ¹H-NMR (DMSO-d₆) δ : 1.28-1.57 (1H, m), 1.69-2.20 (3H, m), 2.59-3.15 (3H, m), 4.74-4.98 (1H, m), 6.62-6.80 (1H, m), 6.86-7.37 (5H, m), 7.50-7.70 (2H, m), 8.95-9.02 (1H, m), 9.03-9.15 (2H, m), 10.85 (1H, s)
- 1H-NMR (CDCl₃) δ : 1.40-1.66 (5H, m), 1.72-2.20 (7H, m), 2.63-3.18 (3H, m), 3.42 (2H, t, J=6.7 Hz), 4.00 (2H, t, J=6.3 Hz), 4.91-5.13 (1H, m), 6.58-6.72 (1H, m), 6.82-7.00 (3H, m), 7.02-7.30 (4H, m), 7.36-7.51 (2H, m), 7.70-7.88 (2H, m), 7.91 (1H, s)
- 47) 1 H-NMR (DMSO- d_{6}) δ : 2.05-2.95 (8H, m), 3.43-3.70 (1H, m), 4.08-4.30 (1H, m), 4.72-5.00 (1H, m),

WO 91/05549 - 353 - PCT/JP90/01340

6.70-8.08 (11H, m), 10.8 (1H, s), 11.1 (1H, brs)

¹H-NMR (DMSO-d₆) 6: 2.10-3.00 (8H, m), 3.47-3.70 (1H, m), 4.07-4.33 (1H, m), 4.75-4.98 (1H, m),

6.78-6.91 (1H, m), 7.05-7.22 (2H, m), 7.30-7.97 (9H, m), 10.75 (1H, s), 10.94 (1H, brs)

Example 377

To a solution of 1-[4-(4-formylbenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline (0.3 g) in methanol (10 ml) is added gradually sodium borohydride (59 mg) under ice-cooling and the mixture is stirred at room temperature for 2 hours. Water is added to the mixture and the solvent is distilled off under reduced pressure. The resulting residue is extracted with dichloromethane, washed with water, and dried over magnesium sulfate. The solvent is distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50 : 1), and recrystallized from methanol to give 1-[4-(4-hydroxymethylbenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (165 mg) as white powder, m.p. 224.5 -225.5°C.

Using the suitable starting materials, the compound of the above Example 37 is obtained in the same manner as in Example 377.

Example 378

To a solution of 1-[4-(4-methoxycarbonylbenzoyl-

amino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.5 g) in methanol (20 ml) is added 5 % aqueous sodium hydroxide solution (10 ml) and the mixture is stirred at room temperature overnight. Methanol is distilled off under reduced pressure and the resulting residue is acidified with diluted aqueous hydrochloric acid solution. The precipitated crystal is collected by filtration to give 1-[4-(4-carboxybenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.4 g) as white powder, m.p. >300°C.

¹H-NMR (DMSO-d₆) δ : 2.05 (2H, quint, J=6.4 Hz), 2.91 (2H, t, J=6.4 Hz), 3.86 (2H, t, J=6.4 Hz), 6.85 (1H, d, J=7.6 Hz), 6.9-7.2 (2H, m), 7.30 (1H, d, J=7.2 Hz), 7.44 (2H, d, J=8.5 Hz), 7.85 (2H, d, J=8.5 Hz), 8.1-8.2 (4H, m), 10.65 (1H, s), 13.2-13.4 (1H, br)

Using the suitable starting materials, the compounds of the above Examples 39, 241, 252, 253 and 362 are obtained in the same manner as in Example 378.

Example 379

To a solution of 1-[4-(3-acetyloxybenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline (1.5 g) in methanol (20 ml) is added 5 % aqueous sodium hydroxide solution (10 ml) and the mixture is stirred at room temperature overnight.

Methanol is distilled off under reduced pressure and the resulting residue is acidified with diluted aqueous hydrochloric acid solution. The precipitated crystal is collected by filtration and recrystallized from methanol to

give 1-[4-(3-hydroxybenzoylamino)benzoyl]-1,2,3,4tetrahydroquinoline (1.22 g) as white powder, m.p. 217 -218°C.

Using the suitable starting materials, the compounds of the above Examples 10, 343, 356, 364 and 365 are obtained in the same manner as in Example 379.

Example 380

To a solution of 1-[4-(3-hydroxybenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline (0.4 g) in acetone (5 ml) are added potassium carbonate (0.22 g) and ethyl iodide (0.34 g), and the mixture is refluxed for 5 hours. Then, acetone is distilled off under reduced pressure and water is added to the residue. The precipitated crystal is collected by filtration, and recrystallized from methanol to give 1-[4-(3-ethoxybenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.36 g) as white powder, m.p. 170.5 - 171.5°C.

Using the suitable starting materials, the compounds of the above Examples 11, 12, 13, 14, 33, 35, 48, 50 - 55, 90 - 92, 97 - 100, 109 - 111, 120 - 122, 136 - 138, 165 - 167, 175 - 177, 192 - 194, 211, 212, 214, 321, 322, 330 - 333, 335, 336, 339 - 342, 344 - 355, 357 - 366 and 370 - 374 are obtained in the same manner as in Example 380.

Example 381

Ethanol (50 ml) is added to 10 % Pd-C (0.1 g) and thereto is added-1-[4-(3-nitrobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.73 g). The mixture is subjected to

WO 91/05549 - 356 - PCT/JP90/01340

catalytic reduction at ordinary temperature under atmospheric pressure of hydrogen. After completion of the reduction, 10 % Pd-C is removed by filtration and the filtrate is concentrated under reduced pressure. The residue is extracted with dichloromethane and the extract is dried over magnesium sulfate. The solvent is distilled off under reduced pressure and recrystallized from methanol to give 1-[4-(3-aminobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.54 g) as white powder, m.p. 205.5 - 206.5°C.

Using the suitable starting materials, the compounds of the above Examples 24, 334 and 338 are obtained in the same manner as in Example 381.

Example 382

To a solution of 1-(4-aminobenzoyl)-1,2,3,4-tetra-hydroquinoline (0.5 g) in dichloromethane (20 ml) is added triethylamine (0.3 g), and thereto is added benzoyl chloride (0.28 g) under ice-cooling. The mixture is stirred at room temperature for 1 hour. To the reaction mixture is added water and extracted with dichloromethane. The extract is dried over magnesium sulfate and the solvent is distilled off under reduced pressure. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50:1) and recrystallized from methanol to give 1-[4-(benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (245 mg) as white powder, m.p. 202.5 - 203.5°C.

Using the suitable starting materials, the compounds of the above Examples 2 - 119, 131 - 373, 375 and 376 are obtained in the same manner as in Example 382.

Example 383

Thionyl chloride (10 ml) is added to 1-(4-carboxybenzoyl)-1,2,3,4-tetrahydroquinoline (0.5 g) and the mixture is refluxed for 1 hour. Thionyl chloride is distilled off under reduced pressure to give 4-[1-(1,2,3,4-tetrahydroquinolyl)carbonyl]benzoyl chloride. Separately, to a solution of m-anisidine (0.27 g) in dichloromethane (20 ml) is added triethylamine (0.34 g), and thereto is added gradually the above obtained 4-[1-(1,2,3,4-tetrahydroquinolyl)carbonyl]benzoyl_chloride_under_ice-cooling-and-themixture is stirred at room temperature for 1 hour. Water is added to the reaction mixture and the mixture is extracted with dichloromethane. The extract is dried over magnesium sulfate. The solvent is distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 50 : 1), and recrystallized from methanol to give 1-[4-(3methoxyanilinocarbonyl)benzoyl]-1,2,3,4-tetrahydroquinoline (203 mg) as colorless needles, m.p. 154 - 155°C.

Using the suitable starting materials, the compounds of the above Examples 120, 122 - 130 and 374 are obtained in the same manner as in Example 383.

Example 384

To 4-oxo-1-[4-(3,5-dichlorobenzoylamino)benzoyl]
1,2,3,4-tetrahydroquinoline (0.7 g) are added tetrahyrdofuran (10 ml) and methanol (10 ml). To the mixture is added
sodium borohydride (0.1 g) in portions and the mixture is
stirred at room temperature for 1 hour. Water is added to
the reaction mixture and the mixture is extracted with
dichloromethane. The solvent is concentrated and the
resulting residue is purified by silica gel column
chromatography (eluent; dichloromethane + dichloromethane :
methanol = 20 : 1), and recrystallized from ethanol to give
4-hydroxy-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4tetrahydroquinoline (0.4 g) as white powder, m.p. 215 217°C.

Example 385

To 3-ethoxycarbonyl-1-[4-(3,5-dichlorobenzoyl-amino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.6 g) are added an aqueous solution of sodium hydroxide (0.1 g) in water (1 ml) and ethanol (5 ml). The mixture is stirred at room temperature for 15 minutes, and acidified with diluted hydrochloric acid, extracted with dichloromethane. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane + dichloromethane : methanol = 50 : 1), and recrystallized from ethanol to give 3-carboxy-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.4 g) as white powder, m.p. 221 - 223°C.

To 3-carboxy-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (3.7 g) are added tetrahydrofuran (50 ml) and thionyl chloride (5 ml). The mixture is reacted at 60°C for 1 hour. The reaction mixture is concentrated and to the residue is added acetone (20 ml). To the mixture is added dropwise a solution of sodium azide (1.0 g) in water (5 ml) under ice-cooling. The reaction mixture is stirred at the same temperature for 30 minutes and extracted with dichloromethane, dried over magnesium sulfate. The solvent is concentrated and to the resulting residue are added anhydrous toluene (30 ml) and benzyl alcohol (1.7 g). The mixture is refluxed for 1 hour. The reaction mixture is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane + dichloromethane : methanol = 50 : 1) to give 3-benzyloxycarbonylamino-1-[4-(3,5-dichlorobenzoylamino)benzoyl)-1,2,3,4tetrahydroquinoline (3.7 g) as colorless amorphous. 1 H-NMR (CDCl₃) δ : 2.80 (1H, dd, J=16.1 Hz, 5.3

 1 H-NMR (CDCl₃) δ : 2.80 (1H, dd, J=16.1 Hz, 5.3 Hz), 3.16 (1H, dd, J=15.8 Hz, 5.3 Hz), 3.75-4.50 (3H, m), 4.87-5.10 (3H, m), 6.80-7.60 (14H, m), 7.74 (2H, d, J=1.9 Hz), 8.47 (1H, brs)

Example 387

benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (3.3 g)

are added acetic acid (40 ml) and 10 % Pd-C (0.4 g) and the reaction mixture is subjected to catalytic reduction at ordinary temperature under atmospheric pressure of hydrogen. One hour thereafter, the catalyst is removed by filtration and the filtrate is concentrated. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 20:1), and recrystallized from ethanol to give 3-amino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (1.6 g) as white powder, m.p. 207 - 210°C.

Example 388

To 3-amino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]1,2,3,4-tetrahydroquinoline (0.5 g) are added methanol (10 ml), 37 % formaline (0.8 ml) and sodium cyanoborohydride
(0.16 g). To the mixture is added acetic acid (0.5 ml)
under ice-cooling and the mixture is stirred at room
temperature for 1 hour. Water is added to the reaction
mixture and the mixture is basified with potassium carbonate
and extracted with dichloromethane. The solvent is
concentrated and the resulting residue is purified by silica
gel column chromatography (eluent; dichloromethane +
dichloromethane: methanol = 20 : 1) to give 3-dimethylamino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4tetrahydroquinoline (0.3 g) as colorless amorphous.

¹H-NMR (CDCl₃) & : 2.35 (6H, s), 2.72-3.10 (3H, m), 3.65-3.78 (1H, m), 4.06-4.18 (1H, m), 6.60-7.62 (9H, m),

7.74 (2H, d, J=1.8 Hz), 8.52 (1H, brs)

Using the suitable starting materials, the compounds of the above Examples 246, 247, 375 and 376 are obtained in the same manner as in Example 388.

Example 389

To 3-amino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]1,2,3,4-tetrahydroquinoline (0.44 g) are added dichloromethane (5 ml) and acetic anhydride (0.12 g) and the mixture
is stirred for 1 hour. The reaction mixture is concentrated
and the resulting residue is purified by silica gel column
chromatography (eluent; dichloromethane + dichloromethane :
methanol = 50 : 1) to give 3-acetylamino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.3 g) as
colorless amorphous.

¹H-NMR (CDCl₃) δ : 1.87 (3H, s), 2.68 (1H, dd, J=5.6 Hz, 16 Hz), 3.14 (1H, dd, J=5.6 Hz, 16 Hz), 3.70-3.95 (2H, m), 4.32-4.50 (1H, m), 6.29 (1H, d, J=7.6 Hz), 6.90-7.80 (11H, m), 9.16 (1H, brs)

Using the suitable starting materials, the compound of the above Example 242 is obtained in the same manner as in Example 389.

Example 390

To $4-\infty-1-[4-(3,5-\text{dichlorobenzoylamino})\text{benzoyl}]-1,2,3,4-tetrahydroquinoline (0.5 g) are added 40 % solution of methylamine in methanol (5 ml), molecular sieves 4A (1 g) and dimethylformamide (6 ml), and the mixture is refluxed$

for 4 hours. After cooling, the reaction mixture is filtered and to the filtrate is added sodium borohydride (80 mg), and the mixture is stirred at room temperature for 1 hour. The reaction mixture is concentrated and water is added to the resulting residue, and extracted with ethyl acetate. The solvent is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 20:1) to give 4-methylamino-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline (0.2 g) as colorless amorphous.

 1 H-NMR (CDCl₃) δ : 1.62 (1H, brs), 1.90-2.25 (2H, m), 2.55 (3H, s), 3.78 (1H, t, J=5.1 Hz), 3.95 (2H, t, J=6.7 Hz), 6.99 (1H, d, J=7.9 Hz), 6.90-7.13 (2H, m)

Using the suitable starting materials, the compounds of the above Examples 238, 239, 244, 247, 375 and 376 are obtained in the same manner as in Example 390.

Example 391

To 3-carboxy-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline (0.7 g) are added dimethylformamide (7 ml), diethyl cyanophosphate (0.3 ml) and dimethylamine hydrochloride (0.15 g). Further thereto is added triethylamine (0.8 ml) and the mixture is stirred at room temperature for 1 hour. Water is added to the reaction mixture and extracted with ethyl acetate. The solvent is concentrated and to the resulting residue is added diethyl ether. The precipitated crystal is collected by filtration

to give 3-dimethylamido-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline (0.5 g) as light yellow powder, m.p. 186 - 187°C.

Example 392

Using the suitable starting materials, the compound of the above Example 253 is obtained in the same manner as in Example 392.

Example 393

1-[4-(3-Carboxypropionylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine (0.5 g) is dissolved in dimethylformamide (1 ml) and thereto is added dropwise diethyl
cyanophosphate (0.25 g) under ice-cooling. The mixture is
stirred at room temperature for 30 minutes and then cooled
again with ice. Thereto are added dropwise a solution of
diethylamine (0.11 g) in dimethylformamide (1 ml) and
triethylamine (0.34 g). The mixture is stirred at room
temperature for 16 hours. The solvent is distilled off

under reduced pressure and water is added to the resulting residue. The mixture is extracted with dichloromethane. The organic layer is washed successively with diluted hydrochloric acid, water, saturated sodium hydrogen carbonate solution, water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate), and recrystallized from n-hexane/ethyl acetate to give 1-[4-(3-diethylaminocarbonylpropionylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.42 g) as colorless scales, m.p. 165 - 167°C.

Using the suitable starting materials, the compounds of the above Examples 255 - 263 are obtained in the same manner as in Example 393.

Example 394

To a solution of 1-[4-(2-chloroacetylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (2.06 g) in dimethylformamide (5 ml) are added sodium iodide (0.90 g), potassium carbonate (1.1 g) and cyclohexylamine (0.89 g), and the mixture is stirred at room temperature for 2 hours. Dimethylformamide is distilled off under reduced pressure and water is added to the resulting residue. The mixture is extracted with dichloromethane. The organic layer is washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is

distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate), and recrystallized from n-hexane/-ethyl acetate to give 1-[4-(2-cyclohexylaminoacetylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (2.03 g) as white powder, m.p. 139 - 142°C.

Using the suitable starting materials, the compounds of the above Examples 271 - 309 and 317 are obtained in the same manner as in Example 394.

Example 395

o-Cresol (0.36 g) is dissolved in dimethylsulfoxide (4 ml) containing sodium hydroxide powder (0.18 g) and thereto is added 1-[4-(2-chloroacetylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.03 g). The mixture is stirred at 90°C for 7.5 hours. The reaction mixture is poured into ice-water (300 ml) and the precipitated crystal is collected by filtration, washed with water, and purified by silica gel column chromatography (eluent; n-hexane: ethyl acetate = 2:1), and recrystallized from ethyl acetate to give 1-{4-[2-(2-methylphenoxy)acetylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (546 mg) as colorless scales, m.p. 172.5 - 175°C.

Using the suitable starting materials, the compounds of the above Examples 310 and 312 - 316 are obtained in the same manner as in Example 395.

Example 396

A mixture of $1-\{4-[2-(6-bromohexyloxy)benzoyl-$

amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (2.00 g), sodium acetate (0.36 g), sodium iodide (0.55 g) and acetic acid (20 ml) is refluxed for 1 day. The solvent is distilled off and the resulting residue is extracted with ethyl acetate. The organic layer is washed successively with 2N aqueous sodium hydroxide solution and saturated saline solution, and dried over magnesium sulfate. The solvent is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; chloroform: methanol = 500: 1), and recrystallized from ethanol to give 1-{4-[2-(6-acetyloxyhexyloxy)benzoylamino]-benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (1.07 g) as white powder, m.p. 145 - 146°C.

Using the suitable starting materials, the compound of the above Example 360 is obtained in the same manner as in Example 396.

Example 397

A mixture of 1-{4-[2-(6-bromohexyloxy)benzoyl-amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (0.70 g), diethylamine (0.16 ml), triethylamine (0.21 ml) and acetonitrile (20 ml) is refluxed overnight. The solvent is distilled off and the resulting residue is dissolved in chloroform, washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent;

chloroform: methanol = 200 : 1 + 50 : 1) and converted into the hydrochloride thereof in methanol. The product is recrystallized from methanol/diethyl ether to give 1-{4-[2-(6-diethylaminohexyloxy)benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine hydrochloride (0.42 g) as white powder, m.p. 91 - 95°C.

Using the suitable starting materials, the compounds of the above Examples 330, 332, 333, 335, 336, 339, 341, 342, 344 - 349, 352 - 355, 357 and 366 are obtained in the same manner as in Example 397.

Example 398

A mixture of 1-{4-[2-(6-bromohexyloxy)benzoyl-amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (4.00 g), potassium phthalimide (2.02 g) and dimethylformamide (100 ml) is stirred at 100°C for 5 hours. The reaction mixture is filtered and the filtrate is distilled off. The resulting residue is extracted with ethyl acetate and the organic layer is washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane), and recrystallized from methanol/diethyl ether to give 1-{4-[2-(6-phthalimidohexyloxy)benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (4.06 g) as white powder, m.p. 145 - 146.5°C.

Using the suitable starting materials, the

compounds of the above Examples 331, 340, 364 and 365 are obtained in the same manner as in Example 398.

Example 399

A mixture of 1-{4-[2-[6-phthalimidohexyloxy)-benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (3.75 g), hydrazine hydrate (0.44 ml) and ethanol (30 ml) is refluxed for 3.5 hours. The precipitated crystal is collected by filtration, dried and purified by silica gel column chromatography (eluent; chloroform: methanol: aqueous ammonia = 100: 10: 1), and recrystallized from methanol/diethyl ether to give 1-{4-[2-(6-aminohexyloxy)-benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (2.52 g) as white powder, m.p. 135.5 - 137.5°C.

Using the suitable starting materials, the compounds of the above Examples 284, 344 and 345 are obtained in the same manner as in Example 399.

Example 400

A mixture of 1-{4-[2-(6-aminohexyloxy)benzoyl-amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (0.70 g), acetic anhydride (20 ml) and two drops of conc. sulfuric acid is stirred at room temperature for 3 hours. To the reaction mixture is added aqueous 2N aqueous sodium hydroxide solution under ice-cooling and the mixture is extracted with chloroform. The organic layer is washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off

WO 91/05549 - 369 - PCT/JP90/01340

and the resulting residue is purified by silica gel column chromatography (eluent; chloroform: methanol = 200:1), and recrystallized from methanol/diethyl ether to give 1-{4-[2-(6-acetylaminohexyloxy)benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (0.60 g) as colorless needles, m.p. 171 - 172°C.

Example 401

A mixture of 1-{4-[2-(6-aminohexyloxy)benzoyl-amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (0.70 g), benzoyl chloride (0.20 ml), triethylamine and dichloromethane (20 ml) is stirred at room temperature for 1 hour. The reaction mixture is washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is concentrated and the resulting residue is recrystallized from ethanol to give 1-{4-[2-(6-benzoylaminohexyloxy)benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine (0.71 g) as white powder, m.p. 178 -178.5°C.

Using the suitable starting materials, the compounds of the above Examples 348 and 357 are obtained in the same manner as in Examples 400 and 401.

Example 402

A mixture of 1-[4-(2-ethoxycarbonylmethoxybenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.00 g), aqueous ammonia (100 ml), ammonium chloride (0.3 g) and methanol (150 ml) is heated at 100°C for 4 hours in a sealed

tube. The solvent is distilled off and the resulting residue is extracted with chloroform, washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; chloroform: methanol = 50:1), and recrystallized from methanol/diethyl ether to give 1-[4-(2-carbamoylmethoxybenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.43 g) as white powder, m.p. 198 - 199°C.

Example 403

A mixture of 1-[4-(2-chloro-4-aminobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.55 g), acetic anhydride (15 ml), acetic acid (5 ml) and a drop of sulfuric acid is stirred at room temperature for 1 hour. To the reaction mixture is added aqueous 2N aqueous sodium hydroxide solution and the mixture is extracted with chloroform. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is concentrated and the resulting residue is recrystallized from methanol/diethyl ether to give 1-[4-(2-chloro-4-acetylaminobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.28 g) as white powder, m.p. 214 - 243°C.

Using the suitable starting materials, the compound of the above Example 44 is obtained in the same manner as in Example 403.

Example 404

A mixture of 1-[4-(1-benzyloxycarbonyl-4piperidinylcarbonylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine (8.00 g), 10 % Pd-C (0.8 g) and ethanol (250 ml) is subjected to catalytic hydrogenation at 50°C under 4 atm. of hydrogen pressure for 6 hours. The catalyst is removed by filtration and the filtrate is evaporated under reduced pressure. The resulting residue is extracted with ethyl acetate and washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; chloroform : methanol : ammonium hydroxide = 50 : 10 : 1) to give 1-{4-[4-(4-piperidinyl)benzoylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine (4.80 g), and a part (0.5 g) thereof is converted into the hydrochloride thereof in methanol. The hydrochloride is recrystallized from methanol/diethyl ether to give $1-\{4-[4-(4-piperidinyl)$ benzoylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine hydrochloride (0.42 g) as white powder, m.p. 177 -181.5°C.

Example 405

٤

Using the suitable starting materials, the following compound is obtained in the same manner as in the above Examples 1, 382 and 388.

l-[4-(4-Dimethylaminobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline, colorless amorphous

 1 H-NMR (DMSO- d_{6}) & : 1.90-2.00 (2H, m), 2.82 (2H, t, J=6.5 Hz), 2.98 (6H, s), 3.77 (2H, t, J=6.5 Hz), 6.70-7.30 (6H, m), 7.32 (2H, d, J=8.6 Hz), 7.73 (2H, d, J=8.6 Hz), 8.00-8.20 (1H, m), 8.39 (1H, d, J=2.2 Hz), 10.37 (1H, s)

Using the suitable starting materials, the following compounds are obtained in the same manner as in Example 1.

Table 2

Example 406

Structure

re
$$N(CH_3)_2$$
 R^1
 N

R²: н

$$R^3: 4-NHC \xrightarrow{O}_{C1}^{C1}$$

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 216 - 218°C

Structure

$$\mathbb{R}^1$$
: $\mathbb{N}^{(CH_3)_2}$

в²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 181 - 183°C

Form: Free

Example 408

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 207 - 208°C

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 213 - 214°C

Form: Free

Example 410

Structure

$$\begin{array}{c}
\text{N(CH}_3)_2 \\
\\
\mathbb{R}^1 \\
\end{array}$$

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 136 - 138°C

Structure

are
$$\frac{N(CH_3)_2}{R^1}$$
:

 $R^2: 3-OCH_3$

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 130 - 132°C

Form: Free

Example 412

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}(CH_{3})_{2}$$

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 143 - 145°C

Structure

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 171 - 173°C

Form: Free

Example 414

Structure

re
$$N(CH_3)_2$$
 R^1 N

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 162 - 164°C

Structure

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 49)

Form: Free

Example 416

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$ $\mathbb{C}^{\mathbb{N}}$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 50)

Structure

$$\mathbb{R}^{1} \quad : \quad \mathbb{N}^{(CH_3)_2}$$

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 51)

Form: Free

Example 418

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}(CH_{3})_{2}$$

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 228.5 - 230°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1} \mathbb{N} \mathbb{N}

г²: н

Crystalline form: White powder

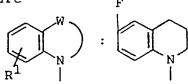
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 205.5 - 206.5°C

Form: Free

Example 420

Structure



 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 210 - 212°C

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 166 - 167°C

Form: Free

Example 422

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 191.5 - 192.5°C

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 209 - 210°C

Form: Free

Example 424

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 52)

Structure

२²: н

$$R^3$$
: 4-NHC CH_3

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 148 - 149°C

Form: Free

Example 426

Structure

re
$$CH_3$$
 CH_3

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 157 - 158°C

Structure

$$\begin{array}{c}
\text{CH} \\
\text{CH}_{3}
\end{array}$$

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 194.5 - 195.5°C

Form: Free

Example 1428

Structure

$$\begin{array}{c}
\text{CH} \\
\text{CH}_{3}
\end{array}$$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 179.5 - 180.5°C

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 190 - 191°C

Form: Free

Example 430

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 159 - 160°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 53)

Form: Hydrochloride

Example 432

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 155 - 156°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

 R^2 : H

Crystalline form:Colorless amorphous

NMR analysis: 54)

Form: Free

Example 434

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \end{array}\right)$$

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 55)

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \end{array}\right)$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 175 - 177°C

Form: Free

Example 436

Structure

$$\begin{array}{c}
\text{re} \\
 & \text{CH}_{1} \\
 & \text{R}^{1} \\
 & \text{I}
\end{array}$$

R²: H

Crystalline form: Colorless amorphous .

NMR analysis: 56)

Structure

$$\mathbb{R}^{1} \quad : \quad \mathbb{N}$$

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 57)

Form: Free

Example 438

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{0} \mathbb{R}^{0}

R². н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 219 - 220°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 215 - 218°C

Form: Free

Example 440

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 128.5 - 129.5°C

Structure

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 58)

Form: Free

Example 442

Structure

$$\begin{array}{c}
\text{re} \\
\text{R}^{1}
\end{array}$$
: $\begin{array}{c}
\text{CH}_{2} \\
\text{N}
\end{array}$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 153 - 154°C

Structure

в². н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 150 - 153°C

Form: Free

Example 444

Structure

ъ². н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 139 - 141°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 59)

Form: Free

Example 446

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array}\right)$$

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 60)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{N} : \mathbb{R}^{1}

R²: н

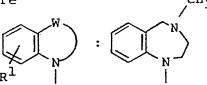
Crystalline form: Colorless amorphous

NMR analysis: 61)

Form: Free

Example 448

Structure



R²: H

Crystalline form: Colorless amorphous

NMR analysis: 62)

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: 3-OCH₃

$$R^3$$
: 4-NHCCH₂N C_2 H₅

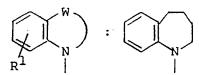
Crystalline form: Colorless amorphous ..

NMR analysis: 63)

Form: Free

Example 450

Structure



R²: H

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 172.5 - 173.5°C

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 122.5 - 123°C

Form: Free

Example 452

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 198 - 199.5°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 118 - 119.5°C

Form: Hydrochloride

Example 454

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 163 - 165°C

Form: Dihydrochloride

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 246 - 248°C

Form: Hydrochloride

Example 456

Structure

Crystalline form: White powder

Recrystallization solvent: Chloroform/ethanol

Melting Point: 204 - 205°C

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 127 - 128°C

Form: Hydrochloride

Example 458

Structure

$$\bigcap_{\mathbb{R}^1} \bigvee_{\mathbb{N}} : \bigcap_{\mathbb{R}^2 \colon \mathbb{H}}$$

$$R^3: 4-NHC \xrightarrow{O (CH_2)_3CONH_2}$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 220 - 221°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

$$R^3$$
: 4-NHC- $(CH_2)_2^{NH_2}$

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 190 - 192°C

Form: Free

Example 460

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 189 - 191°C

Form: Hydrochloride

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

$$R^3$$
: 4-NHC- $\left\langle \begin{array}{c} O & O(CH_2)_2OCOCH_3 \\ \\ \end{array} \right\rangle$

Crystalline form: Colorless needles

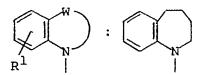
Recrystallization solvent: Ethanol

Melting Point: 173 - 174°C

Form: Free

Example 462

Structure



R²: н

$$R^3$$
: 4-NHC- $OCH_2CO_2C_2H_5$

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethanol

Melting Point: 129 - 130°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 130 - 133°C

Form: Hydrochloride

Example 464

Structure

Crystalline form: Light yellow powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 170.5 - 172°C

Form: Hydrochloride

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 126 - 131°C

Form: Hydrochloride

Example 466

Structure

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 182 - 185°C

Form: Dihydrochloride

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 116 - 121°C

Form: Hydrochloride

Example 468

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 178 - 182.5°C

Structure

$$\mathbb{R}^{1}$$
 :

R²: н

Crystalline form: Colorless particles

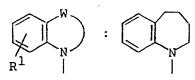
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 185 - 187°C

Form: Free

Example 470

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 215 - 217°C

Structure

R²: н

$$R^3$$
: 4-NHSO₂-CH₃-CH₃

Crystalline form: White powder

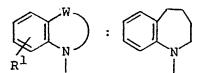
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 176 - 178°C

Form: Free

Example 472

Structure



R²: н

Crystalline form: Light yellow powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 194.5 - 197°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

к²: н

$$R^3$$
: 4-NHCO- N -COCH₂N C_2H_5

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 161.5 - 165.5°C

Form: Hydrochloride

Example 474

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 152 - 153°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 147 - 148°C

Form: Free

Example 476

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}

R²: н

Crystalline form: Light yellow powder

Recrystallization solvent: Ethyl acetate

Melting Point: 215 - 217°C

Structure

$$\mathbb{R}^1$$
 :

R²: н

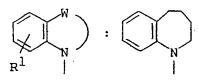
Crystalline form: Colorless amorphous

NMR analysis: 64)

Form: Free

Example 478

Structure



 R^2 : H

$$R^3$$
: 4-NHC-CHNH-CH₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 180 - 181°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 65)

Form: Free

Example 480

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 66)

Structure

$$\mathbb{R}^{1}$$
 :

R²: Н

Crystalline form: Colorless amorphous . . .

NMR analysis: 67)

Form: Free

Example 482

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

Crystalline form: Colorless scales

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 165 - 167°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 68)

Form: Free

Example 484

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 69)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 70)

Form: Free

Example 486

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 71)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

$$R^3: 4-NHCCH_2NH-CH_2NHCCH_3$$

Crystalline form: Colorless amorphous

NMR analysis: 72)

Form: Free

Example 488

Structure

Crystalline form: Colorless amorphous

NMR analysis: 73)

Structure

Crystalline form: Light yellow amorphous

NMR analysis: 74)

Form: Free

Example 490

Structure

$$\mathbb{R}^2$$
: H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 182 - 182.5°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 244 - 245°C

Form: Free

Example 492

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 220 - 221.5°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{R}^{NCH_{3}}$ \mathbb{R}^{2} :

Crystalline form: Light yellow amorphous

NMR analysis: 75)

Form: Free

Example 494

Structure

$$\mathbb{R}^{1} \longrightarrow \mathbb{R}^{NCH_{3}}$$

R²: 3-OCH₃

Crystalline form: Light yellow amorphous

NMR analysis: 76)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

$$R^3$$
: 4-NHC- $O(CH_2)_6$ NHCOCH₃

Crystalline form: Colorless needles

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 171 - 172°C

Form: Free

Example 496

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 178 - 178.5°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Example 498

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Example 499

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

$$R^3$$
: 4-NHC- CH_3

Example 501

Structure

$$\mathbb{R}^1$$
 :

R²: Н

$$R^3$$
: 4-NHC- N -CH₂CONH₂

Example 502

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : H

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Example 504

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R} \mathbb{R} \mathbb{R}

R²: н

R³: 4-NHCCH₂COOH

Crystalline form: Light yellow scales

Recrystallization solvent: Ethanol/water

Melting Point: 129 - 131°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 199 - 201°C

Form: Free

Example 506

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}^{(CH_{3})_{2}}$$

R²: H

Crystalline form: Colorless amorphous

NMR analysis: 77)

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{N}^{(CH_3)_2}$

R²: н

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 187.5 - 189°C

Form: Free

Example 508

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 161 - 164°C

Form: Free

4

Structure

$$\mathbb{R}^{1} \longrightarrow \mathbb{R}^{1}$$

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 242 - 243°C

Form: Free

Example 510

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N} \mathbb{R}^{N}

R²: 3-OCH₃

R³: 4-NHCOCH₂C1

Crystalline form: White powder

Recrystallization solvent: Dichloroethane/diethyl ether

Melting Point: 186 - 188°C

Form: Free

į.

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 78)

Form: Free

Example 512

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 79)

- 1H-NMR(CDCl₃) δ; 1.11 (3H, t, J=7.1 Hz), 1.90-2.25 (2H, m), 2.29 (3H, s), 2.55 (2H, q, J=7.1 Hz), 3.62-3.90 (2H, m), 4.00-4.20 (1H, m), 6.63 (1H, d, J=7.9 Hz), 6.85-7.10 (2H, m), 7.25-7.80 (9H, m), 8.25 (1H, brs)
- 1H-NMR(CDCl₃) δ; 1.10 (3H, t, J=7.1 Hz), 1.90-2.20 (2H, m), 2.28 (3H, s), 3.60-3.90 (2H, m), 3.95-4.20 (1H, m), 6.62 (1H, d, J=7.9 Hz), 6.80-7.10 (2H, m), 7.20 (2H, d, J=8.6 Hz), 7.31-7.55 (4H, m), 7.80 (2H, d, J=1.9 Hz), 9.05 (1H, brs)
- 1H-NMR(CDCl₃) δ; 1.80-2.05 (lH, m), 2.15-2.50 (lH, m), 2.34 (6H, s), 2.51 (3H, s), 3.48-3.62 (lH, m), 3.72 (3H, s), 3.70-3.85 (lH, m), 4.00-4.22 (lH, m), 6.64 (lH, d, J=7.8 Hz), 6.84-7.58 (9H, m), 8.16 (lH, brs), 8.40 (lH, d, J=8.7 Hz)
- ¹H-NMR(CDCl₃) δ; 1.16 (3H, t, J=7.1 Hz), 2.40-2.70 (2H, m), 2.90-3.30 (3H, m), 3.80-4.20 (2H, m), 4.80-5.00 (1H, m), 6.60-6.80 (1H, m), 7.00-7.70 (10H, m), 8.24 (1H, s)
- ¹H-NMR(DMSO-d₆) δ; 1.0-2.5 (10H, m), 2.34 (3H, s), 3.30-3.80 (4H, m), 4.50-5.30 (3H, m), 6.70-7.00 (1H, m), 7.10-7.80 (11H, m), 10.43 (1H, s), 10.5-12.0 (1H, br)
- ¹H-NMR(CDCl₃) δ; 1.10-2.10 (10H, m), 2.40-2.70 (1H, m), 2.80-3.20 (3H, m), 3.92 (2H, s), 4.90-5.20 (1H, m), 6.50-6.70 (1H, m), 6.80-7.60 (8H, m), 7.75

(2H, s), 8.73 (1H, s)

- 1H-NMR(CDCl₃) 6; 1.10-2.20 (10H, m), 2.40-2.70 (1H, m), 2.90-3.30 (3H, m), 3.93 (2H, s), 4.90-5.20 (1H, m), 6.62 (1H, d, J=7.6 Hz), 6.90-7.70 (10H, m), 8.29 (1H, s)
- ¹H-NMR(CDCl₃) δ; 1.50-2.10 (2H, m), 2.38 (6H, s), 2.30-2.70 (1H, m), 2.70-3.00 (2H, m), 3.45 (1H, d, J=13 Hz), 3.81 (1H, d, J=14 Hz), 4.70-5.00 (1H, m), 7.0-7.50 (12H, m), 8.23 (1H, s)
- ¹H-NMR(CDCl₃) ⁶; 1.50-2.10 (2H, m), 2.42 (3H, s), 2.40-2.70 (1H, m), 2.80-3.00 (2H, m), 3.52 (1H, d, J=13 Hz), 3.85 (1H, d, J=13 Hz), 4.70-5.00 (1H, m), 7.00-7.70 (12H, m), 8.54 (1H, s)
- 1H-NMR(CDCl₃) 6; 2.43 (3H, s), 2.47 (3H, s), 3.00-3.30 (3H, m), 3.76 (1H, d, J=14 Hz), 4.06 (1H, d, J=14 Hz), 4.90-5.20 (1H, m), 6.50-6.80 (3H, m), 6.90-7.50 (6H, m), 7.70-8.00 (2H, m), 8.48 (1H, d, J=8 Hz), 10.58 (1H, s)
- 60) $^{1}\text{H-NMR}(CDCl_{3})$ 6; 2.41 (3H, s), 2.80-3.20 (3H, m), 3.73 (1H, d, J=13 Hz), 4.03 (1H, d, J=14 Hz), 6.66 (2H, d, J=7.6 Hz), 6.90-8.00 (10H, m), 8.57 (1H, s)

- 1 1 H-NMR(CDCl₃) δ; 2.40 (3H, s), 2.90-3.20 (3H, m),
 3.73 (1H, d, J=13 Hz), 4.07 (1H, d, J=13 Hz), 4.705.00 (1H, m), 6.60-6.80 (2H, m), 6.90-8.00 (10H, m), 8.54 (1H, s)
- ¹H-NMR(CDCl₃) δ; 2.41 (3H, s), 2.90-3.20 (3H, m), 3.75 (1H, d, J=14 Hz), 4.08 (1H, d, J=14 Hz), 4.80-5.00 (1H, m), 6.67 (1H, d, J=7.6 Hz), 6.82 (1H, d, J=7.6 Hz), 6.90-7.90 (10H, m), 8.08 (1H, s)
- 63)

 H-NMR(CDCl₃) δ; 1.23 (3H, t, J=7 Hz), 1.40-1.70

 (1H, m), 1.90-2.20 (3H, m), 2.70-3.30 (3H, m),

 3.40-3.60 (5H, m), 3.91 (2H, s), 5.00-5.20 (1H, m),

 6.60-7.40 (11H, m), 8.12 (1H, d, J=8 Hz), 8.99 (1H, s)
- 1H-NMR(CDCl₃) δ; 1.35-1.70 (1H, m), 1.80-2.20 (3H, m), 2.25-2.35 (1H, m), 2.65-3.20 (3H, m), 4.01 (2H, s), 4.05-4.17 (2H, m), 4.90-5.10 (1H, m), 6.61 (1H, d, J=7.5 Hz), 6.75-7.50 (12H, m), 8.44 (1H, brs)
- 1H-NMR(CDCl₃) δ; 0.88 (3H, t, J=7.4 Hz), 1.16 (3H, t, J=7.0 Hz), 1.35-2.20 (6H, m), 2.27 (3H, s), 2.60-3.20 (3H, m), 3.20-3.45 (2H, m), 3.85-4.10 (1H, m), 4.90-5.10 (1H, m), 6.63 (1H, d, J=7.4 Hz),

- 6.77 (2H, d, J=8.5 Hz), 6.92 (1H, t, J=8.0 Hz), 7.00-7.45 (8H, m), 8.85 (1H, brs)
- 1H-NMR(CDCl₃) & ; 1.17 (3H, t, J=7.0 Hz), 1.35-1.65 (4H, m), 2.60-3.45 (5H, m), 4.20 (2H, q, J=7.0 Hz), 4.90-5.10 (1H, m), 6.63 (1H, d, J=7.6 Hz), 6.80-7.45 (12H, m), 8.66 (1H, brs)
- 1H-NMR(CDCl₃) δ; 0.96 (6H, d, J=6.6 Hz), 1.35-1.65 (1H, m), 1.80-2.25 (4H, m), 2.65-3.15 (3H, m), 3.19 (2H, d, J=7.3 Hz), 3.99 (2H, s), 4.90-5.10 (1H, m), 6.60 (1H, d, J=7.8 Hz), 6.75-7.05 (4H, m), 7.05-7.40 (8H, m), 8.15 (1H, brs)
- 1H-NMR(CDCl₃) 6; 1.19 (3H, t, J=7.0 Hz), 1.35-1.65 (1H, m), 1.80-2.25 (3H, m), 2.70-3.20 (3H, m), 3.44 (2H, q, J=7.0 Hz), 3.77 (3H, s), 3.87 (2H, s), 4.90-5.10 (1H, m), 6.25-6.50 (3H, m), 6.67 (1H, d, J=7.5 Hz), 6.85-7.45 (8H, m), 8.29 (1H, brs)
- Th-NMR(CDCl₃) δ; 1.05 (3H, t, J=7.1 Hz), 1.35-1.65 (1H, m), 1.85-2.25 (3H, m), 2.65-3.30 (5H, m), 3.74 (2H, s), 4.95-5.15 (1H, m), 6.63 (1H, d, J=7.5 Hz), 6.80-7.55 (11H, m), 9.51 (1H, brs)
- Th-NMR(CDCl₃) δ; 1.30-1.65 (lH, m), 1.80-2.30 (3H, m), 2.65-3.15 (3H, m), 3.75 (2H, s), 3.74 (2H, s), 4.95-5.10 (lH, m), 6.45-6.70 (3H, m), 6.88 (lH, t, J=6.8 Hz), 7.00-7.45 (8H, m), 8.74 (lH, brs)
- 72) ¹H-NMR(CDCl₃) δ; 1.30-1.70 (1H, m), 1.75-2.25 (6H, m), 2.65-3.15 (3H, m), 3.78 (2H, d, J=5.4 Hz), 4.28

- (2H, d, J=5.5 Hz), 4.53 (1H, brs), 4.90-5.10 (1H, m), 5.89 (1H, brs), 6.50-6.70 (3H, m), 6.89 (1H, t, J=7.5 Hz), 7.00-7.40 (8H, m), 8.61 (1H, brs)
- TH-NMR(CDCl₃) δ; 1.35-1.65 (1H, m), 1.70-2.20 (8H, m), 2.65-3.20 (3H, m), 3.25-3.55 (4H, m), 3.88 (2H, s), 4.90-5.10 (1H, m), 5.79 (1H, brs), 6.55-7.40 (13H, m), 8.37 (1H, brs)
- TH-NMR(CDCl₃) δ; 1.35-2.00 (8H, m), 2.65-3.20 (3H, m), 3.30-3.35 (2H, m), 3.60-3.85 (2H, m), 3.90 (2H, s), 4.95-5.15 (1H, m), 6.55-7.00 (5H, m), 7.00-7.40 (8H, m), 7.65-7.90 (4H, m), 8.22 (1H, brs)
- Th-NMR(CDCl₃) δ; 1.16 (3H, t, J=7.0 Hz), 2.39 (3H, s), 2.80-3.20 (3H, m), 3.44 (2H, q, J=7.0 Hz), 3.65-4.20 (4H, m), 4.80-5.05 (1H, m), 6.50-7.45 (13H, m), 8.50 (1H, brs)
- TH-NMR(CDCl₃) δ; 1.23 (3H, t, J=7.0 Hz), 2.41 (3H, s), 2.75-3.20 (3H, m), 3.40-3.60 (5H, m), 3.65-3.90 (1H, m), 3.92 (2H, s), 3.90-4.20 (1H, m), 4.85-5.10 (1H, m), 6.65-7.45 (11H, m), 8.13 (1H, d, J=8.4 Hz), 9.01 (1H, brs)
- 1 H-NMR(CDCl₃) δ; 1.80-1.95 (1H, m), 2.20-2.70
 (10H, m), 3.50-3.60 (1H, m), 3.63-3.80 (1H, m),
 4.00-4.15 (1H, m), 6.60 (1H, d, J=7.6 Hz), 6.92
 (1H, t, J=7.6 Hz), 7.02 (1H, t, J=6.3 Hz), 7.20-7.65 (9H, m), 7.87 (1H, brs)
- 78) $^{1}\text{H-NMR}(CDCl_{3})$ δ ; 1.40-1.62 (1H, m), 1.84-2.22 (3H,

- m), 2.65-3.19 (3H, m), 3.97 (2H, t, J=4.9 Hz), 4.43 (2H, t, J=4.9 Hz), 4.95-5.18 (1H, m), 6.60-6.77 (1H, m), 6.85-7.02 (2H, m), 7.02-7.30 (5H, m), 7.40-7.68 (3H, m), 8.20-8.32 (1H, m), 9.62-9.81 (1H, m)
- The NMR (CDCl₃) δ; 1.38-1.65 (lH, m), 1.84-2.21 (3H, m), 2.64-3.15 (3H, m), 3.81 (2H, t, J=5.7 Hz), 4.25 (2H, t, J=5.7 Hz), 4.90-5.13 (lH, m), 6.58-6.71 (lH, m), 6.82-7.00 (lH, m), 7.00-7.52 (lOH, m), 8.11 (lH, brs)

To a solution of l-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-lH-benzazepine (1.06 g) in dichloromethane (80 ml) is added o-methylphenyl isocyanate (0.66 g) under ice-cooling. The mixture is stirred at room temperature for 4 hours. After completion of the reaction, the solvent is concentrated under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; n-hexane: ethyl acetate = 1:1), and recrystallized from ethyl acetate to give l-[4-(2-methyl-anilinocarbonylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.97 g) as white powder, m.p. 182 - 182.5°C.

Using the suitable starting materials, the compounds of the above Examples 491 - 492 are obtained in the same manner as in Example 513.

Example 514

A mixture of 1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (0.50 g), phenylsulfonyl chloride (0.29 ml), triethylamine (0.32 ml) and dichloromethane (30 ml) is stirred at room temperature overnight. The reaction mixture is washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; chloroform), and recrystallized from methanol/diethyl ether to give 1-(4-phenylsulfonylaminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (0.27 g) as colorless prisms, m.p. 178 - 182.5°C.

Using the suitable starting materials, the compounds of the above Examples 469 - 471, 498, 502 and 503 are obtained in the same manner as in Example 514.

Example 515

To a solution of 1-[4-(4-piperidinylcarbonylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.70 g) in dimethylformamide (20 ml) is added 60 % sodium hydride dispersion in mineral oil (82 mg) and the mixture is stirred at room temperature for 30 minutes. Thereto is added methyl iodide (0.14 ml) and the mixture is stirred ar room temperature overnight. The solvent is distilled off and the resulting residue is extracted with chloroform, and washed successively with water and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off

and the resulting residue is purified by silica gel column chromatography (eluent; chloroform : methanol = 10 : 1), and recrystallized from methanol/n-hexane to give $1-\{4-[N-(1-methyl-4-piperidinylcarbonyl)-N-methylamino]benzoyl\}-2,3,4,5-tetrahydro-lH-benzazepine (0.03 g) as light yellow powder, m.p. <math>194.5 - 197^{\circ}C$.

Using the suitable starting materials, the compounds of the above Examples 497 and 501 are obtained in the same manner as in Example 515.

Example 516

6-Fluoro-1-(4-aminobenzoyl)-1,2,3,4-tetrahydroquinoline (0.15 g) is dissolved in dichloromethane (10 ml) and thereto is added triethylamine (0.31 ml). To the mixture is added dropwise a solution of 3,5-dichlorobenzoyl chloride (0.14 g) in dichloromethane (2.0 ml) under icecooling, and the mixture is stirred for 30 minutes under ice-cooling, and further, at room temperature for 1 hour. To the mixture are added triethylamine (0.31 ml) and 3,5dichlorobenzoyl chloride (0.14 ml). The mixture is stirred at room temperature for 4 hours. The reaction mixture is washed with water, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate : n-hexane = 1 : 5 + 1 : 4), and recrystallized from ethyl acetate/n-hexane to give 6-fluoro-1-[4-(3,5dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline

WO 91/05549 PCT/JP90/01340

(0.12 g) and 6-fluoro-1-{4-[bis-(3,5-dichlorobenzoyl)amino]-benzoyl}-1,2,3,4-tetrahydroquinoline.

The former: White powder, m.p. 205.5 - 206.5°C

The latter: White powder, m.p. 210.5 - 212°C

Using the suitable starting materials, the compounds of the above Examples 450 and 504 are obtained in

Example 518

the same manner as in Example 378.

Example 517

Using the suitable starting materials, the compounds of the above Examples 450 - 467, 495, 496, 499, 500, 511 and 512 are obtained in the same manner as in Example 380.

Example 519

Using the suitable starting materials, the compounds of the above Examples 449, 474 - 489, 493 and 494 are obtained in the same manner as in Example 394.

Example 520

Using the suitable starting materials, the compounds of the above Examples 453, 455, 457, 459, 460, 463 - 467, 495, 496 and 499 are obtained in the same manner as in Example 397.

Example 521

Using the suitable starting materials, the compound of the above Example 461 is obtained in the same manner as

in Example 396.

Example 522

Using the suitable starting materials, the compound of the above Example 456 is obtained in the same manner as in Example 398.

Example 523

Using the suitable starting materials, the compound of the above Example 459 is obtained in the same manner as in Example 399.

Example 524

Using the suitable starting materials, the compounds of the above Examples 495 and 496 are obtained in the same manner as in Examples 400 and 401.

Example 525

Using the suitable starting materials, the compound of the above Example 458 is obtained in the same manner as in Example 402.

Using the suitable starting materials, the compounds of the following Table 3 are obtained in the same manner as in Examples 1 and 382.

Table 3

Example 527

Structure

$$\mathbb{R}^{1} \qquad \mathbb{CH}_{3}$$

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 225 - 226°C

Structure

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 142.5 - 145°C

Form: Free

Example 529

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 213 - 215°C

Structure

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 167 - 167.5°C

Form: Free

Example 531

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 3-СН_З

Crystalline form: Colorless scales

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 217 - 221°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} \mathbb{R}^{N} \mathbb{R}^{N}

R²: 3-СН₃

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 182 - 184°C

Form: Free

Example 533

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right)$$

 R^2 : 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 209 - 210°C

Structure

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 148 - 149°C

Form: Free

Example 535

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 202 - 203°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 218 - 219°C

Form: Free

Example 537

Structure

$$\mathbb{R}^1$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 159 - 160°C

Structure

$$\mathbb{R}^1$$
: $\mathbb{N}(CH_3)_2$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 201 - 202°C

Form: Free

Example 539

Structure

$$\mathbb{R}^{1} \qquad \mathbb{R}^{1}$$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 205 - 207°C

Structure

re
$$N(CH_3)_2$$
 R^1 :

R²: 3-он

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 201.5 - 202.5°C

Form: Free

Example 541

Structure

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 226 - 228°C

Structure

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 218 - 221°C

Form: Free

Example 543

Structure

are
$$N(CH_3)_2$$
 R^1

R²: 2-осн₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 156 - 157°C

Structure

ire
$$N(CH_3)_2$$
 R^1 N

R²: 2-OCH₃

Crystalline form: White powder

NMR analysis: 80)

Form: Free

Example 545

Structure

R²:--3-осн₃

Crystalline form: Colorless amorphous .

NMR analysis: 81)

Structure

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 82)

Form: Free

Example 547

Structure

R²: н

Crystalline form: Light yellow amorphous

NMR analysis: 83)

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 84)

Form: Free

Example 549

Structure

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 85)

Structure

R²: 3-OCH₂CH₃

Crystalline form: White powder

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 135 - 136°C

Form: Free

Example 551

Structure

$$\mathbb{R}^{1} \quad \mathbb{N}^{(CH_{3})_{2}}$$

R²: 3-OCH₂CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 122 - 123°C

Structure

$$\mathbb{R}^1$$
: $\mathbb{N}(CH_3)_2$

R²: 3-OCH₂CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 118 - 119°C

Form: Free

Example 553

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}^{(CH_3)_2}$$

R²: 3-OCH₂-

Crystalline form: Colorless prisms

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 145 - 147°C

Structure

Crystalline form: Light yellow needles

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 169.5 - 170.5°C

Form: Free

Example 555

Structure

$$\begin{array}{c}
\text{N(CH}_3)_2 \\
\\
\mathbb{R}^1 \\
\end{array}$$

R²: 3-OH

Crystalline form: Colorless prisms

Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 194 - 195°C

Structure

к²: 3-он

Crystalline form: Colorless needles

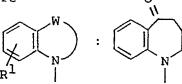
Recrystallization solvent: n-Hexane/ethyl acetate

Melting Point: 202 - 204°C

Form: Free

Example 557

Structure



R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 242 - 243°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: Light yellow powder

NMR analysis: 86)

Form: Free

Example 559

Structure

R⁻²: н

Crystalline form: Light yellow powder

NMR analysis: 87)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 237 - 238°C

Form: Free

Example 561 ---

Structure

re $\frac{NHCH_3}{R^1}$:

 R^2 : H

Crystalline form: Colorless prisms

Recrystallization solvent: Dioxane

Melting Point: 258 - 259°C

Structure

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 182.5 - 183.5°C

Form: Free

Example 563

Structure

re
$$_{\text{NHCH}_3}$$

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 209 - 211°C

Structure

re
$$\frac{NHCH_3}{R^1}$$
 :

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Dioxane

Melting Point: 210 - 211°C

Form: Free

Example 565

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 176 - 178°C

Structure

Crystalline form: Light yellow amorphous

NMR analysis: 88)

Form: Free

Example 567

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Dioxane/water

Melting Point: 272 - 273°C

---Form: Free

Structure

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Dioxane

Melting Point: 253 - 254°C

Form: Free

Example 569

Structure

R²: H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 248.5 - 249.5°C

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 266.5 - 267.5°C

Form: Free

Example 571

Structure

$$\begin{array}{c}
\text{re} \\
\mathbb{R}^{1} \\
\mathbb{R}^{1}
\end{array}$$
: $\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{R}^{1}
\end{array}$

 R^2 : H

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 252 - 253°C

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1

R²: н

Crystalline form: Light yellow powder

NMR analysis: 89)

Form: Free

Example 573

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Light brown powder

NMR analysis: 90)

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}^{12}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether

Melting Point: 198.5 - 199.5°C

Form: Free

Example 575

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N} \mathbb{N} \mathbb{N} \mathbb{N}

к²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 297 - 299°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 2-C1

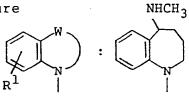
Crystalline form: Colorless amorphous

NMR analysis: 91)

Form: Free

Example 577

Structure



R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 202 - 203°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 92)

Form: Free

Example 579

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 232 - 233°C

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1

R²: н

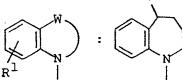
Crystalline form: Colorless amorphous

NMR analysis: 93)

Form: Free

Example 581

Structure



 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 256.5 - 257°C

Form: Free

2

Structure

re
$$OCOCH_3$$
 R^1 :

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 193 - 194°C

Form: Free

Example 583

Structure

re
$$\mathbb{R}^{1}$$
 : \mathbb{R}^{1}

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 227 - 230°C

Structure

are $OCOCH_3$ \mathbb{R}^1 :

R²: н

O C1 R3: 4-NHC-C1

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 199.5 - 202°C

Form: Free

Example 585

Structure

R²: н

R³: 4-NHC

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 219 - 220°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{0CH_3}

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 190 - 191.5°C

Form: Free

Example 587

Structure

$$\begin{array}{c}
\text{N} \\
\text{R}^1
\end{array}$$
:

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 184 - 185°C

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 223 - 224°C

Form: Free

Example 589

Structure

_D2. _U

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 178 - 181°C

Structure

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 168 - 168.5°C

Form: Free

Example 591

Structure .

OCH2CO2CH2CH3

Crystalline form: Colorless amorphous

NMR analysis: 94)

Structure

R²: F

Crystalline form: Colorless amorphous

NMR analysis: 95)

Form: Free

Example 593

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 196 - 197°C

Structure

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 96)

Form: Free

Example 595

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

Melting Point: 188 - 189°C

Structure

осн₂соон

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 97)

Form: Free

Example 597

Structure

OCH₂COOH

$$\mathbb{R}^{1}$$

в²: н

Crystalline form: Colorless amorphous

NMR analysis: 98)

Structure

O CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 203 - 204°C

Form: Free

Example 599

Structure

ore
$$(CH_2CON(CH_3)_2)$$
 $R^2: H$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 196 - 197°C

Structure

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 99)

Form: Free

Example 601

Structure

$$O(CH2)4N(CH3)2$$

$$\left(\begin{array}{c} W \\ N \end{array}\right)$$
:

Crystalline form: Colorless amorphous

NMR analysis: 100)

Structure

Crystalline form: Colorless amorphous

NMR analysis: 101)

Form: Free

Example 603

Structure

Crystalline form: Colorless amorphous .

NMR analysis: 102)

Structure

Crystalline form: Colorless amorphous

NMR analysis: 103)

Form: Free

Example 605

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 104)

Structure

O(CH₂)₃NHCOCH₃

$$\mathbb{Q}^{\mathbb{N}}$$
 : \mathbb{Q}

 R^2 : H

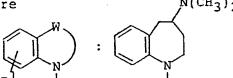
Crystalline form: Colorless amorphous

NMR analysis: 105)

Form: Free

Example 607

Structure



 R^2 : H

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 169 - 171°C

Structure

$$\bigcap_{\mathbb{R}^1} \bigcap_{\mathbb{N}} \mathbb{N} : \bigcap_{\mathbb{N}} \mathbb{N}$$

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 178 - 181°C

Form: Free

Example 609

Structure

$$\begin{array}{c} \text{CH}_2\text{N}(\text{CH}_3)_2 \\ \\ \text{R}^1 \end{array}$$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 187 - 188°C

Structure

are
$$CH_2N(CH_3)_2$$
 R^1 :

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 181 - 183°C

Form: Free

Example 611

Structure

ure
$$(CH_2N(CH_3)_2)$$

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 124 - 127°C

Structure

re
$$CH_2N(CH_3)_2$$

,2. н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 179 - 181°C

Form: Free

Example 613

Structure

re
$$N(CH_3)_2$$
 R^1 N

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 148 - 150°C

Structure ·

re
$$N(CH_3)_2$$
 R^1

$$R^3: 4-CNH$$

Crystalline form: Colorless amorphous

NMR analysis: 106)

Form: Free

Example 615

Structure"

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 219 - 220°C

Free Form:

Structure

are
$$N(CH_3)_2$$

$$R^1 \qquad \qquad N$$

R²: Н

$$R^3: 4-NHC-$$

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 226 - 228°C

Form: Free

Example 617

Structure

re
$$N(CH_3)_2$$

 $R^2: 3-OCH_3$

Crystalline form: Colorless amorphous

NMR analysis: 107)

Structure

R²: 3-осн₃

Crystalline form: Colorless amorphous

NMR analysis: 108)

Form: Free

Example 619

Structure

 R^2 : 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 109)

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{R}^{N(CH_3)_2}$ \mathbb{R}^2 : \mathbb{R}^2

Crystalline form: Colorless amorphous .

NMR analysis: 110)

Form: Free

Example 621

Structure

$$\mathbb{R}^{1} \quad | \quad \mathbb{R}^{2} : \mathbb{R}^{2}$$

Crystalline form: Colorless amorphous

NMR analysis: 111)

Structure

$$(\mathbb{R}^{1}) : \mathbb{R}^{N(CH_{3})_{2}}$$

$$\mathbb{R}^{2} : \mathbb{R}^{2}$$

Crystalline form: Colorless amorphous

NMR analysis: 112)

Form: Free

Example 623

Structure

$$\mathbb{R}^{1} \quad : \quad \mathbb{R}^{N(CH_{3})_{2}} \\ \mathbb{R}^{2} : \quad 3-OCH_{3}$$

Crystalline form: Colorless amorphous

NMR analysis: 113)

Structure

Crystalline form: Colorless amorphous

NMR analysis: 114)

Form: Free

Example 625

Structure

Crystalline form: Colorless amorphous

NMR analysis: 115)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: H

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 183 - 184°C

Form: Free

Example 627

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 219 - 220°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 240 - 241°C

Form: Free

Example 629

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 205 - 206°C

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 238 - 239°C

Form: Free

Example 631

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right)$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 233 - 234°C

Structure

$$\bigcap_{\mathbb{R}^1} \bigvee_{\mathbb{N}} : \bigcap_{\mathbb{N}} \bigvee_{\mathbb{N}}$$

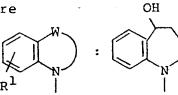
Crystalline form: Colorless amorphous

NMR analysis: 116)

Form: Free

Example 633

Structure



 $R_{..}^{2}$: 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 259.5 - 260.5°C

- 80) $^{1}H-NMR(CDCl_{3})$ & ; 1.24-5.26 (18H, m), 6.39-7.59 (13H, m)
- 1_{H-NMR}(CDCl₃) & ; 1.70-2.10 (m, 2H), 2.15-2.60 (m, 12H), 3.56 (t, J=5.8 Hz, 1H), 3.65-3.95 (m, 4H) 4.05-4.25 (m, 1H), 6.64 (d, J=7.7 Hz, 1H), 6.85-7.50 (m, 9H), 8.11 (brs, 1H), 8.42 (d, J=8.8 Hz, 1H)
- ¹H-NMR(CDCl₃) δ; 2.00-2.90 (m, 3H), 2.49 (s, 3H), 3.70-3.90 (m, 1H), 4.00-4.20 (m, 1H), 4.80-5.00 (m, 1H), 6.89 (d, J=6.3 Hz, 1H), 6.95-7.65 (m, 11H), 7.70 (brs, 1H)
- ¹H-NMR(CDCl₃) δ; 1.95-2.90 (m, 2H), 2.48 (s, 3H), 2.55 (s, 3H), 3.77 (t, J=5.1 Hz, 1H), 3.92 (t, J=6.7 Hz, 2H), 6.72 (d, J=8.0 Hz, 1H), 6.90-7.15 (m, 2H), 7.15-7.70 (m, 9H), 7.81 (brs, 1H)
- ¹H-NMR(CDCl₃) δ; 2.11 (s, 3H), 2.20-2.40 (m, 2H), 2.50 (s, 3H), 3.80-4.10 (m, 1H), 4.12-4.25 (m, 1H), 6.03 (t, J=4.3 Hz, 1H), 6.80-7.65 (m, 12H), 7.80 (brs, 1H)
- 1H-NMR(CDCl₃) & ; 1.80-2.40 (m, 5H), 2.45 (s, 3H), 2.81 (s, 3H), 3.55-3.82 (m, 1H), 4.15-4.40 (m, 1H), 5.90-6.10 (m, 1H), 6.80-7.80 (m, 12H), 8.67 (brs, 1H)
- ¹H-NMR(CDCl₃) δ; 1.95-2.35 (2H, m), 2.75-3.0 (2H, m), 3.0-5.4 (2H, m), 6.55-7.95 (11H, m), 8.09 (1H, s)

- ¹H-NMR(DMSO-d₆) 6; 1.85-2.2 (2H, m), 2.7-2.95 (2H, m), 3.5-5.0 (2H, m), 6.8-7.8 (12H, m), 10.60 (1H, s)
- ¹H-NMR(CDCl₃) δ; 0.8-1.1 (3H, m), 1.2-2.35 (6H, m), 2.35-5.25 (6H, m), 6.63 (1H, d, J=7.7 Hz), 6.8-7.6 (9H, m), 7.67 (1H, d, J=8.2 Hz), 7.9-8.15 (1H, m)
- 89) ¹H-NMR(CDCl₃) δ ; 1.7-2.9 (7H, m), 4.5-6.5 (3H, m), 6.55-6.75 (1H, m), 6.85-7.6 (12H, m)
- 90) ¹H-NMR(CDCl₃) δ; 1.65-3.1 (7H, m), 4.7-6.6 (3H, m), 6.6-6.8 (1H, m), 6.85-7.65 (12H, m)
- 91) ¹H-NMR(CDCl₃) δ; 1.8-2.4 (2H, m), 2.86 (2H, t, J=6 Hz), 3.1-5.15 (2H, m), 6.85-7.5 (8H, m), 7.5-7.85 (3H, m), 8.19 (1H, s)
- 93) ¹H-NMR(CDCl₃) δ; 1.45-1.91 (2H, m), 1.91-2.65 (2H, m), 2.65-2.90 (1H, m), 4.63-5.22 (2H, m), 6.63 (1H, d, J=7.4 Hz), 7.34-8.03 (11H, m), 10.16-10.44 (1H, m)
- 1 H-NMR(CDCl₃) δ; 1.08-1.47 (3H, m), 1.50-1.97 (2H, m), 1.97-2.48 (2H, m), 2.65-3.02 (1H, m), 4.00-4.43 (4H, m), 4.52-5.15 (2H, m), 6.50-6.79 (1H, m), 6.90-7.70 (10H, m), 8.26-8.60 (1H, m)

- 95) ¹H-NMR(CDCl₃) δ; 1.56-2.67 (4H, m), 2.46 (3H, s), 2.67-3.03 (1H, m), 3.82-4.32 (2H, m), 4.45-5.15 (2H, m), 5.43-5.83 (1H, m), 6.20-6.45 (1H, m), 6.50-6.86 (2H, m), 6.86-7.70 (10H, m), 7.76-8.10 (1H, m)

- 98) ¹H-NMR(CDCl₃) δ; 1.52-1.89 (2H, m), 1.89-2.56 (2H, m), 2.65-3.02 (1H, m), 3.90-4.40 (2H, m), 4.40-5.07 (2H, m), 6.58-6.78 (1H, m), 6.90-7.70 (10H, m), 8.57-8.81 (1H, brs)
- 99) ¹H-NMR(CDCl₃) δ; 1.49-1.89 (2H, m), 1.89-2.60 (2H, m), 2.63-3.23 (7H, m), 4.04-4.49 (2H, m), 4.52-5.21 (2H, m), 6.52-6.80 (1H, m), 6.89-7.84 (10H, m), 8.08-8.52 (1H, m)
- 100) ¹H-NMR(CDCl₃) δ; 1.41-1.86 (6H, m), 1.86-2.53 (4H, m), 2.25 (3H, s), 2.29 (3H, s), 2.43 (3H, s), 2.60-2.97 (1H, m), 3.36-3.77 (2H, m), 4.40-5.10 (2H, m), 6.54-6.72 (1H, m), 6.88-7.67 (11H, m), 8.27-8.58

- 108) $^{1}H-NMR(CDCl_{3})$ & ; 1.2-3.0 (10H, m), 3.0-5.2 (6H, m), 6.5-7.7 (8H, m), 8.22 (1H, d, J=8.4 Hz), 8.36 (1H, s)
- 109) $l_{\text{H-NMR}(CDCl_3)} \delta$; 1.2-3.0 (10H, m), 3.0-5.2(6H, m), 6.3-7.7 (10H, m)

- 114)

 1_{H-NMR}(CDCl₃) δ; 1.5-1.7 (1H, m), 2.1-2.3 (1H, m),
 2.41 (6H, s), 2.4-2.6 (1H, m), 2.8-3.0 (3H, m),
 3.71 (3H, s), 5.2-5.4 (1H, m), 6.6-6.8 (2H, m),
 6.9-7.5 (7H, m), 7.7-7.8 (1H, m), 8.27 (1H, d,
 J=8.4 Hz), 8.57 (1H, s)

(1H, m)

- 106) ¹H-NMR(CDCl₃) δ; 1.20-2.53 (13H, m), 2.63-2.82, 3.00-3.13, 3.50-3.67, 4.05-4.23 (total 3H, m), 6.55 8.00 (13H, m)

WO 91/05549 - 495 - PCT/JP90/01340

- ¹H-NMR(CDCl₃) δ; 1.5-1.7 (1H, m), 2.1-2.7 (2H, m), 2.41 (6H, s), 2.7-3.0 (3H, m), 3.71 (3H, s), 5.2-5.4 (1H, m), 6.6-7.6 (8H, m), 7.70 (1H, d, J=8.3 Hz), 8.24 (1H, d, J=8.5 Hz), 8.59 (1H, s)
- 116) ¹H-NMR(CDCl₃) δ; 1.8-2.3 (3H, m), 2.7-2.9 (2H, m), 3.5-3.7 (1H, m), 6.8-8.0 (10H, m), 8.7-9.1 (1H, br) Example 634

To a mixture of 5-oxo-1-[4-(3,5-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (4 g) and pyridine (50 ml) is added hydroxylamine hydrochloride (1.84 g) and the mixture is refluxed for 2.5 hours. The reaction solution is concentrated and water is added to the resulting residue. The precipitated crystal is collected by filtration, and recrystallized from dioxane/water to give 5-hydroxyimino-l-[4-(3,5-dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (2 g) as white powder, m.p. 272 - 273°C.

Example 635

5-Chloro-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (0.8 g) is dissolved in dimethylformamide and thereto is added sodium azide (0.18 g) at room temperature. The mixture is stirred at room temperature overnight, and further reacted with heating at 50°C for 5 hours. Water is added to the reaction mixture and the precipitated crystal is collected by filtration to give 5-azido-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-

tetrahydro-lH-benzazepine (0.68 g) as light brown powder.

 $l_{H-NMR}(CDCl_3)$ & ; 1.65-3.1 (8H, m), 4.7-6.6 (3H, m), 6.6-6.8 (1H, m), 6.85-7.65 (12H, m)

Example 636

5-Azido-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (0.63 g) is dissolved in ethanol and thereto is added 10 % Pd-C (0.1 g). The mixture is subjected to catalytic hydrogenation at room temperature under 1 atm. of hydrogen. Pd-C is removed by filtration and the filtrate is evaporated. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50:1), and recrystallized from diethyl ether to give 5-amino-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.34 g) as white powder, m.p. 198.5 - 199.5°C.

Example 637

To 5-hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.58 g) are added acetic anhydride (8.0 ml) and pyridine (2.0 ml). The mixture is stirred at room temperature for 1 hour. Water is added to the reaction mixture and the precipitated crystal is collected by filtration, and recrystallized from ethyl acetate/n-hexane to give 5-acetyloxy-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.56 g) as white powder, m.p. 193 - 194°C.

Example 638

5-Ethoxycarbonylmethoxy-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.00 g) is dissolved in methanol (35 ml) and thereto are added aqueous ammonia (20 ml) and ammonium chloride (0.50 g). The mixture is heated at 100°C for 3.5 hours in a sealed tube. After cooling, the reaction solution is concentrated under reduced pressure and acidified with hydrochloric acid, and extracted with dichloromethane. The extract is dried over magnesium sulfate and the solvent is distilled off. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 15:1) to give 5-carbamoylmethoxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.68 g) as colorless amorphous.

¹H-NMR(CDCl₃) δ; 1.56-2.67 (4H, m), 2.46 (3H, s),
2.67-3.03 (1H, m), 3.82-4.32 (2H, m), 4.45-5.15 (2H, m),
5.43-5.83 (1H, m), 6.20-6.45 (1H, m), 6.50-6.86 (2H, m),
6.86-7.70 (10H, m), 7.76-8.10 (1H, m)

Using the suitable starting materials, the compounds of the above Examples 593 and 594 are obtained in the same manner as in Example 638.

Example 639

5-Ethoxycarbonylmethoxy-1-[4-(2,4-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.94 g) is dissolved in ethanol (100 ml) and thereto is added 5N agueous sodium hydroxide solution (0.50 ml). The mixture is

stirred at room temperature for 2 hours. The reaction solution is concentrated under reduced pressure and to the resulting residue is added diluted hydrochloric acid and then extracted with dichloromethane. The extract is dried over magnesium sulfate and the solvent is distilled off. The resulting residue is washed with n-hexane and collected by filtration to give 5-carboxymethoxy-1-[4-(2,4-dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.79 g) as colorless amorphous.

 $1_{\text{H-NMR}(\text{CDCl}_3)}$ & ; 1.52-1.89 (2H, m), 1.89-2.56 (2H, m), 2.65-3.02 (1H, m), 3.90-4.40 (2H, m), 4.40-5.07 (2H, m), 6.58-6.78 (1H, m), 6.90-7.70 (10H, m), 8.57-8.81 (1H, brs)

Using the suitable starting materials, the compounds of the above Examples 595 and 596 are obtained in the same manner as in Example 639.

Example 640

5-Carboxymethoxy-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.55 g) is dissolved in dimethylformamide (20 ml) and thereto are added dimethylamine hydrochloride (0.20 g) and diethyl chloro-phosphate (0.33 g). To the mixture is added triethylamine (1.0 ml) under ice-cooling, and the mixture is stirred under ice-cooling for 30 minutes, and at room temperature for more 2 hours. Water is added to the reaction solution and the precipitated crystal is collected by filtration and recrystallized from ethyl acetate/n-bexane to give 5-

dimethylaminocarbonylmethoxy-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.50 g) as colorless prisms, m.p. 203 - 204°C.

Using the suitable starting materials, the compounds of the above Examples 599 and 600 are obtained in the same manner as in Example 640.

Example 641

5-[3-(Phthalimid-1-yl)propoxy]-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.26 g) is dissolved in ethanol (100 ml) and thereto is added hydrazine hydrate (1.0 ml). The mixture is refluxed with stirring for 1 hour. The reaction solution is evaporated under reduced pressure and to the resulting residue is added dichloromethane. The insoluble materials are removed by filtration. The filtrate is purified by silica gel column chromatography (eluent; dichloromethane: methanol: aqueous ammonia = 70:10:1) to give 5-(3-aminopropoxy)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.70 g) as colorless amorphous.

¹H-NMR(CDCl₃) 6; 1.42-2.32 (6H, m), 2.44 (3H, s), 2.57-2.97 (1H, m), 3.12-3.83 (4H, m), 4.39-5.13 (2H, m), 6.50-6.71 (1H, m), 6.90-7.73 (12H, m)

Example 642

A solution of 5-dimethylamino-1-[3-methoxy-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-

benzazepine (0.50 g) in dichloromethane (30 ml) is added dropwise to a solution of 1M boron tribromide in dichloromethane (5.46 ml) at -45°C. After completion of the dropping, the mixture is stirred for 1 day while the temperature of the reaction mixture is gradually raised to room temperature. To the reaction solution is added water and the mixture is neutralized with sodium hydrogen carbonate, and extracted with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; chloroform : methanol = 500 : 1), and recrystallized from methanol/diethyl ether to give 5dimethylamino-1-[3-hydroxy-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.33 g) as white powder, m.p. 201.5 - 202.5°C.

Using the suitable starting materials, the compounds of the above Examples 10, 32, 343, 356, 535, 555 and 556 are obtained in the same manner as in Example 642.

Example 643

To a solution of 4-[4-(2-methylbenzoylamino)-benzoyl]-3,4-dihydro-2H-1,4-benzazepine (0.5 g) in dichloromethane (10 ml) is added m-chloroperbenzoic acid (0.58 g) under ice-cooling, and the mixture is stirred at room temperature for 6 hours. The above reaction solution is poured into an aqueous solution of sodium carbonate (0.6 g) in water (10 ml) and the mixture is extracted with dichloro-

methane. The extract is washed with water, and dried over magnesium sulfate. The solvent is distilled off under reduced pressure and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 100: 1), and recrystallized from diethyl ether/dichloromethane to give 4-[4-(2-methylbenzoylamino)-benzoyl]-3,4-dihydro-2H-1,4-benzothiazine-1,1-dioxide (0.49g) as white powder, m.p. 219 - 220°C.

Using the suitable starting materials, the compound of the above Example 630 is obtained in the same manner as in Example 643.

Example 644

To a suspension of 4-[4-(2-methylbenzoylamino)-benzoyl]-3,4-dihydro-2H-1,4-benzothiazine (0.5 g) in methanol (15 ml) is added an aqueous solution of sodium metaperiodate (0.28 g) in water (2.5 ml) and the mixture is stirred at room temperature for 72 hours. Water is added to the reaction solution and extracted with dichloromethane. The extract is dried over magnesium sulfate and the solvent is distilled off under reduced pressure. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 100: 1), and recrystallized from dichloromethane/diethyl ether to give 4-[4-(2-methylbenzoylamino)benzoyl]-3,4-dihydro-2H-1,4-benzothiazin-1-oxide (0.34 g) as white powder, m.p. 240 - 241°C.

Using the suitable starting materials, the compound of the above Example 631 is obtained in the same manner as in Example 644.

2

Example 645

5-Hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (3.57 g) is dissolved in dichloromethane (30 ml) and pyridine (1.1 ml), and thereto is added dropwise methanesulfonyl chloride (0.9 ml) in small portions at 0°C. Then, the mixture is stirred at room temperature for 3 days. The solvent is distilled off and the resulting residue is poured into ice-water. The precipitated crystal is collected by filtration, washed with water, and dried to give 5-chloro-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (3.10 g) as light yellow powder.

 $^{1}\text{H-NMR}(\text{CDCl}_{3})$ & ; 1.7-2.9 (8H, m), 4.5-6.5 (3H, m), 6.55-6.75 (1H, m), 6.85-7.6 (12H, m)

Example 646

5-Hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (2.69 g) is dissolved in dimethylformamide (30 ml) and thereto are added 60 % sodium hydride dispersion in mineral oil (0.44 g) and ethyl bromoacetate (1.00 ml) under ice-cooling, and the mixture is stirred at room temperature for 4 hours. The reaction solution is poured into an aqueous ammonium chloride solution under ice-cooling, and extracted with ethyl

acetate. The extract is dried over magnesium sulfate and the solvent is distilled off. The resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate: n-hexane = 1:2), and recrystallized from ethyl acetate/n-hexane to give 5-ethoxycarbonylmethoxy-1-[4-(2-methylbenzoylamino)benzoyl-2,3,4,5-tetrahydro-lH-benzazepine (2.10 g) as white powder, m.p. 178 - 181°C.

Using the suitable starting materials, the compounds of the above Examples 585 - 588 and 590 - 606 are obtained in the same manner as in Example 646.

Example 647

الرابات المتعددة المت

Using the suitable starting materials, the compounds of the above Examples 546 and 578 - 581 are obtained in the same manner as in Example 384.

Example 648

Using the suitable starting materials, the compounds of the above Examples 537 - 545, 547, 549 - 556, 561 - 564, 566, 568 - 571, 577, 601 - 603 and 607 - 625 are obtained in the same manner as in Example 388.

Example 649

Using the suitable starting materials, the compounds of the above Examples 549, 568 - 571, 575 and 606 are obtained in the same manner as in Example 389.

Example 650

Using the suitable starting materials, the compounds of the above Examples 537 - 545, 547, 549 - 556,

561 - 566, 568 - 571, 575, 577, 607, 608 and 613 - 625 are obtained in the same manner as in Example 390.

Example 651

Using the suitable starting materials, the compounds of the above Examples 601 - 603, 605 and 606 are obtained in the same manner as in Example 397.

Example 652

Using the suitable starting materials, the compound of the above Example 604 is obtained in the same manner as in Example 398.

Example 653

Using the suitable starting materials, the following compound is obtained in the same manner as in Examples 1, 382, 388 and 390.

5-Methylamino-1-[2-chloro-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 184.5 - 185.5°C (recrystallized from ethanol)

Using the suitable starting materials, the compounds of the following Table 4 are obtained in the same manner as in Examples 1 and 382.

Table 4

Example 654

Structure

R²: 2-ОН

O CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 193.5 - 196°C

Structure

R²: 2-OH

0 C1 R³: 4-NHC-

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 195 - 198°C

Form: Free

Example 656

Structure

R²: 2-OC₂H₅

Crystalline form: White powder

Recrystallization solvent: Methanol

Melting Point: 230.5 - 231.5°C

Structure

 R^2 : 2-OC₂H₅

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 223 - 224.5°C

Form: Free

Example 658

Structure

$$\mathbb{R}^{1} \quad | \quad \mathbb{N}^{(CH_{3})_{2}}$$

 R^2 : 2-OC₂H₅

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 173 - 174°C

Structure

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 174 - 175°C

Form: Free

Example 660

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}^{(CH_{3})_{2}}$$

 $R^2: 3-CH_3$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 198 - 200°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{N(CH}_3)_2 \\
\text{R}^1 \\
\text{N}
\end{array}$$

R²: 3-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 149 - 150.5°C

Form: Free

Example 662

Structure

re
$$N(CH_3)_2$$
 R^1 : N

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 183 - 185°C

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{N}^{(CH_3)_2}$

R²: 2-СН₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 203 - 207°C

Form: Free

Example 664

Structure

$$\mathbb{R}^{1} \qquad \mathbb{N}^{(CH_{3})_{2}}$$

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 221 - 222°C

Structure

re
$$N(CH_3)_2$$
 R^1
 N

R²: 2-F

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 189 - 191°C

Form: Free

Example 666

Structure

 $R^2: 2-F$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 215.5 - 217°C

Structure

re
$$N(CH_3)_2$$
 R^1 N

R²: 2-F

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 192 - 194°C

Form: Free

Example 668

Structure

R²: 3-F

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 195 - 196°C

Structure

R²: 3-F

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 202 - 204.5°C

Form: Free

Example 670

Structure

re
$$N(CH_3)_2$$
 R^1 N

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 183 - 187°C

Structure

are
$$\frac{N(CH_3)_2}{R^1}$$
:

R²: H

$$R^3: 4-NHC \xrightarrow{O (CH_2)_2N(C_2H_5)_2}$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 120 - 122°C

Form: Free

Example 672

Structure

$$\mathbb{R}^{1} \quad | \quad \mathbb{N}$$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Chloroform/diethyl ether

Melting Point: 208 - 210°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 182 - 183°C

Form: Free

Example 674

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 257 - 259°C

Structure

 R^2 : 2-OC₂H₅

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 134 - 135°C

Form: Free

Example 676

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 167 - 169°C

Structure

 R^2 : 2-C1

Crystalline form: Light brown prisms

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 170 - 172°C

Form: Free

Example 678

Structure

re
$$M(CH_3)_2$$
 R^1 $N(CH_3)_2$

0 Cl R²: 3-OC

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 181.5 - 182.5°C

Structure

$$: \qquad \stackrel{\mathsf{N}(\mathsf{CH}_3)_2}{\longrightarrow}$$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

NHCH2CH=CH2

Melting Point: 176.5 - 177°C

Form: Free

Example 680

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{N} :

 $R^2: 2-C1$

Crystalline form: Yellow amorphous

NMR analysis: 117)

Structure

R²: 2-C1

$$R^3: 4-NHC$$

Crystalline form: Yellow amorphous

NMR analysis: 118)

Form: Free

Example 682

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 236 - 239°C

Structure

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 153 - 154°C

Form: Free

Example 684

Structure

R²: F

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 128 - 130°C

Structure

re
$$N-COCH_3$$
 $R^2: H$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 231 - 234°C

Form: Free

Example 686

Structure

 R^2 : F

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate

Melting Point: 246 - 248°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

Melting Point: 248 - 248.5°C

Form: Free

Example 688

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 204 - 205°C

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: >300°C

NMR analysis: 119)

Form: Free

....

Example 690

Structure

$$\mathbb{R}^1$$
 : $\mathbb{S}^{=0}$

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 292 - 294°C

Structure

R²: 2-N(CH₃)₂

Crystalline form: Colorless amorphous --

NMR analysis: 120)

Form: Free

Example 692

Structure

$$\begin{array}{c} \mathbb{R}^1 & \mathbb{N} \\ \mathbb{R}^1 & \mathbb{N} \\ \mathbb{R}^1 & \mathbb{N} \end{array}$$

R²: 2-N(CH₃)₂

Crystalline form: Colorless amorphous

NMR analysis: 121)

Structure

R²: 2-C1

Crystalline form: Colorless amorphous . .

NMR analysis: 122)

Form: Free

Example 694

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR_analysis: 123)

Structure

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 198.5 - 199°C

Form: Free

Example 696

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 168 - 170°C

Structure

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 175 - 176°C

Form: Free

Example 698

Structure

$$\mathbb{R}^1$$
 \mathbb{N} \mathbb{N}

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 177 - 178°C

Structure

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 222 - 223.5°C

Form: Free

Example 700

Structure

$$\mathbb{R}^{1} \quad : \quad \mathbb{N}$$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 243 - 244°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 180 - 181°C

Form: Free

Example 702

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 124)

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 231 - 233°C

Form: Free

Example 704

Structure

к²: н

$$R^3: 4-NHC \longrightarrow_F^F$$

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 196 - 198°C

Structure

re
$$\frac{NHCH_3}{R^1}$$
 :

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 125)

Form: Free

Example 706

Structure

re
$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: 2-Cl

Crystalline form: Yellow amorphous

NMR analysis: 126)

Structure

re
$$N(CH_3)_2$$
 R^1 N

к²: в

O || R³: 4-NHCCH₂Cl

Crystalline form: Yellow powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 146 - 147°C

Form: Free

Example 708

Structure

re
$$N(CH_3)_2$$
 R^1 N

к²: н

$$R^3: 4-NHCCH_2N$$
 CH_2CH_3

Crystalline form: Colorless amorphous

NMR analysis: 127)

Structure

к²: н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 220 - 221°C

Form: Free

Example 710

Structure

R²: F

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 170 - 172°C

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 128)

Form: Free

Example 712

Structure

re
$$N(CH_3)_2$$
 R^1 : N

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 224 - 225°C

Structure

are
$$N(CH_3)_2$$
 R^1 N

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 193 - 196°C

Form: Free

Example 714

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1

R²: 2-C]

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 212 - 214°C

Structure

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 211 - 213°C

Form: Free

Example 716

Structure

are
$$\frac{N(CH_3)_2}{R^1}$$
:

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 213 - 215°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3} & \text{O}_{1} \\
\text{N} \\
\text{N}
\end{array}$$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 199 - 201°C

Form: Free

Example 718

Structure

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 238 - 240°C

Structure

 $N(CH_2)_2N(CH_3)_2$

$$\left(\left(\right) \right)^{N}$$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 188 - 189°C

Form: Free

Example 720

Structure

 $NHCH_2CH_3$



R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Dioxane/water

Melting Point: 135.5 - 137°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^1
\end{array}$$
:

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Isopropyl alcohol/

petroleum ether

Melting Point: 192 - 193°C

Form: Free

Example 722

Structure

re
$$N$$
 CH_2
 CH_2
 N
 R^1

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate

Melting Point: 239 - 240°C

Structure

Crystalline form: Colorless amorphous

NMR analysis: 129)

Form: Free

Example 724

Structure

The
$$CH_2$$
 CH_2 CH_2 CH_2 CH_3 CH_2 CH_3 CH_2 CH_3 CH_2 CH_3 CH_3 CH_4 CH_5 $CH_$

Crystalline form: Colorless amorphous

NMR analysis: 130)

Structure

к²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 193 - 194°C

Form: Free

Example 726

Structure

re
$$N-OH$$

$$R^1$$

$$R$$

R²: Н

Crystalline form: Light yellow prisms

Recrystallization solvent: Ethanol

Melting Point: 245.5 - 247°C

Structure

re
$$N \sim OCOCH_3$$
 R^1 $N \sim OCOCH_3$

к²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 142 - 144°C

Form: Free

Example 728

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Light yellow prisms

Recrystallization solvent: Ethanol

Melting Point: 214 - 217°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

p2. H

Crystalline form: Colorless needles

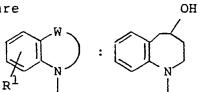
Recrystallization solvent: Ethanol

Melting Point: 205 - 207°C

Form: Free

Example 730

Structure



R²: E

$$R^3$$
: 4-NHC- \mathbb{R}^3

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 201 - 203°C

Structure

re
$$N(CH_3)_2$$

$$OH$$

$$R^1$$

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 180 - 182°C

Form: Free

Example 732

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: H

Crystalline form: Light yellow scales

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 178 - 180°C

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 208 - 213°C

Form: Free

Example 734

Structure

re
$$W$$
 OH OH

R²: н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 175 - 177°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}

R²: Н

Crystalline form: White powder

NMR analysis: 131)

Form: Free

Example 736

Structure

к²: н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 277 - 279°C

Form: Free

•

¢ ;

Structure

re
$$\frac{\text{OCONH}_2}{\mathbb{R}^1}$$
:

R²: H

Crystalline form: Colorless amorphous

NMR analysis: 132)

Form: Free

Example 738

Structure

re
$$OCON(CH_3)_2$$
 R^1 : N

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 133)

Structure

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 134)

Form: Free

Example 740

Structure

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 135)

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 213 - 214°C

Form: Free

Example 742

Structure

$$\mathbb{R}^{1} \longrightarrow \mathbb{R}^{1}$$

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 216 - 217°C

Structure

$$\begin{array}{c}
\text{CH}_2\\
\text{N}\\
\text{R}^1 & |
\end{array}$$

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

CH₂OH

Melting Point: 165 - 167°C

Form: Free

Example 744

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 202 - 206°C

Structure

re
$$\mathbb{C}H_2OH$$
 \mathbb{R}^1 : $\mathbb{C}H_2OH$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 220 - 221.5°C

Form: Free

Example 746

Structure

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: H

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 186 - 186.5°C

Structure

R²: 2-C1

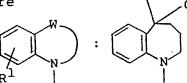
Crystalline form: Colorless amorphous

NMR analysis: 136)

Form: Free

Example 748

Structure



к²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

сн₂он

Melting Point: 136 - 140°C

Form: Free

ė

Structure

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 151 - 153°C

Form: Free

Example 750

Structure

$$\mathbb{R}^{1}$$

CH2OCOCH3

ъ2. u

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 155 - 156°C

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 189 - 190°C

Form: Free

Example 752

Structure

re
$$CH_2N_3$$
 R^1 R^1

p2. н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 188 - 190°C

Form: Free

.

Structure

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 233 - 235°C

Form: Free

Example 754

Structure

CH₂OH

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 137)

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 176 - 179°C

Form: Free

Example 756

Structure

OCH₃

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 183 - 185°C

- ¹H-NMR(CDCl₃) δ; 1.3-2.3 (4H, m), 3.1-3.4 (3H, m), 3.8-4.6 (2H, m), 5.0-5.3 (2H, m), 5.8-6.1 (1H, m), 6.8-8.5 (11H, m)
- ll8) 1 H-NMR(CDCl₃) δ ; 1.6-2.2 (4H, m), 2.46, 2.53 (3H, each s), 3.1-3.5 (3H, m), 3.8-4.6 (2H, m), 5.0-5.3 (2H, m), 5.8-6.1 (1H, m), 6.8-8.0 (11H, m)
- 120) ¹H-NMR(CDCl₃) 6; 1.25-5.05 (22H, m), 6.65-7.65 (11H, m), 7.75-8.25 (1H, m)
- 121) ¹H-NMR(CDCl₃) 6; 1.15-5.05 (19H, m), 6.75-7.85 (11H, m), 7.85-8.25 (1H, m)
- 122) $^{1}H-NMR(CDCl_{3})$ & ; 1.25-2.85 (8H, m), 2.95 4.95 (2H, m), 6.75-7.85 (10H, m), 9.25-9.75 (1H, m)
- 123) ¹H-NMR(CDCl₃) δ; 0.20-0.70 (4H, m), 0.95-2.35 (6H, m), 2.65-5.00 (2H, m), 6.75-7.90 (10H, m), 8.65-9.25 (1H, m)
- 124) ¹H-NMR(CDCl₃) δ; 1.20-3.15 (11H, m), 3.45-3.70 (1H, m), 4.05-5.20 (1H, m), 6.60-7.65 (10H, m), 8.15-8.45 (2H, m)
- 125) ¹H-NMR(CDCl₃) δ; 1.19 (3H, t, J=7 Hz), 1.25-3.25 (8H, m), 3.46 (2H, q, J=7 Hz), 3.40-4.10 (3H, m), 4.45-5.10 (1H, m), 6.65-7.75 (12H, m), 8.30-8.60 (1H, m)

¹ H-NMR(CDCl ₃) 6 ; 1.10-1.30 (3H, m), 1.50-2.35 (4H,	
m), 2.65-3.05 (2H, m), 3.35-3.60 (2H, m), 3.80-4.05	
(2H, m), 4.65-5.15 (2H, m), 6.55-7.85 (12H, m),	,
8.35-8.65 (lH, m)	•
¹ H-NMR(CDCl ₃) & ; 1.20 (3H, t, J=7 Hz), 1.10-3.15	i
	-
_	
_	
m)	
¹ H-NMR(CDCl ₃) & ; 1.10 (6H, t, J=7 Hz), 1.35-5.1	
(23H, m), 6.55-7.8 (13H, m)	
¹ H-NMR(CDCl ₃) 6; 1.94-3.21 (3H, m), 3.30-4.82 (3H,	
8.72 (1H, brs)	
¹ H-NMR(DMSO-d ₆) δ ; 1.57-1.85 (2H, m), 1.85-2.28	
(2H, m), 2.33 (3H, s), 2.64-2.86 (1H, m), 4.53-5.07	
to the control of the	
(12H, m)	•
¹ H-NMR(CDCl ₃) δ ; 1.68-1.97 (2H, m), 2.03-2.53 (2H,	1
	(2H, m), 4.65-5.15 (2H, m), 6.55-7.85 (12H, m), 8.35-8.65 (1H, m) 1H-NMR(CDCl ₃) 6; 1.20 (3H, t, J=7 Hz), 1.10-3.15 (11H, m), 3.45-3.65 (3H, m), 3.88 (2H, s), 3.95-5.15 (1H, m), 6.55-7.65 (13H, m), 8.37 (1H, s) 1H-NMR(CDCl ₃) 6; 2.45 (3H, s), 3.40 (3H, s), 4.01 (2H, m), 4.38 (2H, m), 7.20-7.77 (13H, m) 1H-NMR(CDCl ₃) 6; 1.35-4.55 (22H, m), 6.3-7.8 (13H, m) 1H-NMR(CDCl ₃) 6; 1.10 (6H, t, J=7 Hz), 1.35-5.1 (23H, m), 6.55-7.8 (13H, m) 1H-NMR(CDCl ₃) 6; 1.94-3.21 (3H, m), 3.30-4.82 (3H, m), 6.57 (1H, d, J=7.5 Hz), 6.86-8.10 (11H, m), 8.72 (1H, brs) 1H-NMR(DMSO-d ₆) 6; 1.57-1.85 (2H, m), 1.85-2.28 (2H, m), 2.33 (3H, s), 2.64-2.86 (1H, m), 4.53-5.07 (1H, m), 5.79-5.94 (1H, m), 6.47-7.68 (2H, br), 6.64-6.77 (1H, m), 6.96-7.62 (12H, m) 1H-NMR(CDCl ₃) 6; 1.61-1.97 (2H, m), 2.00=2.54 (2H, m), 2.47 (3H, s), 2.60-3.23 (7H, m), 4.76-5.22 (1H, m), 5.94-6.19 (1H, m), 6.61-6.74 (1H, m), 6.91-7.62

m), 2.61-3.24 (7H, m), 4.76-5.22 (1H, m), 5.97-6.17

(1H, m), 6.59-6.74 (1H, m), 6.92-7.13 (1H, m), 7.13-7.58 (9H, m), 7.66-7.85 (1H, m), 7.85-8.00 (1H, m)

- 135) ¹H-NMR(CDCl₃) δ; 1.57-1.93 (2H, m), 1.93-2.54 (2H, m), 2.54-2.72 (1H, m), 2.79-3.09 (3H, m), 3.90-4.32 (2H, m), 4.49-5.18 (2H, m), 6.31-6.93 (2H, m), 6.96-7.63 (10H, m), 7.63-7.89 (1H, m), 7.89-8.16 (1H, m)
- 136) ¹H-NMR(CDCl₃) δ; 1.44-1.95 (2H, m), 1.95-2.28 (2H, m), 2.40-2.67 (3H, m), 2.73-3.38 (3H, m), 3.40-3.97 (1H, m), 4.50-5.20 (1H, m), 6.67-8.11 (11H, m)

Example 757

A mixture of 5-dimethylamino-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (10 g), methyl iodide (1.7 ml) and chloroform (10 ml) is heated with stirring at 100°C for 3 hours in an autoclave. After completion of the reaction, the solvent is distilled off under reduced pressure and the resulting residue is dissolved in methanol. The mixture is treated with IRA-400 (trade mark; Organo Co., Ltd., OH type). Methanol is

distilled off and the resulting residue is suspended in t-butyl alcohol (90 ml), and thereto is added potassium t-butoxide (2.3 g). The mixture is refluxed for 5 hours. The solvent is distilled off under reduced pressure, and the resulting residue is dissolved in dichloromethane. The mixture is washed successively with water and saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and to the resulting residue is added dichloromethane/diethyl ether. The precipitated crude crystal is recrystallized from ethanol to give 1-[4-(2-chlorobenzoylamino)benzoyl]-2,3-dihydro-1H-benzazepine (5.15 g) as colorless needles, m.p. 205 - 207°C.

Example 758

l-[4-(2-Chlorobenzoylamino)benzoyl]-2,3-dihydro-lH-benzazepine (4.7 g) is dissolved in dichloromethane (50 ml) and thereto is added 80 % m-chloroperbenzoic acid (3 g). The mixture is stirred at room temperature overnight. The dichloromethane layer is washed successively with saturated aqueous sodium hydrogen carbonate solution and saturated saline solution, and the solvent is distilled off under reduced pressure. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50 : 1) to give 4,5-epoxy-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (4.26 g) as white powder.

 $1_{\text{H-NMR}(CDCl_3)}$ & ; 1.94-3.21 (3H, m), 3.30-4.82 (3H,

s s

.

. 2 2 m), 6.57 (1H, d, J=7.5 Hz), 6.86-8.10 (11H, m), 8.72 (1H, brs)

Example 759

A mixture of 4,5-epoxy-1-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.5 g), dimethylamine hydrochloride (2.6 g), triethylamine (4.5 g) and methanol (15 ml) is refluxed for 19 hours. After completion of the reaction, the solvent is distilled off and the resulting residue is dissolved in dichloromethane. The mixture is washed successively with water and saturated saline solution. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 50 :

1), and recrystallized from ethanol/diethyl ether to give trans-5-dimethylamino-4-hydroxy-1-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.38 g) as colorless needles, m.p. 180 - 182°C.

Using the suitable starting materials, the compounds of the above Examples 733 and 734 are obtained in the same manner as in Example 759.

Example 760

Methyltriphenylphosphonium bromide (4.30 g) is suspended in tetrahydrofuran (100 ml) and thereto is added potassium t-butoxide (1.58 g) under ice-cooling. The mixture is stirred at -5°C for 1 hour and thereto is added 5-oxo-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-

tetrahydro-lH-benzazepine (1.60 g) and the mixture is stirred at room temperature for 1 hour. The reaction solution is poured into ice-water (200 ml) and extracted with ethyl acetate. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate: n-hexane = 1:2), and recrystallized from ethyl acetate/n-hexane to give 5-methylidene-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.34 g) as white powder, m.p. 216 - 217°C.

Using the suitable starting materials, the compound of the above Example 743 is obtained in the same manner as in Example 760.

Example 761

5-Methylidene-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (2.84 g) is suspended in tetrahydrofuran (50 ml) and thereto is added 1 M solution of boran-tetrahydrofuran complex in tetrahydrofuran (43 ml).

The mixture is stirred at room temperature for 6 hours.

After completion of the reaction, the reaction solution is cooled with ice, and thereto is added water (70 ml). After termination of the evolution of hydrogen gas, to the reaction solution are added 25 % aqueous sodium hydroxide solution (7.0 ml), and subsequently 31 % aqueous hydrogen peroxide solution (4.7 ml), and the mixture is heated with

stirring at 50°C for 1 hour. After cooling, to the reaction solution is added saturated saline solution and the tetrahydrofuran layer is collected, washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is recrystallized from ethyl acetate/n-hexane to give 5-hydroxymethyl-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.96 g) as white powder, m.p. 202-206°C.

Using the suitable starting materials, the compound of the above Example 745 is obtained in the same manner as in Example 761.

Example 762

5-Methylidene-1-(2-chloro-4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.81 g) is dissolved in dichloromethane (30 ml) and thereto is added m-chloroperbenzoic acid (0.57 g). The mixture is stirred at room temperature for 15 hours. After completion of the reaction, the reaction solution is washed successively with aqueous sodium hydrogensulfite solution, aqueous sodium hydrogen carbonate solution and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified with silica gel column chromatography (eluent; ethyl acetate: n-hexane = 2:3) to give 5,5-epoxy-1-(2-chloro-4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.70 g) as

colorless amorphous.

 1 H-NMR(CDCl₃) $_{\delta}$; 1.44-1.95 (2H, m), 1.95-2.28 (2H, m), 2.40-2.67 (3H, m), 2.73-3.38 (3H, m), 3.40-3.97 (1H, m), 4.50-5.20 (1H, m), 6.67-8.11 (11H, m)

Using the suitable starting materials, the compound of the above Example 746 is obtained in the same manner as in Example 762.

Example 763

To 5-methylidene-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.60 g) are added successively t-butyl alcohol (6.0 ml), water (1.2 ml), pyridine (0.3 ml), osmium tetroxide (1.2 mg) and trimethylamine N-oxide dihydrate (0.22 g), and the mixture is refluxed with stirring for 2.5 hours. After cooling, to the reaction solution is added 20 % aqueous sodium hydrogensulfite solution (10 ml), and the mixture is stirred at room temperature for 1.5 hour. The reaction solution is extracted with a mixture of ethyl acetate/tetrahydrofuran (1:1). The extract is washed successively with diluted hydrochloric acid and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is recrystallized from ethyl acetate/ n-hexane to give 5-hydroxymethyl-5-hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.55 g) as white powder, m.p. 136 - 140°C.

Using the suitable starting materials, the compound

÷

ě

of the above Example 749 is obtained in the same manner as in Example 763.

Example 764

benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.40 g) are added acetic anhydride (4.0 ml) and pyridine (0.5 ml), and the mixture is stirred at room temperature for 5 hours.

After completion of the reaction, the reaction solution is poured into ice-water and extracted with ethyl acetate. The extract is washed successively with diluted hydrochloric acid and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is recrystallized from ethyl acetate/n-hexane to give 5-acetyloxymethyl-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.43 g) as colorless needles, m.p. 155 - 156°C.

Example 765

5-Hydroxymethyl-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.70 g) is dissolved in a mixture (30 ml) of dichloromethane/acetonitrile (1:1) and thereto are added methanesulfonyl chloride (0.8 ml) and pyridine (1.0 ml), and the mixture is refluxed with stirring for 2 hours. After cooling, the reaction solution is evaporated under reduced pressure and to the resulting residue is added water and then extracted with ethyl acetate. The extract is washed successively with

diluted hydrochloric acid and saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is recrystallized from ethyl acetate/n-hexane to give 5-methanesulfonyloxymethyl-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.72 g) as white powder, m.p. 189 - 190°C.

Example 766

5-Methanesulfonyloxymethyl-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.49 g) is dissolved in a mixture (25 ml) of acetonitrile/dimethyl-formamide (4:1) and thereto is added sodium azide (0.11 g). The mixture is refluxed with stirring for 3.5 hours. After cooling, the reaction solution is poured into ice-water (40 ml), extracted with ethyl acetate, washed with saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate: n-hexane = 1 : 2), and recrystallized from ethyl acetate/n-hexane to give 5-azidomethyl-1-[4-(2-methylbenzoylamino)benzoly]-2,3,4,5-tetrahydro-1H-benzazepine (0.29 g) as white powder, m.p. 188 - 189°C.

Example 767

5-Azidomethyl-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine (0.27 g) is suspended in
ethanol (50 ml) and the mixture is subjected to catalytic

hydrogenation at room temperature under 3 kg/cm² for 6 hours by using 10 % Pd-C (27 mg). The catalyst is removed by filtration with celite and the filtrate is distilled off and the resulting residue is recrystallized from ethanol to give 5-aminomethyl-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.12 g) as colorless needles, m.p. 233 - 235°C.

Example 768

To 5,5-epoxy-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine (0.30 g) is added 30 %
solution of methylamine in methanol (30 ml), and the mixture
is refluxed for 14 hours. After compeltion of the reaction,
the reaction solution is evaporated under reduced pressure
and the resulting residue is purified by silica gel column
chromatography (eluent; ethyl acetate: n-hexane = 1:1 +
dichloromethane: methanol: aqueous ammonia = 60:10:1)
to give 5-hydroxymethyl-5-methylamino-l-[4-(2-methylbenzoylamino]benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (A; 35.3
mg) and 5-methylaminomethyl-5-hydroxy-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (B; 109
mg).

(A); Colorless amorphous

¹H-NMR(CDCl₃) δ; 1.50-2.10 (3H, m), 2.10-2.28 (1H, m), 2.36 (3H, s), 2.48 (3H, s), 2.68-2.97 (1H, m), 3.26-3.47 (1H, m), 4.16 (1H, d, J=13.8 Hz), 4.25 (1H, d, J=13.8 Hz), 5.95 (1H, brs), 6.60-6.76 (1H, m), 6.97-7.52 (8H, m), 7.52-

7.73 (2H, m), 7.73-7.97 (2H, m)

m.p. 176 - 179°C

Example 769

5-Methylamino-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine (1 g) is dissolved in dimethylformamide (10 ml) and thereto are added potassium carbonate (0.5 g) and ethyl iodide (0.45 g). The mixture is stirred at room temperature overnight. After completion of the reaction, the reaction solution is poured into ice-water and the precipitated crystal is collected by filtration, and purified by silica gel column chromatography (eluent; dichloromethane: methanol = 90:1), and recrystallized from diisopropyl alcohol/petroleum ether to give 5-(N-methyl-N-ethylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (50 mg) as white powder, m. p. 192 - 193°C.

Using the suitable starting materials, the compounds of the above Examples 244, 246 - 248, 330, 339, 342, 346, 350, 366, 375, 376, 406 - 418, 453, 455, 457, 460, 464, 467, 506, 507, 537 - 545, 547, 549 - 556, 561 - 566, 568 - 571, 577, 601 - 603, 607 - 625, 654 - 672, 675, 677 - 681, 691 - 695, 697, 698, 701 - 705, 707, 708, 712, 713, 715, 716, 719, 720 and 722 - 725 are obtained in the same manner as in Example 769.

• •

*

To a suspension of 5-methylamino-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (3 g) in methanol (30 ml) are added potassium carbonate (1.5 g) and epichlorohydrine (5.7 ml), and the mixture is refluxed for 3 hours. The solvent is distilled off and to the resulting residue is added water and extracted three times with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 80:1) to give 5-(N-methyl-N-oxiranylmethylamino)-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (C; 1.92 g) and 5-[N-methyl-N-(2-hydroxy-3-methoxypropyl)amino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (D; 0.38 g).

(C); Colorless needles (recrystallization from ethylacetate)

m.p. 239 - 240°C

(D); Colorless amorphous

1H-NMR(CDCl₃) & ; 1.35-4.55 (22H, m), 6.3-7.8 (13H, m)

Example 771

5-[N-Methyl-N-oxiranylmethylamino)-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.5 g) is dissolved in methanol (10 ml) and thereto is added

diethylamine (0.13 ml). The mixture is refluxed for 3 hours. After completion of the reaction, the solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 30 : 1 + dichloromethane: methanol : aqueous ammonia = 9 : 1 : 0.1) to give 5-[N-methyl-N-(2-hydroxy-3-diethylaminopropyl)amino]-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.38 g) as colorless amorphous.

 $l_{H-NMR(CDCl_3)}$ δ ; 1.10 (6H, t, J=7 Hz), 1.35-5.1 (23H, m), 6.55-7.8 (13H, m)

Example 772

A solution of 5-hydroxyimino-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.06 g) in acetic anhydride (10 ml) and pyridine (10 ml) is stirred at room temperature overnight. After completion of the reaction, the reaction solution is concentrated. To the resulting residue is added water and the mixture is extracted with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 80:1), and recrystallized from ethanol/petroleum ether to give 5-acetyloxyimino-1-[4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.75 g) as colorless prisms, m.p.

142 - 144°C.

Example 773

Using the suitable starting materials, the compounds of the above Examples 671 and 672 are obtained in the same manner as in Example 380.

Example 774

Using the suitable starting materials, the compounds of the above Examples 674, 699, 700, 706, 718 and 730 are obtained in the same manner as in Example 384.

Example 775

Using the suitable starting materials, the compounds of the above Examples 654 - 672, 675, 677 - 687, 691 - 695, 697, 698, 701 - 705, 707, 708, 712, 713, 715, 716 and 719 - 725 are obtained in the same manner as in Example 390.

Example 776

Using the suitable starting materials, the compounds of the above Examples 654 - 672, 675, 677 - 679, 691 - 693, 698, 701 - 705, 707, 708, 712, 713, 715, 716 and 719 - 725 are obtained in the same manner as in Example 388.

Example_777

Using the suitable starting materials, the compounds of the above Examples 705, 706 and 708 are obtained in the same manner as in Example 394.

Example 778

Using the suitable starting materials, the compound

of the above Example 671 is obtained in the same manner as in Example 397.

Example 779

Using the suitable starting materials, the compound of the above Example 672 is obtained in the same manner as in Example 402.

Example 780

Using the suitable starting materials, the compound of the above Example 726 is obtained in the same manner as in Example 634.

Example 781

Using the suitable starting materials, the compound of the above Example 740 is obtained in the same manner as in Examples 638 and 640.

Example 782

Using the suitable starting materials, the compound of the above Example 689 is obtained in the same manner as in Example 643.

Example 783

Using the suitable starting materials, the compound of the above Example 690 is obtained in the same manner as in Example 644.

Example 784

Using the suitable starting materials, the following compound is obtained in the same manner as in Examples 1, 382, 388 and 390.

5-Dimethylamino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine hydrochloride, colorless needles (recrystallized from ethanol/water), m.p. 233 - 237°C

Reference Example 13

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

5-(2-Chloroacetyloxy)-1-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 156 - 159°C (recrystallized from ethyl acetate/n-hexane)

5-(2-Dimethylaminoacetyloxy)-l-(4-nitrobenzoyl)2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 108 109°C (recrystallized from ethyl acetate/n-hexane)

5-Oxo-7-chloro-1-(4-nitrobenzoy1)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder, m.p. 157.5 - 159.5°C (recrystallized from diethyl ether/dichloromethane)

5-Oxo-8-chloro-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 151.5 - 153.5°C (recrystallized from diethyl ether/dichloromethane)

Reference Example 14

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

5-(2-Dimethylaminoacetyloxy)-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, colorless amorphous

Reference Example 15

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

5-Dimethylaminocarbonylmethoxy-l-(4-nitrobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 129 - 131°C (recrystallized from ethyl acetate/n-hexane)

6-Oxo-l-(4-nitrobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, yellow needles

1H-NMR (CDCl₃) & ; 1.65-2.3 (4H, m), 2.5-5.2 (4H,
m), 6.7-6.9 (1H, m), 7.27-7.5 (4H, m), 7.90-8.15 (3H, m)
6-Chloro-5-oxo-1-(4-nitrobenzoy1)-2,3,4,5tetrahydro-1H-benzazepine, white powder, m.p. 198 - 202°C
(recrystallized from dichloromethane/diethyl ether)

Reference Example 16

Using the suitable starting materials, the

following compounds are obtained in the same manner as in Reference Example 2.

6-0xo-l-(4-aminobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous

¹H-NMR (CDCl₃) δ; 1.7-2.2 (4H, m), 2.5-5.2 (6H, m), 6.42 (2H, d, J=8.7 Hz), 6.75-6.9 (1H, m), 7.05-7.4 (4H, m), 7.95-8.1 (1H, m)

6-Chloro-5-oxo-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder, m.p. 166 - 169°C (recrystallized from dichloromethane/diethyl ether)

9-Chloro-5-oxo-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-1H-benzazepine, yellow powder, m.p. 192.5 - 195°C (recrystallized from dichloromethane/diethyl ether)

Reference Example 17

5-Dimethylamino-1-(2-methyl-4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine (86.0 g) is dissolved in ethanol (800 ml), and thereto is added platinum oxide (10 g). The mixture is subjected to hydrogenation at ordinary temperature under atmospheric pressure of hydrogen for 4 hours. The catalyst is removed by filtration, and the

solvent is distilled off. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 200 : 1 + 100 : 1), and further purified by silica gel thin layer chromatography (developer; chloroform : methanol = 10 : 1), and recrystallized from methanol/diethyl ether to give 5-dimethylamino-1-(2-methyl-4-amino-benzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (G) (Rf: 0.52, 27.4 g) and 5-dimethylamino-1-(2-methyl-4-amino-benzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (H) (Rf: 0.48, 12.3 g).

(G): White powder

M.p. 154 - 156°C $[\alpha]_D^{22} = 0^\circ \text{ (c=1.0, chloroform)}$

 1 H-NMR (CDC1₃) δ ; 1.10-1.50 (1H, m), 1.50-2.00 (1H, m), 2.00-2.35 (11H, m), 2.90-5.18 (5H, m), 6.00-6.76 (3H, m), 6.81-7.64 (4H, m)

(H): White powder

M.p. $169.5 - 170^{\circ}C$ $[\alpha]_{D}^{22} = 0^{\circ} (c=1.5, chloroform)$

 $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.11 - 2.90 (13H, m), 2.91-5.23 (5H, m), 6.15-6.53 (1H, m), 6.57-7.62 (6H, m)

Using the suitable starting materials, the compounds of the following Table 5 are obtained in the same manner as in above Examples 1 and 382.

Table 5

Example 785

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 174 - 175°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N}^{1} \mathbb{N}^{1}

R². н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 176 - 178°C

Form: Free

Example 787

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \end{array} \right)$$

NH(CH₂)₂

к²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/petroleum ether

Melting Point: 154.5 - 155°C

Structure

CH₃-N-CH₂CHCH₂NHCH₃

ОН

$$\mathbb{R}^1$$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 138)

Form: Free

Example 789

Structure

CH₃-N-SO₂CH₃

$$\left(\begin{array}{c} W \\ N \end{array}\right) : \left(\begin{array}{c} W \\ N \end{array}\right)$$

R²: н

Crystalline form: Colorless scales

Recrystallization solvent: Ethanol

Melting Point: 197 - 198°C

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 248 - 249°C

Form: Free

Example 791

Structure

R². н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/n-hexane

Melting Point: 162 - 163°C

Form: Free

è

* G

Structure

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/petroleum ether

CH3

Melting Point: 235 - 236.5°C

Form: Free

Example 793

Structure

re
$$\begin{array}{c}
\text{NHCO}_2\text{C-CH}_3\\
\text{CH}_3
\end{array}$$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 139)

·			

Structure

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Dioxane

Melting Point: 269 - 271°C

Form: Free

Example 795

Structure

NHCONHCH₃

R²: H

Crystalline form: Colorless prisms

Recrystallization solvent: Dimethylformamide

Melting Point: 286 - 287°C .

Free Form:

Structure

re
$$CH_3-N-CH_2CN$$
 R^1

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Acetonitrile

Melting Point: 227 - 228°C

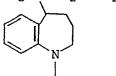
Form: Free

Example 797

Structure

ОН | | | СН₃-N-СН₂СНСН₂ОН

 $\left(\left(\right) \right)$



. . .

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 140)

Structure

$$\text{CH}_3$$
-N- CH_2 CO $_2$ C $_2$ H $_5$

к²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/petroleum ether

Melting Point: 167 - 168°C

Form: Free

Example 799

Structure

$$\mathbb{R}^{1}$$

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 141)

Structure

R²: Н

$$R^3: 4-NHC$$

Crystalline form: Colorless needles

Recrystallization solvent: Diethyl ether

Melting Point: 164 - 171°C

Form: K+

Example 801

Structure

$$\mathbb{R}^1$$
 :

Crystalline form: Colorless amorphous

NMR analysis: 142)

Structure

$$CH_3-N-(CH_2)_3OCOCH_3$$

≀²: н

Crystalline form: Colorless amorphous

NMR analysis: 143)

Form: Free

Example 803

Structure

$$CH_3-N-(CH_2)_3OH$$





к²: н

Crystalline form: Colorless amorphous

NMR analysis: 144)

		•		

Structure

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 145)

Form: Free

Example 805

Structure

R²: н

$$R^3: 4-NHC$$

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 207 - 208°C

Structure

$$\begin{array}{c}
\text{NH} \\
\text{NCH}_2 \\
\text{N}
\end{array}$$

$$\begin{array}{c}
\text{R}^2: \text{ If }
\end{array}$$

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 187 - 189°C

Form: Free

Example 807

Structure

 R^2 : H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 217 - 218°C

Form: Free

£

÷

Structure

NHCH2CH2CH3

$$\mathbb{Q}_{\mathbb{Q}}^{\mathbb{Q}}$$
 : $\mathbb{Q}_{\mathbb{Q}}^{\mathbb{Q}}$

R²: Н

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate

Melting Point: 170 - 171°C

Form: Free

Example 809

Structure

re N-OH

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 239.5 - 241°C

Structure

re
$$N-OCOCH_3$$
 R^1

 R^2 : H

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 190 - 191°C

Form: Free

Example 811

Structure

e $CH_3-N-CH_2CH_2CH_3$

O CH₃
R³: 4-NHC

Crystalline form: Colorless prisms

Recrystallization solvent: Diethyl ether

Melting Point: 163 - 163.5°C

Form: Free

4,14

Structure

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 208 - 210°C

Form: Free

Example 813

Structure

$$\begin{array}{c}
\text{NHCOCH}_2\text{Cl} \\
\\
\downarrow \\
\\
R^1
\end{array}$$

R²: н

Crystalline form: White powder

NMR analysis: 146)

Structure

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 147)

Form: Free

Example 815

Structure

$$\begin{array}{c}
\text{NHCOCH}_{2} \text{N} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{NHCOCH}_{2} \text{N} \\
\text{N} \\
\text{R}^{2} : \text{H}$$

R³: 4-NHC

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/n-hexane

Melting Point: 250 - 252°C

R²: Н

Example 816

Structure

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate

Melting Point: 214 - 216°C

Form: Free

Example 817

Structure

NHCOCH₂NHCH₃ \mathbb{R}^2 :

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/n-hexane

Melting Point: 243 - 245°C

 R^2 : H

Example 818

Structure

$$\begin{array}{c} & \text{CH}_3\text{-N-COCH}_2\text{N} \\ & & \\ &$$

Crystalline form: Colorless prisms

Recrystallization solvent: Diethyl ether

Melting Point: 159 - 162°C

Form: Free

Example 819

Structure

$$\begin{array}{c}
\text{re} \\
\text{W} \\
\text{R}^{1}
\end{array}$$
:
$$\begin{array}{c}
\text{CH}_{2}\text{CON} \\
\text{CH}_{3}
\end{array}$$

$$\text{R}^{2}: H$$

Crystalline form: Colorless amorphous

NMR analysis: 148)

Structure

NHCOCH₂NH₂

$$\mathbb{R}^1$$
 : \mathbb{N}

R²: Н

$$R^3: 4-NHC$$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate

Melting Point: 287 - 289°C

Form: Free

Example 821

Structure ***

 $\begin{array}{c} \text{CH}_3 \\ \text{NHCOCH}_2 \text{NHCO}_2 \text{C-CH}_3 \end{array}$

Crystalline form: Colorless prisms

Recrystallization solvent: Diethyl ether

Melting Point: 170 - 171°C

Structure

R²: н

Crystalline form: White powder

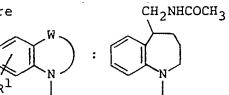
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 204 - 205°C

Form: Free

Example 823

Structure



R²: н

$$R^3$$
: 4-NHC- \mathbb{Z}

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 273 - 273.5°C

Form: Free

3

Structure

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 149)

Form: Free

Example 825

Structure

re CH_2NHCHO R^1 :

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethyl acatete/n-hexane

Melting Point: 240 - 241°C

 R^2 : H

Example 826

Structure

R³: 4-NHC

Crystalline form: White powder

Recrystallization solvent: Acetonitrile/ethanol

Melting Point: 231 - 232°C

Form: Free

Example 827

Structure

ococh₂N
$$CH_3$$
 $R^2: H$

Crystalline form: White powder

Recrystallization solvent: Acetonitrile/ethanol

Melting Point: 222 - 224°C

 R^2 : H

Example 828

Structure

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 235 - 237°C

Form: Free

Example 829

Structure

ore
$$OCH_2CON$$
 $S=O$ R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 150)

Structure

re
$$OCH_2CON$$
 S_{O} R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 151)

Form: Free

Example 831

Structure

ore
$$OCH_2CON$$
 $N-CH_3$ R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 152)

Structure

ore
$$OCH_2CON$$
 $NCOCH_3$

$$R^2: H$$

Crystalline form: Colorless amorphous

NMR analysis: 153)

Form: Free

Example 834

Structure

och₂con o
$$\mathbb{R}^2$$
:

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 247 - 248°C

H CONH₂

Structure

re
$$OCH_2CON$$
 R^2 : H

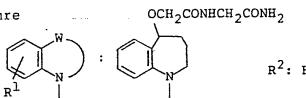
Crystalline form: Colorless amorphous

NMR analysis: 154)

Form: Free

Example 836

Structure



Crystalline form: Colorless amorphous

NMR analysis: 155)

Structure

och₂con
$$CH_3$$
 CH_2CH_2OH
 $R^2: R$

Crystalline form: Colorless amorphous

NMR analysis: 156)

Form: Free

Example 838

Structure

ore
$$OCH_2CONHCH_2 \leftarrow N$$

$$R^2: F$$

Crystalline form: Colorless amorphous

NMR analysis: 157)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{N} \mathbb{N}

 R^2 : H

Crystalline form: White powder

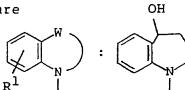
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 234 - 235°C

Form: Free

Example 840

Structure



г²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 234 - 235°C

Form: Free

•

.

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

Crystalline form: White powder

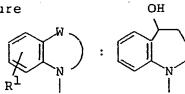
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 226 - 228°C

Form: Free

Example 842

Structure



 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 230 - 231°C

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 186 - 188°C

Form: Free

Example 844

Structure

re
$$CH_3-N$$
 R^1 R^1

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Chloroform/methanol

Melting Point: 286 - 290°C

Form: Free

•

ŧ.

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 186 - 188.5°C

Form: Free

Example 846

Structure

$$\begin{array}{c}
\text{OH} \\
\text{CH}_{3}
\end{array}$$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 220 - 222°C

Structure

R². F

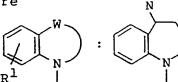
Crystalline form: White powder

NMR analysis: 158)

Form: Free

Example 848

Structure



Ř²: H

R³: 4-NHCOCH₂CONH₂

Crystalline form: Colorless prisms

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 189 - 192°C

Form: Free

3 6

€; (4

Structure

$$\mathbb{R}^1$$
 \mathbb{N} \mathbb{N} \mathbb{N}

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 159)

Form: Free

Example 850

Structure

$$\begin{array}{c}
\text{CH}_{2} \\
\text{R}^{1}
\end{array}$$
:

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 207 - 209°C (decomposed)

Structure

 $R^2: 2-C1$

Crystalline form: White powder

NMR analysis: 160)

Form:

Example 852

Structure

$$\mathbb{R}^1$$
 :

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 193 - 194°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 185.5 - 186°C

Form: Free

Example 854

Structure

$$\mathbb{R}^1$$
 \mathbb{C}^1 \mathbb{C}^N \mathbb{C}^N

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 223.5 - 226°C (decomposed)

Structure

R²: H

Crystalline form: Colorless amorphous

NMR analysis: 161)

Form: Free

Example 856

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 225.5 - 227°C

Form: Free

3

٠

Structure

R²: Н

$$R^3: 4-NHC$$

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 212 - 214°C

Form: Free

Example 858

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 230.5 - 233°C

Structure

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 212.5 - 215°C (decomposed)

Form: Free

Example 860

Structure

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 192 - 194.5°C

Form:

Free

Structure

$$\begin{array}{c}
\text{re} \\
\text{R}^{1} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{2} \\
\text{C1}
\end{array}$$

R²: н

$$R^3: 4-NHC \xrightarrow{CH_3}$$

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 175 - 177°C

Form: Free

Example 862

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1

в². н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 208.5 - 209.5°C

Structure

are
$$\frac{W}{R^1}$$
 : $\frac{W}{C1}$

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 191 - 193.5°C

Form: Free

Example 864

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 204 - 205.5°C

Structure

 R^2 : H

Crystalline form: Light yellow prisms

Recrystallization solvent: Ethanol

Melting Point: 221 - 223°C

Form: Free

Example 866

Structure

$$\mathbb{R}^1$$
 : $\mathbb{C}H_2$

R²: Н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate

Melting Point: 171 - 173°C

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{1}

 R^2 : H

$$R^3$$
: 4-NHC- \sim -CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate

Melting Point: 185 - 187°C

Form: Free

Example 868

Structure

 R^2 : H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 190 - 192°C

Form: Free

á

•

Structure

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 175- 177°C

Form: Free

Example 870

Structure

R²: н

$$R^3: 4-NHC-$$
 OCH₃

Crystalline form: Colorless powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 148 - 151°C

Structure

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 200 - 202°C

Form: Free

Example 872

Structure

 R^2 : H

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 200 - 202°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$
:

R²: H

Crystalline form: Light yellow powder

Recrystallization solvent: Acetone

Melting Point: 235 - 238°C

Form: Free

Example 874

Structure

R²: н

_Crystalline_form:_Light yellow powder

Recrystallization solvent: Acetone

Melting Point: 198 - 201°C

Structure

R²: н

Crystalline form: Light yellow needles

Recrystallization solvent: Chloroform/ethyl acetate

Melting Point: 232 - 237°C

Form: Free

Example 876

Structure

$$\mathbb{R}^{1}$$
 :

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Chloroform/ethyl acetate

Melting Point: 224 - 227°C

Form: Free

•

Structure

$$\mathbb{R}^1$$
 : \mathbb{R}^N :

R²: Н

$$R^3: 4-NHC-$$
NH2

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 211 - 214°C

Form: Free

Example 878

Structure

R.²: H

Crystalline form: Colorless powder

Recrystallization solvent: Dichloromethane/n-hexane

Melting Point: 238 - 243°C

Structure

$$\mathbb{R}^1$$
 : $\mathbb{C}^{\mathbb{N}}$:

R²: н

$$CH_3$$
O
CH₃
 CH_3
 R^3 : 4-NHC-

Crystalline form: Colorless amorphous

NMR analysis: 162)

Form: Free

Example 880

Structure

re
$$\mathbb{C}^{\mathbb{N}}$$
 : $\mathbb{C}^{\mathbb{N}}$

R²: н

$$R^3: 4-NHC \longrightarrow N CH_3$$

Crystalline form: Colorless amorphous

NMR analysis: 163)

Form: Free

⊒ Æ

*

Structure

R²: H

$$R^3$$
: 4-NHC- N -NCH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 198 - 202°C

Form: Free

Example 882

Structure

 R^2 : H

...Crystalline form: Colorless prisms

Recrystallization solvent: Chloroform/ethyl acetate

Melting Point: 226 - 229°C

Structure

R²: 2-СН₃

$$R^3: 4-NHC-CH_3$$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 139 - 140°C

Form: Free

Example 884

Structure

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 149 - 152°C

Structure

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 180.5 - 182°C

Form: Free

Example 886

Structure

re
$$\frac{NH-CH_3}{R^1}$$

 R^2 : 2-CH₃

-- Crystalline form: White powder --

Recrystallization solvent: Chloroform/diethyl ether

Melting Point: 211 - 214°C

Structure

$$\begin{array}{c}
\text{NH} \\
\text{R}^{1} \\
\text{R}^{1}
\end{array}$$

 R^2 : 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Chloroform/diethyl ether

Melting Point: 171 - 174.5°C

Form: Free

Example 888

Structure

R²: н

$$R^3$$
: 4-NHC-OCH₂CH₃

Crystalline form: White powder ----

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 203 - 205°C

Structure

R²: 3-OCH₃

$$R^3$$
: 4-NHC-OCH₃-OCH₂CH₃

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 202 - 202.5

Form: Free

Example 890

Structure

 R^2 : 3-OCH₂CONH₂

Crystalline-form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 130 - 133°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 104.5 - 106°C

Form: Free

Example 892

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 197 - 198°C

Form: Free

•

Structure

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/ethyl acetate

Melting Point: 191 - 192°C

Form: Free

Example 894

Structure

 R^2 : H

Crystalline form: Colorless columnar

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 211 - 213°C

Structure

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 164)

Form: Free

Example 896

Structure

к²: н

O || R³: 4-NHCCH₂CO₂C₂H₅

Crystalline form: Colorless amorphous

NMR analysis: 165)

Form: Free

*

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{R}^{N} : $\mathbb{C}\mathbb{H}_{3}$

R²: н

$$R^3$$
: 4-NHC-NHCOCH₃

Crystalline form: Colorless amorphous

NMR analysis: 166)

Form: Free

Example 898

Structure

$$\mathbb{R}^1$$
 : \mathbb{C}^{N}

 R^2 : H

$$R^3$$
: 4-NHC- \sim -CO₂CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 224 - 228°C

- 139) ${}^{1}H-NMR$ (CDCl₃) δ ; 1.1-2.3 (13H, m), 2.65-3.2 (1H, m), 4.55-5.6 (3H, m), 6.55-6.7 (1H, m), 6.9-7.6 (12H, m)
- 140) 1 H-NMR (CDCl₃) & ; 1.3-4.15 (19H, m), 4.3-5.0 (1H, m), 6.65 (1H, d, J=7.7 Hz), 6.9-8.05 (12H, m)
- 142) $^{1}H-NMR$ (CDCl₃) δ ; 1.3-2.85 (21H, m), 3.2-4.0 (4H, m), 4.3-4.4 (1H, m), 4.45-5.2 (2H, m), 6.61 (1H, d, J=7.6 Hz), 6.9-7.65 (12H, m)
- 143) $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.3-3.45 (17H, m), 3.8-5.7 (5H, m), 6.5-7.65 (13H, m)
- ¹H-NMR (CDCl₃) δ; 0.9-3.25 (16H, m), 3.9-5.9 (2H, m), 6.65 (1H, d, J=7.4 Hz), 6.85-7.5 (11H, m), 7.9-8.3 (1H, m)

- d, J=7.6 Hz), 6.9-7.45 (9H, m), 7.52 (2H, d, J=8.6 Hz), 8.9-9.05 (1H, m), 10.31 (1H, s)
- 147)

 1H-NMR (CDCl₃) 6; 1.5-2.35 (4H, m), 2.45 (3H, s),
 2.6-2.85 (1H, m), 3.32 (3H, s), 4.19 (2H, AB-q,
 J=12.2 Hz, 15.6 Hz), 5.0-5.2 (1H, m), 5.82 (1H, d,
 J=10.3 Hz), 6.69 (1H, d, J=7.8 Hz), 6.75-7.95 (12H, m)
- 148) 1 H-NMR (CDCl₃) $_{\delta}$; 1.2-3.3 (17H, m), 3.45 (2H, AB-q, J=14.7, 22.9 Hz), 3.9-4.35 (2H, m), 6.60 (2H, d, J=7.7 Hz), 6.8-8.0 (11H, m), 8.39 (1H, s)
- 149)

 ¹H-NMR (CDCl₃) δ; 1.45-3.40 (8H, m), 2.23 (3H, s),
 2.33 (3H, s), 2.46 (3H, s), 4.44-5.23 (1H, m),
 6.54-6.78 (1H, m), 6.84-7.94 (12H, m)
- 150) ¹H-NMR (CDCl₃) 6; 1.50-1.92 (3H, m), 1.92-2.05 (1H, m), 2.47 (3H, s), 2.55-3.06 (5H, m), 3.43-5.76 (8H, m), 6.63-6.82 (1H, m), 6.97-8.08 (12H, m)
- 151) ¹H-NMR (CDCl₃) δ; 1.43-2.65 (4H, m), 2.48 (3H, s), 2.69-3.25 (5H, m), 3.90-5.40 (8H, m), 6.64-6.94 (1H, m), 6.94-7.77 (12H, m)

```
4.50-5.20 (2H, m), 6.60-6.80 (1H, m), 6.94-7.64
(11H, m), 8.16 (1H, brs)

1H-NMR (CDCl<sub>3</sub>) δ; 1.48-2.60 (8H, m), 2.46 (3H, s),
2.65-3.01 (1H, m), 3.20-3.74 (2H, m), 3.80-5.14
(4H, m), 5.30-5.84 (1H, m), 6.51-8.14 (13H, m)

1H-NMR (CDCl<sub>3</sub>) δ; 1.54-1.91 (2H, m), 1.91-2.20
(1H, m), 2.22-2.64 (1H, m), 2.44 (3H, s), 2.70-3.13
(1H, m), 3.60-4.40 (4H, m), 4.50-5.20 (2H, m),
6.07-8.00 (13H, m), 9.93 (1H, s)

1H-NMR (CDCl<sub>3</sub>) δ; 1.56-1.92 (2H, m), 1.92-2.19 (1H, m), 2.19-2.60 (1H, m), 2.46 (3H, s), 2.66-3.26 (4H,
```

157)

¹H-NMR (CDCl₃) δ; 1.57-2.17 (3H, m), 2.21-2.68
(1H, m), 2.47 (3H, s), 2.73-3.04 (1H, m), 3.91-4.42
(4H, m), 4.50-5.17 (2H, m), 6.61-6.99 (2H, m),
6.99-8.10 (14H, m), 8.21-8.71 (2H, m)

(1H, m), 6.93-8.21 (12H, m)

m), 3.33-3.95 (4H, m). 4.00-5.20 (4H, m), 6.58-6.82

- 158)

 1 H-NMR (CDCl₃) 6; 1.31 (3H, d, J=6.7 Hz), 1.53
 1.90 (1H, m), 2.29-2.58 (1H, m), 2.47 (3H, s),

 2.94-3.63 (2H, m), 4.57-5.05 (1H, m), 6.68-6.82

 (1H, m), 7.10-7.59 (10H, m), 7.72 (1H, s), 7.78
 7.96 (1H, m)
- 160) 1 H-NMR (DMSO-d₆) $_{6}$; 1.40-1.75 (1H, m), 1.90-2.15 (1H, m), 2.33 (3H, s), 2.50-2.80 (2H, m), 3.10-3.50

```
(1H, m), 4.40-4.65 (1H, m), 6.85-7.60 (10H, m), 7.85 (1H, s), 10.44 (1H, s)
```

- 161) ${}^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.30-2.70 (11H, m), 3.00-5.20 (3H, m), 6.58 (1H, d, J=8 Hz), 6.90-7.05 (1H, m), 7.10-7.70 (10H, m)
- 163) ¹H-NMR (CDCl₃) δ; 1.25-3.00 (4H, m), 2.42 (6H, s), 2.99 (6H, s), 3.40-3.65 (2H, m), 4.01-5.15 (1H, m), 6.58-7.59 (12H, m), 7.94 (1H, brs)

To a solution of 5-acetyloxyimino-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

(0.48 g) in acetic acid (20 ml) is added platinum oxide

WO 91/05549 - 638 - PCT/JP90/01340

(0.05 g) and the mixture is subjected to catalytic reduction under hydrogen atmosphere. After completion of the reaction, the catalyst is removed by filtration, and the filtrate is concentrated. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 20 : 1 + 10 :1), and recrystallized from ethanol/diethyl ether to give 5-amino-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.19 g) as colorless prisms, m.p. 176 - 178°C.

Example 900

198°C.

To a solution of 5-methylamino-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) in dichloromethane (10 ml) is added triethylamine (0.24 ml). Subsequently, thereto is added methanesulfonyl chloride (0.14 ml) under ice-cooling, and then, the mixture is warmed to room temperature and stirred overnight. Water is added to the reaction solution, extracted three times with dichloromethane. The extract is washed with saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 20:1), and recrystallized from ethanol to give 5-(N-methyl-N-methanesulfonylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.48 g) as colorless scales, m.p. 197 -

To a solution of 5-methylamino-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) in dichloromethane is added triethylamine (0.24 ml). Subsequently, thereto is added benzoyl chloride (0.2 ml) under ice-cooling, and the temperature thereof is raised to room temperature, and the mixture is stirred overnight. Water is added to the reaction solution and extracted three times with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 20 : 1), and recrystallized from ethanol to give 5-(N-methyl-N-benzoylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5tetrahydro-1H-benzazepine (0.64 g) as colorless needles, m.p. 248 - 249°C.

Example 902

A mixture of 5-amino-l-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) and ethyl formate (10 ml) is refluxed for 4 hours. The reaction solution is concentrated and the resulting residue is recrystallized from ethanol/petroleum ether to give 5-formylamino-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.38 g) as colorless columnar crystal, m.p. 211 - 213°C.

Using the suitable starting materials, the compounds of above Examples 825 and 894 are obtained in the same manner as in above Example 902.

Example 903

To a solution of 5-amino-l-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) in dichloromethane (10 ml) is added triethylamine (0.22 ml). Subsequently, thereto is added di-tert-butyl dicarbonate (0.34 g) at room temperature and the mixture is stirred for 2 hours. Then, thereto is added additional di-tert-butyl dicarbonate (0.1 g) and the mixture is stirred for 1 hour. The reaction mixture is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; n-hexane: ethyl aceate = 1:1) to give 5-t-butoxycarbonylamino-l-[4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.66 g) as colorless amorphous.

¹H-NMR (CDCl₃) δ ; 1.1-2.3 (13H, m), 2.65-3.2 (1H, m), 4.55-5.6 (3H, m), 6.55-6.7 (1H, m), 6.9-7.6 (12H, m)

Using the suitable starting materials, the compound of above Example 791 is obtained in the same manner as in above Example 903.

Example 904

To a solution of 5-amino-l-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) in dichloromethane (10 ml) is added phenyl isocyanate (0.2 g)

under ice-cooling. The mixture is stirred at the same temperature for 30 minutes, and the temperature thereof is raised to room temperature and then the mixture is stirred overnight. The reaction solution is distilled off and the resulting residue is recrystallized from dioxane to give 5-anilinocarbonylamino-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.65 g) as colorless prisms, m.p. 269 - 271°C.

Using the suitable starting materials, the compound of above Example 795 is obtained in the same manner as in above Example 904.

Example 905

To a solution of 5-methylamino-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzozepine (0.6 g) in methanol (10 ml) is added glycolonitrile (50 %, 0.19 ml) and the mixture is stirred at room temperature for 20 minutes, and then refluxed for 30 minutes. Thereto is added additional glycolonitrile (0.5 ml) and the mixture is refluxed for 5.5 hours. The reaction solution is concentrated and to the resulting residue is added ethyl acetate. The precipitated crystal is collected by filtration, and recrystallized from acetonitrile to give 5-(N-methyl-N-cyanomethylamino)-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.32 g) as colorless needles, m.p. 227 - 228°C.

Example 906

To 5-(N-methyl-N-oxiranylmethylamino)-1-[4-(2methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine (0.62 g) is added trifluoroacetic acid (1.22 ml) under ice-cooling and the mixture is stirred for 4 hours. The reaction solution is neutralized with aqueous sodium carbonate solution, and extracted three times with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is dissolved in methanol (10 ml). Thereto is added 40 % aqueous sodium hydroxide solution (10 ml) and water (10 ml), and the mixture is stirred at room temperature overnight. Methanol is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 30:1) to give 5-[N-methyl-N-(2,3-dihydroxypropyl)amino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.23 g) as colorless amorphous.

 1 H-NMR (CDCl₃) $_{\delta}$; 1.3-4.15 (19H, m), 4.3-5.0 (1H, m), 6.65 (1H, d, J=7.7 Hz), 6.9-8.05 (12H, m)

Example 907

A mixture of 5-methylamino-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (1.64 g), acetonitrile (20 ml), potassium carbonate (0.6 g) and ethyl bromoacetate (0.44 ml) is refluxed for 3 hours. The reaction solution is concentrated and water is added to the

resulting residue, and the mixture is extracted three times with dichloromethane. The extract is washed with saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 30:1), and recrystallized from ethyl acetate/petroleum ether to give 5-(N-methyl-N-ethoxycarbonylmethylamino)-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.82 g) as colorless prisms, m.p. 167 - 168°C.

Using the suitable starting materials, the compounds of above Examples 785, 787, 799, 800, 802 - 806, 808, 811, 819, 824, 826, 827, 845, 848, 849, 850, 852, 855 - 858, 860, 861, 863 - 882, 885 - 893 and 895 - 898 are obtained in the same manner as in above Example 907.

Example 908

5-(N-Methyl-N-ethoxycarbonylmethylamino)-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g) is dissolved in saturated solution of ammonia in methanol (20 ml), and the mixture is heated at 100°C for 8 hours in a sealed tube. The reaction solution is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 30: l) to give 5-(N-methyl-N-carbamoylmethyl-amino)-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.4 g) as colorless amorphous.

¹H-NMR (CDCl₃) δ; 1.4-3.0 (9H, m), 3.05-3.6 (3H, m), 3.9-4.1 (1H, m), 4.35-4.55 (1H, m), 4.9-5.65 (1H, m), 6.67 (1H, d, J=7.4 Hz), 6.85-7.6 (12H, m), 7.6-7.85 (2H, m)

Example 909

To a solution of 5-(N-methyl-N-ethoxycarbonylmethylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine (0.6 g) in dioxane (10 ml) is added aqueous solution (1 ml) of sodium hydroxide (0.07 g) and the mixture is stirred at room temperature for 2 days. The reaction solution is concentrated and to the resulting residue is added water. The insoluble materials are removed by filtration. The filtrate is neutralized with 10 % hydrochloric acid and extracted three times with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. The solvent is distilled off and to the resulting residue is added a solution of potassium ethylhexanoate (0.2 q) in dichloromethane (20 ml). The solvent is distilled off, and diethyl ether is added to the resulting residue. precipitated crystal is collected by filtration, and recrystallized from diethyl ether to give potassium 2-[N $methyl-N-\{1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5$ tetrahydro-1H-benzazepin-5-yl}amino]acetate (0.6 g) as colorless needles, m.p. 164 - 171°C.

Example 910

To a solution of 5-methylamino-1-[4-(2-methyl-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (1.5 g) in dimethyformamide (20 ml) are added potassium carbonate (0.6 g), potassium iodide (0.72 g) and 2-(3-bromopropyloxy)-3,4,5,6-tetrahydro-2H-pyrane (0.97 g) and the mixture is stirred at room temperature overnight. The reaction solution is concentrated and to the resulting residue is added water. The mixture is extracted three times with dichloromethane. The extract is washed with saturated saline solution, and dried over magnesium sulfate. solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane + dichloromethane : methanol = 50 : 1) to give 5-{N-methyl-N-[3-(3,4,5,6-tetrahydro-2H-pyran-2yloxy)propyl amino}-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.3 g) as colorless amorphous.

 1 H-NMR (CDCl₃) ; 1.3-2.85 (21H, m), 3.2-4.0 (4H, m), 4.3-4.4 (1H, m), 4.45-5.2 (2H, m), 6.61 (1H, d, J=7.6 Hz), 6.9-7.65 (12H, m)

Example 911

To 5-{N-methyl-N-[3-(3,4,5,6-tetrahydro-2H-pyran-2-yloxy)propyl}amino}-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.4 g) is added a mixture of acetyl chloride (0.5 ml) and acetic acid (5 ml) at room temperature, and the mixture is stirred overnight. The reaction solution is concentrated and the resulting residue

is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 30 : 1), and further purified again by silica gel column chromatography (eluent; n-hexane : ethyl acetate = 1 : 2) to give 5-[N-methyl-N-(3acetyloxypropyl)amino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.06 g) as colorless amorphous.

 1 H-NMR (CDCl₃) δ ; 1.3-3.45 (17H, m), 3.8-5.7 (5H, m), 6.5-7.65 (13H, m)

Example 912

To a solution of $5-\{N-\text{methyl-N-}[3-(3,4,5,6$ tetrahydro-2H-pyran-2-yloxy)propyl]amino}-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.55 g) in ethanol (10 ml) is added pyridinium ptoluenesulfonate (0.03 g) and the mixture is heated at 60°C overnight. After the mixture is refluxed for more 2 hours, water and pyridinium p-toluenesulfonate (0.03 g) are added thereto. The mixture is refluxed for 4 hours. The reaction solution is concentrated and to the resulting residue is added dichloromethane. The mixture is basified with aqueous sodium hydrogen carbonate solution and extracted three times with dichloromethane. The extract is washed with saturated saline solution and dried over magnesium sulfate. solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane : methanol = 30 : 1) to give 5-[N-methyl-N-methy

(3-hydroxypropyl)amino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.26 g) as colorless amorphous.

 1 H-NMR (CDCl₃) 6 ; 1.25-3.1 (14H, m), 3.3-4.0 (4H, m), 4.15-4.4 (1H, m), 4.45-5.2 (1H, m), 6.64 (1H, d, J=7.4 Hz), 6.9-7.7 (12H, m)

Example 913

To a solution of 5-amino-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.6 g) in acetic acid (10 ml) is added dropwise 2,5-dimethoxytetra-hydrofuran (0.19 ml), and the mixture is refluxed for 1 hour. The reaction solution is concentrated and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50:

1), and recrystallized from ethyl acetate/n-hexane to give 5-(1-pyrrolyl)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.31 g) as colorless prisms, m.p. 208 - 210°C.

Example 914

To a solution of 5-amino-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (2.5 g) in dichloromethane (30 ml) is added triethylamine (0.96 ml) and further thereto is added dropwise chloroacetyl chloride (0.55 ml) under ice-cooling. The mixture is stirred for 5 minutes. The reaction solution is concentrated and to the resulting residue is added water. The precipitated crystal

is collected by filtration, washed with water, and dried to give 5-(2-chloroacetylamino)-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.4 g) as white powder.

 1 H-NMR (DMSO-d₆) $_{6}$; 1.3-2.15 (4H, m), 2.32 (3H, s), 2.8-3.05 (1H, m), 4.24 (2H, AB-q, J=12.8, 15.4 Hz), 4.35-4.55 (1H, m), 4.9-5.25 (1H, m), 6.68 (1H, d, J=7.6 Hz), 6.9-7.45 (9H, m), 7.52 (2H, d, J=8.6 Hz), 8.9-9.05 (1H, m), 10.31 (1H, s)

Using the suitable starting materials, the compound of above Example 814 is obtained in the same manner as in above Example 914.

Example 915

A mixed solution of 5-(2-chloroacetylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.6 g), imidazole (0.1 g) and potassium carbonate (0.19 g) in acetonitrile (30 ml) is refluxed for 8 hours. The reaction solution is concentrated and the resulting residue is washed with water and separated by decantation. The remainder is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 20:1 + 15:1), and recrystallized from ethanol/n-hexane to give 5-[2-(1-imidazolyl)acetylamino]-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.15 g) as colorless needles, m.p. 250 - 252°C.

Using the suitable starting materials, the compound

of above Example 818 is obtained in the same manner as in above Example 915.

Example 916

To a solution of 5-(2-chloroacetylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.6 g) in dimethylformamide (20 ml) are added dimethylamine hydrochloride (0.2l g) and potassium carbonate (0.54 g), and the mixture is stirred at room temperature for 2 days. The reaction solution is concentrated and water is added to the resulting residue. The precipitated crystal is collected by filtration, and recrystallized from ethyl acetate to give 5-(2-dimethylaminoacetylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.24 g) as colorless prisms, m.p. 214 - 216°C.

Using the suitable starting materials, the compounds of above Examples 816, 817, 820, 821, 826 and 827 are obtained in the same manner as above Example 916.

Example 917

A mixture of 5-methylamino-l-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (0.6 g), N,N-dimethyl-2-chloroacetamide (0.19 g) and potassium carbonate (0.22 g) is refluxed for 24 hours. The reaction solution is concentrated and water is added to the resulting residue. The mixture is extracted three times with dichloromethane. The extract is washed with saturated

saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 30:1) to give 5-[N-methyl-N-(dimethylaminocarbonylmethyl)amino]-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.05 g) as colorless amorphous.

¹H-NMR (CDCl₃) 6; 1.2-3.3 (17H, m), 3.45 (2H, AB-q, J=14.7, 22.9 Hz), 3.9-4.35 (2H, m), 6.60 (1H, d, J=7.7 Hz), 5.8-8.0 (11H, m), 8.39 (1H, s)

Example 918

To a solution of t-butoxycarbonylglycine (0.84 g) in dimethylformamide (20 ml) are added diethyl cyanophosphate (0.73 ml) and 5-amino-l-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (1.74 g), and further thereto is added triethylamine (1.8 ml) under ice-cooling. The mixture is stirred for 30 minutes, and then stirred at room temperature overnight. The reaction solution is concentrated and water is added to the resulting residue. The precipitated crystal is collected by filtration, washed with water, and recrystallized from ethyl acetate to give 5-(2-aminoacetylamino)-l-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (E) (0.16 g). Separately, the filtrate is concentrated and purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50:1), and recrystallized

from diethyl ether to give 5-[2-(t-butoxycarbonylamino)-acetylamino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (F) (0.19 g).

(E): Colorless prisms, m.p. 287 - 289°C

(F): Colorless prisms, m.p. 170 - 171°C

Example 919

5-Oxo-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.50 g) is suspended in tetrahydrofuran (20 ml), and thereto is added dropwise a 3.0 M solution of methyl magnesium bromide in diethyl ether (1.5 ml) at room temperature. The mixture is stirred at room temperature for 1 hour. The reaction solution is poured into ice-water (20 ml), and extracted with ethyl acetate. The extract is dried over magnesium sulfate, and the solvent is distilled off. The resulting residue is purified by silica gel column chromatography (eluent; ethyl acetate: n-hexane = 2:3+1:1), and recrystallized from ethyl acetate/n-hexane to give 5-methyl-5-hydroxy-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.23 g) as white powder, m.p. 204 - 205°C.

Example 920

To a solution of 5-carboxymethoxy-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (1.50 g) in dimethylformamide (60 ml) are added successively thiomorpholine (0.66 ml), diethyl cyanophosphate (0.89 g) and triethylamine (1.37 ml) with stirring under ice-

cooling. The mixture is stirred for 30 minutes under icecooling, and at room temperature for 20 minutes. Water (60
ml) is added to the reaction solution, and extracted with
dichloromethane. The extract is dried over magnesium
sulfate, and the solvent is distilled off. The resulting
residue is purified by silica gel column chromatography
(eluent; ethyl acetate: n-hexane = 5: 2 + 3: 1), and
recrystallized from ethyl acetate/n-hexane to give 5(thiomorpholinocarbonylmethoxy)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (1.60 g) as white
powder, m.p. 235 - 237°C.

Using the suitable starting materials, the compounds of above Examples 829 - 838 are obtained in the same manner as in above Example 920.

Example 921

To a solution of 5-(thiomorpholinocarbonylmethoxy)1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine (0.40 g) in dichloromethane (40 ml) is added 80
% m-chloroperbenzoic acid (175 mg) with stirring at -8°C,
and the mixture is stirred at -8°C for 1 hour. To the
reaction solution is added 20 % aqueous sodium
hydrogensulfite solution (40 ml) and the mixture is stirred
at room temperature for 30 minutes. The dichloromethane
layer is collected, washed with saturated saline solution
and dried over magnesium sulfate. The solvent is distilled
off and the resulting residue is purified by silica gel

column chromatography (eluent; dichloromethane : methanol = 20 : 1) to give 5-[(1-oxothiomorpholino)carbonylmethoxy]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.32 g) as colorless amorphous.

¹H-NMR (CDCl₃) δ ; 1.50-1.92 (3H, m), 1.92-2.05 (1H, m), 2.47 (3H, s), 2.55-3.06 (5H, m), 3.43-5.76 (8H, m), 6.63-6.82 (1H, m), 6.97-8.08 (12H, m)

Example 922

To a solution of 5-(thiomorpholinocarbonylmethoxy)1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lHbenzazepine (0.40 g) in dichloromethane (40 ml) is added 80
% m-chloroperbenzoic acid (0.35 g), and the mixture is
stirred at room temperature for l hour. The reaction
solution is washed successively with an aqueous sodium
hydrogensulfite solution and saturated saline solution, and
dried over magnesium sulfate. The solvent is distilled off
to give 5-[(l,l-dioxothiomorpholino)carbonylmethoxy]-l-[4(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lHbenzazepine (0.41 g) as colorless amorphous.

¹H-NMR (CDCl₃) δ; 1.43-2.65 (4H, m), 2,48 (3H, s), 2.69-3.25 (5H, m), 3.90-5.40 (8H, m), 6.64-6.94 (1H, m), 6.94-7.77 (12H, m)

Example 923

To a solution of 5-oxo-1-[4-(2-hydroxybenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (400 mg) in acetone (20 ml) are added potassium carbonate (210 mg), potassium

iodide (250 mg) and 2-chloroacetamide (120 mg), and the mixture is refluxed for 2 hours. The insoluble materials are removed by filtration, and the filtrate is distilled off. Dichloromethane is added to the resulting residue, and the mixture is washed with saturated saline solution, and dried over magnesium sulfate. The solvent is distilled off and the resulting residue is recrystallized from ethyl acetate/n-hexane to give 5-oxo-l-[4-(2-carbamoylmethoxy-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine (436 mg) as white powder, m.p. 226 - 228°C.

Using the suitable starting materials, the compound of above Example 842 is obtained in the same manner as above Example 923.

Example 924

A mixture of 5-methylamino-4-hydroxy-1-[4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.13 g), ethyl a-bromoacetate (58 mg), diisopropylethylamine (49 mg) and acetonitrile (5 ml) is refluxed for 10 hours. Acetonitrile is distilled off under reduced pressure, and the resulting residue is dissolved in dichloromethane, washed with water, dried over magnesium sulfate, and distilled off under reduced pressure. The resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol = 50:1), and recrystallized from chloroform/methanol to give 7-[4-(2-chlorobenzoylamino)benzoyl]-1-methyl-1,2,3,4a,5,6,7,11b-

octahydro-3-oxo[l]benzazepino[4,5-b][l,4]oxazine (80 mg) as colorless prisms, m.p. 286 - 290°C.

Example 925

To a solution of 5-oxo-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (1 g) in methanol (20 ml) and dichloromethane (20 ml) is added hydroxylamine-O-sulfonic acid (0.28 g) with stirring at room temperature, and the mixture is stirred at the same temperature for 1 hour. Subsequently, to the reaction solution is added with stirring an aqueous solution of patassium carbonate (0.34 g) in water (1 ml) at room temperature, and the mixture is stirred at the same temperature for 2 hours. The precipitated crystal is removed by filtration, and the filtrate is concentrated under reduced pressure. The resulting residue is purified by silica gel column chromatography to give potassium $\{1-[2$ chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydrolH-benzazepin-5-yl}imino-O-sulfonate (0.4 g) as white powder.

 1 H-NMR (DMSO- 1 d) $_{6}$; 1.40-1.75 (1H, m), 1.90-2.15 (1H, m), 2.33 (3H, s), 2.50-2.80 (2H, m), 3.10-3.50 (1H, m), 4.40-4.65 (1H, m), 6.85-7.60 (10H, m), 7.85 (1H, s), 10.44 (1H, s)

Example 926

Using the suitable starting materials, the compounds of above Examples 841 - 843, 868 - 870, 888 and

889 are obtained in the same manner as in above Example 380.

Example 927

Using the suitable starting materials, the compounds of above Examples 876 - 878 are obtained in the same manner as in above Example 381.

Example 928

Using the suitable starting materials, the compounds of above Examples 840, 842 and 846 are obtained in the same manner as in above Example 384.

Example 929

Using the suitable starting materials, the compounds of above Examples 788 - 790, 796 - 804, 805, 808, 811, 814, 818, 819, 824, 826, 827, 837, 845, 848, 850, 852, 855, 856 - 858, 860, 861, 863 - 882, 885, 886, 888 - 893 and 895 - 898 are obtained in the same manner as in above Example 388.

Example 930

Using the suitable starting materials, the compound of above Example 848 is obtained in the same manner as in above Example 393.

Example 931

Using the suitable starting materials, the compounds of above Examples 841 and 842 are obtained in the same manner as in above Example 402.

Example 932

Using the suitable starting materials, the

*

compounds of above Examples 882 and 897 are obtained in the same manner as in above Example 403.

Example 933

Using the suitable starting materials, the compound of above Example 809 is obtained in the same manner as in above Example 634.

Example 934

Using the suitable starting materials, the compounds of above Examples 828 - 838 are obtained in the same manner as in above Example 640.

Example 935

Using the suitable starting materials, the compound of above Example 810 is obtained in the same manner as in above Example 772.

Example 936

Using the suitable starting materials, the compound of above Example 788 is obtained in the same manner as in above Example 771.

Example 937

Using the suitable starting materials, the compounds of above Examples 785, 787, 788 - 790, 796 - 805, 806, 807, 808, 811, 814, 818, 819, 845, 848, 849, 850, 852, 855, 856 - 858, 860, 861, 863 - 882, 885, 886, 888 - 893 and 896 - 898 are obtained in the same manner as in above Example 390.

Example 938

ere ere with the constitution of the erection of the continuous eneme.

To 5-methanesulfonyloxymethyl-1-[4-(2-methyl-benzoylamino)benozyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.50 g) is added a 30 % solution of methylamine in methanol (50 ml), and the mixture is heated at 100°C for 3 hours in a sealed tube. After cooling, the reaction solution is evaporated under reduced pressure, and the resulting residue is purified by silica gel column chromatography (eluent; dichloromethane: methanol: aqueous ammonia = 100:10:1) to give 5-methylaminomethyl-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.07 g).

¹H-NMR (DMSO-d₆) 6; 1.40-2.18 (4H, m), 2.34 (3H, s), 2.47 (3H, s), 2.54-3.50 (4H, m), 4.30-5.08 (1H, m), 6.56-6.82 (1H, m), 6.87-7.48 (10H, m), 7.48-7.75 (2H, m), 10.35 (1H, s)

Using the suitable starting materials, the compounds of above Examples 823 - 825 are obtained in the same manner as in above Example 938.

•

Using the above suitable starting materials, the compounds of the following Table 6 are obtained in the same manner as in Examples 1 and 382.

Table 6

$$\mathbb{R}^1$$
 $\mathbb{C}=0$
 \mathbb{R}^2

Example 939

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 208 - 211°C

gate, alle a transcription on the comment of galaxies and the comment of the comment of the compact

Structure

$$\mathbb{R}^{1} \qquad : \qquad \mathbb{R}^{1}$$

 \mathbb{R}^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 171.5 - 172.5°C

Form: Free

Example 941

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 151 - 154°C

Form: Free

and property and a second contracting the first of the first contracting the second contrac

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N} $\mathbb{C}H_{2}$

$$R^3$$
: 4-NHC- \mathbb{R}^3

Crystalline form: Colorless amorphous

NMR analysis: 167)

Form: Free

Example 943

Structure

$$R^3$$
: 4-NHC- \sim -O(CH₂)₃NHCOCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 180 - 183°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 109 - 110°C

Form: Free

Example 945

Structure

$$\mathbb{R}^{1}$$
 : \mathbb{N}' :

R²: н

Crystalline form: Colorless oil

NMR analysis: 168)

Structure

R²: H

$$R^3$$
: 4-NHCCH₂N- CH_3

Crystalline form: Colorless oil

NMR analysis: 169)

Form: Free

Example 947

Structure

г²: н

$$R^3$$
: 4-NHSO₂-

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 177 - 178.5°C

Structure

$$\begin{array}{c}
\text{re} \\
\text{CH}_{3} \\
\text{R}^{1}
\end{array}$$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 170)

Form: Free

Example 949

Structure

$$\mathbb{Q}_{\mathbb{R}^1} \mathbb{Q}_{\mathbb{R}^1} = \mathbb{Q}_{\mathbb{R}^1}$$

R²: Н

$$R^3$$
: 4-NHC- $(CH_2)_3NH_2$

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 162 - 165°C

Form: Free

•

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 212 - 215°C

Form: Free

Example 951

Structure

$$\mathbb{R}^{1} = \mathbb{N}$$
 :
$$\mathbb{R}^{N}$$

R²: н

$$R^3$$
: 4-NHCCH₂NH $-$ CH₃

Crystalline form: Colorless oil

NMR analysis: 171)

Structure

$$\mathbb{R}^{1} \longrightarrow \mathbb{R}^{N} \longrightarrow \mathbb{R}^{N}$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 112 - 114°C

Form: Free

Example 953

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: H

$$R^3$$
: 4-NHCCH₂N-CH₃

Crystalline form: Colorless oil

NMR analysis: 172)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{C} \mathbb{N} \mathbb{C} \mathbb{N}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 173)

Form: Free

Example 955

Structure

R²: н

$$R^3: 4-NHC \xrightarrow{CO_2C_2H_5}$$

Crystalline form: Light yellow amorphous

NMR analysis: 174)

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether

Melting Point: 189 - 193°C

Form: Free

Example 957

Structure

$$\mathbb{R}^1$$
 : \mathbb{N} CH

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 175)

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{CH}_{3}
\end{array}$$

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 234 - 238°C

Form: Free

Example 959

Structure

$$\begin{array}{c}
C1 & O \\
\downarrow \\
R^{1}
\end{array}$$

$$\begin{array}{c}
C1 & O \\
\downarrow \\
N
\end{array}$$

R²: н

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 183 - 184.5°C

Structure

$$\begin{array}{c}
\text{C1} & \text{O} \\
\text{R1} & \text{N}
\end{array}$$
:

R²: н

Crystalline form: Brown oil

NMR analysis: 176)

Form: Free

Example 961

Structure

$$\begin{array}{c}
\text{C1} & \text{NHCH}_3 \\
\text{R}^1 & | & & \\
\end{array}$$

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 177)

Form: Free

3

•

Structure

$$\begin{array}{c}
\text{Cl} & \text{N} \\
\text{CH}_{3}
\end{array}$$

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 178)

Form: Free

Example 963

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 202.5 - 204.5°C

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 199.5 - 201°C

Form: Free

Example 965

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 196.5 - 197°C

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 204 - 205°C

Form: Free

Example 967

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$
: C1 N CH

R²: H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 175 - 177°C

Structure

к²: н

Crystalline form: Pink amorphous

NMR analysis: 179)

Form: Free

Example 969

Structure

re
$$\frac{NHCH_3}{R^1}$$
 : $\frac{N}{C_1}$

к²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 186 - 189°C

Structure

re
$$\begin{pmatrix}
N \\
C_2H_5
\end{pmatrix}$$
 $\begin{pmatrix}
N \\
N
\end{pmatrix}$

 R^2 : 2-C1

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 211 - 212°C

Form: Free

Example 971

Structure

R²: E

$$R^3: 4-NHC$$

Crystalline form: Colorless amorphous

NMR analysis: 180)

Structure

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 206 - 207°C

Form: Free

Example 973

Structure

R²: E

Crystalline form: Colorless amorphous

NMR analysis: 181)

Form: Free

•

R²: н

R²: н

Example 974

Structure

O | C2H5

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 152 - 154°C

Form: Free

Example 975

Structure

Crystalline form: Colorless amorphous

NMR analysis: 182)

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 204 - 206°C

Form: Free

Example 977

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1}

_в2. н

$$^{O}_{R^3}: 4-NHC$$
 $^{O}_{I}$
 $^{O}_{A}$
 $^{O}_{A}$

Crystalline form: Colorless needles

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 162 - 163°C

- 167)

 ¹H-NMR (CDCl₃) δ; 1.14-2.83 (13H, m), 2.43 (3H, s), 2.95-5.19 (4H, m), 4.12 (2H, t, J=6.2 Hz), 6.27-6.83 (2H, m), 6.83-7.36 (6H, m), 7.36-7.67 (4H, m), 7.93-8.11 (1H, m), 9.77 (1H, brs)
- 168) 1 H-NMR (CDCl₃) $_{\delta}$; 1.11-2.98 (11H, m), 2.80 (3H, s), 3.69 (2H, s), 2.98-5.24 (2H, m), 6.50-7.71 (12H, m), 9.37 (1H, brs)
- 170) ¹H-NMR (CDCl₃) δ; 1.10-1.98 (3H, m), 1.98-2.82 (10H, m), 2.82-3.20 (2H, m), 3.34-5.15 (2H, m), 6.48-7.68 (15H, m), 7.86 (1H, brs)

- 173)

 ¹H-NMR (DMSO-d₆) δ; 1.28-2.62 (4H, m), 2.07 (3H, s), 2.34 (6H, s), 3.04-3.57 (2H, m), 3.99-4.86 (1H, m), 6.62-7.88 (12H, m), 10.12-10.20 (2H, m)
- 174)

 1H-NMR (CDCl₃) 6; 1.39 (3H, t, J=7.1 Hz), 1.64
 2.68 (4H, m), 2.42 (6H, s), 3.04-3.58 (2H, m),

 3.98-5.01 (1H, m), 4.38 (2H, q, J=7.1 Hz), 6.57
 8.57 (13H, m)

175)	¹ H-NMR (DMSO-d ₆) δ ; 1.67-5.02 (7H, π), 3.35 (6H,	
	s), 6.75-8.17 (12H, m), 8.46 (1H, s), 10.54 (1H, s)	
176)	$^{1}\text{H-NMR}$ (CDCl ₃) δ ; 1.21 (3H, t, J=7.1 Hz), 1.95-	3
	2.30 (2H, m), 2.88 (2H, t, J=6.2 Hz), 3.40-3.65	
	(2H, m), 3.70-4.50 (2H, m), 3.91 (2H, s), 6.66 (1H,	
	d, J=8.5 Hz), 6.70-7.00 (3H, m), 7.10-7.50 (7H, m),	
	7.81 (lH, d, J=2.5 Hz), 8.44 (lH, s)	
177)	1 H-NMR (CDCl ₃) δ ; 1.21 (3H, t, J=7 Hz), 1.30-5.20	
	(11H, m), 3.48 (2H, q, J=7 Hz), 3.90 (2H, s), 6.53	
	(1H, d, J=8.3 Hz), 6.65-7.00 (4H, m), 7.00-7.40	
	(6H, m), 7.51 (1H, d, J=2.5 Hz), 8.40 (1H, s)	
178)	$^{1}\text{H-NMR}$ (CDCl ₃) δ ; 1.21 (3H, t, J=7 Hz), 1.20-5.20	
	(15H, m), 3.90 (2H, s), 6.48 (1H, d, J=8.3 Hz),	
	6.50-7.70 (11H, m), 8.39 (1H, s)	
179)	1 H-NMR (CDCl ₃) δ ; 1.60-2.20 (1H, m), 2.10-2.35	
	(1H, m), 2.45 (3H, s), 2.70-2.95 (2H, m), 3.25-3.45	
	(1H, m), 4.60-4.85 (1H, m), 7.10-7.80 (12H, m)	
180)	¹ H-NMR (CDCl ₃) δ ; 1.65-2.15 (4H, m), 2.46 (3H, s),	
	2.6-5.15 (4H, m), 6.75-6.95 (1H, m), 7.15-7.55	
	(10H, m), 7.61 (1H, s), 7.95-8.1 (1H, m)	
181)	¹ H-NMR (CDCl ₃) δ ; 1.60-2.15 (3H, m), 2.15-2.90	
	(2H, m), 2.90-3.22 (6H, m), 4.00-4.50 (2H, m), 4.13	
	(2H, s), 4.58-5.22 (2H, m), 6.53-6.80 (1H, m),	₹
	6.90-7.90 (7H, m), 8.48 (1H, s)	4
182)	1 H-NMR (CDC1 ₃) δ ; 1.48-2.20 (3H, m), 2.20-2.85	

Committee and the committee of the second

(2H, m), 2.85-3.27 (6H, m), 4.05-4.47 (2H, m),

4.47-5.22 (2H, m), 6.50-6.76 (1H, m), 6.76-6.91 (1H, m), 6.91-7.69 (9H, m), 7.69-8.13 (1H, m), 9.28 (1H, s), 11.87 (1H, brs)

Example 978

5-Dimethylamino-1-(2-methyl-4-aminobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine (H) (1.00 g) is dissolved in dichloromethane (30 ml), and thereto is added triethyl-amine (0.48 ml) under ice-cooling, and further added dropwise 2-methylbenzoyl chloride (0.44 ml). The mixture is stirred at room temperature for 1 hour. The reaction solution is washed with water, and dried over magnesium sulfate. The solvent is distilled off, and the resulting residue is crystallized by adding thereto ethyl acetate. The precipitated crystal is recrystallized from dichloromethane/ethyl acetate to give 5-dimethylamino-1-[2-methyl-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.92 g) as white powder, m.p. 191 - 192°C.

HPLC retention time: 7.5 minutes

Column; Wakosil II 5C₁₈ (trade mark; Wako Pure Chemical Co., Ltd.)

Solvent; acetonitrile : 50 mN aqueous Na_2SO_4 solution : acetic acid = 27 : 73 : 1

Rate; 1.0 ml/min.

 $[\alpha]_D^{22} = 0^{\circ} \text{ (c=1.0, chloroform)}$

a para santi anti di la companya di kacamatan kana di kacamatan kana di kacamatan kana di kacamatan kana di ka

 1 H-NMR (CDCl₃) δ ; 1.15-3.25 (17H, m), 3.35-5.14

¥

(2H, m), 6.62-8.05 (12H, m)

Charts of ^1H-NMR (CDCl $_3$) of the starting compound (H) and the compound obtained in Exmaple 978 are shown in Fig. 1 and Fig. 2, respectively.

Example 979

Using 5-dimethylamino-1-(2-methyl-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine (G) (1.00 g), 5-dimethyl-amino-1-[2-methyl-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (0.48 g) is obtained in the same manner as in Example 978 except that methanol/diethyl ether is used instead of ethyl acetate as recrystallization solvent, as white powder, m.p. 183 - 185°C.

HPLC retention time : 8.1 minutes

(the conditions of HPLC are same as those in Example 978) $\left[\alpha\right]_{D}^{22} = 0^{\circ} \text{ (c=1.3, chloroform)}$

 1 H-NMR (CDCl₃) δ ; 1.10-3.20 (17H, m), 3.35-5.15 (2H, m), 6.50-6.80 (1H, m), 6.86-7.62 (10H, m), 7.65-8.09 (1H, m)

Charts of $^1\text{H-NMR}$ (CDCl $_3$) of the starting compound (G) and the compound obtained in Exmaple 979 are shown in Fig. 3 and Fig. 4, respectively.

Reference Example 18

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

and the first of the control of the

7-Methoxy-5-oxo-1-(4-nitrobenzoy1)-2,3,4,5-tetra-hydro-1H-benzazepine, colorless needles, m.p. 178 - 178.5°C (recrystallized from ethyl acetate/n-hexane)

7-Methoxy-5-oxo-1-(2-chloro-4-nitrobenzoy1)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 150 151°C (recrystallized from ethyl acetate/n-hexane)

2,3,4,5-tetrahydro-lH-benzazepine, white powder, m.p. 116 - 118°C (recrystallized from ethyl acetate/n-hexane)

7-Methoxy-5-oxo-1-(3-methoxy-4-nitrobenzoyl)-

7-Chloro-5-oxo-1-(3-methoxy-4-nitrobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, yellow powder, m.p. 156 158°C (recrystallized from diethyl ether/dichloromethane)

Reference Example 19

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

7-Methoxy-5-oxo-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-lH-benzazepine, white powder, m.p. 172.5 - 173.5°C (recrystallized from ethyl acetate/n-hexane)

7-Methoxy-5-oxo-1-(2-chloro-4-aminobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, white powder, m.p. 153 155°C (recrystallized from ethyl acetate/n-hexane)

7-Methoxy-5-oxo-1-(3-methoxy-4-aminobenzoyl)2,3,4,5-tetrahydro-1H-benzazepine, colorless needles, m.p.
170 - 171°C (recrystallized from ethyl acetate/n-hexane)
7-Chloro-5-oxo-1-(3-methoxy-4-aminobenzoyl)-

for state

2,3,4,5-tetrahydro-1H-benzazepine, yellow oil

¹H-NMR (CDCl₃) δ; 2.05-2.30 (2H, m), 2.85-3.00 (2H, m), 3.70 (3H, s), 3.85-4.30 (4H, m), 6.42 (1H, d, J=8.1 Hz), 6.64 (1H, dd, J=1.7 Hz, 8.1 Hz), 6.72 (1H, d, J=8.5 Hz), 6.80 (1H, d, J=1.8 Hz), 7.19 (1H, dd, J=2.6 Hz, 8.5 Hz), 7.81 (1H, d, J=2.5 Hz)

.

. .

Using the suitable starting materials, the compounds of the following Table 7 are obtained in the same manner as in above Examples 1 and 382.

Table 7

Example 980

Structure

$$\mathbb{R}^1$$
 : \mathbb{C}^{H_3}

к²: н

Crystalline form: Colorless amorphous

NMR analysis: 183)

Structure

$$\mathbb{R}^1$$
 : \mathbb{R}^N :

R²: Н

Crystalline form: Colorless amorphous

NMR analysis: 184)

Form: Free

Example 982

Structure

$$\mathbb{R}^{1}$$
 : $\mathbb{C}^{\mathbb{N}}$: $\mathbb{C}^{\mathbb{N}}$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 185)

Form: Free

e d

*

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^1
\end{array}$$

R²: н

$$R^3$$
: 4-NHC-CHNH-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 186)

Form: Free

Example 984

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^N : \mathbb{R}^N

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 187)

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 188)

Form: Free

Example 986

Structure

$$\mathbb{R}^1$$
 \mathbb{C}^{H_30} \mathbb{C}^{N}

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 189)

Form: Free

ī

· ·

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3}\text{O} \\
\text{O} \\
\text{N}
\end{array}$$

R²: Н

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

Melting Point: 267 - 268°C

Form: Free

Example 988

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Ethanol/water

Melting Point: 264 - 266°C

Structure

R²: н

Crystalline form: White powder

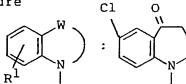
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 218 - 220°C

Form: Free

Example 990

Structure



 R^2 : 3-OCH₃

Crystalline form: Yellow oil

NMR analysis: 190)

Form: Free

2

Structure

$$\begin{array}{c}
C1 & O \\
C1 & O \\
N & O
\end{array}$$

R²: 3-осн₃

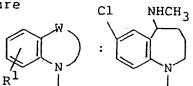
Crystalline form: Yellow oil

NMR analysis: 191)

Form: Free

Example 992

Structure



R²: 3-OCH₃

Crystalline form: Yellow powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 174 - 177°C

Structure

R²: 3-OCH₃

Crystalline form: Yellow amorphous

NMR analysis: 192)

Form: Free

Example 994

Structure

R²: 3-OCH

Crystalline form: Colorless amorphous

NMR analysis: 193)

Form: Free

.

?

Structure

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 163 - 165°C

Form: Free

Example 996

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 194)

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 195)

Form: Free

•

*

- 183) $^{1}\text{H-NMR}$ (CDCl₃) 5; 1.10-2.83 (11H, m), 2.96-5.21 (2H, m), 4.55 (2H, s), 6.48-7.72 (13H, m), 8.30 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.10-2.85 (11H, m), 1.58 (3H, d, J=6.8 Hz), 2.23 (3H, s), 2.95-5.19 (4H, m), 6.38-7.70 (12H, m), 8.69 (1H, brs)
- 185) $^{1}\text{H-NMR}$ (CDCl₃) & ; 1.10-2.85 (14H, m), 2.26 (3H, s), 2.96-5.19 (4H, m), 6.36-7.68 (12H, m), 8.72 (1H, brs)
- 186) $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.09-2.72 (11H, m), 1.53 (3H, d, J=6.9 Hz), 2.24 (3H, s), 2.93-5.21 (4H, m), 6.30-7.78 (12H, m), 8.76 (1H, brs)

- 189)

 ¹H-NMR (CDCl₃) δ; 1.64-2.43 (2H, m), 2.67-2.97

 (2H, m), 3.00-3.70 (1H, m), 3.77 (3H, s), 4.20-5.10

 (1H, m), 6.60-7.75 (10H, m), 8.51 (1H, s)
- 190)

 1H-NMR (CDCl₃) & ; 2.00-2.35 (2H, m), 2.49 (3H, s),
 2.89 (2H, t, J=6.2 Hz), 3.72 (3H, s), 3.40-4.80
 (2H, m), 6.74 (2H, d, J=8.5 Hz), 6.80-7.00 (2H, m),
 7.25-7.60 (5H, m), 7.80 (1H, d, J=2.6 Hz), 8.16
 (1H, s), 8.37 (1H, d, J=8.6 Hz)

٠

- 1 H-NMR (CDCl₃) δ ; 1.90-2.40 (2H, m), 2.90 (2H, t, 191) J=6.2 Hz), 3.75 (3H, s), 3.40-4.80 (2H, m), 6.74 (1H, d, J=8.5 Hz), 6.80-7.00 (2H, m), 7.10-7.50(4H, m), 7.73 (1H, dd, J=2.3 Hz, 6 Hz), 7.80 (1H, d, J=2.5 Hz), 8.38 (1H, d, J=8.8 Hz), 8.65 (1H, s)
- 1 H-NMR (CDCl₃) δ ; 1.10-2.10 (13H, m), 2.90-5.20 192) (6H, m), 6.56 (1H, d, J=8.4 Hz), 6.69 (1H, d, J=7)Hz), 6.85-7.70 (7H, m), 8.15 (1H, s), 8.31 (1H, d, J=8.4 Hz)
- 1 H-NMR (CDCl₃) δ ; 0.30-0.65 (4H, m), 1.20-2.50 193) (7H, m), 2.50 (3H, s), 3.10-5.20 (2H, m), 3.75 (3H, s), 6.60 (1H, d, J=8.3 Hz), 6.70-7.60 (8H, m), 8.14 (1H, s), 8.20-8.40 (1H, m)
- 1 H-NMR (CDCl₃) δ ; 0.80-2.50 (10H, m), 2.90-4.10 194) (6H, m), 6.50-7.80 (9H, m), 8.32 (1H, d, J=8 Hz),8.62 (lH, s)
- 1 H-NMR (CDCl₃) δ ; 0.30-0.65 (4H, m), 0.70-2.40 195) (6H, m), 2.60-5.20 (6H, m), 6.50-7.80 (9H, m), 8.30 (1H, d, J=8 Hz), 8.62 (1H, s)

Reference Example 20

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

7-Methyl-5-oxo-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-lH-benzazepine, white needles

 $^{1}\text{H-NMR}$ (CDCl₃) δ ; 2.20 (2H, brs), 2.32 (3H, s),

and the control of th

2.88 (2H, t, J=6.3 Hz), 3.40-4.79 (2H, m), 6.57 (1H, d, J=8.0 Hz), 7.04 (1H, d, J=7.7 Hz), 7.36 (2H, d, J=8.6 Hz), 7.62 (1H, d, J=1.7 Hz), 8.04 (2H, d, J=8.7 Hz)

7-Dimethylamino-5-oxo-1-(4-nitrobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, red brown prisms (recrystallized from dichloromethane/diethyl ether)

¹H-NMR (CDCl₃) δ; 1.75-2.47 (2H, m), 2.60-3.62, 4.51-4.92 (total 4H, m), 2.93 (6H, s), 6.46 (1H, dd, J=2.2 Hz, 7.0 Hz), 6.52 (1H, d, J=7.0 Hz), 7.33 (2H, d, J=7.0 Hz), 8.00 (2H, d, J=7.0 Hz)

7-Bromo-5-oxo-1-(4-nitrobenzoy1)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder (recrystallized from dichloromethane/diethyl ether), m.p. 177 - 182°C

7-Chloro-5-oxo-1-(2-methyl-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder (recrystallized from dichloromethane/diethyl ether)

¹H-NMR (CDCl₃) 6;1.78-2.37 (2H, m), 2.48 (3H, s), 2.88 (2H, t, J=6.1 Hz), 3.30-5.12 (2H, m), 6.47-6.82 (1H, m), 6.82-7.09 (1H, m), 7.09-7.27 (1H, m), 7.48-8.35 (3H, m)

6-Oxo-1-(2-chloro-4-nitrobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, yellow amorphous

 1 H-NMR (CDCl₃) δ ; 1.7-2.1 (4H, m), 2.85-4.7 (4H, m), 7.12 (1H, d, J=8.4 Hz), 7.17-7.51 (4H, m), 7.89 (1H, dd, J=7.8 Hz, 2.1 Hz), 8.11 (1H, d, J=2.2 Hz)

8-Chloro-6-oxo-1-(2-chloro-4-nitrobenzoyl)-

1,2,3,4,5,6-hexahydrobenzazocine, yellow amorphous $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.7-2.15 (4H, m), 2.85-4.8 (4H, m), 7.14 (1H, d, J=8.5 Hz), 7.16 (1H, d, J=8.4 Hz), 7.34(1H, dd, J=8.3 Hz, 2.5 Hz), 7.85 (1H, d, J=2.5 Hz), 7.94 (1H, dd, J=8.4 Hz, 2.2 Hz), 8.13 (1H, d, J=2.1 Hz) 8-Methyl-6-oxo-1-(2-chloro-4-nitrobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, yellow amorphous 1 H-NMR (CDCl₃) δ ; 1.65-2.2 (4H, m), 2.33 (3H, s), 2.7-5.0 (4H, m), 7.0-7.25 (3H, m), 7.67 (1H, d, J=2.0 Hz), 7.89 (1H, dd, J=8.4 Hz, 2.2 Hz), 8.10 (1H, d, J=2.1 Hz) 8-Methoxy-6-oxo-1-(2-chloro-4-nitrobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous $^{1}\text{H-NMR}$ (CDCl₃) & ; 1.6-2.05 (4H, m), 2.8-5.2 (4H, m), 3.78 (3H, s), 6.88 (1H, dd, J=8.6 Hz, 3.1 Hz), 7.11 (1H, d, J=8.4 Hz), 7.12 (1H, d, J=8.6 Hz), 7.38 (1H, d, J=3.0Hz), 7.90 (1H, dd, J=8.4 Hz, 2.2 Hz), 8.11 (1H, d, J=2.2 Hz) 7-Chloro-5-oxo-1-(2-chloro-4-nitrobenzoyl)-2,3,4,5tetrahydro-lH-benzazepine, yellow powder (recrystallized from diethyl ether/dichloromethane), m.p. 125 - 126.5°C

Reference Example 21

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

7-Methyl-5-oxo-1-(4-aminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine, yellow powder

 $^{1}\text{H-NMR}$ (CDCl₃) 6 ; 2.13 (2H, brs), 2.32 (3H, s),

عتقافي فالمالف الإناف المعالي الما

2.86 (2H, t, J=6.2 Hz), 2.89-5.29 (2H, m), 3.86 (2H, brs), 6.41 (2H, m), 6.65 (1H, d, J=8.1 Hz), 7.06 (3H, m), 7.65 (1H, d, J=1.7 Hz)

7-Dimethylamino-5-oxo-1-(4-aminobenzoy1)-2,3,4,5-tetrahydro-1H-benzazepine, yellow needles (recrystallized from dichloromethane/diethyl ether)

 1 H-NMR (CDCl $_{3}$) $_{\delta}$; 1.78-2.49 (2H, m), 2.64-3.78, 4.07-5.02 (total 4H, m), 2.93 (6H, m), 3.96 (2H, m), 6.38 (2H, d, J=8.7 Hz), 6.55 (1H, dd, J=2.7, 8.7 Hz), 6.62 (1H, d, J=8.7 Hz), 6.96-7.18 (3H, m)

7-Bromo-5-oxo-1-(4-aminobenzoyl)-2,3,4,5-tetra-hydro-1H-benzazepine, white powder (recrystallized from methanol/diethyl ether)

¹H-NMR (CDCl₃) 6; 1.98-2.37 (2H, m), 2.88 (2H, t, J=6.3 Hz), 3.52-4.55 (4H, m), 6.28-6.57 (2H, m), 6.57-6.76 (1H, m), 6.92-7.20 (2H, m), 7.28-7.42 (1H, m), 7.90-8.09 (1H, m)

7-Chloro-5-oxo-1-(2-methyl-4-aminobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, white powder (recrystallized from dichloromethane/diethyl ether), m.p. 190 - 191°C

6-Oxo-1-(2-chloro-4-aminobenzoy1)-1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous

 $^{1}\text{H-NMR}$ (CDCl₃) $_{6}$; 1.3-2.25 (4H, m), 2.8-4.4 (6H, m), 6.1-6.9 (3H, m), 6.95-7.75 (3H, m), 7.8-8.3 (1H, m)

8-Chloro-6-oxo-1-(2-chloro-4-aminobenzoyl)1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous

 1 H-NMR (CDCl₃) δ ; 1.59-2.2 (4H, m), 2.6-4.4 (6H, m), 6.1-6.9 (3H, m), 6.95-7.5 (2H, m), 7.8-8.05 (1H, m)

7-Chloro-5-oxo-1-(2-chloro-4-aminobenzoyl)-2,3,4,5tetrahydro-1H-benzazepine, yellow powder (recrystallized from diethyl ether/dichloromethane), m.p. 188 - 191.5°C

Using the suitable starting materials, the compounds of the following Table 8 are obtained in the same manner as in above Examples 1 and 382.

Table 8

Example 998

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} \mathbb{R}^{1} \mathbb{N}

Crystalline form: White powder

NMR analysis: 196)

Structure

$$\begin{array}{c}
\text{CH}_3 & \text{O}_1 \\
\text{R}_1 & \text{I}
\end{array}$$

$$: \begin{array}{c}
\text{CH}_3 & \text{O}_1 \\
\text{II}
\end{array}$$

R²: н

Crystalline form: White powder

NMR analysis: 197)

Form: Free

Example 1000

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 200 - 205°C

Structure

Crystalline form: Colorless amorphous

NMR analysis: 198)

Form: Free

Example 1002

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 234 - 238°C

Structure

$$\begin{array}{c}
\text{CH}_{3} \\
\text{CH}_{2}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3}
\end{array}$$

 R^2 : F

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 174 - 178°C

Form: Free

Example 1004

Structure

R²: 2-C1

Crystalline form: Light yellow amorphous

NMR analysis: 199)

Structure

$$\begin{array}{c} \mathbb{C} \\ \mathbb{C} \\ \mathbb{R}^1 \end{array} \qquad \mathbb{C}^{\mathbb{H}_3} \qquad \mathbb{C}^{\mathbb{N}} \\ \mathbb{C} \\ \mathbb{R}^1 \qquad \mathbb{C} \\ \mathbb{C} \\ \mathbb{C} \\ \mathbb{R}^1 \qquad \mathbb{C} \\ \mathbb{C$$

R²: 2-Cl

Crystalline form: Light yellow amorphous

NMR analysis: 200)

Form: Free

Example 1006

Structure

R²: 2-C1

Crystalline form: Light yellow amorphous

NMR analysis: 201)

Structure

CH₃ NHCH₃

$$R^{1}$$

R²: 2-Cl

Crystalline form: Light yellow amorphous

NMR analysis: 202)

Form: Free

Example 1008

Structure

R²: 2-C1

Crystalline form: Light yellow amorphous

MNR analysis: 203)

Structure

R²: 2-Cl

Crystalline form: Light yellow amorphous

NMR analysis: 204)

Form: Free

Example 1010

Structure

R²: н

$$CH_3$$
 CH_3
 CH_3
 CH_3

Crystalline form: Colorless amorphous

NMR analysis: 205)

Form: Free

7 4

ê

Structure

$$\begin{array}{c}
\text{re} \\
\text{W} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3}\text{O} \\
\text{O} \\
\text{N}
\end{array}$$

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 153 - 155°C

Form: Free

Example 1012

Structure

$$\begin{array}{c}
\text{re} \\
\text{CH}_3\text{O} \\
\text{N}
\end{array}$$

$$\begin{array}{c}
\text{CH}_3\text{O} \\
\text{N}
\end{array}$$

R²: 3-ОСН₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 142 - 143°C

Structure

$$\begin{array}{c}
\text{CH}_{3}\text{O} & \text{NHCH}_{3} \\
\text{R}_{1} & & & \\
\end{array}$$

к²: н

Crystalline form: White powder

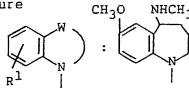
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 176 - 178°C

Form: Free

Example 1014

Structure



R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 186 - 188°C

Form: Free

•

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 206)

Form: Free.

Example 1016

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 207)

Structure

R²: 3-OCH₃

Crystalline form: White powder

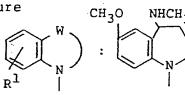
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 191 - 191.5°C

Form: Free

Example 1018

Structure



R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 210 - 212°C

Structure CH₃O CH

 R^2 : H

R³: 4-NHCO

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 196 - 198°C

Form: Free

Example 1020

Structure CH₃O CH₃CH₃

R²: н

R³: 4-NHCO

Crystalline form: Colorless amorphous

NMR analysis: 208)

Structure сн30

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 209)

Form: Free

Example 1022

Structure СН30

Crystalline form: Colorless amorphous

NMR analysis: 210)

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 211)

Form: Free

Example 1024

Structure

re
$$CH_3O$$
 CH_3 CH_3

R²: 3-OCH₃

R³: 4-NHCO-

Crystalline form: Colorless amorphous

NMR analysis: 212)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: 2-C1

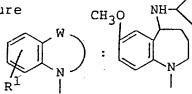
Crystalline form: Colorless amorphous

NMR analysis: 213)

Form: Free

Example 1026

Structure



R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 214)

Form: Free

•

•

Structure

$$\mathbb{R}^1$$
 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1

R²: 2-C1

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 207 - 208°C

Form: Free

Example 1028

Structure

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 201 - 202°C

Structure

$$\begin{array}{c}
\text{Cl} & \text{OH} \\
\text{R}^{1} & \text{N}
\end{array}$$

R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 193 - 194°C

Form: Free

Example 1030

Structure W : C1 N

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 205 - 208°C

Form: Free

.

7

Structure

R²: H

Crystalline form: White powder

Recrystallization solvent: Ethanol

Melting Point: 214 - 216°C

Form: Free

Example 1032

Structure

 R^2 : H

Crystalline form: Yellow needles

Recrystallization solvent: Ethanol

Melting Point: 223 - 226°C

Structure CH_3 NHCH₃ CH_3 $CH_$

R.²: Н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 203 - 206°C

Form: Free

Example 1034

Structure CH₃ N CH₃

CH₃ N CH₃

CH₃ N CH₃

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether/n-hexane

Melting Point: 168 - 171°C

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 206 - 208°C

Form: Free

Example 1036

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 229 - 232°C

Structure

R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 220 - 222°C

Form: Free

Example 1038

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 232 - 233.5°C

Form: Free

*

Structure

R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 215)

Form: Free

Example 1040

Structure

R²: 2-CH₂

Crystalline form: Colorless amorphous

NMR analysis: 216)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 147 - 151°C

Form: Free

Example 1042

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 127 - 129°C

Form: Free

Ţ

Ī

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 109 - 112°C

Form: Free

Example 1044

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 198 - 200°C

Structure

$$\begin{array}{c}
\text{Br} & \text{O} \\
\text{R}^{1} & \text{I}
\end{array}$$

 R^2 : H

Crystalline form: White powder

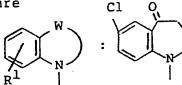
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 210 - 211°C

Form: Free

Example 1046

Structure



 $R^2: 2-CH_3$

Crystalline form: Colorless amorphous

NMR analysis: 217)

Structure

R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 218)

Form: Free

Example 1048

Structure

$$\mathbb{R}^{1}$$
 \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1} \mathbb{R}^{1}

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 219)

Structure

$$\begin{array}{c}
\text{NHCH}_{3} \\
\text{NHCH}_{3}
\end{array}$$

 R^2 : I

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol

Melting Point: 243 - 243.5°C

Form: Free

Example 1050

Structure

R²: н

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 207 - 209°C

Form: Free

3

Structure

R²: F

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 239 - 241°C

Form: Free

Example 1052

Structure

$$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right) : \left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \end{array}\right)$$

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 220)

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 221)

Form: Free

Example 1054

Structure

$$\begin{array}{c}
\text{c1} & \text{c1} & \text{o} \\
\text{R}^1 & \text{N}
\end{array}$$

 $R^2: 2-C1$

$$R^3$$
: 4-NHCO-

Crystalline form: Light yellow amorphous

NMR analysis: 222)

Form: Free

*

*

Structure

$$\begin{array}{c}
\text{C1} & \text{O} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{C1} & \text{O} \\
\text{N} \\
\text{N}
\end{array}$$

 R^2 : 2-C1

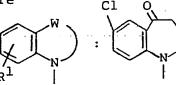
Crystalline form: Light yellow amorphous

NMR analysis: 223)

Form: Free

Example 1056

Structure



R²: 2-C1

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/dichloromethane

Melting Point: 169.5 - 173°C

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 224)

Form: Free

Example 1058

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 225)

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 226)

Form: Free

Example 1060

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 227)

Structure

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 228)

Form: Free

Example 1062

Structure

$$\begin{array}{c}
\text{C1} & \text{CH}_2\text{CH}=\text{CH}_2 \\
\text{R}^1 & \text{N}
\end{array}$$

Crystalline form: Colorless amorphous

NMR analysis: 229)

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 230)

Form: Free

Example 1064

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 231)

Structure

$$\begin{array}{c}
\text{C1} & \text{NHCH}_3 \\
\text{R}^1 & \text{N}
\end{array}$$

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 232)

Form: Free

Example 1066

Structure

re
$$C1$$
 CH_{2} CH_{3}

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 233)

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 234)

Form: Free

Example 1068

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 235)

Structure

R²: 2-C

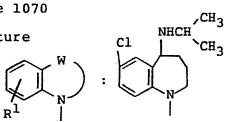
Crystalline form: Colorless amorphous

NMR analysis: 236)

Form: Free

Example 1070

Structure



R²: 2-ci

R³: 4-NHCO-

Crystalline form: Colorless amorphous

NMR analysis: 237)

```
^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 2.14 (2H, brs), 2.33 (3H, s),
196)
           2.46 (3H, s), 2.85 (2H, t, J=6.1 Hz), 4.83 (2H,
           brs), 6.64 (1H, d, J=8.1 Hz), 7.07 (1H, d, J=8.0 Hz),
           7.21-7.48 (8H, m), 7.65 (1H, m), 7.74 (1H, brs)
           <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ ; 2.12 (2H, brs), 2.33 (3H, s),
197)
           2.85 (2H, t, J=6.2 Hz), 2.88-5.28 (2H, m), 6.63
           (1H, d, J=8.1 Hz), 7.06 (1H, dd, J=1.7 Hz, 8.1 Hz),
           7.19-7.69 (9H, m), 8.26 (1H, brs)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 0.49 (4H, m), 1.25-5.13 (9H, m),
198)
           2.33 (3H, s), 2.45 (3H, s), 6.53 (1H, m), 6.79 (1H,
          m), 7.07-7.42 (9H, m), 7.73 (1H, m)
          ^{1}\text{H-NMR} (CDCl<sub>3</sub>) \delta ; 2.04 (2H, brs), 2.29 (3H, s),
199)
          2.82 (2H, t, J=5.9 Hz), 2.85-5.29 (2H, m), 6.82-
          7.69 (10H, m), 8.31 (1H, brs)
          <sup>1</sup>H-NMR (CDCl<sub>3</sub>) 6; 2.05 (2H, brs), 2.29 (3H, s),
200)
          2.44 (3H, s), 2.79 (2H, t, J=5.5 Hz), 2.82-5.28
          (2H, m), 6.82-8.12 (11H, m)
          ^{1}H-NMR (CDCl<sub>3</sub>) & ; 1.40-4.85 (11H, m), 2.51 (3H,
201)
          s), 6.78-7.63 (10H, m), 8.64 (1H, brs)
          ^{1}H-NMR (CDCl<sub>3</sub>) 6 ; 1.40-4.85 (11H, m), 2.45 (3H, s),
202)
```

2.50 (3H, s), 6.78-7.55 (10H, m), 8.10 (1H, brs) 1 H-NMR (CDCl₃) 5 ; 0.49 (4H, m), 1.25-4.85 (9H, m),

203) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 0.49 (4H, m), 1.25-4.85 (9H, m), 2.28 (3H, s), 6.77-7.62 (10H, m), 8.64 (1H, brs)

204) ¹H-NMR (CDCl₃) δ; 0.48 (4H, m), 1.26-4.85 (9H, m), 2.29 (3H, s), 2.44 (3H, s), 6.78-7.58 (10H, m), 8.18 (1H, brs)

- ¹H-NMR (CDCl₃) δ; 1.14 (6H, d, J=6.3 Hz), 1.52-2.20 (7H, m), 2.20-2.60 (1H, m), 2.64-3.66 (10H, m), 4.00-4.50 (4H, m), 4.50-5.23 (2H, m), 6.57-7.90 (11H, m), 8.10-8.30 (1H, m), 9.97 (1H, s)
- 206) ¹H-NMR (CDCl₃) δ; 1.24-2.08 (4H, m), 2.08-2.26 (3H, m), 2.26-3.16 (4H, m), 3.47-4.03 (4H, m), 4.18-4.92 (1H, m), 6.40-7.94 (10H, m), 8.45-9.03 (1H, m)
- 207) ¹H-NMR (CDCl₃) δ; 1.26-2.10 (4H, m), 2.10-2.28 (3H, m), 2.28-3.20 (1H, m), 3.43-4.06 (4H, m), 4.20-4.93 (1H, m), 6.40-8.00 (10H, m), 8.78-9.30 (1H, m)
- 208) ¹H-NMR (CDCl₃) δ; 1.10-1.98 (4H, m), 1.98-3.10 (7H, m), 3.30-3.90 (4H, m), 3.90-5.10 (1H, m), 6.45-8.25 (12H, m)
- 209) 1 H-NMR (CDCl₃) $_{\delta}$; 1.06-1.94 (4H, m), 1.94-3.19 (10H, m), 3.19-3.90 (4H, m), 3.90-5.10 (1H, m), 6.44-8.60 (11H, m)
- 210) 1 H-NMR (CDCl₃) & ; 1.06-1.97 (4H, m), 1.97-3.20 (7H, m), 3.20-3.92 (4H, m), 3.92-5.10 (1H, m), 6.44-8.55 (11H, m)
- 211) ¹H-NMR (CDCl₃) δ; 1.07-1.98 (4H, m), 1.98-3.10 (10H, m), 3.37-5.20 (8H, m), 6.44-6.86 (3H, m), 6.97-7.60 (6H, m), 8.13 (1H, s), 8.19-8.38 (1H, m)
- 212) ${}^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.08-1.99 (4H, m), 1.99-3.13 (7H, m), 3.33-5.14 (8H, m), 6.40-6.90 (3H, m),

- 6.95-7.56 (5H, m), 7.63-7.87 (1H, m), 8.17-8.37 (1H, m), 8.60 (1H, s)
- 213) $^{1}\text{H-NMR}$ (CDCl₃) $_{6}$; 0.30-0.64 (4H, m), 0.70-3.42 (9H, m), 3.42-5.10 (5H, m), 6.40-8.70 (11H, m)
- ¹H-NMR (CDCl₃) δ; 0.30-0.76 (4H, m), 0.80-3.43 (6H, m), 3.50-5.00 (5H, m), 6.40-9.04 (11H, m)
- ¹H-NMR (CDCl₃) δ; 1.25-3.25 (14H, m), 3.55-5.06 (2H, m), 6.43-7.00 (2H, m), 7.00-7.71 (8H, m), 7.91-8.45 (1H, m)
- 216) ¹H-NMR (CDCl₃) δ; 1.11-3.20 (17H, m), 3.28-5.12 (2H, m), 6.41-7.01 (2H, m), 7.02-7.63 (8H, m), 7.76-8.21 (1H, m)
- 217) ¹H-NMR (CDCl₃) δ; 1.92-2.29 (2H, m), 2.36 (3H, s), 2.45 (3H, s), 2.84 (2H, t, J=6.3 Hz), 3.32-4.64 (2H, m), 6.40-8.10 (11H, m)
- 218) ¹H-NMR (CDCl₃) δ; 1.92-2.25 (2H, m), 2.34 (3H, s), 2.83 (2H, t, J=6.3 Hz), 3.21-4.52 (2H, m), 6.39-7.97 (10H, m), 8.43 (1H, brs)
- 219) ¹H-NMR (CDCl₃) δ; 1.7-2.15 (4H, m), 2.5-5.2 (4H, m), 6.75-6.9 (1H, m), 7.27-7.6 (9H, m), 7.65-7.85 (1H, m), 7.9-8.15 (2H, m)
- ¹H-NMR (CDCl₃) δ ; 1.65-2.1 (4H, m), 2.44 (3H, s), 2.8-4.5 (4H, m), 6.75-8.0 (12H, m)
- 221) ¹H-NMR (CDCl₃) δ ; 1.65-2.3 (4H, m), 2.7-4.8 (4H, m), 6.75-8.4 (12H, m)
- 222) ¹H-NMR (CDCl₃) & ; 1.45-2.15 (4H, m), 2.45-2.55

```
(3H, m), 2.85-4.6 (4H, m), 6.8-8.25 (11H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.5-2.2 (4H, m), 2.8-4.7 (4H,
223)
           m), 6.8-8.4 (11H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.75-2.25 (2H, m), 2.30-2.70
224)
           (3H, m), 2.70-2.95 (2H, m), 3.20-5.10 (2H, m),
           6.70-8.40 (11H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.20-2.60 (8H, m), 2.60-5.10
225)
           (3H, m), 6.80-7.90 (10H, m), 8.20-8.60 (1H, m)
           <sup>1</sup>H-NMR (CDCl<sub>3</sub>) \delta; 1.20-2.60 (10H, m), 2.60-5.10
226)
           (3H, m), 6.80-8.15 (11H, m)
           ^{1}\text{H-NMR} (CDCl<sub>3</sub>) \delta ; 0.30-0.70 (4H, m), 1.20-2.45
227) -
           (6H, m), 2.60-5.10 (3H, m), 6.80-7.95 (10H, m),
           8.15-8.50 (1H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta; 1.20-2.40 (5H, m), 2.60-5.35
228)
           (7H, m), 5.80-6.15 (1H, m), 6.75-7.95 (10H, m),
           8.20-8.70 (1H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.20-2.55 (7H, m), 2.60-5.35
229)
           (7H, m), 5.85-6.05 (1H, m), 6.70-7.10 (2H, m),
           7.10-7.90 (8H, m), 8.15-8.60 (1H, m)
           ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.00-1.20 (6H, m), 1.00-2.40
230)
           (5H, m), 2.60-5.10 (4H, m), 6.80-8.00 (10H, m),
           8.15-8.65 (1H, m)
           <sup>1</sup>H-NMR (CDCl<sub>3</sub>) \delta; 0.80-2.50 (13H, m), 2.60-5.10
231)
           (4H, m), 6.70-8.85 (10H, m), 8.25-8.60 (1H, m)
           <sup>1</sup>H-NMR (CDCl<sub>3</sub>) \delta ; 1.30-2.60 (11H, m), 2.60-5.10
232)
```

(3H, m), 6.80-8.15 (11H, m)

```
233) ^{1}\text{H-NMR} (CDCl<sub>3</sub>) _{\delta} ; 1.10-2.50 (13H, m), 2.50-5.10 (3H, m), 6.75-8.40 (11H, m)
```

- ¹H-NMR (CDCl₃) δ; 0.30-0.65 (4H, m), 1.20-2.30 (6H, m), 2.35-2.55 (3H, m), 2.60-5.10 (3H, m), 6.75-8.35 (11H, m)
- 235) $^{1}\text{H-NMR}$ (CDCl₃) & ; 1.20-2.60 (8H, m), 2.60-5.40 (7H, m), 5.80-6.15 (1H, m), 6.80-8.20 (11H, m)
- 236) ¹H-NMR (CDCl₃) δ; 1.25-2.60 (10H, m), 2.60-5.40 (7H, m), 5.75-6.10 (1H, m), 6.75-7.10 (2H, m), 7.10-8.40 (9H, m)

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 1.

8-Chloro-6-oxo-1-(4-nitrobenzoy1)-1,2,3,4,5,6-hexahydrobenzazocine, yellow prisms

 $^{1}\text{H-NMR}$ (DMSO- $^{1}\text{d}_{6}$) $_{\delta}$; 1.3-2.2 (4H, m), 2.6-5.0 (4H, m), 7.05-8.5 (7H, m)

5-Oxo-7-methyl-1-(2-chloro-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, light yellow amorphous

¹H-NMR (CDCl₃) δ; 1.71-2.32 (2H, m), 2.29 (3H, s), 2.86 (2H, t, J=6.3 Hz), 3.10-5.30 (2H, m), 6.84-8.38 (6H, m) 5-Oxo-7-methyl-l-(3-methoxy-4-nitrobenzoyl)- 2,3,4,5-tetrahydro-1H-benzazepine, light yellow amorphous

¹H-NMR (CDCl₃) δ; 2.17 (2H, brs), 2.34 (3H, s),

2.84 (2H, t, J=6.0 Hz), 3.10-5.29 (2H, m), 3.77 (3H, s),

6.67 (1H, d, J=7.9 Hz), 6.85 (2H, m), 7.10 (1H, d, J=8.0 Hz), 7.57-7.65 (2H, m)

5-Oxo-7-dimethylamino-1-(2-chloro-4-nitrobenzoyl)-2,3,4,5-tetrahydro-1H-benzazepine, yellow powder

¹H-NMR (CDCl₃) & ; 1.66-2.38 (2H, m), 2.65-2.88 (2H, m), 2.92 (6H, s), 3.08-3.64, 4.58-5.01 (total 2H, m), 6.49 (1H, dd, J=3.1, 8.7 Hz), 6.82 (1H, d, J=8.7 Hz), 6.90 (1H, d, J=3.1 Hz), 7.02-7.37 (1H, m), 7.94 (1H, dd, J=1.9, 8.4 Hz), 8.08 (1H, d, J=1.9 Hz)

Reference Example 23

Using the suitable starting materials, the following compounds are obtained in the same manner as in Reference Example 2.

8-Chloro-6-oxo-1-(4-aminobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous

¹H-NMR (CDCl₃) δ; 1.7-2.2 (4H, m), 2.3-4.8 (6H, m), 6.4-6.6 (2H, m), 6.74 (1H, d, J=8.5 Hz), 7.1-7.4 (3H, m), 7.99 (1H, d, J=2.6 Hz)

8-Methyl-6-oxo-(2-chloro-4-aminobenzoyl)
1,2,3,4,5,6-hexahydrobenzazocine, colorless amorphous

¹H-NMR (CDCl₃) δ; 1.4-2.1 (4H, m), 2.15-2.6 (3H, m), 2.7-4.4 (6H, m), 6.15-6.35 (1H, m), 6.51 (1H, s), 6.6-6.85 (1H, m), 6.9-7.25 (2H, m), 7.72 (1H, s)

```
8-Methoxy-6-oxo-(2-chloro-4-aminobenzoyl)-
1,2,3,4,5,6-hexahydrobenzazocine, light yellow amorphous
          ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.4-2.2 (4H, m), 2.7-5.0 (9H,
m), 6.25 (1H, dd, J=8.3 Hz, 2.2 Hz), 6.51 (1H, d, J=2.2 Hz);
6.66 (1H, d, J=8.3 Hz), 6.88 (1H, dd, J=8.6 Hz, 3.0 Hz),
7.23 (1H, d, J=8.6 Hz), 7.43 (1H, d, J=3.0 Hz)
          5-0xo-7-chloro-1-(2-methoxy-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, colorless particles
(recrystallized from methanol/diethyl ether), m.p. 206 -
208°C
          5-0xo-7-methyl-1-(2-chloro-4-aminobenzoyl)-2,3,4,5-
tetrahydro-lH-benzazepine, light yellow amorphous
         ^{1}\text{H-NMR} (CDCl<sub>3</sub>) 6; 2.09 (2H, brs), 2.29 (3H, s),
3.10-5.00 (2H, m), 3.78 (2H, brs), 6.34-7.54 (6H, m)
          5-0xo-7-methyl-1-(3-methoxy-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, light yellow amorphous
         ^{1}\text{H-NMR} (CDCl<sub>3</sub>) \delta ; 2.12 (2H, brs), 2.32 (3H, s),
2.85 (2H, t, J=5.9 Hz), 3.30-5.00 (2H, m), 3.65 (3H, s),
3.98 (2H, brs), 6.40 (1H, d, J=8.1 Hz), 6.64-6.76 (3H, m),
7.06 (lH, dd, J=1.6, 8.1 Hz), 7.63 (lH, d, J=2.0 Hz)
         5-0xo-7-dimethylamino-1-(2-chloro-4-aminobenzoyl)-
2,3,4,5-tetrahydro-lH-benzazepine, yellow amorphous
         ^{1}H-NMR (CDCl<sub>3</sub>) \delta ; 1.60-2.32 (2H, m), 2.67-5.13
(4H, m), 2.92 (6H, s), 3.75 (2H, s), 6.31 (1H, dd, J=2.1,
8.3 Hz), 6.46 (1H, d, J=2.1 Hz), 6.48 (1H, dd, J=3.1, 8.7
```

Hz), 6.66-6.89 (2H, m), 6.95 (1H, d, J=3.1 Hz)

Using the stuitable starting materials, the compounds of the following Table 9 are obtained in the same manner as in above Examples 1 and 382.

Table 9

Example 1071

Structure

 R^2 : 2-Cl

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 227 - 230°C

Structure

re
$$C1$$
 NHCH₃ R^1 R^1

R²: 2-Cl

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/petroleum ether

Melting Point: 216 - 218°C

Form: Free

Example 1073

Structure

R²: 2-C1

Crystalline form: Colorless prisms

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 227 - 228°C

Structure

$$\left(\begin{array}{c} C1 & 0 \\ N \\ N \end{array}\right) : \left(\begin{array}{c} C1 & 0 \\ N \\ N \end{array}\right)$$

R²: н

Crystalline form: White powder

NMR analysis: 238)

Form: Free

Example 1075

Structure

$$\begin{array}{c}
C1 & O \\
N & O
\end{array}$$

$$\begin{array}{c}
C1 & O \\
N & O
\end{array}$$

R²: н

Crystalline form: White powder

NMR analysis: 239)

Structure

re
$$_{\text{CH}_3}$$
 $_{\text{N}}$ $_{\text{R}_1}$ $_{\text{N}}$ $_{\text{N}}$

R²: 2-C1

Crystalline form: Light yellow amorphous

NMR analysis: 240)

Form: Free

Example 1077

Structure

$$(H_3)$$

R²: 2-Cl

Crystalline form: Light yellow amorphous

NMR analysis: 241)

Structure

R²: 2-C1

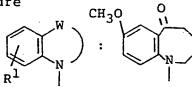
Crystalline form: Colorless amorphous

NMR analysis: 242)

Form: Free

Example 1079

Structure



 $R^2: 2-C$

Crystalline form: Colorless amorphous

NMR analysis: 243)

Structure

R²: 2-Cl

Crystalline form: Light yellow powder

Recrystallization solvent: Ethyl acetate/diethyl ether

Melting Point: 179 - 181°C

Form: Free

Example 1081

Structure

$$\begin{array}{c}
\text{CH}_{3} & \text{NHCH}_{3} \\
\text{N} & \text{NHCH}_{3}
\end{array}$$

 $R^2: 2-01$

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/diethyl ether

Melting Point: 213 - 216°C

Structure

R²: 2-C1

Crystalline form: Light yellow powder

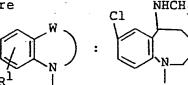
Recrystallization solvent: Ethyl acetate/diethyl ether

Melting Point: 185 - 187°C

Form: Free

Example 1083

Structure



 \mathbb{R}^2 : F

$$R^3$$
: 4-NHCO-CH₃

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol

Melting Point: 249 - 251°C

Structure

$$\begin{array}{c}
\text{C1} \\
\text{NHCH}_{3}
\end{array}$$

R²: н

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol.

Melting Point: 239 - 241°C

Form: Free

Example 1085

Structure

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/diethyl ether

Melting Point: 208 - 210°C

Structure

R²: 2-CH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 178 - 180.5°C

Form: Free

Example 1087

Structure

$$\begin{array}{c}
\text{C1} & \text{NHCH}_3 \\
\text{R}^1 & \text{N}
\end{array}$$

 $R^2: 2-CH_3$

Crystalline form: Colorless amorphous

NMR analysis: 244)

Structure

R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 245)

Form: Free

Example 1089

Structure

$$(10^{\circ})^{\circ} \cdot (10^{\circ})^{\circ} \cdot (10^$$

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

NMR analysis: 246)

Structure

$$(1) \quad (1) \quad (1)$$

R²: 2-осн₃

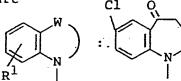
Crystalline form: Colorless amorphous

NMR analysis: 247)

Form: Free

Example 1091

Structure



R²: 2-OCH₂

Crystalline form: Colorless amorphous

NMR analysis: 248)

Structure

R²: 2-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 249)

Form: Free

Example 1093

Structure

R²: 2-OCH

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 205 - 206°C

Structure

 R^2 : 2-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 250)

Form: Free

Example 1095

Structure

R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 172.5 - 174°C

Structure

$$\begin{array}{c}
\text{Cl} & \text{CH}_3 \\
\text{R}^1 & \text{N}
\end{array}$$

R²: 2-OCH₃

Crystalline form: White powder

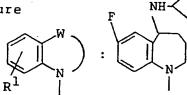
Recrystallization solvent: Methanol/diethyl ether

Melting Point: 215 - 216.5°C

Form: Free

Example 1097

Structure



R²: Н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 133 - 136°C

Structure

 $R^2: 2-C1$

Crystalline form: Colorless amorphous

NMR analysis: 251)

Form: Free

Example 1099

Structure

 R^2 : 2-C1

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 179 - 180°C

Structure

R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/n-hexane

Melting Point: 167.5 - 169.5°C

Form: Free

Example 1101

Structure

 R^2 : 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 176 - 178°C

Structure

R²: 2-OCH₃

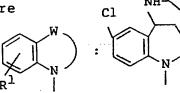
Crystalline form: Colorless amorphous

NMR analysis: 252)

Form: Free

Example 1103

Structure



R²: 2-OCH₃

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 185 - 188°C

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 180 - 181.5°C

Form: Free

Example 1105

Structure

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 181 - 184°C

Structure

к²: н

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 186.5 - 187°C

Form: Free

Example 1107

Structure

R³: 4-NHCO-

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 183 - 184°C

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Methanol/diethyl ether

Melting Point: 151 - 153°C

Form: Free

Example 1109

Structure

$$\begin{array}{c}
\text{CH}_3 & \text{O}_{1} \\
\text{R}_1 & \text{I}
\end{array}$$

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 253)

Structure

R²: н

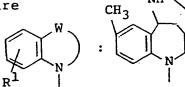
Crystalline form: Colorless amorphous

NMR analysis: 254)

Form: Free

Example 1111

Structure



R²: Н

Crystalline form: White needles

Recrystallization solvent: Ethanol/n-hexane

Melting Point: 191 - 195°C

Structure

$$\mathbb{C}^{H_3}$$
 \mathbb{C}^{H_3} \mathbb{C}^{H_3} \mathbb{C}^{H_3}

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/n-hexane

Melting Point: 227 - 230°C

Form: Free

Example 1113

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 289)

Structure

 R^2 : 2-Cl

Crystalline form: Light yellow amorphous

NMR analysis: 255)

Example 1115

Structure

R²: 3-осн₃

Crystalline form: White powder

Recrystallization solvent: Diethyl ether/n-hexane

Melting Point: 172 - 174°C

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 305)

Form: Free

Example 1117

Structure

R²: 2-СН₃

Crystalline form: Colorless amorphous

NMR analysis: 290)

Structure

re
$$C1$$
 CH_2 CH_3

R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 291)

Form: Free

Example 1119

Structure

R²: H

$$R^3$$
: 4-NHCO- $\stackrel{\text{CH}_3}{\longrightarrow}$

Crystalline form: Colorless amorphous

NMR analysis: 264)

Structure

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 265)

Form: Free

Example 1121

Structure

re
$$C1$$
 $NHCH_3$ R^1 N

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 266)

Structure

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 267)

Form: Free

Example 1123

Structure

$$\begin{array}{c}
\text{Cl} \\
\text{CH}_{3}
\end{array}$$

в². н

Crystalline form: Colorless amorphous

NMR analysis: 268)

Structure

$$\begin{array}{c}
\text{C1} \\
\text{CH}_{3}
\end{array}$$

R²: Е

Crystalline form: Colorless amorphous

NMR analysis: 269)

Form: Free

Example 1125

Structure

re
$$C1$$
 CH_3 R^1 N $C1$ N CH_3

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 270)

Structure

re
$$C1$$
 CH

 R^2 : 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 271)

Form: Free

Example 1127

Structure

 R^2 : H

Crystalline form: Colorless amorphous

NMR analysis: 272)

Form: Free

мен и рудит на видел на на возморения на видел и мене на видел возморения видел возморения видел в возморения

Structure

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 273)

Form: Free

Example 1129

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 274)

Structure

$$\begin{array}{c}
\text{CH}_{3} \\
\text{NHCH}_{3}
\end{array}$$

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 275)

Form: Free

Example 1131

Structure

re
$$CH_3$$
 CH_3 CH_3

R²: н

$$R^3$$
: 4-NHCO-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 276)

Structure

$$\begin{array}{c}
\text{CH}_{3} \\
\text{CH}_{3}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3}
\end{array}$$

R²: H

Crystalline form: Colorless amorphous

NMR analysis: 277)

Form: Free

Example 1133

Structure

$$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \\ \text{R}^1 \end{array} \right) : \begin{array}{c} \text{CH}_3 \\ \text{N} \end{array}$$

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 278)

Structure

$$\begin{array}{c} \text{CH}_{3} \\ \text{CH}_{3} \\ \text{CH}_{3} \end{array}$$

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 279)

Form: Free

Example 1135

Structure

R²: Н

NMR analysis: 280)

Structure

$$\begin{array}{c}
\text{C1} & \text{NHCH}_3 \\
\text{C1} & \text{CH}_3
\end{array}$$

R²: н

NMR analysis: 281)

Form: Free

Example 1137

Structure

 R^2 : 2-C1

NMR analysis: 282)

Structure

 R^2 : 2-C1

NMR analysis: 283)

Form: Free

Example 1139

Structure

$$\begin{array}{c} \text{C1} & \text{CH}_3 \\ \text{R}^1 & \text{N} \end{array}$$

R²: 2-C1

NMR analysis: 306)

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 284)

Form: Free

Example 1141

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 285)

Structure

re
$$C1$$
 CH_3 R^1 $|$

 R^2 : 2-C1

R³: 4-NHCO-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 286)

Form: Free

Example 1143

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 287)

Structure

R²: н

Crystalline form: White powder

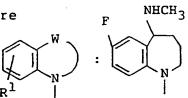
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 203 - 207°C

Form: Free

Example 1145

Structure



 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 199 - 203°C

Structure

 R^2 : H

$$R^3: 4-NHCO-CH_3$$

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 210 - 212°C

Form: Free

Example 1147

Structure

 R^2 : H

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 211 - 214°C

Structure

R²: 2-C1

Crystalline form: White powder

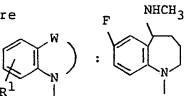
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 186 - 189°C

Form: Free

Example 1149

Structure



R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 288)

Structure

re
$$\frac{1}{1}$$
 $\frac{1}{1}$ \frac

R²: 2-C1

R³: 4-NHCO-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 292)

Form: Free

Example 1151

Structure

 R^2 : 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 293)

Structure

R²: 3-OCH₃

Crystalline form: White powder

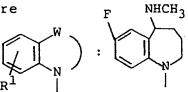
Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 144 - 145°C

Form: Free

Example 1153

Structure



R²: 3-осн₃

ar in the same

Crystalline form: White powder

Recrystallization solvent: Dichloromethane/diethyl ether

Melting Point: 149 - 150°C

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 294)

Form: Free

Example 1155

Structure

$$\begin{array}{c}
\text{re} \\
\text{W} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{F} \\
\text{N} \\
\text{CH}_{1}$$

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 295)

Structure

R²: 2-CH₃

Crystalline form:

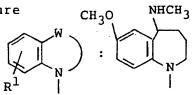
Colorless amorphous

NMR analysis: 301)

Form: Free

Example 1157

Structure



R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 302)

Structure

R²: 2-СН_э

Crystalline form: Colorless amorphous

NMR analysis: 303)

Form: Free

Example 1159

Structure

R²: 2-CH₃

Crystalline form: Colorless amorphous

NMR analysis: 304)

Structure

R²: H.

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/diisopropyl ether

Melting Point: 191 - 193°C

Form: Free

Example 1161

Structure

R²: н

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/diisopropyl ether

Melting Point: 221 - 223°C

Structure

R²: 3-OCH₃

Crystalline form: White powder

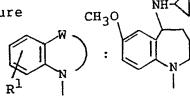
Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 159 - 161°C

Form: Free

Example 1163

Structure



R²: 3-OCH₃

Crystalline form: White powder

Recrystallization solvent: Ethyl acetate/n-hexane

Melting Point: 174 - 175°C

Structure

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 256)

Form: Free

Example 1165

Structure

$$\begin{array}{c}
\text{re} \\
\text{CH}_{3}\text{O} \\
\text{R}^{1}
\end{array}$$

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 257)

Structure

R²: H

Crystalline form: Colorless amorphous

NMR analysis: 258)

Form: Free

Example 1167

Structure

$$\begin{array}{c}
\text{UITE} \\
\text{CH}_{3}\text{O} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{NHCH}_{3}
\end{array}$$

R²: 2-Cl

Crystalline form: Colorless amorphous

NMR analysis: 259)

Structure

$$\mathbb{R}^{1}$$
 $\mathbb{C}^{H_{3}}$ $\mathbb{C}^{H_{3}}$ $\mathbb{C}^{H_{3}}$

R²: н

Crystalline form: Colorless amorphous

NMR analysis: 260)

Form: Free

Example 1169

Structure

R²: 2-C1

Crystalline form: Colorless amorphous

NMR analysis: 261)

Structure -

re
$$CH_3$$
 OH CH_3 N OH

R². н

$$R^3$$
: 4-NHCO- $\left\langle \begin{array}{c} CH_3 \\ \end{array} \right\rangle$

Crystalline form: Colorless amorphous

NMR analysis: 296)

Form: Free

Example 1171

Structure

$$\mathbb{R}^1$$
 : \mathbb{C}^{H_3} \mathbb{C}^{H_3} \mathbb{C}^{H_3}

R²: 2-Cl

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether/n-hexane

Melting Point: 159 - 162°C

Form: Free

•

•

Structure

R²: 2-Cl

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether/n-hexane

Melting Point: 221 - 224°C

Form: Free

Example 1173

Structure

 $R^2: 2-C1$

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 199 - 202°C

Structure

$$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \\ \text{CH}_3 \end{array}$$

R²: 2-Cl

Crystalline form: Colorless prisms

Recrystallization solvent: Ethanol/diethyl ether

Melting Point: 215 - 218°C

Form: Free

Example 1175

Structure

R²: 2-Cl

Crystalline form: Colorless needles

Recrystallization solvent: Ethanol/diethyl ether/n-hexane

Melting Point: 167 - 170°C

Structure

 $R^2: 2-C1$

Crystalline form: White powder

Recrystallization solvent: Ethanol/diethyl ether/n-hexane

Melting Point: 191 - 193°C

Form: Free

Example 1177

Structure

$$\begin{array}{c}
\text{CH}_{3} \\
\text{N}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3}
\end{array}$$

R²: 2-C1

Crystalline form: Light yellow amorphous

NMR analysis: 262)

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$

$$\begin{array}{c}
\text{CH}_{3} \\
\text{N} \\
\text{N}
\end{array}$$

R²: 2-C1

R³: 4-NHCO-

Crystalline form: Light yellow amorphous

NMR analysis: 263)

Form: Free

Example 1179

Structure

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 297)

Structure

$$\begin{array}{c}
\text{re} \\
\text{N} \\
\text{R}^{1}
\end{array}$$
: C1 C2^{H}

R²: 3-осн₃

Crystalline form: Colorless amorphous

NMR analysis: 298)

Form: Free

Example 1181

Structure

C1 NHCH2CH=CH2

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 299)

Structure

$$\left(\begin{array}{c} W \\ X \end{array} \right)$$

C1 NHCH2CH=CH2

R²: 3-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 300)

Form: Free

Example 1183

Structure

$$\left(\left(\right) \right)^{W}$$

C1 CH₂CH=CH₂

R²: 3-ОСН₃

Crystalline form: Colorless amorphous

NMR analysis: 307)

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

R²: 3-осн₃

Crystalline form: Colorless amorphous

NMR analysis: 308)

Form: Free

Example 1185

Structure

$$\mathbb{R}^{1}$$
 \mathbb{N} :

NHCH3

R²: 2-осн₃

Crystalline form: Colorless amorphous

NMR analysis: 309)

Structure

$$\left(\left(\right) \right)^{W}$$

инсн3 сн3о́

R²: 2-OCH₃

Crystalline form: Colorless amorphous

NMR analysis: 310)

Form: Free

Example 1187

Structure

$$\begin{array}{c}
\text{CH}_30 \\
\text{CH}_3^{0}
\end{array}$$

Crystalline form: Colorless amorphous

NMR analysis: 311)

Structure

R²: 2-осн₃

Crystalline form: Colorless amorphous

NMR analysis: 312)

Form: Free

...

. .

- 238) $^{1}H-NMR$ (DMSO- $^{1}d_{6}$) $_{6}$; 1.4-2.1 (4H, m), 2.34 (3H, s), 2.8-5.4 (4H, m), 7.09 (1H, d, J=8.4 Hz), 7.15-7.7 (9H, m), 7.76 (1H, d, J=2.6 Hz), 10.41 (1H, s)
- 239) ${}^{1}\text{H-NMR} \text{ (DMSO-d}_6) \ \delta \ ; \ 1.5-2.2 \ (4\text{H, m}), \ 2.8-5.2 \ (4\text{H, m}), \ 7.09 \ (1\text{H, d, J=8.4 Hz}), \ 7.2-7.7 \ (9\text{H, m}), \ 7.76 \ (1\text{H, d, J=2.6 Hz}), \ 10.63 \ (1\text{H, s})$
- ¹H-NMR (CDCl₃) δ ; 1.65-2.2 (4H, m), 2.25-2.65 (6H, m), 2.75-4.6 (4H, m), 6.8-8.15 (11H, m)
- 241) 1 H-NMR (CDCl₃) δ ; 1.4-2.25 (4H, m), 2.25-2.55 (3H, m), 2.7-4.8 (4H, m), 6.8-8.3 (11H, m)
- ¹H-NMR (CDCl₃) δ ; 1.4-2.1 (4H, m), 2.35-2.6 (3H, m), 2.8-5.2 (7H, m), 6.8-8.05 (11H, m)
- ¹H-NMR (CDCl₃) δ; 1.4-2.15 (4H, m), 2.4-5.2 (7H, m), 6.8-7.85 (10H, m), 7.9-8.3 (1H, m)
- ¹H-NMR (CDCl₃) δ; 1.20-2.38 (11H, m), 2.98-5.10 (3H, m), 6.45-7.04 (2H, m), 7.05-7.86 (8H, m), 8.00-8.50 (1H, m)
- 245) $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.05-2.78 (14H, m), 2.78-5.18 (2H, m), 6.36-7.03 (2H, m), 7.06-7.90 (8H, m), 7.98-8.39 (1H, m)
- ¹H-NMR (CDCl₃) δ; 1.75-2.54 (2H, m), 2.60-4.03 (4H, m), 3.37 (3H, brs), 5.17 (2H, s), 6.60-6.83 (3H, m), 6.90-7.07 (1H, m), 7.07-7.20 (1H, m), 7.22-7.50 (6H, m), 7.73-7.84 (1H, m)
- 247) $^{1}\text{H-NMR}$ (CDCl₃) $_{6}$; 1.60-2.40 (2H, m), 2.45 (3H, s), 2.65-3.06 (2H, m), 3.06-5.28 (2H, m), 3.35 (3H,

- brs), 6.59-7.60 (9H, m), 7.67-7.88 (1H, m), 8.12 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.60-2.52 (2H, m), 2.64-5.32 (4H, m), 3.37 (3H, brs), 6.60-7.98 (10H, m), 8.50 (1H, brs)
- 249) ¹H-NMR (CDCl₃) 5; 1.18-3.15 (12H, m), 3.40-4.38 (5H, m), 6.58-7.75 (10H, m), 8.30-8.71 (1H, m)
- 250) ¹H-NMR (CDCl₃) δ; 0.26-0.71 (4H, m), 1.15-3.29 (10H, m), 3.40-4.95 (5H, m), 6.60-7.85 (10H, m), 8.18-8.68 (1H, m)
- 251) ¹H-NMR (CDCl₃) 6; 0.25-0.72 (4H, m), 1.16-2.35 (6H, m), 2.35-3.30 (4H, m), 3.43-4.98 (2H, m), 6.57-7.94 (10H, m), 8.22-8.89 (1H, m)
- 252) ¹H-NMR (CDCl₃) & ; 0.69-2.90 (9H, m), 2.90-5.10 (5H, m), 6.40-7.85 (10H, m), 8.25-8.54 (1H, m)
- 253)

 ¹H-NMR (CDCl₃) δ; 2.17 (2H, brs), 2.34 (3H, s),
 2.49 (3H, s), 2.87 (2H, t, J=6.0 Hz), 3.10-5.00
 (2H, m), 3.70 (3H, s), 6.67 (1H, d, J=8.0 Hz),
 6.85-6.88 (2H, m), 7.09 (1H, dd, J=1.5, 8.0 Hz),
 7.21-7.50 (4H, m), 7.64 (1H, d, J=1.9 Hz), 8.11
 (1H, m), 8.33 (1H, d, J=8.8 Hz)
- ¹H-NMR (CDCl₃) δ; 1.42-5.06 (13H, m), 6.51 (1H, d, J=7.8 Hz), 6.76 (1H, m), 7.01-7.63 (10H, m), 8.53 (1H, m)
- 255) ¹H-NMR (CDCl₃) 6; 1.26-4.93 (16H, m), 6.69-7.73 (10H, m), 8.62-8.84 (1H, m)

- 256) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 1.45-1.90 (2H, m), 1.90-2.33 (2H, m), 2.33-3.25 (4H, m), 3.60-3.93 (3H, m), 4.45-5.15 (2H, m), 6.40-8.25 (11H, m)
- ¹H-NMR (CDCl₃) δ ; 1.49-1.97 (2H, m), 1.97-3.10 (3H, m), 3.58-3.98 (3H, m), 4.60-5.26 (2H, m), 6.44-8.36 (11H, m)
- 258) ¹H-NMR (CDCl₃) δ; 1.82-2.13 (1H, m), 2.13-2.43 (1H, m), 2.50 (3H, s), 2.57 (3H, s), 3.69-4.06 (3H, m), 3.78 (3H, s), 6.45-6.80 (2H, m), 6.85-7.00 (1H, m), 7.18-7.80 (9H, m)
- ¹H-NMR (CDCl₃) δ; 1.71-2.05 (1H, m), 2.07-2.32 (1H, m), 2.33 (6H, s), 2.47 (3H, s), 3.50-3.80 (2H, m), 3.76 (3H, s), 3.95-4.17 (1H, m), 6.40-6.70 (2H, m), 6.90-7.03 (1H, m), 7.14-7.77 (8H, m), 7.90-8.14 (1H, m)
- ¹H-NMR (CDCl₃) δ; 1.68-2.35 (2H, m), 2.36-5.11 [13H, m, 2.45 (3H, s), 2.92 (6H, s)], 6.56 (1H, dd, J=3.1, 8.7 Hz), 6.78-7.06 (2H, m), 6.82 (1H, d, J=8.7 Hz), 7.11-7.68 (6H, m), 7.97 (1H, brs)

- 263) ¹H-NMR (CDCl₃) δ; 1.69-2.30 (2H, m), 2.59-5.10 [10H, m, 2.92 (6H, s)], 6.56 (1H, dd, J=3.1, 8.8 Hz), 6.72-7.90 (9H, m), 8.42 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.49 (1H, brs), 1.82-2.01 (1H, m), 2.03-2.26 (1H, m), 2.46 (3H, s), 2.54 (3H, s), 3.67-3.76 (1H, m), 3.86 (2H, t, J=6.8 Hz), 6.67 (1H, d, J=8.6 Hz), 6.93 (1H, dd, J=8.6, 2.5 Hz), 7.13-7.43 (9H, m), 8.15 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.58 (1H, brs), 1.86-2.03 (1H, m), 2.08-2.30 (1H, m), 2.56 (3H, s), 3.69-3.78 (1H, m), 3.91 (2H, t, J=6.5 Hz), 6.69 (1H, d, J=8.7 Hz), 6.94 (1H, dd, J=8.6, 2.5 Hz), 7.33-7.47 (6H, m), 7.54-7.63 (2H, m), 7.67-7.77 (1H, m), 8.16 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.50 (lH, brs), 1.76-2.23 (2H, m), 2.42 (3H, s), 2.47 (3H, s), 3.55-3.94 (3H, m), 6.28-7.78 (10H, m), 8.91 (lH, brs)
- ¹H-NMR (CDCl₃) δ; 1.46 (1H, brs), 1.82-2.28 (2H, m), 2.50 (3H, s), 3.52-4.08 (3H, m), 6.34-7.75 (10H, m), 8.61 (1H, brs)
- 268) ¹H-NMR (CDCl₃) δ; 1.80-2.31 (2H, m), 2.32 (3H, s), 2.48 (3H, s), 3.51-3.82 (2H, m), 3.95-4.15 (1H, m), 6.59 (1H, d, J=8.6 Hz), 6.90 (1H, dd, J=8.6, 2.5 Hz), 7.16-7.61 (9H, m), 7.88 (1H, brs)
- 269) $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.86-2.04 (1H, m), 2.13-2.31 (1H, m), 2.33 (3H, s), 3.53-3.62 (1H, m), 3.76 (1H,

```
dt, J=12.8, 6.4 Hz), 6.60 (lH, d, J=8.7 Hz), 6.91 (lH, dd, J=8.7, 2.5 Hz), 7.33-7.52 (6H, m), 7.54-7.66 (2H, m), 7.73-7.82 (lH, m), 8.07 (lH, brs)
```

- 270) ¹H-NMR (CDCl₃) δ; 1.65-2.27 (2H, m), 2.28 (6H, s), 2.48 (3H, s), 3.37-4.07 (3H, m), 6.33-7.91 (10H, m), 8.20 (1H, brs)
- 271) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 1.71-2.26 (2H, m), 2.28 (6H, s), 3.36-4.10 (3H, m), 6.35-7.95 (10H, m), 8.59 (1H, brs)
- 272) ¹H-NMR (CDCl₃) δ; 2.02-2.23 (2H, m), 2.28 (3H, s), 2.47 (3H, s), 2.56 (3H, s), 3.73-4.07 (3H, m), 4.68 (1H, brs), 6.61 (1H, d, J=8.1 Hz), 6.72-6.83 (1H, m), 7.17-7.63 (11H, m), 8.03 (1H, brs)
- 273) ¹H-NMR (CDCl₃) δ; 1.61 (1H, brs), 1.87-2.25 (2H, m), 2.29 (3H, s), 2.56 (3H, s), 3.67-3.78 (1H, m), 3.91 (2H, t, J=6.9 Hz), 6.52-6.79 (2H, m), 7.09-7.15 (1H, m), 7.30-7.90 (8H, m), 8.23 (1H, brs)
- ¹H-NMR (CDCl₃) δ; 1.58 (1H, brs), 1.82-2.23 (2H, m), 2.27 (3H, s), 2.48 (3H, s), 2.50 (3H, s), 3.47-4.05 (3H, m), 6.23-6.83 (2H, m), 7.00-7.50 (7H, m), 7.53-7.74 (1H, m), 8.28 (1H, brs)
- ¹H-NMR (CDCl₃) 6; 1.60 (1H, brs), 1.82-2.35 (5H, m), 2.49 (3H, s), 3.41-4.08 (3H, m), 6.30-6.80 (1H, m), 6.98-7.68 (8H, m), 7.31-7.82 (1H, m), 8.77 (1H, brs)
- 276) $^{1}H-NMR$ (CDCl₃) & ; 1.76-2.03 (2H, m), 2.27 (3H, s),

2.32 (6H, s), 2.47 (3H, s), 3.48-3.58 (1H, m), 3.66 (1H, dt, J=12.7, 6.1 Hz), 3.97-4.14 (1H, m), 6.48 (1H, d, J=8.2 Hz), 6.65-6.77 (1H, m), 7.14-7.59 (9H, m), 7.96 (1H, brs)

- ¹H-NMR (CDCl₃) δ; 1.75-2.04 (2H, m), 2.27 (3H, s), 2.33 (6H, s), 3.48-3.58 (1H, m), 3.67 (1H, dt, J=12.7, 6.1 Hz), 3.98-4.16 (1H, m), 6.48 (1H, d, J=8.2 Hz), 6.72 (1H, dd, J=8.2, 1.9 Hz), 7.16 (1H, d, J=1.9 Hz), 7.27-7.91 (8H, m), 8.31 (1H, brs)
- 278) ¹H-NMR (CDCl₃) δ; 1.72-2.05 (2H, m), 2.28 (9H, s), 2.47 (3H, s), 3.16-4.34 (3H, m), 6.38-7.79 (10H, m), 8.37 (1H, brs)
- ¹H-NMR (CDCl₃) 6; 1.65-2.07 (2H, m), 2.28 (9H, s), 3.26-4.38 (3H, m), 6.34-8.06 (10H, m), 8.53 (1H, brs)
- 280) Two stereoisomers: # Both colorless amorphous $\frac{\text{Isomer } A:}{\left[\alpha\right]_{D}^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}$

¹H-NMR (CDCl₃) δ; 1.04 (3H, d, J=6.9 Hz), 1.59 (1H, brs), 2.25-2.45 (1H, m), 2.49 (3H, s), 2.52 (3H, s), 3.53-3.69 (2H, m), 3.91 (1H, abq, J=7.2, 12.9 Hz), 6.60 (1H, d, J=8.6 Hz), 6.93 (1H, dd, J=8.6, 2.5 Hz), 7.18-7.60 (9H, m), 7.76 (1H, brs)

Isomer B:

 $[\alpha]_D^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}$

¹H-NMR (CDCl₃) δ; 1.06 (3H, d, J=6.9 Hz), 1.60 (1H, brs), 2.21-2.43 (1H, m), 2.47 (3H, s), 2.52 (3H, s), 3.51-3.66 (2H, m), 3.93 (1H, abq, J=7.5, 12.9 Hz), 6.60-6.68 (1H, m), 6.95 (1H, dt, J=7.5, 1.8 Hz), 7.03 (1H, dt, J=7.4, 1.4 Hz), 7.17-7.55 (8H, m), 7.81 (1H, brs)

281) Two stereoisomers: Both colorless amorphous

Isomer A:

 $\left[\alpha\right]_{D}^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}$

¹H-NMR (CDCl₃) δ; 1.04 (3H, d, J=6.9 Hz), 1.55 (1H, brs), 2.23-2.46 (1H, m), 2.53 (3H, s), 3.53-3.67 (2H, m), 3.91 (1H, abq, J=7.1, 12.9 Hz), 6.61 (1H, d, J=8.6 Hz), 6.93 (1H, dd, J=8.6, 2.5 Hz), 7.28-7.52 (6H, m), 7.54-7.65 (2H, m), 7.70-7.79 (1H, m), 8.16 (1H, brs)

Isomer B:

 $[\alpha]_D^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}$

1_{H-NMR} (CDCl₃) & ; 1.06 (3H, d, J=6.9 Hz), 1.61
(1H, brs), 2.21-2.42 (1H, m), 2.51 (3H, s), 3.483.67 (2H, m), 3.90 (1H, abq, J=7.4, 12.9 Hz), 6.596.67 (1H, m), 6.94 (1H, dt, J=7.5, 1.9 Hz), 7.03
(1H, dt, J=7.4, 1.4 Hz), 7.23-7.75 (8H, m), 8.41
(1H, brs)

282) Two stereoisomers: Both colorless amorphous Isomer A:

283)

¹H-NMR (CDCl₃) δ; 1.03 (3H, d, J=6.5 Hz), 1.39 (1H, brs), 2.14-2.39 (1H, m), 2.45 (3H, s). 3.34-3.98 (3H, m), 6.53-7.98 (10H, m), 8.78 (1H, brs) ¹H-NMR (CDCl₃) δ; 1.05-1.25 (3H, m), 1.25-2.80 (10H, m), 3.00-5.10 (3H, m), 6.75-8.40 (11H, m)

- 285) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 1.00-2.80 (12H, m), 3.00-5.10 (3H, m), 6.70-7.80 (10H, m), 8.30-8.80 (1H, m) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 0.95-2.80 (15H, m), 2.80-5.15
- 286) ¹H-NMR (CDCl₃) & ; 0.95-2.80 (15H, m), 2.80-5.15 (3H, m), 6.70-7.05 (2H, m), 7.10-7.80 (10H, m), 7.95-8.45 (1H, m)
- 287) 1 H-NMR (CDC1₃) $_{\delta}$; 0.80-2.60 (16H, m), 2.60-5.05 (4H, m), 6.70-7.70 (10H, m), 7.85-8.40 (1H, m)
- 288) $^{1}\text{H-NMR}$ (CDCl₃) δ ; 1.30-2.60 (8H, m), 2.60-5.10 (3H, m), 6.60-7.95 (10H, m), 8.25-8.70 (1H, m)
- 289) $^{1}H-NMR$ (CDCl₃) $_{\delta}$; 1.27-4.91 (19H, m), 6.68-7.73 (10H, m), 8.40-8.71 (1H, m)
- 290) 1 H-NMR (CDCl₃) $_{6}$; 1.81-2.54 (6H, m), 2.15 (3H, s), 2.41 (3H, s), 2.46 (3H, s), 3.61-3.71 (3H, m), 6.91-7.43 (10H, m), 8.60 (1H, s)
- 291) $^{1}H-NMR$ (CDCl₃) & ; 1.86-2.50 (3H, m), 2.28 (9H, 5), 2.49 (3H, s), 6.60-7.47 (10H, m), 7.75 (1H, m)
- 292) $^{1}\text{H-NMR}$ (CDCl₃) $^{\delta}$; 1.15-2.55 (13H, m), 2.55-5.10 (3H, m), 6.60-8.40 (11H, m)
- 293) $^{1}\text{H-NMR}$ (CDCl₃) $_{\delta}$; 1.15-2.45 (10H, m), 2.55-5.10 (3H, m), 6.60-7.80 (10H, m), 8.30-8.70 (1H, m)
- 295) 1_{H-NMR} (CDCl₃) δ ; 1.15-2.50 (4H, m), 2.41 (6H, s), 2.60-5.20 (3H, m), 3.77 (3H, s), 6.50-7.50 (8H, m),

- 7.65-7.80 (lH, m), 8.31 (lH, d, J=8.4 Hz), 8.61 (lH, s)
- ¹H-NMR (CDCl₃) δ; 1.01-2.88, 3.22-4.41, 4.90-5.28 [total 18H, 1.17 (3H, t, J=7.2 Hz), 2.40 (3H, s), 3.77 (3H, s)], 6.55 (1H, d, J=8.1 Hz), 6.60-7.98 (8H, m), 8.23-8.75 (2H, m)
- 298)

 1 H-NMR (CDCl₃) 6; 1.00-3.04, 3.24-4.45. 4.91-5.27

 [total 21H, m, 1.17 (3H, t, J=7.0 Hz), 2.39 (3H, s), 2.50 (3H, s), 3.75 (3H, s)], 8.56 (1H, d, J=8.3 Hz), 6.69 (1H, d, J=8.3 Hz), 6.82-7.75 (7H, m), 8.05-8.49 (2H, m)
- ¹H-NMR (CDCl₃) 6; 1.21-4.62, 4.90-5.43 (total 15H, m), 5.70-6.11 (lH, m), 6.35-7.90 (9H, m), 8.07-8.92 (2H, m)
- 301) ${}^{1}\text{H-NMR} (CDCl_{3}) \delta$; 1.25-2.80 (14H, m), 3.00-5.10 (6H, m), 6.40-8.00 (11H, m)
- 302) ${}^{1}\text{H-NMR}$ (CDCl₃) & ; 1.30-2.90 (11H, m), 3.00-5.10 (6H, m), 6.40-7.80 (10H, m), 8.00-8.35 (1H, m)
- 1 H-NMR (CDCl₃) 6; 1.10-2.80 (16H, m), 2.85-5.15

```
(6H, m), 6.40-7.80 (11H, m)
          1_{H-NMR} (CDC1<sub>3</sub>) & ; 1.10-2.80 (13H, m), 2.90-5.10
304)
          (6H, m), 6.40-7.85 (10H, m), 7.90-8.20 (1H, m)
          1_{H-NMR} (CDCl<sub>3</sub>) \delta ; 1.27-5.28 (19H, m), 3.75 (3H,
305)
          s), 6.51 (1H, d, J=7.9 Hz), 6.69-6.81 (2H, m),
          7.05-7.49 (6H, m), 8.14 (1H, m), 8.27 (1H, d, J=8.4
          Hz)
          Two stereoisomers: Both colorless amorphous
306)
           Isomer_A:
           [\alpha]_D^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}
           1_{\text{H-NMR}} (CDC1<sub>3</sub>) & ; 0.78-1.02 (3H, m), 2.23-2.52
           (1H, m), 2.39 (6H, s), 2.48 (3H, s), 3.17-4.30 (3H,
           m), 6.85-7.84 (10H, m), 8.17 (1H, brs)
           Isomer B:
           [\alpha]_{0}^{22} = 0^{\circ} \text{ (chloroform, c=1.0)}
           1_{\text{H-NMR}} (CDCl<sub>3</sub>) \delta ; 0.73-1.00 (3H, m), 2.17-2.52
           (1H, m), 2.39 (6H, s), 2.49 (3H, s), 3.15-4.33 (3H,
           m), 6.36-7.55 (8H, m), 7.58-7.83 (2H, m), 8.19 (1H,
           brs)
           1_{H-NMR} (CDCl<sub>3</sub>) & ; 1.25-4.44, 4.98-5.41 [total 17H,
307)
           m, 2.40 (3H, s), 3.76 (3H, s)], 5.72-6.13 (1H, m),
           6.56 (lH, d, J=8.4 Hz), 6.69 (lH, d, J=7.9 Hz),
           6.77-7.93 (7H, m), 8.32 (1H, d, J=8.3 Hz), 8.49-
```

 l_{H-NMR} (CDCl₃) δ ; 1.23-5.42 (20H, m), 5.78-6.09

8.95 (1H, m)

308)

(1H, m), 6.56 (1H, d, J=8.3 Hz), 6.61-7.82 (8H, m), 8.14 (1H, s), 8.30 (1H, d, J=8 Hz)

- 309) ¹H-NMR (CDCl₃) δ; 1.20-2.70 (11H, m), 2.80-4.90 (9H, m), 6.40-7.70 (10H, m), 8.30-8.70 (1H, m)
- 310) ${}^{1}\text{H-NMR} (CDCl_{3}) \delta$; 1.20-2.80 (8H, m), 2.85-5.05 (9H, m), 6.40-7.80 (10H, m), 8.10-8.50 (1H, m)
- 311) ${}^{1}\text{H-NMR} (CDCl_{3}) \delta$; 1.20-2.75 (13H, m), 2.80-5.10 (9H, m), 6.40-8.00 (11H, m)
- 312) ¹H-NMR (CDCl₃) δ; 1.20-2.80 (10H, m), 2.90-5.10 (9H, m), 6.40-7.80 (10H, m), 8.00-8.40 (1H, m) Example 1189

By using di-p-toluoyl-L-tartaric acid monohydride or di-p-toluoyl-D-tartaric acid monohydride, the compound obtained in above Example 408 is optically resorbed to give the following compounds.

(+)-5-Dimethylamino-l-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine hydrochloride White amorphous

$$[\alpha]_{D}^{25} = +234^{\circ}$$
 (methanol, c=0.2)

Purity; more than 99 % ee, determined by HPLC using an optical acitive column

HPLC conditions;

Mobile phase; n-hexane : ethanol : diethylamine = 950 : 50 : 1

Flow rate; 1.0 ml/min.

Column; CHIRALCEL OD, 25 cm x 0.46 cm

(manufactured by Daicel Chemical Ind. Ltd.)

Concentration of sample; 0.1 % in methanol

Retention time; 34 minutes

 l_{H-NMR} (DMSO- d_6) & ; 0.85-1.20, 1.56-4.06, 4.94-5.21 (total 13H, m), 2.36 (3H, s), 6.79 (1H, d, J=7.6 Hz), 7.12-7.60 (8H, m), 7.62 (2H, d, J=8.4 Hz), 8.00 (1H, d, J=7.6 Hz), 10.43 (1H, s), 11.80 (1H, brs)

(-)-5-Dimethylamino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine hydrochloride White amorphous

$$[\alpha]_D^{25} = -23.1^{\circ} \text{ (methanol, c=0.2)}$$

Purity; more than 99 % ee, determined by HPLC using an optical active column, and the conditions are the same as above except that the retention time is 40 minutes.

 $l_{\rm H-NMR}$ (DMSO- d_6) δ ; 0.83-1.19, 1.55-4.06, 4.94-5.20 (total 13H, m), 2.36 (3H, s), 6.80 (1H, d, J=7.8 Hz), 7.12-7.60 (8H, m), 7.63 (2H, d, J=8.5 Hz), 8.00 (1H, d, J=7.8 Hz), 10.44 (1H, s), 11.74 (1H, brs)

Pharmacological Test

Experiment $1 : V_1$ receptor binding assay

Using rat liver plasma membrane preparations prepared according to Ichihara's method [cf: Akira Ichihara, J. Bio. Chem., 258, 9283 (1983)], the plasma membrane $(50000 \text{dpm}, 2 \times 10^{-10} \text{ M}) \text{ of } [^{3}\text{H}] - \text{Arg-vasopressin and a test}$ compound (60 μ g, 10^{-8} - 10^{-4} M) are incubated at 37°C for 10 minutes in 100 mM Tris-HCl buffer (pH: 8.0, 250 µl) containing 5 mM MgCl₂, 1 mM EDTA and 0.1 % BSA. After incubation, the mixture is filtered three times using the glass filter (GF/F) so as to separate the membrane preparation combined with vasopressin and then washed with the buffer (5 ml). This glass filter is taken out and mixed with liquid scintillation cocktail. The amount of [3H]vasopressin combined with the membrane is-measured by liquid scintillation counter and the rate of the inhibitory effect of the test compound is estimated according to the following equation.

Rate of the inhibitory effect (%) = $100 - \frac{c_1 - B_1}{c_0 - B_1} \times 100$

- C^1 : The amount of [3 H]-vasopressin combined with the membrane in the presence of the test compound (in prescribed amount).
- C^0 : The amount of [3 H]-vasopressin combined with the membrane in the absence of the test compound.

 B^1 : The amount of [3H]-vasopressin combined with the membrane in the presence of the excess amount of vasopressin (10^{-6} M).

The results are expressed as IC_{50} values, which is the concentration of the test compound required to achieve the inhibitory effect in the rate of 50 %.

The results are shown in the following Table 5. Test compound

- 1. 1-(4-Benzoylaminobenzoyl)-1,2,3,4-tetrahydroquinoline
- 2. 1-[4-(3-Chlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroguinoline
- 3. 1-[4-(3-Methoxybenzoylamino)benzoyl]-1,2,3,4tetrahydroquinoline
- 4. 1-[4-(3-Cyanobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 5. l-[4-(3-Aminobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 6. 1-[4-(2,3-Dimethylbenzoylamino)benzoyl]-1,2,3,4-tetrahydroguinoline
- 7. 1-[4-(2-Methylbenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 8. 1-[4-(2-Trifluoromethylbenzoylamino)benzoyl]1,2,3,4-tetrahydroquinoline
- 9. 1-[4-(2-Nitrobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline

- 10. 1-[4-(3,5-Dichlorobenzoylamino)benzoyl]1,2,3,4-tetrahydroquinoline
- 11. 1-[4-(3,3-Dimethylbutyrylamino)benzoyl]1,2,3,4-tetrahydroquinoline
- 12. 1-[4-(2-Cyclohexylacetylamino)benzoy1]-1,2,3,4-tetrahydroquinoline
- 13. 1-[4-(2-Phenylacetylamino)benzoyl]-1,2,3,4tetrahydroguinoline
- 14. 1-(4-Cyclohexylcarbonylaminobenzoyl)-1,2,3,4-tetrahydroguinoline
- 15. l-(4-Cycloheptylcarbonylaminobenzoyl)-1,2,3,4-tetrahydroquinoline
- 16. l-(4-Cyclooctylcarbonylaminobenzoyl)-1,2,3,4tetrahydroquinoline
- 17. 1-(4-Tricyclo[3.3.1.1]decanylcarbonylaminobenzoyl)-1,2,3,4-tetrahydroquinoline
- 18. l-[4-(α -Naphthylcarbonylamino)benzoyl]-1,2,3,4-tetrahydroguinoline
- 19. l-{4-(3-Thenoyl)benzoyl}-1,2,3,4-tetrahydroquinoline
- 20. l-[2-(β -Naphthylcarbonylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 21. 1-[4-(4-Methoxyanilinocarbonyl)benzoyl]1,2,3,4-tetrahydroquinoline
- 22. 1-[4-(2-Methylanilinocarbonyl)benzoyl]-1,2,3,4-tetrahydroquinoline

23.	1-[4-(3-Chloroanilinocarbonyl)benzoyl]-1,2,3,4-
tetrahydroqui	noline

24. 1-[4-(3,5-Dichloroanilinocarbonyl)benzoyl]1,2,3,4-tetrahydroquinoline

25. l-(4-Cyclohexylaminocarbonylbenzoyl)-1,2,3,4-tetrahydroquinoline

26. l-(4-Cyclohexylcarbonylaminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine

27. l-(4-Benzoylaminobenzoyl)-2,3,4,5-tetrahydrolH-benzazepine

28. l-[4-(2-Methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

29. 1-[4-(3-Methoxybenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

30. 1-[4-(3-Chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

31. 1-[4-(3-Cyanobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

32. 1-[4-(3,5-Dichlorobenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine

33. 1-[4-(2,3-Dimethylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine

34. 1-(4-Cyclohexylcarbonylaminobenzoyl)-

35. 1-(4-Benzoylaminobenzoyl)-1,2,3,4,5,6-hexahydrobenzazocine

المنافع والمراجع والمراجع والمنافع والمنطوب

المتعدد والمراجين

1,2,3,4,5,6-hexahydrobenzazocine

```
36. 1-[4-(2-Methylbenzoylamino)benzoyl]-
1,2,3,4,5,6-hexahydrobenzazocine
         37. 1-[4-(3-Methoxybenzoylamino)benzoyl]-
1,2,3,4,5,6-hexahydrobenzazocine
       38. l-[4-(2,3-Dimethylbenzoylamino)benzoyl]-
1,2,3,4,5,6-hexahydrobenzazocine
        39. 1-[4-(3,5-Dichlorobenzoylamino)benzoyl}-
1,2,3,4,5,6-hexahydrobenzazocine
        40. 1-(4-Cyclohexylcarbonylaminobenzoyl)-1,2,3,5-
tntrahydro-4,1-benzoxazepine
        41. 1-[4-(3-Methylbenzoylamino)benzoyl]-1,2,3,5-
tetrahydro-4,1-benzoxazepine
        42. l-[4-(2,3-Dimethylbenzoylamino)benzoyl]-
1,2,3,5-tetrahydro-4,1-benzoxazepine
        43. 1-[4-(3,5-Dichlorobenzoylamino)benzoyl]-
1,2,3,5-tetrahydro-4,1-benzoxazepine
        44. 3-Methyl-1-(4-cyclohexylcarbonylaminobenzoyl)-
1,2,3,4-tetrahydroquinoline
        45. 3-Methyl-1-(4-benzoylaminobenzoyl)-1,2,3,4-
tetrahydroquinoline
        46. 3-Methyl-1-[4-(2-methylbenzoylamino)benzoyl]-
1,2,3,4-tetrahydroquinoline
        47. 3-Methyl-1-[4-(3-methoxybenzoylamino)benzoyl}-
1,2,3,4-tetrahydroquinoline
```

48. 3-Methyl-1-[4-(2,3-dimethylbenzoylamino)-

benzoyl]-1,2,3,4-tetrahydroquinoline

49. 3-Methyl-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroguinoline 50. 4-Methyl-1-(4-cyclohexylcarbonylaminobenzoyl)-1,2,3,4-tetrahydroquinoxaline 51. 4-Methyl-1-[4-(2-methylbenzoylamino)benzoyl}-1,2,3,4-tetrahydroquinoxaline 52. 4-Methyl-1-[4-(2,3-dimethylbenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoxaline 53. 4-Methyl-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoxaline 54. 2-Methyl-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline 55. 4-Methyl-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline 56. 1-[4-(2-Bromobenzoylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine 57. l-[4-(3-Nitrobenzoylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine 58. 1-[4-(3-Trifluoromethylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine 59. 1-[4-(3-Ethoxybenzoylamino)benzoyl]-2,3,4,5tetrahydro-lH-benzazepine 60. 1-[4-(3,5-Dimethoxybenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 61. 1-[4-(2-Chloro-4-nitrobenzoylamino)benzoyl]-

2,3,4,5-tetrahydro-1H-benzazepine

- 62. l-[4-(2,4-Dichlorobenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine
- 63. l-[4-(2-Chloro-6-fluorobenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine
- 64. 1-[4-(2,6-Dimethylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine
- 65. l-[4-(2-Chloro-4-aminobenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine
- 66. 1-[4-(2-Chloro-4-acetylaminobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 67. l-[4-(3-Aminobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 68. l-{4-[2-(4-Isopropylaminobutoxy)benzoylamino}-benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine hydrochloride
- 69. 1-[4-(3-Hydroxybenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 70. 1-{4-[2-(4-Aminobutoxy)benzoylamino]benzoyl}2,3,4,5-tetrahydro-lH-benzazepine
- 71. 1-{4-[2-(2-Diethylaminoethoxy)benzoylamino}-
- benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine hydrochloride
- 72. 1-{4-[2-(4-Acetylaminobutoxy)benzoylamino]-
- benzoy1}-2,3,4,5-tetrahydro-1H-benzazepine
 - 73. l-{4-[2-(6-Phthalimidohexyloxy)benzoylamino]-
- benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
- 74. l-{4-[2-(6-Morpholinohexyloxy)benzoylamino]-benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine

- 75. 1-{4-[2-(6-[4-Methyl-1-piperazinyl]hexyloxy)-benzoylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine-dihydrochloride
- 76. l-(3-Methoxy-4-cyclohexylcarbonylaminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine
- 77. 1-(3-methoxy-4-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 78. 1-[3-Methyl-4-(2-methylbenzoylamino)benzoy]2,3,4,5-tetrahydro-lH-benzazepine
- 79. 4-Methyl-1-(4-cyclohexylcarbonylaminobenzoyl)2,3,4,5-tetrahydro-lH-1,4-benzodiazepine hydrochloride
- 80. 4-Methyl-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine hydrochloride
- 8I. 4-Methyl-I-[4-(2,3-dimethylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine hydrochloride
- 82. 4-Methyl-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-1,4-benzodiazepine hydrochloride
- 83. 4-Methyl-1-[4-(3-methoxybenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-1,4-benzodiazepine
- 84. 4-Methyl-1-[4-(3-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine
- 85. 4-Methyl-1-[4-(2,3,5-trichlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine
 - 86. 4-Propyl-1-[4-(2,3-dimethylbenzoylamino)-

```
benzoyl]-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine hydrochloride
```

- 87. 5-Methyl-1-(4-benzoylaminobenzoyl)-2,3,4,5-tetrahydro-1H-1,5-benzodiazepine
- 88. 5-Methyl-1-(4-cyclohexylcarbonylaminobenzoyl)
 -2,3,4,5-tetrahydro-1H-1,5-benzodiazepine
- 89. 5-Methyl-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-1,5-benzodiazepine
- 90. 5-Methyl-1-[4-(2-methylbenzoylamino)benzoyl]
- -2,3,4,5-tetrahydro-1H-1,5-benzodiazepine
- 91. 5-Methyl-1-[4-(2,3-dimethylbenzoylamino)-
- benzoyl]-2,3,4,5-tetrahydro-lH-1,5-benzodiazepine

 92. 4-Methyl-1-[3-methoxy-4-(3,5-dichlorobenzoyl-
- amino)benzoyl]-2,3,4,5-tetrahydro-1H-1,4-benzodiazepine
- 93. 3-(1-Pyrrolidinyl)-1-[4-(2,3-dimethylbenzoyl-amino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 94. 6-Methyl-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline
- 95. 6-Methoxy-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline
- 96. 3-Hydroxymethyl-1-[4-(3,5-dichlorobenzoyl-amino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 97. 4-Methylamino-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline
- 98. 3-Amino-1-[4-(3,5-dichlorobenzoylamino)-benzoyl]-1,2,3,4-tetrahydroquinoline

```
99. 3-Acetylamino-1-[4-(3,5-dichlorobenzoylamino)-
benzoyl]-1,2,3,4-tetrahydroquinoline
         100. 4-Dimethylamino-1-[4-(3,5-dichlorobenzoyl-
amino)benzoyl]-1,2,3,4-tetrahydroquinoline
         101. 1-[4-(2-t-Butylaminoacetylamino)benzoyl]-
2,3,4,5-tetrahydroquinoline-lH-benzazepine
         102. 1-\{4-[2-(N-Cyclohexyl-N-ethyl)acetylamino]-
benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
         103. 1-{4-[2-(1-Piperidinyl)acetylamino]benzoyl}-
2,3,4,5-tetrahydro-lH-benzazepine
         104. 1-[4-(2-Phenoxyacetylamino)benzoy1]-2,3,4,5-
tetrahydro-lH-benzazepine
         105. l-[4-(2-Phthalimidoacetylamino)benzoyl]-
2,3,4,5-tetrahydro-lH-benzazepine
         106. 1-{4-[2-(1,1-Dimethyl-2-phenoxyethyl)amino-
acetylamino]benzoy1}-2,3,4,5-tetrahydro-1H-benzazepine
         107. 1-{4-(2-(3-Methylphenoxy)acetylamino)benzoyl}-
2,3,4,5-tetrahydro-lH-benzazepine
         108. 1-{4-[2-(3-Methoxyanilino)acetylamino]-
benzoy1}-2,3,4,5-tetrahydro-1H-benzazepine
         109. 1-{4-[2-(8-Naphthyloxy)acetyamino]benzoyl}-
2,3,4,5-tetrahydro-lH-benzazepine
         110. 1-{4-[2-(4-Methylanilino)acetylamino]benzoyl}-
2,3,4,5-tetrahydro-lH-benzazepine
         111. 1-\{4-[2-(3-Methoxyphenoxy)acetylamino]-
benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
```

- 112. 1-[4-(4-Pyridylcarbonylaminobenzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
 - 113. $1-\{4-[2-(2,4-Dimethylanilino)acetylamino]-$
- benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine
 - 114. 1-{4-[2-(N-Ethylanilino)acetylamino]benzoyl}-
- 2,3,4,5-tetrahydro-1H-benzazepine
 - 115. 1-{4-[2-(N-Allylanilino)acetylamino]benzoyl}-
- 2,3,4,5-tetrahydro-lH-benzazepine
 - 116. 1-{4-{2-(2-Chloroanilino)acetylamino}benzoy1}-
- 2,3,4,5-tetrahydro-lH-benzazepine
 - 117. 1-{4-[2-(4-Acetyloxybutoxy)benzoylamino}-
- benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
 - 118. 1-[4-(2-Carboxymethoxybenzoylamino)benzoyl]-
- 2,3,4,5-tetrahydro-lH-benzazepine
- 119. 1-[4-(2-Carbamoylmethoxybenzoylamino)benzoyl]-
- 2,3,4,5-tetrahydro-lH-benzazepine
 - 120. 1-{4-[2-(4-Hydroxybutoxy)benzoylamino]-
- benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
 - 121. 1-[4-(2-Ethoxycarbonylmethoxybenzoylamino)-
- benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
 - 122. 6-Fluoro-1-[4-(3,5-dichlorobenzoylamino)-
- benzoyl]-1,2,3,4-tetrahydroguinoline
 - 123. 6-Fluoro-1-{4-[di-(3,5-dichlorobenzoyl)amino]-
- benzoyl}-1,2,3,4-tetrahydroquinoline
 - 124. 1-[4-(2-Diethylaminocarbonylmethoxybenzoyl-
- amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

125. $1-\{4-[2-(2-[(N-(2-hydroxyethyl)-N-methyl-amino]ethoxy)benzoylamino]benzoyl\}-2,3,4,5-tetrahydro-lH-benzoazepine hydrochloride$

126. 1-[4-(2-Methylanilinocarbonylamino)benzoyl]-.
2,3,4,5-tetrahydro-lH-benzazepine

127. l-[4-(2-Chlorophenylsulfonylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine

128. 1-{4-[2-(4-Aminomethylanilino)acetylamino}-benzoyl}-2,3,4,5-tetrahydro-lH-benzázepine

129. 1-{4-[2-(N-Phenyl-N-(3-acetylaminopropyl)-amino)acetylamino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine

130. 1-{4-{2-(N-Phenyl-N-propargylamino)acetyl-amino]benzoyl}-2,3,4,5-tetrahydro-lH-benzazepine

131. 4-(N-Methyl-N-ethylamino)-1-[4-(3,5-dichloro-benzoylamino)benzoyl]-1,2,3,4-tetrahydroguinoline

132. 5-Dimethylamino-1-[4-(2,4-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

133. 4-Dimethylamino-1-[3-methoxy-4-(2-methyl-benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline

134. 5-Dimethylamino-1-[3-methoxy-4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

135. 1-[4-(2,3-Dimethylbenzoylamino)benzoyl]-4-ethyl-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine

136. l-[4-(3,5-Dichlorobenzoylamino)benzoyl]-4isopropyl-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine
137. l-[4-(2-Methylbenzoylamino)benzoyl]-5-methyl-

- 1,2,3,4,5,6-hexahydro-1,5-benzodiazocine
- 138. l-[4-(2-Methylbenzoylamino)benzoyl]-1,2,3,4-tetrahydro-5,1-benzoxazepine
- 139. 5-Oxo-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 140. 4-Methyl-1-[2-chloro-4-(3,5-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine
- 141. 5-Methylamino-1-[4-(3,5-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 142. 5-(N-Acetyl-N-methylamino)-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 143.5-Hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 144. 4-Dimethylamino-1-[3-methoxy-4-(2,3-dimethyl-benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 145. 4-Dimethylaminomethyl-1-[4-(2-methylbenzoyl-amino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 146. 4-Dimethylaminomethyl-1-[4-(3,5-dichloro-benzoylamino)benzoyl]-1,2,3,4-tetrahydroquinoline
- 147. 5-Methoxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 148. 4-Methyl-1-[3-methyl-4-(2,4-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine
- 149. 5-Methoxy-1-[4-(2,4-dichlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
 - 150. 4-Dimethylamino-1-[4-(2-chlorobenzoylamino)-

benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

151. 4-Acetyloxy-1-[4-(2-methylbenzoylamino)-

benzoyl]-1,2,3,4-tetrahydroquinoline

152. 5-Hydroxyimino-1-[4-(3,5-dichlorobenzoyl-

amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

153. 5-Acetyloxy-1-[4-(2-chlorobenzoylamino)-

benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

154. 5-Ethoxycarbonylmethoxy-1-[4-(2,4-dichloro-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

155. 4-Allylamino-1-[4-(3,5-dichlorobenzoylamino)-

benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

156. 5-Dimethylamino-1-[3-methoxy-4-(2,3,5-

trichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-

benzazepine

157. 4-[4-(2-Methylbenzoylamino)benzoyl]-3,4-

dihydro-2H-1,4-benzothiazine

158. 5-Dimethylamino-1-[2-chloro-4-(2-methyl-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

159. 5-Dimethylamino-l-[4-(2-methylanilino-

carbonyl)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

160. 5-Ethoxycarbonylmethoxy-l-[4-(2-methyl-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

161. 5-(4-dimethylaminobutoxy)-1-{4-(2-methyl-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

162. 5-Carboxymethoxy-1-[4-(2-chlorobenzoyl-

amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

- 163. 5-Dimethylaminocarbonylmethoxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 164. 5-Carbamoylmethoxy-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 165. 5-Dimethylamino-1-[3-ethoxy-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 166. 5-[4-(2-Methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1,5-benzothiazepine
- 167. 5-Amino-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine
- 168. 5-Dimethylamino-1-[3-hydroxy-4-[2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 169. 5-n-Propylamino-l-[4-(2,4-dichlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 170. 5-Dimethylamino-1-[3-benzyloxy-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 171. 5-[4-(2-Methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1,5-benzothiazepin-1-oxide
- 172. 5-[3-(Phthalimid-1-yl)-propoxy]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 173. 5-(3-Aminopropoxy)-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 174. 5-(3-Acetylaminopropoxy)-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

175. 5-Dimethylamino-1-[2-chloro-4-(2-t-butylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 176. 5-Methylamino-1-[2-chloro-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 177. 5-Dimethylamino-l-[2-methoxy-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 178. 5-Hydroxy-1-[4-(3,5-dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 179. 5-Dimethylamino-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 180. 5-Dimethylamino-1-[4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 181. 5-Methylamino-1-{4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine 182. 5-Methylamino-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 183. 5-Dimethylamino-1-[2-methoxy-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 184. 5-Dimethylamino-1-[2-chloro-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine 185. 5-Methylamino-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl-2,3,4,5-tetrahydro-lH-benzazepine 186. 5-Cyclopropylamino-1-[2-chloro-4-(2,4dichlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine

187. 5-Dimethylaminocarbonyloxy-1-[4-(2-methyl-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

188. 5-Dimethylamino-1-[4-(2-trifluoromethylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

189. 5-Dimethylamino-1-[3-(2-chlorobenzoyloxy)-4-.
(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine

190. 5-(N-Methyl-N-Allylamino)-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

- 191. 5-Carbamoyloxy-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 192. l-[4-(2-Methylbenzoylamino)benzoyl]-1,2,3,5-tetrahydro-4,1-benzothiazepine
- 193. 4-Oxo-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine
- 194. l-[4-(2-Methylbenzoylamino)benzoyl]-1,2,3,5-tetrahydro-4,1-benzothiazepine-1,1-dioxide
- 195. 5-Methylaminocarbonylmethoxy-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 196. 5-Methylaminocarbonyloxy-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahyro-1H-benzazepine
- 197. 5-Dimethylamino-1-[2-dimethylamino-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 198. 5-Methylamino-l-[2-chloro-4-(2-trifluoro-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-

```
benzazepine
```

199. 5-Cycloropropylamino-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

200. 5-Cyclopropylamino-1-[2-chloro-4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

201. 5-Allylamino-1-[2-chloro-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

202. 5-(1-Piperidinyl)-1-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

203. 5-(4-Benzyl-1-piperazinyl)-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

204. 5-(1-Pyrrolidinyl)-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

205. 5-(4-Acetyl-1-piperazinyl)-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

206. 5-(4-Methyl-1-piperazinyl)-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

207. 1-[4-(2-Chlorobenzoylamino)benzoyl]-2,3-dihydro-lH-benzazepine

208. 5-Methyl-1-[4-(2-methylbenzoylamino)benzoyl]2,3,4,5-tetrahydro-lH-benzazepine

209. 5-Methylidene-l-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

210. 5-Hydroxy-l-[2-chloro-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

- 211. 5-(1-Morpholino)-1-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 212. 5-Dimethylamino-1-[4-(2-fluorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 213. 5-Dimethylamino-l-[4-(2,4-difluorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 214. 4-Hydroxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 215. 5-Hydroxymethyl-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 216. 5-Dimethylamino-4-hydroxy-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 217. 1-[4-(2-Methylbenzoylamino)benzoyl]-1,2-dihydroquinoline
- 218. 5-Dimethylamino-1-[2-methyl-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine (the compound of Example 979)
- 219. 5-Dimethylamino-1-[2-methyl-4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 220. 5-Dimethylamino-1-[2-methyl-4-(2,4-dichloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 221. 5-Methylamino-1-{2-chloro-4-{2-(N-ethyl-anilino)acetylamino}benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
- 222. 5-Hydroxy-1-{2-chloro-4-[2-(N-ethylanilino)-acetylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
 223. 5-Dimethylamino-1-[2-fluoro-4-(2-chloro-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

224. 5-Methylamino-4-hydroxy-1-[2-chloro-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine hydrochloride

225. 5-Hydroxymethyl-5-hydroxy-l-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

226. 5-Dimethylamino-1-[2-fluoro-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

227. 5-Dimethylamino-1-[3-methyl-4-(2,4-dichloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

228. 5-(N-Methyl-N-ethylamino)-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

229. 5-Ethylamino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

230. 5-Dimethylamino-1-[4-(3,5-difluorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

231. 5-Acetyloxymethyl-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

232. 5-Dimethylamino-1-[3-fluoro-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

233. 4,4-Dimethoxy-1-[4-(2-methylbenzoylamino)-

benzoy1]-2,3,4,5-tetrahydro-1H-benzazepine

A production of the contract of

234. 5-Acetyloxyimino-1-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

235. 5-Methylsulfonyloxymethyl-1-[4-(2-methyl-

- benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

 236. 5,5-Epoxy-1-[4-(2-methylbenzoylamino)benzoyl]
 2,3,4,5-tetrahydro-1H-benzazepine
- 237. 5-Hydroxymethyl-5-hydroxy-l-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 238. 5-Hydroxy-1-[2-methoxy-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 239. 5-Dimethylamino-1-[4-(2-carbamoylmethoxy-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 240. 5-Hydroxy-6-methyl-1-[2-chloro-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 241. 5-(2-Dimethylaminoethyl)amino-1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 242. 5-Hydroxymethyl-5-methylamino-l-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 243. 5-Methylaminomethyl-5-hydroxy-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 244. 5-Aminomethyl-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 245. 5-[N-Methyl-N-(3-methoxy-2-hydroxypropyl)-amino]-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetra-hydro-lH-benzazepine
- 246. 5-[N-Methyl-N-(3-diethylamino-2-hydroxy-propyl)amino]-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

[1,4]oxazine

247. 5-Dimethylamino-1-[3-methoxy-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

248. 5-Dimethylamino-1-[3-methoxy-4-(2,4-dichloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

249. 5-Dimethylamino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine · hydrochloride

250. 5-Azidomethyl-l-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

251. 7-[4-(2-Chlorobenzoylamino)benzoyl]-l-methyl-1,2,3,4a,5,6,7,llb-octahydro-3-oxo[1]benzazepino[4,5-b]-

252. 5-Benzylamino-l-[4-(2-chlorobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

253. 5-Amino-1-[4-(2-chlorobenzoylamino)benzoyl]2,3,4,5-tetrahydro-1H-benzazepine

254. 5-Dimethylamino-4-methyl-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

255. 5-Acetylaminomethyl-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

256. 5-Hydroxy-4-methyl-l-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine

257. 5-[2-(2-Pyridyl)ethylamino]-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

258. 5-(N-Methyl-N-methanesulfonylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-

benzazepine

- 259. 5-(N-Methyl-N-benozylamino)-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 260. 5-Ethoxycarbonylamino-l-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 261. 5-Methyl-5-hydroxy-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 262. 5-(N-Methyl-N-ethoxycarbonylmethylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 263. 5-Cyclopentylamino-1-[2-chloro-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 264. 5-[N-Methyl-N-(2,3-dihydroxypropyl)amino]-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 265. 5-(N-Methyl-N-cyanomethylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 266. 5-(N-Methyl-N-carbamoylmethylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 267. 5-{N-Methyl-N-[3-(3,4,5,6-tetrahydro-2H-pyran-2-yloxy)propyl]amino}-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 268. 5-Dimethylaminomethyl-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

- 269. 5-Formylaminomethyl-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 270. 5-[N-Methyl-N-(3-acetyloxypropyl)amino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine
- 271. 5-[N-Methyl-N-(3-hydroxypropyl)amino]-1-[4-(2methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 272. Potassium {1-[2-chloro-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepin-5-yl}iminoo-sulfonate
- 273. 5-Dimethylamino-l-(4-benzoylaminobenzoyl)-2,3,4,5-tetrahydro-lH-benzazepine
- 274. 5-(1-Benzyl-4-piperidinyl)amino-1-[4-(2methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine
- 275. 5-(2-Dimethylaminoacetyloxy)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 276. 5-Dimethylamino-1-[4-(3-methoxybenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 277. 5-[(4-Methyl-1-piperazinyl)carbonylmethoxy]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lHbenzazepine
- 278. 5-Morpholinocarbonylmethoxy-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 279. 5-Thiomorpholinocarbonylmethoxy-1-[4-(2methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-

benzazepine

- 280. 5-Anilinocarbonylamino-l-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 281. 5-(1-Oxothiomorpholino)carbonylmethoxy-1-[4-. (2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 282. 5-Hydrazino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 283. 5-Methylaminocarbonylamino-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 284. $5-[(2-\alpha-Carbamoyl-1-pyrrolidinyl)carbonyl-methoxy]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine$
- 285. 5-(Carbamoylmethylaminocarbonylmethoxy)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 286. 5-(1,1-Dioxothiomorpholino)carbonylmethoxy-1[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine
- 287. 7-Chloro-5-methylamino-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 288. 5-[(4-Acetyl-1-piperazinyl)carbonylmethoxy]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 289. 5-Dimethylamino-1-[4-(3-nitrobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

- 290. 5-[(4-Pyridyl)methylaminocarbonylmethoxy]-1[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1Hbenzazepine
- 291. 5-[2-(Methylamino)acetylamino]-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 292. 5-Dimethylamino-1-[4-(3-aminobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 293. $5-\{[N-Methyl-N-(2-hydroxyethyl)amino\}carbonyl-methoxy\}-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetra-hydro-lH-benzazepine$
- 294. 5-Dimethylamino-1-[3-(2-diethylaminoethoxy)-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 295. 5-[N-Methyl-N-(dimethylaminocarbonylmethyl)-amino]-l-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 296. Potassium 2-[N-methyl-N-{1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepin-5-yl}amine]acetate
- 297. 5-{N-Methyl-N-[2-(1-imidazolyl)acetyl]amino}1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-lHbenzazepine
- 298. 5-Dimethylamino-1-[4-(2-dimethylaminobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
 - 299. 5-[(2-Aminoacetyl)amino]-1-[4-(2-methyl-

- benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

 300. 5-Dimethylamino-1-[4-(3-acetylaminobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 301. 5-(2-t-Butoxycarbonylaminoacetylamino)-1-[4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 302. 5-Methylamino-7-chloro-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 303. 5-Dimethylamino-7-chloro-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 304. 5-Dimethylamino-7-chloro-1-[4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 305. 5-Dimethylamino-1-[4-(phenylacetylamino)-benzoy1]-2,3,4,5-tetrahydro-1H-benzazepine
- 306. 5-Dimethylamino-1-[4-(3-phenylpropionylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
- 307. 5-Methylamino-7-chloro-1-{4-[(N-ethylanilino)-acetylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine
- 308. 5-Dimethylamino-7-chloro-1- $\{4-[(N-ethyl-anilino)acetylamino]benzoy1\}-2,3,4,5-tetrahydro-1H-benzazepine$
- 309. 5-Dimethylamino-l-[4-(2-bromobenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine
- 310. 5-Cyclopropylamino-7-chloro-1-[4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine
 311. 5-Cyclopropylamino-7-chloro-1-[4-(2-chloro-

benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

312. 5-hydroxy-1-{4-[2-(4-isopropylaminobutoxy)-benzoylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine

313. 5-Dimethylaminocarbonylmethoxy-1-{4-{(N-ethyl-anilino)acetylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine

314. 5-(N-Methyl-N-ethylamino)-1-[2-chloro-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

315. 5-Dimethylamino-1-{4-[(2-chloroanilino)acetyl-amino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine

316. 5-Dimethylamino-1-{4-{(2-methylanilino)acetyl-amino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine

317. 5-Dimethylamino-1-{4-[(N-methyl-2-methyl-anilino)acetylamino]benzoyl}-2,3,4,5-tetrahydro-1H-benzazepine

318. 5-Methylamino-9-chloro-1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

319. 5-Dimethylamino-1-[4-(phenoxyacetylamino)-benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

320. 6-Methylamino-l-[4-(2-methylbenzoylamino)-benzoyl]-1,2,3,4,5,6-hexahydrobenzazocine

321. 5-Methylamino-7-chloro-1-[3-methoxy-4-(2-chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

322 5-Cyclopropylamino-7-chloro-1-[3-methoxy-4-(2-

chlorobenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

323. 5-Methylamino-7-chloro-1-[3-methoxy-4-(2-methylbenzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine

,

Table 10

Test Comp.	IC ₅₀	Test Comp.	IC ₅₀
No.	(µ M)	No.	(µM)
26	0.071	78	0.10
27	0.095	88	0.34
28	0.056	90	0.38
29	0.15	114	0.011
30	0.15	. 115	0.012
32	0.30	116	0.04
33	0.092	117	0.22
36	0.41	119	0.049
46	0.40	120	0.29
56	0.025	121	0.45
57	0.46	124	0.15
58	0.40	125	0.091
59	0.31	130	0.023
60	0.18	143	0.15
62	0.098	147	0.28
63	0.14	161	0.14
64	0.069	163	0.22
67	0.34	164	0.15
68	0.013	172	0.26
69	0.066	173	0.15
70	0.041	174	0.14
71	0.18	187	0.45
72	0.12	188	0.47
74	0.10	192	0.054
75	0.069	193	0.17
76	0.042	195	0.17
77	0.085	196	0.40

2

Test	IC ₅₀	Test	1C ₅₀
Comp. No.	(Mu)	Comp. No.	(µM)
207	0.16	284	0.29
208	0.11	285	0.18
209	0.074	286	0.40
214	0.27	287	0.064
215	0.13	288	0.26
222	0.096	290	0.21
231	0.16	293	0.19
235	0.088	298	0.29
236	0.16	302	0.071
238	0.39	303	0.19
244	0.23	304	0.21
250	0.19	307	0.024
252	0.36	308	0.11
255	0.046	309	0.43
256	0.049	310	0.065
266	0.29	311	0.078
269	0.48	312	0.056
274	0.11	313	0.032
275	0.18	315	0.38
277	0.23	316	0.47
278	0.30	321	0.059
279	0.15	322	0.044
280	0.47	323	0.064
281	0.18		

manage, and and one

Pharmacological Test

Experiment 2 : V₂ receptor binding assay

Using rat kidney plasma membrane preparations prepared according to O. Hechter's method [cf: J. Bio. Chem., 253, 3211 (1978)], the plasma membrane (100000dpm, 4×10^{-10} M) of [3 H]-Arg-vasopressin and a test compound (0.6 mg, 10^{-10} - 10^{-5} M) are incubated at 4°C for 3 hours in 100 mM Tris-HCl buffer (pH: 8.0, 250 μ l) containing 5 mM MgCl₂, 1 mM EDTA and 0.1 % BSA. After incubation, the mixture is filtered using the glass filter (GF/F) so as to separate the membrane preparation combined with vasopressin and then washed twice with the buffer (5 ml). This glass filter is taken out and mixed with liquid scintillation cocktail. The amount of [3 H]-vasopressin combined with the membrane is measured by liquid scintillation counter and the rate of the inhibitory effect of the test compound is estimated according to the following equation.

Rate of the inhibitory effect (%) = $100 - \frac{C_1 - B_1}{C_0 - B_1} \times 100$

- C¹: The amount of [³H]-vasopressin combined with the membrane in the presence of the test compound (in prescribed amount).
- C⁰: The amount of [³H]-vasopressin combined with the membrane in the absence of the test compound.
- B^1 : The amount of [3H]-vasopressin combined

with the membrane in the presence of the excess amount of vasopressin (10^{-6} M).

The results are expressed as ${\rm IC}_{50}$ values, which is the concentration of the test compound required to achieve the inhibitory effect in the rate of 50 %.

The results are shown in the following Table 6.

Table 11

Test Comp.	1C ₅₀	Test Comp.	IC ₅₀	
No.	(µM)	No.	(µM)	
1	0.98	28	0.018	
2	0.20	29	0.069	
3	0.40	30	0.029	
4	0.58	31	0.098	
5	1.2	32	0.016	
6	0.076	33	0.007	
7	0.20	34	0.049	
8	0.32	35	0.20	
9	0.53	36	0.028	
10	0.082	37	0.16	
11	1.05	38 -	0.029	
12	1.97	39	0.071	
13	1.02	40	0.33	
14	0.23	41	0.20	
15	0.13	42	0.063	
16	0.17	43	0.17	
17	0.23	44	0.050	
18	1.0	45	0.19	
19	1.7	46	0.018	
20	1.4	47	0.20	
21	1	48	0.021	
22	0.33	49	0.063	
23	1.07	50	1.3	
24	1.09	51	0.40	
25	1.67	52	0.32	
26	0.025	53	1.6	
27	0.070	54	0.11	
			* '	

Test Comp.	IC ₅₀	Test Comp.	1C ₅₀
No.	(µM)	No.	(M M)
55	0.091	86	0.58
56	0.037	87	0.046
57	0.16	88	0.021
58	0.14	89	0.035
59	0.24	90	0.014
60	0.15	91	0.005
61	0.090	92	0.41
62	0.023	93	0.52
63	0.046	94	0.095
64	0.007	95	0.089
65	0.081	96	0.039
66	0.45	97	0.024
67	0.050	98	0.45
68	0.19	99	1.6
69	0.12	100	0.011
70	0.012	101	0.60
71	0.085	102	0.29
72	0.16	103	0.54
74	0.51	104	0.37
75	0.30	105	0.72
76	0.017	106	0.44
77	0.090	107	0.032
78	0.084	108	0.12
79	0.53	109	0.49
80	0.070	110	0.044
81	0.15	111	0.087
82	0.17	112	0.29
83	0.73	113	0.28
84	0.11	114	0.006
85	0.068	115	0.006

.

Test Comp.	IC ₅₀	Test Comp.	IC ₅₀	
No.	(µM)	No.	(µM)	
116	0.039	146	0.056	<u> </u>
117	0.24	147	0.009	
118	0.55	148	0.34	
119	0.059	149	0.004	
120	0.28	150	0.14	
121	0.18	151	0.18	
122	0.10	152	0.039	
123	0.10	153	0.063	
124	0.13	154	0.063	
125	0.28	155	0.028	-
126	0.062	156	0.15	
127	0.99	157	0.38	
128	0.23	158	0.018	
129	0.29	159	0.020	
130	0.007	160	0.020	
131	0.027	161	0.009	
132	0.013	162	0.059	
133	0.022	163	0.009	
134	0.048	164	0.010	
135	0.081	165	0.098	
136	0.18	166	0.070	
137	0.41	167	0.032	
138	0.11	168	0.083	
139	0.10	169	0.071	
140	0.024	170	0.25	
141	0.010	171	0.87	
142	0.008	172	0.023	
143	0.008	173	0.008	
144	0.02	174	0.007	
145	0.06	175	0.038	

ř

s a grande

Test Comp.	1C ₅₀	Test Comp.	IC ₅₀
No.	(µM)	No.	(µM)
176	0.004	206	0.088
177	0.15	207	0.045
178	0.012	208	0.007
179	0.040	209	0.004
180	0.034	210	0.004
181	0.038	211	0.12
182	0.005	212	0.035
183	0.26	213	0.033
184	0.023	214	0.058
185	0.005	215	0.006
186	0.030	216	0.91
187	0.029	217	0.37
188	0.039	218	0.022
1,89	0.087	219	0.023
190	0.082	220	0.026
191	0.009	221	0.024
192	0.011	222	0.010
193	0.036	223	0.022
194	0.21	224	0.38
195	0.010	225	0.030
196	0.013	226	0.019
197	0.99	227	0.029
198	0.040	228	0.029
199	0.019	229	0.029
200	0.024	230	0.020
201	0.023	231	0.007
202	0.14	232	0.020
203	0.070	233	0.15
204	0.11	234	0.14
205	0.074	235	0.006

Test	IC ₅₀	Test	IC ₅₀
Comp. No.	(µM)	Comp. No.	(µM)
236	0.006	267	0.12
237	0.041	268	0.018
238	0.020	269	0.003
239	0.17	270	0.046
240	0.022	271	0.030
241	0.006	272	0.40
242	0.17	273	0.027
243	0.40	274	0.024
244	0.018	275	0.018
245	0.059	276	0.032
246	0.027	277	0.016
247	0.048	278	0.013
248	0.060	279	0.008
250	0.12	280	0.045
251	0.094	281	0.011
252	0.063	282	0.38
253	0.052	283	0.096
254	0.016	284	0.019
255	0.005	285	0.008
256	0.004	286	0.019
257	0.045	287	0.007
258	0.20	288	0.015
259	0.25	289	0.071
260	0.13	290	0.021
261	0.011	291	0.13
262	0.029	292	0.18
263	0.053	293	0.065
264	0.030	294	0.33
265	0.025	295	0.026
266	0.013	296	0.25

È

Test Comp.	1C ₅₀	Test Comp.	^{IC} 50
No.	(µM)	No.	(µM)
297	0.051	311	0.013
298	0.10	312	0.29
299	0.22	313	0.012
300	0.48	314	0.096
301	0.14	315	0.025
302	0.011	316	0.060
303	0.025	317	0.072
304	0.024	318	0.060
305	0.038	319	0.058
306	0.077	320	0.039
307	0.010	321	0.012
308	0.023	322	0.025
309	0.015	323	0.014
310	0.008		

Experiment 3: Anti-antidiuretic activity (effect on endogenous ADH)

A test compound or solvent (dimethylformamide) is administered into a caudal vein of untreated, unrestrained SD rats (male, weight: 300 - 350 g) and the amount of urine, which is spontaneously excreted for a period of 2 hours thereafter, is collected and measured by using a metabolic gauge. During this measurement, the rats are allowed to take water and feed freely.

The amount of urine of control rats (solvent-treated group) is regarded as 100 %, and the results are expressed as ED_3 value, which is the dose of the test compound to be required to excrete the urine by three times than that of the control rats. The results are shown in the following Table 7.

Table 12

ED ₃ (mg/kg)	
10	
1.9	
4.2	
0.4 *)	
	10 1.9 4.2

^{*):} Physicological saline solution was used as a solvent instead of dimethylformamide.

What is claimed is:

1. A benzoheterocyclic compound of the following formula:

$$\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{R}^{2} \\
\mathbb{R}^{3}
\end{array}$$

wherein R¹ is hydrogen atom, a halogen atom, a lower alkyl, an amino having optionally a lower alkyl substituent, or a lower alkoxy,

R² is hydrogen atom, a halogen atom, a lower alkoxy, a phenyl(lower)alkoxy, hydroxy, a lower alkyl, an amino having optionally a lower alkyl substituent, a carbamoyl-substituted lower alkoxy, an amino-substituted lower alkoxy having optionally a lower alkyl substituent, or a benzoyloxy which has optionally a halogen substituent on the phenyl ring,

 ${\rm R}^3$ is a group of the formula: $-{\rm N}_{\rm R}^{4}$ or a group of the formula: $-{\rm C}_{\rm N}^{0}$

 \mathbb{R}^4 is hydrogen atom, a benzoyl which has optionally a halogen substituent on the phenyl ring, or a lower alkyl,

 R^5 is a group of the formula: $-CO \longrightarrow (R^{16})_m$ [wherein R^{16} is a halogen atom; a lower alkyl which has

optionally a substituent selected from a halogen atom and hydroxy; hydroxy; a lower alkoxy; a lower alkanoyloxy; a lower alkylthio; a lower alkanoyl; carboxy; a lower alkoxycarbonyl; cyano; nitro; an amino which has optionally a substituent selected from a lower alkyl and a lower alkanoyl; phenyl; a cycloalkyl; a lower alkanoyloxy-substituted lower alkoxy; a carboxy-substituted lower alkoxy; a carbamoyl-substituted lower alkoxy; a hydroxy-substituted lower alkoxy; a hydroxy-substituted lower alkoxy; a phthalimido-substituted lower alkoxy; an aminocarbonyl-lower alkoxy having a lower alkyl substituent; or a group of the formula: -O-A-N R⁶ (A is a lower alkylene, and R⁶ and R⁷ are

the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a lower alkanoyl, or benzoyl, or R⁶ and R⁷ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocyclic group has optionally a substituent selected from piperidinyl and a lower alkyl); and m is an integer of 0 to 3], a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkyl-carbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthyl-carbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substi-

tuents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxy-carbonyl-lower alkanoyl, a carboxy-lower alkanoyl, a naphthyloxy-lower alkanoyl, a halogen-substituted lower alkanoyl, a group of the formula: -CO—N-R⁸ (wherein R⁸ is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula:

-CO-B-(CO) $_{n}$ -N $_{R^{10}}^{R^{9}}$ (wherein B is a lower alkylene, n is an integer of 0 or 1, and $_{R^{9}}^{9}$ and $_{R^{10}}^{10}$ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyl-lower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxy-lower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl having optionally a lower alkyl having optionally a lower alkyl substituent, or $_{R^{9}}^{9}$ and $_{R^{10}}^{10}$ may bind together

with nitrogen atom to which they bond to form a 5- or 6membered saturated heterocyclic group with or without being
intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a
lower alkyl, a lower alkoxycarboyl and piperidinyl),

R¹¹ is hydrogen atom or a lower alkyl,

R¹² is a cycloalkyl, or a phenyl which has optionally 1 to 3 substituents selected from a lower alkoxy, a lower alkyl and a halogen atom,

W is a group of the formula: $-(CH_2)_p$ - (p is an integer of 3 to 5), or a group of the formula: $-CH=CH-(CH_2)_q$ - (q is an integer of 1 to 3), the carbon atom of these groups: $-(CH_2)_p$ - and $-CH=CH-(CH_2)_q$ - being optionally replaced by oxygen atom, sulfur atom, sulfinyl, sulfonyl, or a group of

the formula: -N- (R¹³ is hydrogen atom, a cycloalkyl, or a lower alkyl), and further said -(CH₂)_p- and -CH=CH-(CH₂)_q- groups having optionally 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxy, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkyl sulfonyloxy-lower alkyl, an azido-lower alkyl, a group of the formula: O, an aminocarbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-

substituted lower alkoxy, a carboxy-substituted lower alkoxy, an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, an amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxy-imino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula:

R⁸¹-N-CH₂COO- (R⁸¹ is hydrogen atom or a lower alkyl),

hydrazino, pyrrolyl, an amino-lower alkanoyloxy having optionally a lower alkyl substituent, a group of the

formula: $-O-A-CO-N_{R83}^{82}$ (A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, or a pyridyl-lower alkyl, or R^{82} and R^{83} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula:

-(CO) $_{\rm n}^{\rm -N}$ $_{\rm R}^{\rm 15}$ (wherein n is as defined above, and $_{\rm R}^{\rm 14}$ and $_{\rm R}^{\rm 15}$

are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower

alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, benzöyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoylsubstituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyllower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or ${
m R}^{14}$ and ${
m R}^{15}$ may bind together with nitrogen atom to which they bond to form a 5or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen, wherein the heterocyclic group may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl or a lower alkanoyl),

and a salt thereof.

- 2. The compound according to claim 1, wherein \mathbb{R}^1 in the formula (1) is hydrogen atom, or a salt thereof.
 - 3. The compound according to claim 1, wherein R1

in the formula (1) is a halogen atom, and a salt thereof.

- 4. The compound according to claim 1, wherein R¹ in the formula (1) is a lower alkyl, an amino having optionally a lower alkyl substituent, or a lower alkoxy, and a salt thereof.
- 5. The compound according to claim 2, wherein \mathbb{R}^2 is hydrogen atom, and a salt thereof.
- 6. The compound according to claim 2, wherein \mathbb{R}^2 is a halogen atom, a lower alkoxy, or a lower alkyl, and a salt thereof.
- 7. The compound according to claim 2, wherein R² is a phenyl-lower alkoxy; hydroxy; an amino having optionally a lower alkyl substituent; a carbamoyl-substituted lower alkoxy; an amino-substituted lower alkoxy having optionally a lower alkyl substituent; or a benzoyloxy having optionally a halogen substituent on the phenyl ring thereof, and a salt thereof.
- 8. The compound according to claim 3, wherein ${\ensuremath{\mathsf{R}}}^2$ is hydrogen atom, and a salt thereof.
- 9. The compound according to claim 3, wherein \mathbb{R}^2 is a halogen atom, a lower alkoxy, or a lower alkyl, and a salt thereof.
- is a phenyl-lower alkoxy; hydroxy; an amino having optionally a lower alkyl substituent; a carbamoyl-substituted lower alkoxy; an amino-substituted lower alkoxy having optionally a lower alkyl substituent; or a benzoyloxy

having optionally a halogen substituent on the phenyl ring thereof, and a salt thereof.

- ll. The compound according to claim 4, wherein \mathbb{R}^2 is hydrogen atom, and a salt thereof.
- 12. The compound according to claim 4, wherein \mathbb{R}^2 is a halogen atom, a lower alkoxy, or a lower alkyl, and a salt thereof.
- is a phenyl-lower alkoxy; hydroxy; an amino having optionally a lower alkyl substituent; a carbamoyl-substituted lower alkoxy; an amino-substituted lower alkoxy having optionally a lower alkyl substituent; or a benzoyloxy having optionally a halogen substituent on the phenyl ring thereof, and a salt thereof.
- 14. The compound according to claim 5, wherein \mathbb{R}^3 is a group of the formula: $-N\mathbb{R}^4\mathbb{R}^5$ (\mathbb{R}^4 and \mathbb{R}^5 are as defined in claim 1), and a salt thereof.
- 15. The compound according to claim 5, wherein \mathbb{R}^3 is a group of the formula: $-\text{CO-NR}^{11}\mathbb{R}^{12}$ (\mathbb{R}^{11} and \mathbb{R}^{12} are as defined in claim 1), and a salt thereof.
- 16. The compound according to claim 6, wherein \mathbb{R}^3 is a group of the formula: $-N\mathbb{R}^4\mathbb{R}^5$ (\mathbb{R}^4 and \mathbb{R}^5 are as defined in claim 1), and a salt thereof.
- 17. The compound according to claim 6, wherein \mathbb{R}^3 is a group of the formula: $-\text{CO-NR}^{11}\mathbb{R}^{12}$ (\mathbb{R}^{11} and \mathbb{R}^{12} are as defined in claim 1), and a salt thereof.
 - 18. The compound according to claim 8, wherein ${ t R}^3$

is a group of the formula: $-NR^4R^5$ (R^4 and R^5 are as defined in claim 1), and a salt thereof.

- 19. The compound according to claim 8, wherein \mathbb{R}^3 is a group of the formula: $-\text{CO-NR}^{11}\mathbb{R}^{12}$ (\mathbb{R}^{11} and \mathbb{R}^{12} are as defined in claim 1), and a salt thereof.
- 20. The compound according to claim 9, wherein R^3 is a group of the formula: $-NR^4R^5$ (R^4 and R^5 are as defined in claim 1), and a salt thereof.
- 21. The compound according to claim 9, wherein \mathbb{R}^3 is a group of the formula: $-\text{CO-NR}^{11}\mathbb{R}^{12}$ (\mathbb{R}^{11} and \mathbb{R}^{12} are as defined in claim 1), and a salt thereof.
- 22. The compound according to claim 14, wherein ${\bf R}^4$ is hydrogen atom, and ${\bf R}^5$ is a group of the formula:
- $-CO \xrightarrow{(R^{16})_m}$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanyl-carbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxy-lower alkanoyl, a naphthyloxy-lower alkanoyl, a halogen-

substituted lower alkanoyl, a group of the formula: $N-R^8$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula: $-CO-B-(CO)_n-N_{n10}^{R^9}$ (wherein B is a lower alkylene, n is an integer of 0 or 1, and R^9 and R^{10} are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R9 and R^{10} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and

WO 91/05549 - 867 - PCT/JP90/01340

piperidinyl), and a salt thereof.

- 24. The compound according to claim 14, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 25. The compound according to claim 16, wherein ${\bf R}^4$ is hydrogen atom, and ${\bf R}^5$ is a group of the formula:
- $-CO \xrightarrow{(R^{16})_m}$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- 26. The compound according to claim 16, wherein \mathbb{R}^4 is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: $-CO-(N-R^8)$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl

on the phenyl ring, quinolylsulfonyl, or a group of the formula: -CO-B-(CO)_n-N_R^9 (wherein B is a lower alkylene, n

is an integer of 0 or 1, and ${\ensuremath{\mathtt{R}}}^9$ and ${\ensuremath{\mathtt{R}}}^{10}$ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R9 and R^{10} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 27. The compound according to claim 16, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 28. The compound according to claim 7, wherein ${\bf R}^4$ is hydrogen atom, and ${\bf R}^5$ is a group of the formula:

 $-CO \xrightarrow{(R^{16})_m}$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.

The compound according to claim 7, wherein R4 is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanyl- . carbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: $N-R^8$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower_alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula: $-CO-B-(CO)_n-N_{n+10}^{R^9}$ (wherein B is a lower alkylene, n is an integer of 0 or 1, and R^9 and R^{10} are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3

substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R⁹ and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 30. The compound according to claim 7, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 31. The compound according to claim 18, wherein \mathbb{R}^4 is hydrogen atom, and \mathbb{R}^5 is a group of the formula:
- $-CO \xrightarrow{(R^{16})_m}$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- 32. The compound according to claim 18, wherein R⁴ is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanyl-carbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower

alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxy-lower alkanoyl, a naphthyloxy-lower alkanoyl, a halogen-substituted lower alkanoyl, a group of the formula:

-CO-\N-R^8 (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the

formula: $-CO-B-(CO)_{n}-N_{R}$ (wherein B is a lower alkylene, n

is an integer of 0 or 1, and R⁹ and R¹⁰ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxyllower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R⁹ and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with

or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 33. The compound according to claim 18, whrein ${\ensuremath{\mathsf{R}}}^4$ is a lower alkyl, and a salt thereof.
- 34. The compound according to claim 20, wherein \mathbb{R}^4 is hydrogen atom, and \mathbb{R}^5 is a group of the formula:
- $(R^{16})_m$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- The compound according to claim 20, wherein ${\ensuremath{\mathtt{R}}}^4$ is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: $-CO \sim N-R^8$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which

has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the

formula: $-CO-B-(CO)_{n}-N_{R}^{9}$ (wherein B is a lower alkylene, n

is an integer of 0 or 1, and ${\ensuremath{\mathsf{R}}}^9$ and ${\ensuremath{\mathsf{R}}}^{10}$ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or \mathbb{R}^9 and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 36. The compound according to claim 20, whrein ${\ensuremath{R^4}}$ is a lower alkyl, and a salt thereof.
- 37. The compound according to claim 10, wherein \mathbb{R}^4 is hydrogen atom, and \mathbb{R}^5 is a group of the formula:

 $-CO \longrightarrow (R^{16})_m$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.

The compound according to claim 10, wherein ${\ensuremath{\mathtt{R}}}^4$ is hydrogen atom and R^5 is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: -CO- $\sqrt{N-R^8}$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula: $-CO-B-(CO)_n-N$ (wherein B is a lower alkylene, n is an integer of 0 or 1, and ${\ensuremath{\mathtt{R}}}^9$ and ${\ensuremath{\mathtt{R}}}^{10}$ are the same or

is an integer of 0 or 1, and R⁹ and R¹⁰ are the same or different and are each hydrogen atom, a lower alkyl having

optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkyl substituent, or R⁹ and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 39. The compound according to claim 10, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 40. The compound according to claim 11, wherein ${\bf R}^4$ is hydrogen atom, and ${\bf R}^5$ is a group of the formula:
- $-CO \xrightarrow{(R^{16})_m}$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- 41. The compound according to claim 11, wherein R⁴ is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanyl-carbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl,

WO 91/05549 - 876 - PCT/JP90/01340

thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: -CO- $N-R^8$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula: $-CO-B-(CO)_{n}-N_{n}$ (wherein B is a lower alkylene, n

is an integer of 0 or 1, and R⁹ and R¹⁰ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower

alkyl having optionally a lower alkyl substituent, or R^9 and R^{10} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 42. The compound according to claim 11, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 43. The compound according to claim 12, wherein \mathbb{R}^4 is hydrogen atom, and \mathbb{R}^5 is a group of the formula:
- -CO (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanyl-carbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxy-lower alkanoyl, a naphthyloxy-lower alkanoyl, a halogen-substituted lower alkanoyl, a group of the formula:

phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an

amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the formula: $-CO-B-(CO)_n-N_{n+10}$ (wherein B is a lower alkylene, n is an integer of 0 or 1, and ${ extsf{R}}^9$ and ${ extsf{R}}^{10}$ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R9 and R^{10} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom

45. The compound according to claim 12, whrein R4

wherein the heterocylic group has optionally a substituent

selected from a lower alkyl, a lower alkoxycarboyl and

piperidinyl), and a salt thereof.

is a lower alkyl, and a salt thereof.

- 46. The compound according to claim 13, wherein \mathbb{R}^4 is hydrogen atom, and R^5 is a group of the formula:
- $-CO \sim (R^{16})_m$ (wherein R^{16} and m are as defined in claim 1), and a salt thereof.
- The compound according to claim 13, wherein R^4 is hydrogen atom and R⁵ is a phenyl-lower alkoxycarbonyl, a lower alkanoyl, a phenyl-lower alkanoyl, a cycloalkyl-lower alkanoyl, a cycloalkylcarbonyl, tricyclo[3.3.1.1]decanylcarbonyl, naphthylcarbonyl, pyridylcarbonyl, furoyl, thenoyl, a phenoxy-lower alkanoyl which phenyl ring has optionally 1 to 3 substituents selected from a lower alkyl, a lower alkoxy and an amino having optionally a lower alkanoyl substituent, a phthalimido-substituted lower alkanoyl, a lower alkoxycarbonyl-lower alkanoyl, a carboxylower alkanoyl, a naphthyloxy-lower alkanoyl, a halogensubstituted lower alkanoyl, a group of the formula: $-CO \sim N-R^8$ (wherein R^8 is hydrogen atom, a lower alkyl, a phenyl-lower alkoxycarbonyl, a carbamoyl-lower alkyl, an amino-lower alkanoyl having optionally a lower alkyl substituent, or a lower alkanoyl), an anilinocarbonyl which has optionally a lower alkyl substituent on the phenyl ring, phenoxycarbonyl, a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, quinolylsulfonyl, or a group of the

formula: $-CO-B-(CO)_n-N_{pl0}^{R^9}$ (wherein B is a lower alkylene, n

is an integer of 0 or 1, and ${\ensuremath{\mathtt{R}}}^9$ and ${\ensuremath{\mathtt{R}}}^{10}$ are the same or different and are each hydrogen atom, a lower alkyl having optionally a hydroxy substituent, a cycloalkyl, a phenyllower alkyl, a lower alkanoyl, a lower alkenyl, a phenoxylower alkyl, a phenyl which has optionally 1 to 3 substituents selected from an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkyl, a lower alkoxy and a halogen atom, a phthalimido-substituted lower alkyl, an amino-lower alkyl having optionally a lower alkanoyl substituent, a lower alkynyl, or an amino-lower alkyl having optionally a lower alkyl substituent, or R9 and R¹⁰ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen atom wherein the heterocylic group has optionally a substituent selected from a lower alkyl, a lower alkoxycarboyl and piperidinyl), and a salt thereof.

- 48. The compound according to claim 13, whrein \mathbb{R}^4 is a lower alkyl, and a salt thereof.
- 49. The compound according to claim 22, wherein \mathbb{R}^{16} is a halogen atom, or a lower alkyl having optionally a substituent selected from a halogen atom and hydroxy, and a salt thereof.
- 50. The compound according to claim 25, wherein R^{16} is a halogen atom, or a lower alkyl having optionally a substituent selected from a halogen atom and hydroxy, and a salt thereof.

- 51. The compound according to claim 31, wherein \mathbb{R}^{16} is a halogen atom, or a lower alkyl having optionally a substituent selected from a halogen atom and hydroxy, and a salt thereof.
- 52. The compound according to claim 34, wherein \mathbb{R}^{16} is a halogen atom, or a lower alkyl having optionally a substituent selected from a halogen atom and hydroxy, and a salt thereof.
- 53. The compound according to claim 1, wherein W is a group of the formula: $-(CH_2)_p$ wherein p is an integer of 3 to 5, and the carbon atom of said group is optionally replaced by oxygen atom, sulfur atom, sulfinyl, sulfonyl, or

a group of the formula: -N- (R¹³ is hydrogen atom, a cycloalkyl, or a lower alkyl), and further said -(CH₂)_p-group has optionally 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxy, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkyl sulfonyloxy-lower alkyl, an azido-lower alkyl, a group of the formula: O, an amino-carbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-substituted lower alkoxy, a carboxy-substituted lower alkoxy, an amino-carbonyl-lower alkoxy having optionally a lower alkyl

substituent, an amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxy-imino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula: R^{81} -N-CH₂COO- (R^{81} is hydrogen atom or a lower alkyl),

hydrazino, pyrrolyl, an amino-lower alkanoyloxy having optionally a lower alkyl substituent, a group of the

formula: $-0-A-CO-N_{R83}^{82}$ (A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, or a pyridyl-lower alkyl, or R^{82} and R^{83} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula:

-(CO) $_{n}^{-N}$ \, R¹⁴ (wherein n is as defined above, and R¹⁴ and R¹⁵

are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a

WO 91/05549 - 883 - PCT/JP90/01340

lower alkylsulfonyl, benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoylsubstituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanovl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyllower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or R¹⁴ and R¹⁵ may bind together with nitrogen atom to which they bond to form a 5or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen, wherein the heterocyclic group may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl or a lower alkanovl), and a salt thereof.

54. The compound according to claim 1, wherein W is a group of the formula: $-CH=CH-(CH_2)_q$ — wherein q is an integer of 1 to 3, and the carbon atom of said group is optionally replaced by oxygen atom, sulfur atom, sulfinyl,

sulfonyl, or a group of the formula: -N- (R¹³ is hydrogen atom, a cycloalkyl, or a lower alkyl), and further said

-CH=CH-(CH $_2$) $_q$ - group has optionally 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxy, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkyl sulfonyloxy-lower alkyl, an azido-lower alkyl, a group of the formula: O, an aminocarbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-substituted lower alkoxy, a carboxy-substituted lower alkoxy, an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, an amino-lower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxy-imino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula: R^{81} -N-CH₂COO- (R^{81} is hydrogen atom or a lower alkyl), hydrazino, pyrrolyl, an amino-lower alkanoyloxy having optionally a lower alkyl substituent, a group of the formula: $-O-A-CO-N_{-83}$ (A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxysubstituted lower alkyl, or a pyridyl-lower alkyl, or R82 and R⁸³ may bind together with nitrogen atom to which they

8

bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula:

-(CO) $_{n}$ -N $_{R}^{14}$ (wherein n is as defined above, and R 14 and R 15

are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoylsubstituted lower alkyl, a carboxy-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxycarbonyl, an aminocarbonyllower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or \mathbb{R}^{14} and \mathbb{R}^{15} may bind

together with nitrogen atom to which they bond to form a 5or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen, wherein the heterocyclic group may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl or a lower alkanoyl), and a salt thereof.

The compound according to claim 53, wherein W is a group of the formula: $-(CH_2)_p$ - (p is an integer of 3 to 5) and said $-(CH_2)_p$ - group has optionally 1 to 3 substituents selected from a lower alkyl having optionally a hydroxy substituent, a lower alkoxycarbonyl, carboxy, hydroxy, oxo, a lower alkanoyloxy having optionally a halogen substituent, an amino-lower alkyl having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a lower alkanoyloxysubstituted lower alkyl, a lower alkyl sulfonyloxy-lower alkyl, an azido-lower alkyl, a group of the formula: 0, an aminocarbonyloxy having optionally a lower alkyl substituent, a lower alkoxy, a lower alkoxycarbonyl-substituted lower alkoxy, a carboxy-substituted lower alkoxy, an aminocarbonyl-lower alkoxy having optionally a lower alkyl substituent, an aminolower alkoxy having optionally a substituent selected from a lower alkyl and a lower alkanoyl, a phthalimido-substituted lower alkoxy, hydroxyimino, a lower alkanoyloxy-imino, a lower alkylidene, a halogen atom, azido, sulfoxyimino, a group of the formula: R^{81} -N-CH₂COO- (R^{81} is hydrogen atom or a lower alkyl); hydrazino, pyrrolyl, an amino-lower alkanoyloxy having

optionally a lower alkyl substituent, a group of the formula: $-0-A-CO-N {R^{82} \over R^{83}} \ (\mbox{A is as defined above, and R^{82} and R^{83} are the same or different and are each hydrogen atom, a lower alkyl, a carbamoyl-substituted lower alkyl, a hydroxy-substituted lower alkyl, or a pyridyl-lower alkyl, or <math display="inline">R^{82}$ and R^{83} may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen, oxygen or sulfur atom wherein the heterocyclic group has optionally a substituent selected from oxo, a lower alkyl, a lower alkanoyl, and carbamoyl), and a group of the formula: $-(CO)_{n}-N_{p15}^{R14} \ (\mbox{wherein n is as defined}$

above, and R¹⁴ and R¹⁵ are the same or different and are each hydrogen atom, a lower alkyl, a lower alkenyl, a lower alkanoyl, a cycloalkyl, an oxiranyl-substituted lower alkyl, a lower alkyl having 1 to 2 substituents selected from a lower alkoxy, hydroxy and an amino having optionally a lower alkyl substituent, a phenyl-lower alkyl, a pyridyl-lower alkyl, a lower alkylsulfonyl, benzoyl, a lower alkoxycarbonyl, anilinocarbonyl, an aminocarbonyl having optionally a lower alkyl substituent, a cyano-substituted lower alkyl, a lower alkoxycarbonyl-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a carbamoyl-substituted lower alkyl, a tetrahydropyranyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a lower alkanoyloxy-substituted lower alkyl, a piperidinyl having optionally a phenyl-lower alkyl substituent on the piperidinyl ring, a

halogen-substituted lower alkanoyl, an imidazolyl-substituted lower alkanoyl, an amino-lower alkanoyl having optionally a substituent selected from a lower alkyl and a lower alkoxy-carbonyl, an aminocarbonyl-lower alkyl having optionally a lower alkyl substituent, or a phenyl-lower alkoxycarbonyl, or R¹⁴ and R¹⁵ may bind together with nitrogen atom to which they bond to form a 5- or 6-membered saturated heterocyclic group with or without being intervened with nitrogen or oxygen, wherein the heterocyclic group may optionally have a substituent selected from a lower alkyl, a phenyl-lower alkyl or a lower alkanoyl), and a salt thereof.

56. The compound according to claim 53, wherein the carbon atom of the group of the formula: $-(CH_2)_p$ is replaced by oxygen atom, sulfur atom, sulfinyl, sulfonyl, or a group of

 R^{13} the formula: -N- (R^{13} is hydrogen atom, a cycloalkyl, or a lower alkyl), and a salt thereof.

57. The compound according to claim 55, wherein p in the group: $-(CH_2)_p^-$ is 3 and the group has no substituent, and a salt thereof.

58. The compound according to claim 55, wherein p in the group: $-(CH_2)_p$ is 3 and the group has a substituent of a group of the formula: $-(CO)_n$ (wherein R^{14} , R^{15} , and n is as defined above), and a salt thereof.

59. The compound according to claim 55, wherein p in $^{\circ}$ the group: $-(CH_2)_p-$ is 4 and the group has no substituent, and

a salt thereof.

- 60. The compound according to claim 55, wherein p in the group: $-(CH_2)_p$ is 4 and the group has a substituent of a group of the formula: $-(CO)_n$ -N R^{14} (wherein R^{14} , R^{15} , and n is as defined above), and a salt thereof.
- 61. The compound according to claim 55, wherein p in the group: $-(CH_2)_D$ is 5, and a salt thereof.
- 62. The compound according to claim 56, wherein p in the group: $-(CH_2)_p$ is 3 and the carbon atom of this group is

replaced by a group of the formula: -N- (wherein R^{13} is as defined above), and a salt thereof.

the group: $-(CH_2)_p$ is 4 and the carbon atom of this group is

replaced by a group of the formula: -N- (wherein R^{13} is as defined above), and a salt thereof.

- 64. The compound according to claim 56, wherein p in the group: $-(CH_2)_p$ is 5 and the carbon atom of this group is
- replaced by a group of the formula: -N- (wherein R¹³ is as defined above), and a salt thereof.
 - 65. The compound according to claim 56, wherein the carbon atom of the group: $-(CH_2)_p$ is replaced by oxygen atom, sulfur atom, sulfinyl, or sulfonyl, and a salt thereof.
 - 66. The compound according to claim 54, wherein q in

the group: $-CH=CH-(CH_2)_q$ is 1, and a salt thereof.

- 67. The compound according to claim 54, wherein q in the group: -CH=CH-(CH $_2$) $_q$ is 2, and a salt thereof.
- 68. The compound according to claim 54, wherein q in , the group: $-CH=CH=(CH_2)_q$ is 3, and a salt thereof.
- 69. The compound according to claim 58 or 60, wherein n in the substituent: $-(CO)_{n}-N_{R15}$ is 0, and R^{14} and R^{15} are the same or different and are each hydrogen atom, a lower alkyl, or a cycloalkyl, and a salt thereof.
- 70. The compound according to claim 63 wherein the heterocyclic group of the formula: (M) is 2,3,4,5-

tetrahydro-lH-1,4-benzodiazepine, and a salt thereof.

71. The compound according to claim 67 wherein the heterocyclic group of the formula: (W) is 2,3-dihydro-

1H-benzazepine, and a salt thereof.

- 72. l-[4-(2-Methylbenzoylamino)benzoyl]-4-methyl-2,3,4,5-tetrahydro-lH-1,4-benzodiazepine.
- 73. 5-Dimethylamino-1-[4-(2-methylbenzoylamino)-benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine.
- 74. 5-Dimethylamino-1-[2-chloro-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine.
- 75. 5-Methylamino-l-[2-chloro-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine.

- 76. 5-Cyclopropylamino-1-[2-chloro-4-(2-methyl-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine.
- 77. 5-Cyclopropylamino-1-[2-chloro-4-(2-chloro-benzoylamino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine.
- 78. 5-Dimethylamino-1-[2-methyl-4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine.
- 79. 4-Dimethylamino-1-[3-methoxy-4-(2-methylbenzoyl-amino)benzoyl]-1,2,3,4-tetrahydroquinoline.
- 80. 7-Chloro-5-methylamino=1-[4-(2-methylbenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-lH-benzazepine.
- 81. 7-Chloro-5-methylamino-1-[4-(2-chlorobenzoyl-amino)benzoyl]-2,3,4,5-tetrahydro-1H-benzazepine.
- 82. A vasopressin antagonistic composition which comprises as an active ingredient a compound of the formula (1) as set forth in claim 1, or a pharmaceutically acceptable salt thereof in admixture with a pharmaceutically acceptable carrier or diluent.
- 83. A process for preparing a compound of the formula (1) as set forth in claim 1, which comprises the following steps of
 - (a) reacting a compound of the formula (2):

$$R^1 \longrightarrow W$$
 (2)

wherein R¹ and W are the same as defined in claim 1, with a compound of the formula (3):

$$R^2$$

wherein \mathbb{R}^2 and \mathbb{R}^3 are the same as defined in claim 1, to give a compound of the formula (1),

(b) reacting a compound of the formula (2b):

wherein R^1 , R^2 , R^4 and W are as defined in claim 1, with a compound of the formula (4):

$$R^{5a}OH$$
 (4)

wherein R^{5a} is the same as R^{5} as defined in claim 1 except excluding an anilinocarbonyl having optionally a lower alkyl substituent on the phenyl ring, a phenylsulfonyl having optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring and quinolylsulfonyl to give a compound of the formula (lb):

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 and W are as defined in claim 1, and \mathbb{R}^{5a} is as defined above,

(c) reacting a compound of the formula (5):

$$R^1$$
 CO
 R^2
 $COOH$
 (5)

wherein \mathbb{R}^1 , \mathbb{R}^2 , and W are as defined in claim 1, with a compound of the formula (6):

$$HN_{R^{12}}$$
 (6)

wherein R^{11} and R^{12} are as defined in claim 1, to give a of the formula (lc):

$$\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{N} \\
\mathbb{C}O \\
\mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{R}^{2} \\
\mathbb{R}^{11} \\
\mathbb{R}^{12}$$

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^{11} , \mathbb{R}^{12} and W are as defined in claim 1, (d) reacting a compound of the formula (7):

$$R^1$$
 N
 CO
 R^2
 N
 R^5
 R^5

wherein R^1 , R^2 , R^5 and W are as defined in claim 1, with a compound of the formula (8) or (9):

 $R^{4a}X$

(8) or

 $R^{17}COR^{18}$ (9)

wherein R^{4a} is a lower alkyl, X is a halogen atom, and R^{17} and R^{18} are each hydrogen atom or a lower alkyl, to give a compound of the formula (ld):

$$R^1$$
 N
 CO
 R^2
 R^4a
 N
 R^5

wherein R^1 , R^2 , R^5 and W are as defined in claim 1, and R^{4a} is as defined above,

(e) reacting a compound of the formula (10):

$$R^{1}$$
 N
 CO
 R^{2}
 $CO-N$
 R^{12}

wherein R^1 , R^2 , R^{12} , and W are as defined in claim 1, with a compound of the formula (11) or (9):

$$R^{11a}X$$
 (11) or $R^{17}COR^{18}$ (9)

wherein R^{11a} is a lower alkyl, and X, R^{17} and R^{18} are as defined above, to give a compound of the formula (le):

$$R^1$$
 R^2
 $CO-N$
 R^{11a}
 R^2

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^{12} and W are as defined in claim 1, and \mathbb{R}^{11a} is as defined above,

(f) Reacting a compound of the formula (12):

$$R^{1}$$
 N
 CO
 R^{1}
 R^{2}
 $CO-N$
 H
 (12)

wherein R^1 , R^2 , R^{11} , and W are as defined in claim 1, with a compound of the formula (13):

$$R^{12a}X$$
 (13)

wherein \mathbb{R}^{12a} is a cycloalkyl and X is as defined above, to give a compound of the formula (lf):

$$R^1$$
 N
 CO
 R^2
 R^{11}
 $CO-N$
 $R^{12}a$

wherein R^1 , R^2 , R^{11} , and W are as defined above, and R^{12a} is as defined above,

(g) reacting a compound of the formula (2b):

$$R^1$$
 N
 CO
 R^2
 $N-R^4$
 H

wherein R^1 , R^2 , R^4 , and W are as defined in claim 1, with a compound of the formula (38):

$$R^{46}N=C=O$$
 (38)

wherein \mathbb{R}^{46} is a phenyl having optionally a lower alkyl substituent, to give a compound of the formula (lcc):

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 , and W are as defined in claim 1, and \mathbb{R}^{46} is as defined above,

(h) reacting a compound of the formula (2b):

$$R^1$$
 N
 CO
 R^2
 $NH-R^4$
 $(2b)$

wherein R^1 , R^2 , R^4 , and W are as defined in claim 1, with a compound of the formula (39):

$$R^{47}X$$
 (39)

wherein \mathbb{R}^{47} is a phenylsulfonyl which has optionally a substituent selected from a halogen atom and a lower alkyl on the phenyl ring, or quinolylsulfonyl, and X is as defined

above, to give a compound of the formula (ldd):

$$R^1$$
 N
 CO
 R^2
 $N-R^4$
 R^{47}

wherein \mathbb{R}^1 , \mathbb{R}^2 , \mathbb{R}^4 , and W are as defined in claim 1, and \mathbb{R}^{47} is as defined above, or

(i) reacting a compound of the formula (7):

$$R^1$$
 N
 CO
 R^2
 NH
 R^2
 NH
 R^5

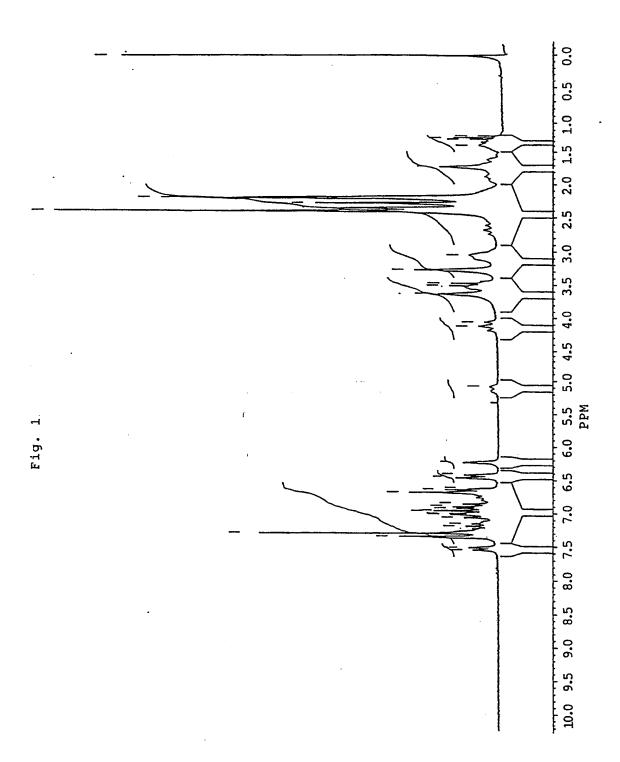
wherein R^1 , R^2 , R^5 , and W are as defined in claim 1, with a compound of the formula (42):

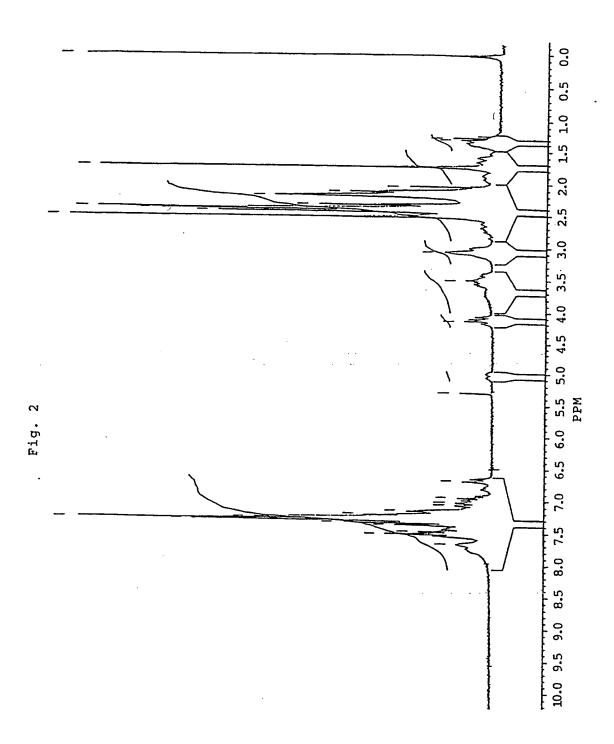
$$R^{50}OH$$
 (42)

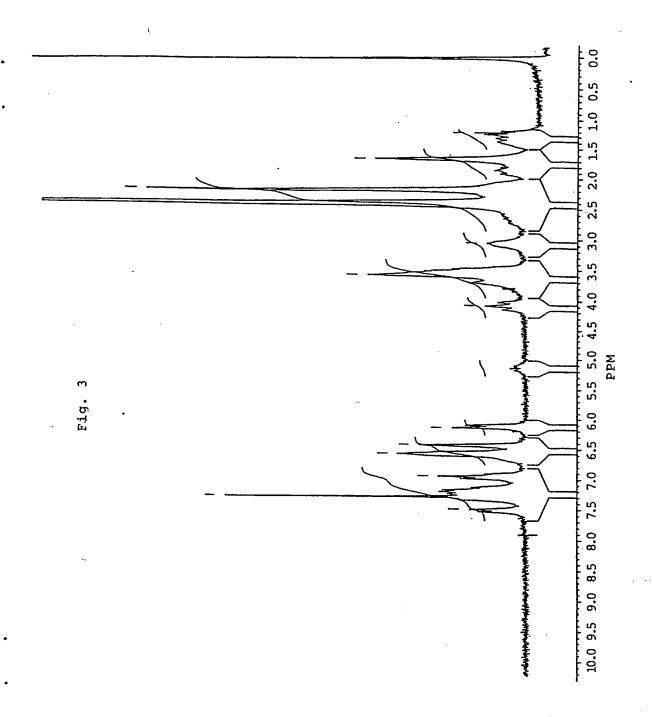
wherein R^{50} is a benzoyl having optionally a halogen substituent on the phenyl ring, to give a compound of the formula (lhh):

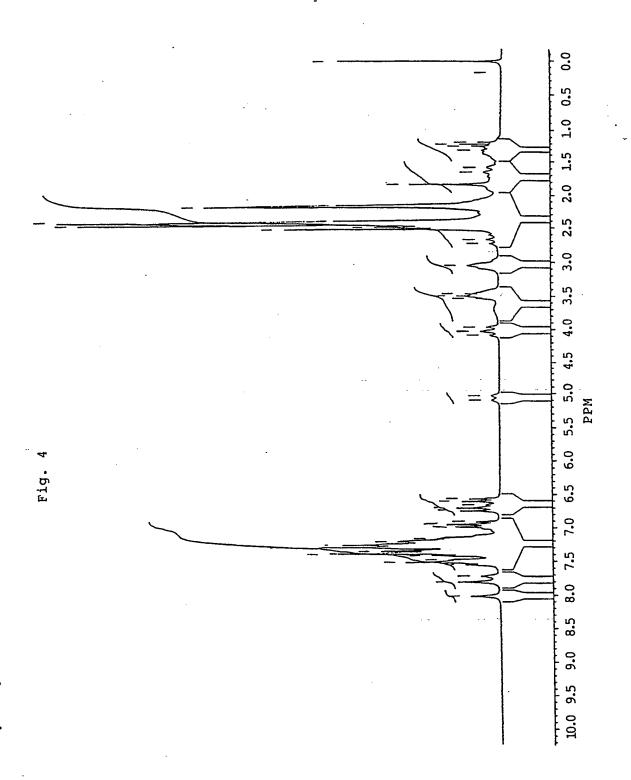
$$R^1$$
 N
 CO
 R^2
 $N-R^{50}$
 R^5

wherein ${\bf R}^1$, ${\bf R}^2$, ${\bf R}^5$, and W are as defined in claim 1, and ${\bf R}^{50}$ is as defined above.









INTERNATIONAL SEARCH REPORT

International Application No PCT/JP 90/01340

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) 6								
	nt Classification (IPC) or to both Na							
		12, 223/16, 225/06, 24	1/42					
243/14, 265/	36. 267/14							
II. FIELDS SEARCHED	Minimum Brancon	tation Convoluted 7						
Minimum Documentation Searched Classification System Classification Symbols								
Classification System								
			İ					
IPC5 A 61	K; C 07 D		İ					
	Documentation Searched other	the Misious Designation	····					
	to the Extent that such Documents							
İ		`						
III. DOCUMENTS CONSIDEREI	D TO BE RELEVANT 9							
Category Citation of Docu	ment, ¹¹ with indication, where appr	ropriate, of the relevant passages ¹²	Relevant to Claim No. 13					
A US, A, 34584	98 (CHARLES M.C. KOO	ET AL.)	1-83					
29 July	1969,	•						
see the	whole document							
								
A US. A. 43351	23 (OTTO GRÄWINGER E	T AL.)	1,82					
15 June		,	1,02					
	whole document							
Charical Ab-	tracts, volume 102,	25 24 hims	1 02					
A Chemical Abs	1,82							
abstract	Columbus, Ohio, US), 220763r, & JP, A, 6	5004170						
(4-Quino	linone derivatives)	10 January 1985						
		•						
	1							
A Special extension of site	d documents: 10	FT later desument with the state of	Abo later discrete distance de la constitución de l					
Special categories of cited documents: 10 "A" document defining the general state of the art which is not considered to be of particular relevance. "Cansidered to be of particular relevance."								
I considered to be of partic	cular relevance	invention						
filing date cannot be considered novel or cannot be considered to								
"L" document which may throw which is cited to establis	Le document which may throw doubts on priority claim(s) or involve an inventive step							
which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or								
other means	other means in the art.							
later than the priority date claimed "a" document member of the same patent family								
IV. CERTIFICATION Date of the Actual Completion of	the International Search	Date of Mailing of this International !	Search Report					
1								
3rd January 1991		2 1 JAN 1998						
International Searching Authorit	у	Signature of Authorized Offices						
FIRNDFAN DA	SUPPLIENT DEFECT							
EUROPEAN PATENT OFFICE								

Form PCT/ISA/210 (second sheet) (January 1985)

III. DOCU	MENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to Claim No
A	DE, A, 2314392 (A.H. ROBINS CO. INC.) 27 September 1973, see particularly pages 12, 14 and the claims	1
A	DE, A, 1906593 (UCB, UNION CHIMIQUE-CHEMISCHE BEDRIJVEN S.A.) 18 September 1969, see the claims	1
A	DE, A, 1595863 (KNOLL AG) 12 February 1970, see particularly pages 7-8	1
A	FR, A, 1405271 (LEPETIT S.P.A.) 4 July 1969, see the whole document	1
		
		,

The additional search fees were accompanied by applicant's protest.

No protest accompanied the payment of additional search fees.

ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.PCT/JP 90/01340

SA

41012

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
US-A-	3458498	29/07/69 15/06/82	US-A- US-A-	3516987 3542760	23/06/70 24/11/70 15/06/83 24/11/83 05/03/81 20/09/83 12/03/81
US-A-	4335123		AT-T- AU-B- AU-D- CA-A- DE-A-	AU-B- 533484 AU-D- 6179380 CA-A- 1154015	
			EP-A-B- JP-A-	0025864 56034668	01/04/81 06/04/81
DE-A-	2314392	27/09/73	AU-D- FR-A- JP-A-	5280373 2182886 49011893	05/09/74 14/12/73 01/02/74
DE-A-	1906593	18/09/69	BE-A- FR-A- GB-A- NL-A- SE-B-	728220 2001730 1193534 6902057 356746	11/08/69 03/10/69 03/06/70 14/08/69 04/06/73
DE-A-	1595863	12/02/70	BE-A- CH-A- CH-A- CH-A- GB-A- NL-A- US-A-	702360 501658 501659 502367 1137796 6711108 3547915	07/02/68 15/01/71 15/01/71 31/01/71 00/00/00 12/02/68 15/12/70
FR-A-	1405271	04/07/69	CH-A- CH-A- DE-A- DE-A- FR-E- GB-A- US-A-	429731 433313 1470008 1470009 94101 1090611 3346565	00/00/00 00/00/00 30/01/69 16/01/69 04/07/69 00/00/00